

# Alain Manceau

## List of Publications by Year in descending order

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88  
papers

7,934  
citations

41344

49  
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51608

86  
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88  
all docs

88  
docs citations

88  
times ranked

6356  
citing authors

#	ARTICLE	IF	CITATIONS
1	Response to Comment on "Mercury Isotope Fractionation by Internal Demethylation and Biomineralization Reactions in Seabirds: Implications for Environmental Mercury Science" Principles and Limitations of Source Tracing and Process Tracing with Stable Isotope Signatures. <i>Environmental Science &amp; Technology</i> , 2022, 56, 2065-2068.	10.0	1
2	Comment on "New insights into the biomineralization of mercury selenide nanoparticles through stable isotope analysis in giant petrel tissues". <i>Journal of Hazardous Materials</i> , 2022, 431, 128583.	12.4	2
3	Chemical Information in the $L_{2,3}$ X-ray Absorption Spectra of Molybdenum Compounds by High-Energy-Resolution Detection and Density Functional Theory. <i>Inorganic Chemistry</i> , 2022, 61, 869-881.	4.0	3
4	Crystal Chemistry of Thallium in Marine Ferromanganese Deposits. <i>ACS Earth and Space Chemistry</i> , 2022, 6, 1269-1285.	2.7	9
5	Fossil Bioapatites with Extremely High Concentrations of Rare Earth Elements and Yttrium from Deep-Sea Pelagic Sediments. <i>ACS Earth and Space Chemistry</i> , 2022, 6, 2093-2103.	2.7	6
6	In Vivo Formation of HgSe Nanoparticles and Hg-Tetraselenolate Complex from Methylmercury in Seabirds" Implications for the Hg-Se Antagonism. <i>Environmental Science &amp; Technology</i> , 2021, 55, 1515-1526.	10.0	75
7	Nature of High- and Low-Affinity Metal Surface Sites on Birnessite Nanosheets. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 66-76.	2.7	11
8	Acute Toxicity of Divalent Mercury to Bacteria Explained by the Formation of Dicysteinate and Tetracysteinate Complexes Bound to Proteins in <i>Escherichia coli</i> and <i>Bacillus subtilis</i> . <i>Environmental Science &amp; Technology</i> , 2021, 55, 3612-3623.	10.0	9
9	Mercury in the tissues of five cephalopods species: First data on the nervous system. <i>Science of the Total Environment</i> , 2021, 759, 143907.	8.0	9
10	Isotope Fractionation from <i>In Vivo</i> Methylmercury Detoxification in Waterbirds. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 990-997.	2.7	18
11	Chemical Forms of Mercury in Blue Marlin Billfish: Implications for Human Exposure. <i>Environmental Science and Technology Letters</i> , 2021, 8, 405-411.	8.7	21
12	Chemical Forms of Mercury in Pilot Whales Determined from Species-Averaged Mercury Isotope Signatures. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 1591-1599.	2.7	8
13	Cellular remains in a ~3.42-billion-year-old subseafloor hydrothermal environment. <i>Science Advances</i> , 2021, 7, .	10.3	34
14	Mercury Isotope Fractionation by Internal Demethylation and Biomineralization Reactions in Seabirds: Implications for Environmental Mercury Science. <i>Environmental Science &amp; Technology</i> , 2021, 55, 13942-13952.	10.0	19
15	Demethylation of Methylmercury in Bird, Fish, and Earthworm. <i>Environmental Science &amp; Technology</i> , 2021, 55, 1527-1534.	10.0	61
16	TEXS: in-vacuum tender X-ray emission spectrometer with 11 Johansson crystal analyzers. <i>Journal of Synchrotron Radiation</i> , 2020, 27, 813-826.	2.4	19
17	The Mode of Incorporation of As(-I) and Se(-I) in Natural Pyrite Revisited. <i>ACS Earth and Space Chemistry</i> , 2020, 4, 379-390.	2.7	18
18	The chemical species of mercury accumulated by <i>Pseudomonas idrijaensis</i> , a bacterium from a rock of the Idrija mercury mine, Slovenia.. <i>Chemosphere</i> , 2020, 248, 126002.	8.2	9

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19	Evidence for syngenetic micro-inclusions of As <sup>3+</sup> - and As <sup>5+</sup> -containing Cu sulfides in hydrothermal pyrite. <i>American Mineralogist</i> , 2019, 104, 300-306.	1.9	4
20	Revealing the Chemical Form of "Invisible" Gold in Natural Arsenian Pyrite and Arsenopyrite with High Energy-Resolution X-ray Absorption Spectroscopy. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 1905-1914.	2.7	39
21	Comment on "Roles of Hydration and Magnetism on the Structure of Ferrihydrite from First Principles" <i>ACS Earth and Space Chemistry</i> , 2019, 3, 1576-1580.	2.7	8
22	Frontispiece: Mercury(II) Binding to Metallothionein in <i>Mytilus edulis</i> revealed by High Energy-Resolution XANES Spectroscopy. <i>Chemistry - A European Journal</i> , 2019, 25, .	3.3	0
23	Divalent Mercury in Dissolved Organic Matter Is Bioavailable to Fish and Accumulates as Dithiolate and Tetrathiolate Complexes. <i>Environmental Science &amp; Technology</i> , 2019, 53, 4880-4891.	10.0	30
24	Thiols in Natural Organic Matter: Molecular Forms, Acidity, and Reactivity with Mercury(II) from First-Principles Calculations and High Energy-Resolution X-ray Absorption Near-Edge Structure Spectroscopy. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 2795-2807.	2.7	9
25	Mercury(II) Binding to Metallothionein in <i>Mytilus edulis</i> revealed by High Energy-Resolution XANES Spectroscopy. <i>Chemistry - A European Journal</i> , 2019, 25, 997-1009.	3.3	23
26	Mercury Trithiolate Binding (HgS <sub>3</sub> ) to a de Novo Designed Cyclic Decapeptide with Three Preoriented Cysteine Side Chains. <i>Inorganic Chemistry</i> , 2018, 57, 2705-2713.	4.0	14
27	Biogenesis of Mercury "Sulfur Nanoparticles in Plant Leaves from Atmospheric Gaseous Mercury. <i>Environmental Science &amp; Technology</i> , 2018, 52, 3935-3948.	10.0	75
28	Chemical Forms of Mercury in Pyrite: Implications for Predicting Mercury Releases in Acid Mine Drainage Settings. <i>Environmental Science &amp; Technology</i> , 2018, 52, 10286-10296.	10.0	37
29	Comment on "Crystal growth and aggregation in suspensions of $\gamma$ -MnO <sub>2</sub> nanoparticles: implications for surface reactivity" by F. F. Marafatto, B. Lanson and J. PeÅ±a, <i>Environ. Sci.: Nano</i> , 2018, 5, 497. <i>Environmental Science: Nano</i> , 2018, 5, 2198-2200.	4.3	2
30	High energy-resolution x-ray spectroscopy at ultra-high dilution with spherically bent crystal analyzers of 0.5 m radius. <i>Review of Scientific Instruments</i> , 2017, 88, 013108.	1.3	62
31	Nucleation of mercury sulfide by dealkylation. <i>Scientific Reports</i> , 2016, 6, 39359.	3.3	21
32	Chemical Forms of Mercury in Human Hair Reveal Sources of Exposure. <i>Environmental Science &amp; Technology</i> , 2016, 50, 10721-10729.	10.0	53
33	Structure, Bonding, and Stability of Mercury Complexes with Thiolate and Thioether Ligands from High-Resolution XANES Spectroscopy and First-Principles Calculations. <i>Inorganic Chemistry</i> , 2015, 54, 11776-11791.	4.0	57
34	Comment on "Molecular controls on Cu and Zn isotopic fractionation in Fe "Mn crusts" by Little et al.. <i>Earth and Planetary Science Letters</i> , 2015, 411, 310-312.	4.4	9
35	Formation of Mercury Sulfide from Hg(II) "Thiolate Complexes in Natural Organic Matter. <i>Environmental Science &amp; Technology</i> , 2015, 49, 9787-9796.	10.0	111
36	Estimating the number of pure chemical components in a mixture by X-ray absorption spectroscopy. <i>Journal of Synchrotron Radiation</i> , 2014, 21, 1140-1147.	2.4	39

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37	PDF analysis of ferrihydrite: Critical assessment of the under-constrained akdalaite model. <i>American Mineralogist</i> , 2014, 99, 102-108.	1.9	27
38	High-level ab initio calculation of the stability of mercury–thiolate complexes. <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	14
39	Mineralogy and crystal chemistry of Mn, Fe, Co, Ni, and Cu in a deep-sea Pacific polymetallic nodule. <i>American Mineralogist</i> , 2014, 99, 2068-2083.	1.9	106
40	Short-range and long-range order of phylломanganate nanoparticles determined using high-energy X-ray scattering. <i>Journal of Applied Crystallography</i> , 2013, 46, 193-209.	4.5	70
41	Incorporation of Ge in ferrihydrite: Implications for the structure of ferrihydrite. <i>American Mineralogist</i> , 2013, 98, 848-858.	1.9	19
42	Determination of Mn valence states in mixed-valent manganates by XANES spectroscopy. <i>American Mineralogist</i> , 2012, 97, 816-827.	1.9	256
43	Zn sorption modifies dynamically the layer and interlayer structure of vernadite. <i>Geochimica Et Cosmochimica Acta</i> , 2012, 85, 302-313.	3.9	110
44	Quantitative analysis of sulfur functional groups in natural organic matter by XANES spectroscopy. <i>Geochimica Et Cosmochimica Acta</i> , 2012, 99, 206-223.	3.9	146
45	Critical evaluation of the revised akdalaite model for ferrihydrite–Reply. <i>American Mineralogist</i> , 2012, 97, 255-256.	1.9	17
46	Mercury–Sequestering Pseudopeptides with a Tris(cysteine) Environment in Water. <i>European Journal of Inorganic Chemistry</i> , 2012, 2012, 3835-3843.	2.0	26
47	Metallothionein-Like Multinuclear Clusters of Mercury(II) and Sulfur in Peat. <i>Environmental Science &amp; Technology</i> , 2011, 45, 7298-7306.	10.0	59
48	Structure of nanocrystalline phylломanganates produced by freshwater fungi. <i>American Mineralogist</i> , 2010, 95, 1608-1616.	1.9	138
49	The nature of Cu bonding to natural organic matter. <i>Geochimica Et Cosmochimica Acta</i> , 2010, 74, 2556-2580.	3.9	162
50	Zinc distribution and speciation in <i>Arabidopsis halleri</i> – <i>Arabidopsis lyrata</i> progenies presenting various zinc accumulation capacities. <i>New Phytologist</i> , 2009, 184, 581-595.	7.3	114
51	Formation of Zn–Ca phylломanganate nanoparticles in grass roots. <i>Geochimica Et Cosmochimica Acta</i> , 2008, 72, 2478-2490.	3.9	74
52	Formation of Metallic Copper Nanoparticles at the Soil–Root Interface. <i>Environmental Science &amp; Technology</i> , 2008, 42, 1766-1772.	10.0	221
53	Relationships between Hg(II)–S bond distance and Hg(II) coordination in thiolates. <i>Dalton Transactions</i> , 2008, , 1421.	3.3	73
54	Natural speciation of Ni, Zn, Ba, and As in ferromanganese coatings on quartz using X-ray fluorescence, absorption, and diffraction. <i>Geochimica Et Cosmochimica Acta</i> , 2007, 71, 95-128.	3.9	204

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55	Chemical and structural control of the partitioning of Co, Ce, and Pb in marine ferromanganese oxides. <i>Geochimica Et Cosmochimica Acta</i> , 2007, 71, 984-1008.	3.9	249
56	Ba and Ni speciation in a nodule of binary Mn oxide phase composition from Lake Baikal. <i>Geochimica Et Cosmochimica Acta</i> , 2007, 71, 1967-1981.	3.9	73
57	Formation of todorokite from vernadite in Ni-rich hemipelagic sediments. <i>Geochimica Et Cosmochimica Acta</i> , 2007, 71, 5698-5716.	3.9	145
58	Structure of the synthetic K-rich phyllomanganate birnessite obtained by high-temperature decomposition of KMnO <sub>4</sub> . <i>Microporous and Mesoporous Materials</i> , 2007, 98, 267-282.	4.4	72
59	Structural model for the biogenic Mn oxide produced by <i>Pseudomonas putida</i> . <i>American Mineralogist</i> , 2006, 91, 489-502.	1.9	288
60	Zinc sorption to biogenic hexagonal-birnessite particles within a hydrated bacterial biofilm. <i>Geochimica Et Cosmochimica Acta</i> , 2006, 70, 27-43.	3.9	177
61	Nanometer-sized, divalent-Mn, hydrous silicate domains in geothermal brine precipitates. <i>American Mineralogist</i> , 2005, 90, 371-381.	1.9	13
62	Zinc mobility and speciation in soil covered by contaminated dredged sediment using micrometer-scale and bulk-averaging X-ray fluorescence, absorption and diffraction techniques. <i>Geochimica Et Cosmochimica Acta</i> , 2005, 69, 1173-1198.	3.9	89
63	Natural speciation of Mn, Ni, and Zn at the micrometer scale in a clayey paddy soil using X-ray fluorescence, absorption, and diffraction. <i>Geochimica Et Cosmochimica Acta</i> , 2005, 69, 4007-4034.	3.9	109
64	Zinc Sorption by a Bacterial Biofilm. <i>Environmental Science &amp; Technology</i> , 2005, 39, 8288-8294.	10.0	105
65	Beamline 10.3.2 at ALS: a hard X-ray microprobe for environmental and materials sciences. <i>Journal of Synchrotron Radiation</i> , 2004, 11, 239-247.	2.4	245
66	Natural speciation of Zn at the micrometer scale in a clayey soil using X-ray fluorescence, absorption, and diffraction. <i>Geochimica Et Cosmochimica Acta</i> , 2004, 68, 2467-2483.	3.9	156
67	Mn, Fe, Zn and As speciation in a fast-growing ferromanganese marine nodule. <i>Geochimica Et Cosmochimica Acta</i> , 2004, 68, 3125-3136.	3.9	142
68	Analysis of the Major Fe Bearing Mineral Phases in Recent Lake Sediments by EXAFS Spectroscopy. <i>Aquatic Geochemistry</i> , 2003, 9, 1-17.	1.3	22
69	Structure of Synthetic K-rich Birnessite Obtained by High-Temperature Decomposition of KMnO <sub>4</sub> . I. Two-Layer Polytype from 800 Å°C Experiment. <i>Chemistry of Materials</i> , 2003, 15, 4666-4678.	6.7	169
70	Molecular-Scale Speciation of Zn and Ni in Soil Ferromanganese Nodules from Loess Soils of the Mississippi Basin. <i>Environmental Science &amp; Technology</i> , 2003, 37, 75-80.	10.0	171
71	Structure of heavy-metal sorbed birnessite: Part 1. Results from X-ray diffraction. <i>American Mineralogist</i> , 2002, 87, 1631-1645.	1.9	115
72	Structure of heavy-metal sorbed birnessite: Part 2. Results from electron diffraction. <i>American Mineralogist</i> , 2002, 87, 1646-1661.	1.9	42

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73	Structure of synthetic Na-birnessite: Evidence for a triclinic one-layer unit cell. <i>American Mineralogist</i> , 2002, 87, 1662-1671.	1.9	152
74	Deciphering Ni sequestration in soil ferromanganese nodules by combining X-ray fluorescence, absorption, and diffraction at micrometer scales of resolution. <i>American Mineralogist</i> , 2002, 87, 1494-1499.	1.9	102
75	Arsenic(III) Oxidation by Birnessite and Precipitation of Manganese(II) Arsenate. <i>Environmental Science &amp; Technology</i> , 2002, 36, 493-500.	10.0	294
76	Quantitative Zn speciation in a contaminated dredged sediment by $\hat{1}/4$ -PIXE, $\hat{1}/4$ -SXRF, EXAFS spectroscopy and principal component analysis. <i>Geochimica Et Cosmochimica Acta</i> , 2002, 66, 1549-1567.	3.9	154
77	Structure of heavy metal sorbed birnessite. Part III: Results from powder and polarized extended X-ray absorption fine structure spectroscopy. <i>Geochimica Et Cosmochimica Acta</i> , 2002, 66, 2639-2663.	3.9	242
78	7. Quantitative Speciation of Heavy Metals in Soils and Sediments by Synchrotron X-ray Techniques. , 2002, , 341-428.		103
79	Accumulation Forms of Zn and Pb in <i>Phaseolus vulgaris</i> in the Presence and Absence of EDTA. <i>Environmental Science &amp; Technology</i> , 2001, 35, 2854-2859.	10.0	185
80	Structure of H-exchanged hexagonal birnessite and its mechanism of formation from Na-rich monoclinic busserite at low pH. <i>American Mineralogist</i> , 2000, 85, 826-838.	1.9	191
81	Structure of synthetic monoclinic Na-rich birnessite and hexagonal birnessite; II, Results from chemical studies and EXAFS spectroscopy. <i>American Mineralogist</i> , 1997, 82, 962-978.	1.9	256
82	Structure of synthetic monoclinic Na-rich birnessite and hexagonal birnessite; I, Results from X-ray diffraction and selected-area electron diffraction. <i>American Mineralogist</i> , 1997, 82, 946-961.	1.9	353
83	Removal of Selenocyanate in Water by Precipitation: Characterization of Copper-Selenium Precipitate by X-ray Diffraction, Infrared, and X-ray Absorption Spectroscopy. <i>Environmental Science &amp; Technology</i> , 1997, 31, 968-976.	10.0	42
84	Structure and Mechanisms of Formation of FeOOH(NO <sub>3</sub> ) Oligomers in the Early Stages of Hydrolysis. <i>Langmuir</i> , 1997, 13, 3240-3246.	3.5	59
85	Structural mechanism of Co (super 2+) oxidation by the phyllosulfate busserite. <i>American Mineralogist</i> , 1997, 82, 1150-1175.	1.9	235
86	Reaction Rates and Products of Manganese Oxidation at the Sediment-Water Interface. <i>Advances in Chemistry Series</i> , 1995, , 111-134.	0.6	61
87	Structure and Stability of Cd <sup>2+</sup> Surface Complexes on Ferric Oxides. <i>Journal of Colloid and Interface Science</i> , 1994, 168, 73-86.	9.4	215
88	Structure and mechanisms of formation of iron oxide hydroxide (chloride) polymers. <i>Langmuir</i> , 1994, 10, 316-319.	3.5	147