

# Prasant Kumar Nayak

## List of Publications by Year in descending order

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49  
papers

4,275  
citations

172207

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197535

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51  
docs citations

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times ranked

5340  
citing authors

#	ARTICLE	IF	CITATIONS
1	From Lithium-ion to Sodium-ion Batteries: Advantages, Challenges, and Surprises. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 102-120.	7.2	1,547
2	Review on Challenges and Recent Advances in the Electrochemical Performance of High Capacity Li-rich and Mn-rich Cathode Materials for Li-ion Batteries. <i>Advanced Energy Materials</i> , 2018, 8, 1702397.	10.2	475
3	Al Doping for Mitigating the Capacity Fading and Voltage Decay of Layered Li and Mn-rich Cathodes for Li-ion Batteries. <i>Advanced Energy Materials</i> , 2016, 6, 1502398.	10.2	360
4	Improving Energy Density and Structural Stability of Manganese Oxide Cathodes for Na-Ion Batteries by Structural Lithium Substitution. <i>Chemistry of Materials</i> , 2016, 28, 9064-9076.	3.2	191
5	Von Lithium-zu Natriumionenbatterien: Vorteile, Herausforderungen und Æerraschendes. <i>Angewandte Chemie</i> , 2018, 130, 106-126.	1.6	125
6	Understanding the Role of Minor Molybdenum Doping in $\text{LiNi}_{0.5}\text{Co}_{0.2}\text{Mn}_{0.3}\text{O}_2$ Electrodes: from Structural and Surface Analyses and Theoretical Modeling to Practical Electrochemical Cells. <i>ACS Applied Materials &amp; Interfaces</i> , 2018, 10, 29608-29621.	4.0	97
7	Structural and Electrochemical Evidence of Layered to Spinel Phase Transformation of Li and Mn Rich Layered Cathode Materials of the Formulae $x\text{Li}_{1/3}\text{Mn}_{2/3}\text{O}_2 \cdot (1-x)\text{LiMn}_{1/3}\text{Ni}_{1/3}\text{Co}_{1/3}\text{O}_2$	1.3	93
8	Electrochemical and structural characterization of carbon coated $\text{Li}_{1.2}\text{Mn}_{0.56}\text{Ni}_{0.16}\text{Co}_{0.08}\text{O}_2$ and $\text{Li}_{1.2}\text{Mn}_{0.6}\text{Ni}_{0.2}\text{O}_2$ as cathode materials for Li-ion batteries. <i>Electrochimica Acta</i> , 2014, 137, 546-556.	2.6	91
9	A Co- and Ni-free P2/O3 Biphase Lithium Stabilized Layered Oxide for Sodium-ion Batteries and its Cycling Behavior. <i>Advanced Functional Materials</i> , 2020, 30, 2003364.	7.8	80
10	Effect of Fe in suppressing the discharge voltage decay of high capacity Li-rich cathodes for Li-ion batteries. <i>Journal of Solid State Electrochemistry</i> , 2015, 19, 2781-2792.	1.2	71
11	In situ real-time gravimetric and viscoelastic probing of surface films formation on lithium batteries electrodes. <i>Nature Communications</i> , 2017, 8, 1389.	5.8	69
12	Study of Cathode Materials for Lithium-Ion Batteries: Recent Progress and New Challenges. <i>Inorganics</i> , 2017, 5, 32.	1.2	68
13	Understanding the influence of Mg doping for the stabilization of capacity and higher discharge voltage of Li- and Mn-rich cathodes for Li-ion batteries. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 6142-6152.	1.3	65
14	Exfoliated $\text{MoS}_2$ as Electrode for All-Solid-State Rechargeable Lithium-Ion Batteries. <i>Journal of Physical Chemistry C</i> , 2019, 123, 12126-12134.	1.5	57
15	In Situ Monitoring of Gravimetric and Viscoelastic Changes in 2D Intercalation Electrodes. <i>ACS Energy Letters</i> , 2017, 2, 1407-1415.	8.8	56
16	Electrochemical Performance of Li- and Mn-Rich Cathodes in Full Cells with Prelithiated Graphite Negative Electrodes. <i>ACS Energy Letters</i> , 2017, 2, 544-548.	8.8	49
17	Understanding the Effect of Lithium Bis(oxalato) Borate (LiBOB) on the Structural and Electrochemical Aging of Li and Mn Rich High Capacity $\text{Li}_{1.2}\text{Ni}_{0.16}\text{Mn}_{0.56}\text{Co}_{0.08}\text{O}_2$ Cathodes. <i>Journal of the Electrochemical Society</i> , 2015, 162, A596-A602.	1.3	47
18	Effect of cycling conditions on the electrochemical performance of high capacity Li and Mn-rich cathodes for Li-ion batteries. <i>Journal of Power Sources</i> , 2016, 318, 9-17.	4.0	47

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19	Mesoporous MnO <sub>2</sub> synthesized by using a tri-block copolymer for electrochemical supercapacitor studies. <i>Microporous and Mesoporous Materials</i> , 2011, 143, 206-214.	2.2	46
20	TEM and Raman spectroscopy evidence of layered to spinel phase transformation in layered LiNi <sub>1/3</sub> Mn <sub>1/3</sub> Co <sub>1/3</sub> O <sub>2</sub> upon cycling to higher voltages. <i>Journal of Electroanalytical Chemistry</i> , 2014, 733, 6-19.	1.9	46
21	Electrochemical Performance of a Layered-Spinel Integrated Li[Ni <sub>1/3</sub> Mn <sub>2/3</sub> ]O <sub>2</sub> as a High Capacity Cathode Material for Li-Ion Batteries. <i>Chemistry of Materials</i> , 2015, 27, 2600-2611.	3.2	46
22	Remarkably Improved Electrochemical Performance of Li- and Mn-Rich Cathodes upon Substitution of Mn with Ni. <i>ACS Applied Materials &amp; Interfaces</i> , 2017, 9, 4309-4319.	4.0	39
23	Cobalt Hydroxide as a Capacitor Material: Tuning Its Potential Window. <i>Journal of the Electrochemical Society</i> , 2008, 155, A855.	1.3	37
24	Rapid sonochemical synthesis of mesoporous MnO <sub>2</sub> for supercapacitor applications. <i>Materials Science and Engineering B: Solid-State Materials for Advanced Technology</i> , 2012, 177, 849-854.	1.7	36
25	Collective Phase Transition Dynamics in Microarray Composite Li <sub>x</sub> FePO <sub>4</sub> Electrodes Tracked by in Situ Electrochemical Quartz Crystal Admittance. <i>Journal of Physical Chemistry C</i> , 2013, 117, 15505-15514.	1.5	35
26	Electrochemical performance of Na <sub>0.6</sub> [Li <sub>0.2</sub> Ni <sub>0.2</sub> Mn <sub>0.6</sub> ]O <sub>2</sub> cathodes with high-working average voltage for Na-ion batteries. <i>Journal of Materials Chemistry A</i> , 2017, 5, 5858-5864.	5.2	35
27	Structural Aspects of P2-Type Na <sub>0.67</sub> Mn <sub>0.6</sub> Ni <sub>0.2</sub> Li <sub>0.2</sub> O <sub>2</sub> (MNL) Stabilization by Lithium Defects as a Cathode Material for Sodium-Ion Batteries. <i>Advanced Functional Materials</i> , 2021, 31, 2102939.	7.8	35
28	Simultaneous Electrodeposition of MnO <sub>2</sub> and Mn(OH) <sub>2</sub> for Supercapacitor Studies. <i>Electrochemical and Solid-State Letters</i> , 2009, 12, A115.	2.2	33
29	Improved capacity and stability of integrated Li and Mn rich layered-spinel Li <sub>1.17</sub> Ni <sub>0.25</sub> Mn <sub>1.08</sub> O <sub>3</sub> cathodes for Li-ion batteries. <i>Journal of Materials Chemistry A</i> , 2015, 3, 14598-14608.	5.2	29
30	Electrochemical and Diffusional Investigation of Na <sub>2</sub> Fe <sup>II</sup> PO <sub>4</sub> F Fluorophosphate Sodium Insertion Material Obtained from Fe <sup>III</sup> Precursor. <i>ACS Applied Materials &amp; Interfaces</i> , 2017, 9, 34961-34969.	4.0	28
31	Mesoporous MnO <sub>2</sub> synthesized by hydrothermal route for electrochemical supercapacitor studies. <i>Journal of Solid State Electrochemistry</i> , 2012, 16, 2739-2749.	1.2	24
32	Sonochemical synthesis of LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub> and its electrochemical performance as a cathode material for 5 V Li-ion batteries. <i>Ultrasonics Sonochemistry</i> , 2015, 26, 332-339.	3.8	23
33	Reaching Highly Stable Specific Capacity with Integrated 0.6Li <sub>2</sub> MnO <sub>3</sub> •0.4LiNi <sub>0.6</sub> Co <sub>0.2</sub> Mn <sub>0.2</sub> O <sub>2</sub> Cathode Materials. <i>ChemElectroChem</i> , 2018, 5, 1137-1146.		
34	Reversible Insertion of a Trivalent Cation onto MnO <sub>2</sub> Leading to Enhanced Capacitance. <i>Journal of the Electrochemical Society</i> , 2011, 158, A585.	1.3	21
35	High-Capacity Layered-Spinel Cathodes for Li-Ion Batteries. <i>ChemSusChem</i> , 2016, 9, 2404-2413.	3.6	17
36	An EQCM Investigation of Electrochemical Precipitation of Mn(OH) <sub>2</sub> and Its Capacitance Behavior. <i>Electrochemical and Solid-State Letters</i> , 2010, 13, F29.	2.2	16

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37	Multiphase $\text{LiNi}_{0.33}\text{Mn}_{0.54}\text{Co}_{0.13}\text{O}_2$ Cathode Material with High Capacity Retention for Li-ion Batteries. ChemElectroChem, 2015, 2, 1957-1965.	1.7	16
38	Investigation of $\text{Li}_{1.17}\text{Ni}_{0.20}\text{Mn}_{0.53}\text{Co}_{0.10}\text{O}_2$ as an Interesting Li- and Mn-Rich Layered Oxide Cathode Material through Electrochemistry, Microscopy, and In-Situ Electrochemical Dilatometry. ChemElectroChem, 2019, 6, 2812-2819.	1.7	16
39	The effect of synthesis and zirconium doping on the performance of nickel-rich NCM622 cathode materials for Li-ion batteries. Journal of Solid State Electrochemistry, 2021, 25, 1513-1530.	1.2	14
40	Porous, hollow $\text{Li}_{1.2}\text{Mn}_{0.53}\text{Ni}_{0.13}\text{Co}_{0.13}\text{O}_2$ microspheres as a positive electrode material for Li-ion batteries. Journal of Solid State Electrochemistry, 2017, 21, 437-445.	1.2	12
41	Studies of a layered-spinel $\text{Li}[\text{Ni}_{1/3}\text{Mn}_{2/3}]\text{O}_2$ cathode material for Li-ion batteries synthesized by a hydrothermal precipitation. Materials Science and Engineering B: Solid-State Materials for Advanced Technology, 2016, 213, 131-139.	1.7	11
42	Electrochemical Performance and Ageing Mechanisms of Sol-Gel Synthesized $\text{Na}_{2/3}[\text{Mn}_{3/5}\text{Fe}_{2/5}]\text{O}_2$ for Sodium-ion Batteries. Batteries and Supercaps, 2019, 2, 104-111.	2.4	11
43	Synergistic-effect of high Ni content and Na dopant towards developing a highly stable Li-Rich cathode in Li-ion batteries. Chemical Engineering Journal, 2022, 444, 136503.	6.6	11
44	High Li storage capacity of poorly crystalline porous $\gamma\text{-MnO}_2$ prepared by hydrothermal route. Journal of Electroanalytical Chemistry, 2013, 703, 126-134.	1.9	8
45	An EQCM investigation of capacitance of $\text{MnO}_2$ in electrolytes containing multivalent cations. Journal of Electroanalytical Chemistry, 2012, 685, 37-40.	1.9	7
46	Temperature and potential dependence electrochemical impedance studies of $\text{LiMn}_2\text{O}_4$ . Journal of Applied Electrochemistry, 2014, 44, 61-71.	1.5	4
47	Effect of sonochemistry: Li- and Mn-rich layered high specific capacity cathode materials for Li-ion batteries. Journal of Solid State Electrochemistry, 2016, 20, 1683-1695.	1.2	4
48	Al-Doped Co-Free Layered-Spinel Mn/Ni Oxides as High-Capacity Cathode Materials for Advanced Li-Ion Batteries. ACS Applied Energy Materials, 2022, 5, 4279-4287.	2.5	3
49	Electrochemical insertion of $\text{Sr}^{2+}$ ions onto nano $\gamma\text{-MnO}_2$ particles. Materials Letters, 2010, 64, 1319-1321.	1.3	0