

# Carmen Fernández-González

## List of Publications by Year in descending order

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45  
papers

2,100  
citations

279487

23  
h-index

243296

44  
g-index

45  
all docs

45  
docs citations

45  
times ranked

2658  
citing authors

#	ARTICLE	IF	CITATIONS
1	Advanced Oxidation Processes for the Removal of Antibiotics from Water. An Overview. Water (Switzerland), 2020, 12, 102.	1.2	381
2	Cherry stones as precursor of activated carbons for supercapacitors. Materials Chemistry and Physics, 2009, 114, 323-327.	2.0	180
3	Preparation of activated carbon from cherry stones by chemical activation with ZnCl <sub>2</sub> . Applied Surface Science, 2006, 252, 5967-5971.	3.1	165
4	Preparation of activated carbon from cherry stones by physical activation in air. Influence of the chemical carbonisation with H <sub>2</sub> SO <sub>4</sub> . Journal of Analytical and Applied Pyrolysis, 2012, 94, 131-137.	2.6	89
5	Preparation of activated carbons from cherry stones by activation with potassium hydroxide. Applied Surface Science, 2006, 252, 5980-5983.	3.1	81
6	Preparation of high-quality activated carbon from polyethyleneterephthalate (PET) bottle waste. Its use in the removal of pollutants in aqueous solution. Journal of Environmental Management, 2016, 181, 522-535.	3.8	78
7	FT-IR Analysis of Pyrone and Chromene Structures in Activated Carbon. Energy & Fuels, 2014, 28, 4096-4103.	2.5	76
8	Adsorption of mercury by carbonaceous adsorbents prepared from rubber of tyre wastes. Journal of Hazardous Materials, 2005, 119, 231-238.	6.5	70
9	Preparation and textural characterisation of activated carbon from vine shoots (Vitis vinifera) by H <sub>3</sub> PO <sub>4</sub> Chemical activation. Applied Surface Science, 2006, 252, 5961-5966.	3.1	69
10	Electrical conductivity of activated carbon-metal oxide nanocomposites under compression: a comparison study. Physical Chemistry Chemical Physics, 2014, 16, 25161-25175.	1.3	65
11	Thermal behaviour of lignocellulosic material in the presence of phosphoric acid. Influence of the acid content in the initial solution. Carbon, 2006, 44, 2347-2350.	5.4	64
12	Porous Structure of Activated Carbon Prepared from Cherry Stones by Chemical Activation with Phosphoric Acid. Energy & Fuels, 2007, 21, 2942-2949.	2.5	57
13	Development of adsorbents from used tire rubber. Fuel Processing Technology, 2011, 92, 206-212.	3.7	50
14	Preparation of mesoporous and macroporous materials from rubber of tyre wastes. Microporous and Mesoporous Materials, 2004, 67, 35-41.	2.2	44
15	Preparation of activated carbons from olive-tree wood revisited. I. Chemical activation with H <sub>3</sub> PO <sub>4</sub> . Fuel Processing Technology, 2011, 92, 261-265.	3.7	44
16	Temperature dependence of the electrical conductivity of activated carbons prepared from vine shoots by physical and chemical activation methods. Microporous and Mesoporous Materials, 2015, 209, 90-98.	2.2	44
17	Development of activated carbon from vine shoots by physical and chemical activation methods. Some insight into activation mechanisms. Adsorption, 2011, 17, 621-629.	1.4	43
18	The development of an activated carbon from cherry stones and its use in the removal of ochratoxin A from red wine. Food Control, 2009, 20, 298-303.	2.8	42

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19	Preparation of activated carbon-metal oxide hybrid catalysts: textural characterization. <i>Fuel Processing Technology</i> , 2014, 126, 95-103.	3.7	40
20	Preparation of Activated Carbon-SnO <sub>2</sub> , TiO <sub>2</sub> , and WO <sub>3</sub> Catalysts. Study by FT-IR Spectroscopy. <i>Industrial &amp; Engineering Chemistry Research</i> , 2016, 55, 5200-5206.	1.8	38
21	Adsorption of cadmium on carbonaceous adsorbents developed from used tire rubber. <i>Journal of Environmental Management</i> , 2011, 92, 2193-2200.	3.8	37
22	Preparation of activated carbons from olive-tree wood revisited. II. Physical activation with air. <i>Fuel Processing Technology</i> , 2011, 92, 266-270.	3.7	35
23	Activated carbon surface chemistry: Changes upon impregnation with Al(III), Fe(III) and Zn(II)-metal oxide catalyst precursors from NO <sub>3</sub> <sup>-</sup> aqueous solutions. <i>Arabian Journal of Chemistry</i> , 2019, 12, 3963-3976.	2.3	34
24	Adsorption of bisphenol A by activated carbon developed from PET waste by KOH activation. <i>Environmental Science and Pollution Research</i> , 2021, 28, 24342-24354.	2.7	27
25	Devulcanization and Demineralization of Used Tire Rubber by Thermal Chemical Methods: A Study by X-ray Diffraction. <i>Energy &amp; Fuels</i> , 2010, 24, 3401-3409.	2.5	24
26	Development of Activated Carbon Using Vine Shoots ( <i>Vitis Vinifera</i> ) and Its Use for Wine Treatment. <i>Journal of Agricultural and Food Chemistry</i> , 2005, 53, 644-650.	2.4	23
27	Preparation and Microstructural Characterization of Activated Carbon-Metal Oxide Hybrid Catalysts: New Insights into Reaction Paths. <i>Journal of Materials Science and Technology</i> , 2015, 31, 806-814.	5.6	22
28	Uptake of lead by carbonaceous adsorbents developed from tire rubber. <i>Adsorption</i> , 2008, 14, 591-600.	1.4	20
29	Adsorption kinetics of zinc in multicomponent ionic systems. <i>Journal of Colloid and Interface Science</i> , 2004, 277, 292-298.	5.0	19
30	On the use of a natural peat for the removal of Cr(VI) from aqueous solutions. <i>Journal of Colloid and Interface Science</i> , 2012, 386, 325-332.	5.0	19
31	Catalysis by alkali and alkaline-earth metals of the gasification in CO <sub>2</sub> and steam of chars from a semi-anthracite with high inorganic matter content. <i>Fuel</i> , 1987, 66, 216-222.	3.4	16
32	The characterization of surface properties and steam reactivities of two Spanish coals of high ash content. <i>Fuel</i> , 1986, 65, 991-996.	3.4	14
33	Temperature dependence of dc electrical conductivity of activated carbon-metal oxide nanocomposites. Some insight into conduction mechanisms. <i>Journal of Physics and Chemistry of Solids</i> , 2015, 87, 259-270.	1.9	14
34	Adsorption of mercury from single and multicomponent metal systems on activated carbon developed from cherry stones. <i>Adsorption</i> , 2008, 14, 601-610.	1.4	12
35	Activated carbon from cherry stones by chemical activation: Influence of the impregnation method on porous structure. <i>Journal of Wood Chemistry and Technology</i> , 2017, 37, 148-162.	0.9	11
36	Preparation of activated carbon-metal (hydr) oxide materials by thermal methods. Thermogravimetric-mass spectrometric (TG-MS) analysis. <i>Journal of Analytical and Applied Pyrolysis</i> , 2015, 116, 243-252.	2.6	8

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37	Particle size distribution and morphological changes in activated carbon-metal oxide hybrid catalysts prepared under different heating conditions. <i>Journal of Microscopy</i> , 2016, 261, 227-242.	0.8	8
38	The influence of the impregnation method on yield of activated carbon produced by H <sub>3</sub> PO <sub>4</sub> activation. <i>Materials Letters</i> , 2011, 65, 1423-1426.	1.3	7
39	Electrical conductivity of metal (hydr)oxide-activated carbon composites under compression. A comparison study. <i>Materials Chemistry and Physics</i> , 2015, 152, 113-122.	2.0	7
40	Physico-chemical characterization of activated carbon-metal oxide photocatalysts by immersion calorimetry in benzene and water. <i>Journal of Thermal Analysis and Calorimetry</i> , 2016, 125, 65-74.	2.0	7
41	Catalysis by alkali and alkaline earth metals of the gasification in CO <sub>2</sub> and in steam of chars from a bituminous coal with high inorganic matter content. <i>Thermochimica Acta</i> , 1988, 125, 79-88.	1.2	6
42	Monitoring of Zn(II) and Cd(II) adsorption on activated carbon from aqueous multicomponent solutions by differential pulse polarography (DPP). <i>International Journal of Environmental Analytical Chemistry</i> , 2005, 85, 1051-1063.	1.8	4
43	Adsorption Isotherms of Methylene Blue in Aqueous Solution onto Activated Carbons Developed from Vine Shoots ( <i>Vitis Vinifera</i> ) by Physical and Chemical Methods. <i>Adsorption Science and Technology</i> , 2010, 28, 751-759.	1.5	4
44	Surface morphological characterization of activated carbon-metal (hydr)oxide composites: some insights into the role of the precursor chemistry in aqueous solution. <i>Journal of Dispersion Science and Technology</i> , 2020, 41, 1743-1753.	1.3	2
45	Shock Resistance and Compression Analysis of Concrete in Expanded Polystyrene Formworks (EPSFWs). <i>Materials Science Forum</i> , 0, 636-637, 287-292.	0.3	0