

Eduard Matito

List of Publications by Year in descending order

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115
papers

5,299
citations

94433

37
h-index

91884

69
g-index

127
all docs

127
docs citations

127
times ranked

3149
citing authors

#	ARTICLE	IF	CITATIONS
1	The aromatic fluctuation index (FLU): A new aromaticity index based on electron delocalization. <i>Journal of Chemical Physics</i> , 2005, 122, 014109.	3.0	396
2	Quantifying aromaticity with electron delocalisation measures. <i>Chemical Society Reviews</i> , 2015, 44, 6434-6451.	38.1	335
3	On the performance of some aromaticity indices: A critical assessment using a test set. <i>Journal of Computational Chemistry</i> , 2008, 29, 1543-1554.	3.3	261
4	Electron sharing indexes at the correlated level. Application to aromaticity calculations. <i>Faraday Discussions</i> , 2007, 135, 325-345.	3.2	203
5	Comparison of the AIM Delocalization Index and the Mayer and Fuzzy Atom Bond Orders. <i>Journal of Physical Chemistry A</i> , 2005, 109, 9904-9910.	2.5	169
6	Nucleus-independent chemical shift (NICS) profiles in a series of monocyclic planar inorganic compounds. <i>Journal of Organometallic Chemistry</i> , 2006, 691, 4359-4366.	1.8	155
7	The role of electronic delocalization in transition metal complexes from the electron localization function and the quantum theory of atoms in molecules viewpoints. <i>Coordination Chemistry Reviews</i> , 2009, 253, 647-665.	18.8	141
8	Electron localization function at the correlated level. <i>Journal of Chemical Physics</i> , 2006, 125, 024301.	3.0	135
9	Properties of Aromaticity Indices Based on the One-Electron Density Matrix. <i>Journal of Physical Chemistry A</i> , 2007, 111, 6521-6525.	2.5	118
10	Electron delocalization and aromaticity in low-lying excited states of archetypal organic compounds. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 20690.	2.8	116
11	A Critical Assessment of the Performance of Magnetic and Electronic Indices of Aromaticity. <i>Symmetry</i> , 2010, 2, 1156-1179.	2.2	115
12	Electron Localization Function at the Correlated Level: A Natural Orbital Formulation. <i>Journal of Chemical Theory and Computation</i> , 2010, 6, 2736-2742.	5.3	115
13	Nine questions on energy decomposition analysis. <i>Journal of Computational Chemistry</i> , 2019, 40, 2248-2283.	3.3	113
14	An electronic aromaticity index for large rings. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 11839-11846.	2.8	110
15	Metalloaromaticity. <i>Wiley Interdisciplinary Reviews: Computational Molecular Science</i> , 2013, 3, 105-122.	14.6	105
16	Aromaticity of Distorted Benzene Rings: Exploring the Validity of Different Indicators of Aromaticity. <i>Journal of Physical Chemistry A</i> , 2007, 111, 4513-4521.	2.5	102
17	New electron delocalization tools to describe the aromaticity in porphyrinoids. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 2787-2796.	2.8	86
18	Separation of dynamic and nondynamic correlation. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 24015-24023.	2.8	85

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19	A Test to Evaluate the Performance of Aromaticity Descriptors in All-Metal and Semimetal Clusters. An Appraisal of Electronic and Magnetic Indicators of Aromaticity. <i>Journal of Chemical Theory and Computation</i> , 2010, 6, 1118-1130.	5.3	84
20	Aromaticity Measures from Fuzzy-Atom Bond Orders (FBO). The Aromatic Fluctuation (FLU) and the para-Delocalization (PDI) Indexes. <i>Journal of Physical Chemistry A</i> , 2006, 110, 5108-5113.	2.5	76
21	New Solids Based on B ₁₂ N ₁₂ Fullerenes. <i>Journal of Physical Chemistry C</i> , 2007, 111, 13354-13360.	3.1	72
22	On the existence and characterization of molecular electrides. <i>Chemical Communications</i> , 2015, 51, 4865-4868.	4.1	68
23	An analysis of the changes in aromaticity and planarity along the reaction path of the simplest Diels-Alder reaction. Exploring the validity of different indicators of aromaticity. <i>Computational and Theoretical Chemistry</i> , 2005, 727, 165-171.	1.5	59
24	All-Metal Antiaromaticity in Sb ₄ -Type Lanthanocene Anions. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 5531-5535.	13.8	59
25	Performance of DFT functionals for calculating the second-order nonlinear optical properties of dipolar merocyanines. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 16579-16594.	2.8	58
26	Aromaticity and electronic delocalization in all-metal clusters with single, double, and triple aromatic character. <i>Theoretical Chemistry Accounts</i> , 2011, 128, 419-431.	1.4	57
27	Peculiar All-Metal If-Aromaticity of the [Au ₂ Sb ₁₆] ⁴⁺ Anion in the Solid State. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 15344-15346.	13.8	52
28	Toward a Unique Definition of the Local Spin. <i>Journal of Chemical Theory and Computation</i> , 2012, 8, 1270-1279.	5.3	51
29	Local Descriptors of Dynamic and Nondynamic Correlation. <i>Journal of Chemical Theory and Computation</i> , 2017, 13, 2705-2711.	5.3	51
30	Electron delocalization and aromaticity measures within the Hückel molecular orbital method. <i>Computational and Theoretical Chemistry</i> , 2007, 811, 3-11.	1.5	46
31	Electron Fluctuation in Pericyclic and Pseudopericyclic Reactions. <i>ChemPhysChem</i> , 2006, 7, 111-113.	2.1	45
32	Scalar and Spin-Orbit Relativistic Corrections to the NICS and the Induced Magnetic Field: The case of the E ₁₂ ²⁺ Spherenes (E = Ge, Sn, Pb). <i>Journal of Chemical Theory and Computation</i> , 2010, 6, 2701-2705.	5.3	44
33	C-O Bond Formation Mediated by a Hexanuclear Iron Complex Supported on a Stannoxane Core. <i>Chemistry - A European Journal</i> , 2012, 18, 2787-2791.	3.3	44
34	Tuning aromaticity in trigonal alkaline earth metal clusters and their alkali metal salts. <i>Journal of Computational Chemistry</i> , 2009, 30, 2764-2776.	3.3	43
35	How do the Hückel and Baird Rules Fade away in Annulenes?. <i>Molecules</i> , 2020, 25, 711.	3.8	43
36	Bonding in Methylalkalimetals (CH ₃ M) _n (M = Li, Na, K; n = 1, 4). Agreement and Divergences between AIM and ELF Analyses. <i>Journal of Physical Chemistry B</i> , 2006, 110, 7189-7198.	2.6	39

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37	Using Electronic Energy Derivative Information in Automated Potential Energy Surface Construction for Vibrational Calculations. <i>Journal of Chemical Theory and Computation</i> , 2010, 6, 3162-3175.	5.3	39
38	Analysis of Hückel's $4n + 2$ Rule through Electronic Delocalization Measures. <i>Journal of Physical Chemistry A</i> , 2008, 112, 13231-13238.	2.5	38
39	Patterns of π -electron delocalization in aromatic and antiaromatic organic compounds in the light of Hückel's $4n + 2$ rule. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 7126.	2.8	38
40	How Aromatic Are Molecular Nanorings? The Case of a Six-Porphyrin Nanoring**. <i>Angewandte Chemie - International Edition</i> , 2021, 60, 24080-24088.	13.8	38
41	A theoretical study of the aromaticity in neutral and anionic borole compounds. <i>Dalton Transactions</i> , 2015, 44, 6740-6747.	3.3	37
42	Comprehensive benchmarking of density matrix functional approximations. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 24029-24041.	2.8	37
43	Performance of 3D-based atoms-in-molecules methods for electronic delocalization aromaticity indices. <i>Journal of Computational Chemistry</i> , 2011, 32, 386-395.	3.3	36
44	The aromaticity of dicupra[10]annulenes. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 9669-9675.	2.8	33
45	Overcoming the Failure of Correlation for Out-of-Plane Motions in a Simple Aromatic: Rovibrational Quantum Chemical Analysis of C_3H_2 . <i>Journal of Chemical Theory and Computation</i> , 2018, 14, 2155-2164.	5.3	33
46	Properties of harmonium atoms from FCI calculations: Calibration and benchmarks for the ground state of the two-electron species. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 6712.	2.8	31
47	New Approximation to the Third-Order Density. Application to the Calculation of Correlated Multicenter Indices. <i>Journal of Chemical Theory and Computation</i> , 2014, 10, 3055-3065.	5.3	31
48	A hierarchy of potential energy surfaces constructed from energies and energy derivatives calculated on grids. <i>Journal of Chemical Physics</i> , 2009, 130, 134104.	3.0	30
49	Understanding Conjugation and Hyperconjugation from Electronic Delocalization Measures. <i>Journal of Physical Chemistry A</i> , 2011, 115, 13104-13113.	2.5	30
50	New Link between Conceptual Density Functional Theory and Electron Delocalization. <i>Journal of Physical Chemistry A</i> , 2011, 115, 12459-12462.	2.5	30
51	Local spins: improved Hilbert-space analysis. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 15291.	2.8	30
52	Vibrational coupled cluster theory with full two-mode and approximate three-mode couplings: The VCC[2pt3] model. <i>Journal of Chemical Physics</i> , 2009, 131, 034115.	3.0	29
53	Exploring the Relation Between Intramolecular Conjugation and Band Dispersion in One-Dimensional Polymers. <i>Journal of Physical Chemistry C</i> , 2017, 121, 27118-27125.	3.1	29
54	Analysis of Electron Delocalization in Aromatic Systems: Individual Molecular Orbital Contributions to Para-Delocalization Indexes (PDI). <i>Journal of Physical Chemistry A</i> , 2006, 110, 11569-11574.	2.5	28

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55	Robust validation of approximate 1-matrix functionals with few-electron harmonium atoms. <i>Journal of Chemical Physics</i> , 2015, 143, 214101.	3.0	28
56	Metal Cluster Electrides: A New Type of Molecular Electride with Delocalised Polyattractor Character. <i>Chemistry - A European Journal</i> , 2018, 24, 9853-9859.	3.3	28
57	Singling Out Dynamic and Nondynamic Correlation. <i>Journal of Physical Chemistry Letters</i> , 2019, 10, 4032-4037.	4.6	28
58	A new tuned range-separated density functional for the accurate calculation of second hyperpolarizabilities. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 11871-11880.	2.8	28
59	The three-electron harmonium atom: The lowest-energy doublet and quadruplet states. <i>Journal of Chemical Physics</i> , 2012, 136, 194112.	3.0	27
60	Benchmark calculations on the lowest-energy singlet, triplet, and quintet states of the four-electron harmonium atom. <i>Journal of Chemical Physics</i> , 2014, 141, 044128.	3.0	26
61	Choosing Bad versus Worse: Predictions of Two-Photon-Absorption Strengths Based on Popular Density Functional Approximations. <i>Journal of Chemical Theory and Computation</i> , 2022, 18, 1046-1060.	5.3	26
62	Performance of PNOF5 Natural Orbital Functional for Radical Formation Reactions: Hydrogen Atom Abstraction and C–C and O–O Homolytic Bond Cleavage in Selected Molecules. <i>Journal of Chemical Theory and Computation</i> , 2012, 8, 2646-2652.	5.3	24
63	The Electronic Structure of the Al ₃ ⁺ Anion: Is it Aromatic?. <i>Chemistry - A European Journal</i> , 2015, 21, 9610-9614.	3.3	23
64	Calculation of local spins for correlated wave functions. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 11308.	2.8	22
65	All-metal π -antiaromaticity in dimeric cluster anion $\{[\text{CuGe}_9\text{Mes}]_2\}^{4-}$. <i>Chemical Communications</i> , 2020, 56, 6583-6586.	4.1	22
66	Approximate Inclusion of Triple Excitations in Combined Coupled Cluster/Molecular Mechanics: Calculations of Electronic Excitation Energies in Solution for Acrolein, Water, Formamide, and N-Methylacetamide. <i>Journal of Chemical Theory and Computation</i> , 2010, 6, 839-850.	5.3	21
67	H4: A challenging system for natural orbital functional approximations. <i>Journal of Chemical Physics</i> , 2015, 143, 164112.	3.0	21
68	Can Density Functional Theory Be Trusted for High-Order Electric Properties? The Case of Hydrogen-Bonded Complexes. <i>Journal of Chemical Theory and Computation</i> , 2019, 15, 3570-3579.	5.3	21
69	Ab Initio Design of Chelating Ligands Relevant to Alzheimer's Disease: Influence of Metalloaromaticity. <i>Journal of Physical Chemistry A</i> , 2011, 115, 12659-12666.	2.5	20
70	A Novel Exploration of the Hartree–Fock Homolytic Bond Dissociation Problem in the Hydrogen Molecule by Means of Electron Localization Measures. <i>Journal of Chemical Education</i> , 2006, 83, 1243.	2.3	19
71	Exploring the Potential Energy Surface of E ₂ P ₄ Clusters (E=Group 1–13 Element): The Quest for Inverse Carbon-Free Sandwiches. <i>Chemistry - A European Journal</i> , 2014, 20, 4583-4590.	3.3	19
72	Peculiar All-Metal π -Aromaticity of the [Au ₂ Sb ₁₆] ⁴⁻ Anion in the Solid State. <i>Angewandte Chemie</i> , 2016, 128, 15570-15572.	2.0	19

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73	Molecular structures of M_2N_2 (M and N = B, Al, and Ga) clusters using the gradient embedded genetic algorithm. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 14850.	2.8	18
74	Transition-Metal Ligand of a Tetrahalodiborane. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 412-416.	13.8	18
75	Comment on the "Nature of Bonding in the Thermal Cyclization of (Z)-1,2,4,6-Heptatetraene and Its Heterosubstituted Analogues". <i>Journal of Physical Chemistry B</i> , 2005, 109, 7591-7593.	2.6	17
76	Benchmark Full Configuration Interaction Calculations on the Lowest-Energy 2P and 4P States of the Three-Electron Harmonium Atom. <i>Journal of Chemical Theory and Computation</i> , 2011, 7, 915-920.	5.3	17
77	Two new constraints for the cumulant matrix. <i>Journal of Chemical Physics</i> , 2014, 141, 234101.	3.0	17
78	Fermi and Coulomb correlation effects upon the interacting quantum atoms energy partition. <i>Theoretical Chemistry Accounts</i> , 2016, 135, 1.	1.4	17
79	Guidelines for Tuning the Excited State Baird Hybrid Aromatic Character of Pro-Aromatic Quinoidal Compounds**. <i>Angewandte Chemie - International Edition</i> , 2021, 60, 10255-10265.	13.8	17
80	Tuning the affinity of catechols and salicylic acids towards $Al(III)$: characterization of Al -chelator interactions. <i>Dalton Transactions</i> , 2018, 47, 9592-9607.	3.3	14
81	Impact of van der Waals interactions on the structural and nonlinear optical properties of azobenzene switches. <i>Physical Chemistry Chemical Physics</i> , 2021, 23, 21227-21239.	2.8	14
82	Bonding description of the Harpoon mechanism. <i>Molecular Physics</i> , 2016, 114, 1345-1355.	1.7	13
83	Electron correlation effects in third-order densities. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 4522-4529.	2.8	13
84	Aromaticity descriptors based on electron delocalization. , 2021, , 235-259.		13
85	Note: The weak-correlation limit of the three-electron harmonium atom. <i>Journal of Chemical Physics</i> , 2011, 134, 116101.	3.0	12
86	How Reliable Are Modern Density Functional Approximations to Simulate Vibrational Spectroscopies?. <i>Journal of Physical Chemistry Letters</i> , 2022, 13, 5963-5968.	4.6	12
87	All-Metal Antiaromaticity in Sb_4 -Type Lanthanocene Anions. <i>Angewandte Chemie</i> , 2016, 128, 5621-5625.	2.0	11
88	Salient signature of van der Waals interactions. <i>Physical Review A</i> , 2017, 96, .	2.5	10
89	Benchmark calculations of metal carbonyl cations: relativistic vs. electron correlation effects. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 20080.	2.8	9
90	Cycloreversion of the CO_2 trimer: a paradigmatic pseudopericyclic [2 + 2 + 2] cycloaddition reaction. <i>Organic and Biomolecular Chemistry</i> , 2017, 15, 435-441.	2.8	9

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91	The vibrational auto-adjusting perturbation theory. Theoretical Chemistry Accounts, 2009, 123, 41-49.	1.4	8
92	Three-center bonding analyzed from correlated and uncorrelated third-order reduced density matrices. Computational and Theoretical Chemistry, 2015, 1053, 173-179.	2.5	8
93	Reply to the Correspondence on "How Aromatic Are Molecular Nanorings? The Case of a Six-Membered Porphyrin Nanoring". Angewandte Chemie - International Edition, 2022, 61, .	13.8	8
94	Electron Localization Function at the Correlated Level: A Natural Orbital Formulation. Journal of Chemical Theory and Computation, 2011, 7, 1231-1231.	5.3	7
95	Rules of Aromaticity. Challenges and Advances in Computational Chemistry and Physics, 2016, , 321-335.	0.6	7
96	Überbergangsmetalle-Komplexierung eines Tetrahalogenidborans. Angewandte Chemie, 2018, 130, 419-423.	2.0	7
97	How Many Electrons Does a Molecular Electride Hold?. Journal of Physical Chemistry A, 2021, 125, 4819-4835.	2.5	7
98	How Aromatic Are Molecular Nanorings? The Case of a Six-Membered Porphyrin Nanoring". Angewandte Chemie, 2021, 133, 24282.	2.0	7
99	Bond centred functions in relativistic and non-relativistic calculations for diatomics. Chemical Physics, 2006, 321, 277-284.	1.9	6
100	Aromaticity Analysis by Means of the Quantum Theory of Atoms in Molecules. , 0, , 399-423.		6
101	Electron-Pair Distribution in Chemical Bond Formation. Journal of Physical Chemistry A, 2018, 122, 1916-1923.	2.5	6
102	The electronic structure and stability of germanium tubes Ge ₃₀ H ₁₂ and Ge ₃₃ H ₁₂ . Physical Chemistry Chemical Physics, 2018, 20, 23467-23479.	2.8	6
103	The Coulomb Hole of the Ne Atom. ChemistryOpen, 2019, 8, 411-417.	1.9	6
104	The electron-pair density distribution of the ^{1,3} Î ₂ excited states of H ₂ . Canadian Journal of Chemistry, 2016, 94, 998-1001.	1.1	5
105	Partition of optical properties into orbital contributions. Physical Chemistry Chemical Physics, 2019, 21, 15380-15391.	2.8	5
106	Aromaticity and Chemical Reactivity. , 2009, , .		5
107	Guidelines for Tuning the Excited State H ^{1/4} ckel "Baird Hybrid Aromatic Character of Pro-Aromatic Quinoidal Compounds". Angewandte Chemie, 2021, 133, 10343-10353.	2.0	3
108	Natural range separation of the Coulomb hole. Journal of Chemical Physics, 2022, 156, 184106.	3.0	3

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109	Comment to "A new population analysis: Dipole-moment-conserving charge-set"™ by H. Sato, S. Skaki [Chem. Phys. Lett. 434 (2007) 165]. Chemical Physics Letters, 2008, 451, 169-170.	2.6	0
110	Editorial [Hot Topic: Electron Delocalization in Organic Chemistry (Guest Editors: Dr. Eduard Matito) Tj ETQq0 0 0 rgBT /Overlock 10 Tf 5	2.6	0
111	Exploring the Potential Energy Surface of E2P4Clusters (E=Group...13 Element): The Quest for Inverse Carbon-Free Sandwiches. Chemistry - A European Journal, 2014, 20, 4497-4497.	3.3	0
112	Frontispiece: The Electronic Structure of the Al3 ⁻ Anion: Is it Aromatic?. Chemistry - A European Journal, 2015, 21, n/a-n/a.	3.3	0
113	Frontispiece: How Aromatic Are Molecular Nanorings? The Case of a Six ⁻ Porphyrin Nanoring. Angewandte Chemie - International Edition, 2021, 60, .	13.8	0
114	Frontispiz: How Aromatic Are Molecular Nanorings? The Case of a Six ⁻ Porphyrin Nanoring. Angewandte Chemie, 2021, 133, .	2.0	0
115	Reply to the Correspondence on "How Aromatic Are Molecular Nanorings? The Case of a Six ⁻ Porphyrin Nanoring". Angewandte Chemie, 0, , .	2.0	0