

Wernfried Mayr-Schmüller

List of Publications by Year in descending order

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Version: 2024-02-01

12

papers

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citations

1163117

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docs citations

14

times ranked

453

citing authors

#	ARTICLE	IF	CITATIONS
1	Characterization of an Al-Cu-Mg-Zn multi principal element alloy by experimental and computational screening methods. <i>Acta Materialia</i> , 2022, 224, 117510.	7.9	3
2	Heterogeneous Adsorption and Local Ordering of Formate on a Magnetite Surface. <i>Journal of Physical Chemistry Letters</i> , 2021, 12, 3847-3852.	4.6	7
3	Many-electron calculations of the phase stability of ZrO_2 polymorphs. <i>Physical Review Research</i> , 2020, 2, .		
4	Adsorption of CO on the $\text{Ca}_3\text{Ru}_2\text{O}_7(001)$ surface. <i>Surface Science</i> , 2019, 680, 18-23.	1.9	2
5	Adsorption of a superoxo O_2 -species on the pure and Ca-doped $\text{Sr}_3\text{Ru}_2\text{O}_7(001)$ surface. <i>Surface Science</i> , 2019, 680, 24-31.	1.9	2
6	A full monolayer of superoxide: oxygen activation on the unmodified $\text{Ca}_{3-x}\text{Ru}_{2-x}\text{O}_{7-x}(001)$ surface. <i>Journal of Materials Chemistry A</i> , 2018, 6, 5703-5713.	10.3	17
7	Water adsorption at zirconia: from the $\text{ZrO}_{2}(111)/\text{Pt}_{3}/\text{Zr}(0001)$ model system to powder samples. <i>Journal of Materials Chemistry A</i> , 2018, 6, 17587-17601.	10.3	24
8	Ordered hydroxyls on $\text{Ca}_3\text{Ru}_2\text{O}_7(001)$. <i>Nature Communications</i> , 2017, 8, 23.	12.8	12
9	Metal Adatoms and Clusters on Ultrathin Zirconia Films. <i>Journal of Physical Chemistry C</i> , 2016, 120, 9920-9932.	3.1	18
10	Adsorption of water at the SrO surface of Ru Ruthenates. <i>Nature Materials</i> , 2016, 15, 450-455.	27.5	63
11	Growth of an Ultrathin Zirconia Film on Pt_3/Zr Examined by High-Resolution X-ray Photoelectron Spectroscopy, Temperature-Programmed Desorption, Scanning Tunneling Microscopy, and Density Functional Theory. <i>Journal of Physical Chemistry C</i> , 2015, 119, 2462-2470.	3.1	46
12	Pt Zr : A substrate for growing well-ordered ultrathin zirconia films by oxidation. <i>Physical Review B</i> , 2012, 86, .	3.2	41