

João Brandão

List of Publications by Year in descending order

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29
papers

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759233

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#	ARTICLE	IF	CITATIONS
19	Quasiclassical and capture studies on the O (1D)+H ₂ →OH+H reaction using a new potential energy surface for H ₂ O. <i>Chemical Physics Letters</i> , 2003, 377, 523-529.	2.6	20
20	Double-valued potential energy surface for H ₂ O derived from accurate ab initio data and including long-range interactions. <i>Journal of Chemical Physics</i> , 2003, 119, 3148-3159.	3.0	25
21	Quasiclassical trajectory calculations of the thermal rate coefficients for the reactions H(D)+O ₂ →OH(D)+O and O+OH(D)→O ₂ +H(D) as a function of temperature. <i>Journal of Chemical Physics</i> , 1992, 96, 5137-5150.	3.0	125
22	Recalibration of a single-valued double many-body expansion potential energy surface for ground-state hydroperoxy and dynamics calculations for the oxygen atom + hydroxyl .fwdarw. oxygen + hydrogen atom reaction. <i>The Journal of Physical Chemistry</i> , 1990, 94, 8073-8080.	2.9	244
23	Accurate diatomic curves for Ne ₂ , Ar ₂ , Kr ₂ and Xe ₂ from the extended Hartree-Fock approximate correlation energy model. <i>Journal of the Chemical Society, Faraday Transactions 2</i> , 1989, 85, 1851-1875.	1.1	19
24	A realistic HFACE potential function for Kr ₂ (X ¹ Σ ^{g+}) from spectroscopic and thermophysical data. <i>Computational and Theoretical Chemistry</i> , 1988, 166, 187-192.	1.5	2
25	A realistic hydroperoxy(X ² A ¹) potential energy surface from the double many-body expansion method. <i>The Journal of Physical Chemistry</i> , 1988, 92, 3732-3742.	2.9	83
26	The rational fraction representation of diatomic potentials. <i>Theoretica Chimica Acta</i> , 1987, 71, 459-465.	0.8	8
27	A double many-body expansion of molecular potential energy functions. <i>Molecular Physics</i> , 1986, 57, 387-414.	1.7	57
28	A simple semi-empirical approach to the intermolecular potential of van der Waals systems. <i>Molecular Physics</i> , 1982, 45, 857-875.	1.7	67
29	Internal energy and temperature of a carbon nanotube. <i>Fullerenes Nanotubes and Carbon Nanostructures</i> , 0, , 1-5.	2.1	0