

# Artur Terzyk

## List of Publications by Year in descending order

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207  
papers

5,719  
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docs citations

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times ranked

5536  
citing authors

#	ARTICLE	IF	CITATIONS
1	Hedgehog-like structure, PVDF- carbon nanohorn hybrid membranes for improved removal of VOCs from water. Chemical Engineering Journal, 2022, 438, 135574.	12.7	14
2	Biomimetically Inspired Highly Homogeneous Hydrophilization of Graphene with Poly( <i>l</i> -DOPA): Toward Electroconductive Coatings from Water-Processable Paints. ACS Sustainable Chemistry and Engineering, 2022, 10, 6596-6608.	6.7	1
3	Are nanohedgehogs thirsty? Toward new superhydrophobic and anti-icing carbon nanohorn-polymer hybrid surfaces. Chemical Engineering Journal, 2022, 446, 137126.	12.7	11
4	MOF materials as therapeutic agents, drug carriers, imaging agents and biosensors in cancer biomedicine: Recent advances and perspectives. Progress in Materials Science, 2021, 117, 100743.	32.8	120
5	Recent Developments in the Electrophoretic Deposition of Carbon Nanomaterials. Engineering Materials, 2021, , 113-137.	0.6	1
6	Nitric-Acid Oxidized Single-Walled Carbon Nanohorns as a Potential Material for Bio-Applications – Toxicity and Hemocompatibility Studies. Materials, 2021, 14, 1419.	2.9	7
7	A New Approach to Obtaining Nano-Sized Graphene Oxide for Biomedical Applications. Materials, 2021, 14, 1327.	2.9	5
8	Ultra-long carbon nanotube-paraffin composites of record thermal conductivity and high phase change enthalpy among paraffin-based heat storage materials. Journal of Energy Storage, 2021, 36, 102396.	8.1	52
9	Revisiting Wetting, Freezing, and Evaporation Mechanisms of Water on Copper. ACS Applied Materials & Interfaces, 2021, 13, 37893-37903.	8.0	17
10	Insight into the Mechanisms of Low Coverage Adsorption of N-Alcohols on Single Walled Carbon Nanohorn. Materials, 2021, 14, 4001.	2.9	2
11	Carbon nanohorn improved durable PVDF membranes - The future of membrane distillation and desalination. Desalination, 2021, 511, 115117.	8.2	11
12	Liquid phase adsorption induced nanosizing of graphene oxide. Carbon, 2021, 183, 948-957.	10.3	6
13	Linking the Defective Structure of Boron-Doped Carbon Nano-Onions with Their Catalytic Properties: Experimental and Theoretical Studies. ACS Applied Materials & Interfaces, 2021, 13, 51628-51642.	8.0	5
14	Chasing the Critical Wetting Transition. An Effective Interface Potential Method. Materials, 2021, 14, 7138.	2.9	3
15	Opening the internal structure for transport of ions: improvement of the structural and chemical properties of single-walled carbon nanohorns for supercapacitor electrodes. RSC Advances, 2020, 10, 38357-38368.	3.6	6
16	Reconstructing the fractal clusters of detonation nanodiamonds from small-angle X-ray scattering. Carbon, 2020, 169, 349-356.	10.3	8
17	Cytotoxic or Not? Disclosing the Toxic Nature of Carbonaceous Nanomaterials through Nano-Bio Interactions. Materials, 2020, 13, 2060.	2.9	18
18	Atomic-scale molecular models of oxidized activated carbon fibre nanoregions: Examining the effects of oxygen functionalities on wet formaldehyde adsorption. Carbon, 2020, 165, 67-81.	10.3	19

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19	Non-thermal plasma-assisted catalytic CO <sub>2</sub> conversion over Zn-TCPP 2D catalyst. <i>Adsorption</i> , 2020, 26, 1165-1171.	3.0	5
20	Electrophoretic deposition of spherical carbon nanoobjects – A comparison of different biocompatible surfaces. <i>Medical Devices &amp; Sensors</i> , 2020, 3, e10075.	2.7	2
21	Electrophoretic Deposition of Layer-by-Layer Unsheathed Carbon Nanotubes – A Step Towards Steerable Surface Roughness and Wettability. <i>Materials</i> , 2020, 13, 595.	2.9	6
22	What Is the Value of Water Contact Angle on Silicon?. <i>Materials</i> , 2020, 13, 1554.	2.9	27
23	Mechanistic aspects of water adsorption-desorption in porphyrin containing MOFs. <i>Microporous and Mesoporous Materials</i> , 2019, 290, 109649.	4.4	9
24	Testing the self-cleaning properties of a coordination polymer surface. <i>Adsorption</i> , 2019, 25, 33-39.	3.0	1
25	Stability of coordination polymers in water: state of the art and towards a methodology for nonporous materials. <i>Adsorption</i> , 2019, 25, 1-11.	3.0	10
26	New strategy of controlled, stepwise release from novel MBioF and its potential application for drug delivery systems. <i>Adsorption</i> , 2019, 25, 383-391.	3.0	3
27	Correlation between the catalytic and electrocatalytic properties of nitrogen-doped carbon nanooxions and the polarity of the carbon surface: Experimental and theoretical investigations. <i>Carbon</i> , 2019, 151, 120-129.	10.3	11
28	Ullmann Reactions of Carbon Nanotubes – Advantageous and Unexplored Functionalization toward Tunable Surface Chemistry. <i>Nanomaterials</i> , 2019, 9, 1619.	4.1	9
29	Selective carboxylation versus layer-by-layer unsheathing of multi-walled carbon nanotubes: new insights from the reaction with boiling nitrating mixture. <i>RSC Advances</i> , 2019, 9, 37608-37613.	3.6	14
30	Water Nanodroplet on a Hydrocarbon – Carpet – The Mechanism of Water Contact Angle Stabilization by Airborne Contaminations on Graphene, Au, and PTFE Surfaces. <i>Langmuir</i> , 2019, 35, 420-427.	3.5	17
31	Super-sieving effect in phenol adsorption from aqueous solutions on nanoporous carbon beads. <i>Carbon</i> , 2018, 135, 12-20.	10.3	34
32	Nanoscale Water Contact Angle on Polytetrafluoroethylene Surfaces Characterized by Molecular Dynamics – Atomic Force Microscopy Imaging. <i>Langmuir</i> , 2018, 34, 4526-4534.	3.5	37
33	Cystine-based MBioF for Maintaining the Antioxidant – Oxidant Balance in Airway Diseases. <i>ACS Medicinal Chemistry Letters</i> , 2018, 9, 1280-1284.	2.8	6
34	Phenol Molecular Sheets Woven by Water Cavities in Hydrophobic Slit Nanospaces. <i>Langmuir</i> , 2018, 34, 15150-15159.	3.5	1
35	Switchable hydrophobicity/hydrophilicity of a HOPG surface - Comment on the paper by Y. Wei and C.Q. Jia, <i>Carbon</i> , 87 (2015) 10-17. <i>Carbon</i> , 2017, 115, 571-573.	10.3	10
36	New forcefield for water nanodroplet on a graphene surface. <i>Chemical Physics Letters</i> , 2017, 674, 98-102.	2.6	21

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37	Controlling enzymatic activity by immobilization on graphene oxide. <i>Die Naturwissenschaften</i> , 2017, 104, 36.	1.6	37
38	Molecular simulation aided nanoporous carbon design for highly efficient low-concentrated formaldehyde capture. <i>Carbon</i> , 2017, 124, 152-160.	10.3	30
39	Water Adsorption Property of Hierarchically Nanoporous Detonation Nanodiamonds. <i>Langmuir</i> , 2017, 33, 11180-11188.	3.5	28
40	Nanoscale Insight into the Mechanism of a Highly Oriented Pyrolytic Graphite Edge Surface Wetting by Interferencing Water. <i>Langmuir</i> , 2017, 33, 8562-8573.	3.5	4
41	CO <sub>2</sub> - Reinforced nanoporous carbon potential energy field during CO <sub>2</sub> /CH <sub>4</sub> mixture adsorption. A comprehensive volumetric, in-situ IR, and thermodynamic insight. <i>Carbon</i> , 2017, 122, 185-193.	10.3	5
42	Morphologically disordered pore model for characterization of micro-mesoporous carbons. <i>Carbon</i> , 2017, 111, 358-370.	10.3	25
43	To what extent can mutual shifting of folded carbonaceous walls in slit-like pores affect their adsorption properties?. <i>Journal of Physics Condensed Matter</i> , 2016, 28, 015002.	1.8	1
44	New findings on the influence of carbon surface curvature on energetics of benzene adsorption from gaseous phase. <i>Chemical Physics Letters</i> , 2016, 645, 157-163.	2.6	4
45	Carbon Nanohorns. , 2016, , 75-114.		1
46	Cubic Carbon Polymorphs. , 2016, , 141-156.		0
47	The influence of geometric heterogeneity of closed carbon nanotube bundles on benzene adsorption from the gaseous phase-Monte Carlo simulations. <i>Adsorption</i> , 2016, 22, 639-651.	3.0	8
48	Using in-situ adsorption dilatometry for assessment of micropore size distribution in monolithic carbons. <i>Carbon</i> , 2016, 103, 263-272.	10.3	36
49	Dynamics of effusive and diffusive gas separation on pillared graphene. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 17018-17023.	2.8	14
50	Nano-Structured Carbon Matrixes Obtained from Chitin and Chitosan by a Novel Method. <i>Journal of Nanoscience and Nanotechnology</i> , 2016, 16, 2623-2631.	0.9	12
51	Phenol adsorption on different nano-sized carbon materials: first comparative study. <i>Adsorption</i> , 2016, 22, 437-444.	3.0	4
52	Water nanodroplet on a graphene surface—a new old system. <i>Journal of Physics Condensed Matter</i> , 2016, 28, 495002.	1.8	13
53	The Chemistry of Bioconjugation in Nanoparticles-Based Drug Delivery System. <i>Advances in Condensed Matter Physics</i> , 2015, 2015, 1-27.	1.1	75
54	Conscious Changes of Carbon Nanotubes Cytotoxicity by Manipulation with Selected Nanofactors. <i>Applied Biochemistry and Biotechnology</i> , 2015, 176, 730-741.	2.9	12

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55	New application of carbon nanotubes in haemostatic dressing filled with anticancer substance. <i>Biomedicine and Pharmacotherapy</i> , 2015, 69, 349-354.	5.6	10
56	Water at Curved Carbon Surface: Mechanisms of Adsorption Revealed by First Calorimetric Study. <i>Journal of Physical Chemistry C</i> , 2015, 119, 2703-2715.	3.1	10
57	New insights into the ideal adsorbed solution theory. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 7232-7247.	2.8	25
58	Intrinsic $D_2/H_2$ Selectivity of NaX Zeolite: Interplay between Adsorption and Kinetic Factors. <i>Journal of Physical Chemistry C</i> , 2015, 119, 15373-15380.	3.1	16
59	Effects of Critical Fluctuations on Adsorption-Induced Deformation of Microporous Carbons. <i>Journal of Physical Chemistry C</i> , 2015, 119, 6111-6120.	3.1	8
60	Nuclear Quantum Effects in the Layering and Diffusion of Hydrogen Isotopes in Carbon Nanotubes. <i>Journal of Physical Chemistry Letters</i> , 2015, 6, 3367-3372.	4.6	15
61	Properties of Phenol Confined in Realistic Carbon Micropore Model: Experiment and Simulation. <i>Journal of Physical Chemistry C</i> , 2015, 119, 19987-19995.	3.1	14
62	New findings on the influence of carbon surface curvature on energetics of benzene adsorption from aqueous solutions. <i>Chemical Physics Letters</i> , 2015, 619, 219-222.	2.6	8
63	Nanovehicles as a novel target strategy for hyperthermic intraperitoneal chemotherapy: a multidisciplinary study of peritoneal carcinomatosis. <i>Oncotarget</i> , 2015, 6, 22776-22798.	1.8	18
64	Folding of graphene slit like pore walls—a simple method of improving $CO_2$ separation from mixtures with $CH_4$ or $N_2$ . <i>Journal of Physics Condensed Matter</i> , 2014, 26, 485006.	1.8	7
65	Synthesis of carbon nanotubes and nanotube forests on copper catalyst. <i>Materials Research Express</i> , 2014, 1, 035040.	1.6	11
66	MD simulation of organics adsorption from aqueous solution in carbon slit-like pores. Foundations of the pore blocking effect. <i>Journal of Physics Condensed Matter</i> , 2014, 26, 055008.	1.8	10
67	Nanotube-mediated efficiency of cisplatin anticancer therapy. <i>Carbon</i> , 2014, 70, 46-58.	10.3	22
68	Toward in silico modeling of palladium–hydrogen–carbon nanohorn nanocomposites. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 11763-11769.	2.8	5
69	Carbon Molecular Sieves: Reconstruction of Atomistic Structural Models with Experimental Constraints. <i>Journal of Physical Chemistry C</i> , 2014, 118, 12996-13007.	3.1	21
70	Surface to volume ratio of carbon nanohorn – A crucial factor in $CO_2/CH_4$ mixture separation. <i>Chemical Physics Letters</i> , 2014, 595-596, 67-72.	2.6	7
71	Carbon nanotubes as potential material for drug delivery—experiment and simulation. <i>Adsorption</i> , 2013, 19, 269-272.	3.0	5
72	Synergetic effect of carbon nanopore size and surface oxidation on $CO_2$ capture from $CO_2/CH_4$ mixtures. <i>Journal of Colloid and Interface Science</i> , 2013, 397, 144-153.	9.4	42

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73	Porosity of closed carbon nanotubes compressed using hydraulic pressure. <i>Adsorption</i> , 2013, 19, 785-793.	3.0	4
74	Carbon materials as new nanovehicles in hot-melt drug deposition. <i>Journal of Physics Condensed Matter</i> , 2013, 25, 355002.	1.8	9
75	To the pore and through the pore: thermodynamics and kinetics of helium in exotic cubic carbon polymorphs. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 17366.	2.8	6
76	The first atomistic modelling-aided reproduction of morphologically defective single walled carbon nanohorns. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 1232-1240.	2.8	10
77	Influence of activated carbon surface oxygen functionalities on SO <sub>2</sub> physisorption – Simulation and experiment. <i>Chemical Physics Letters</i> , 2013, 578, 85-91.	2.6	32
78	Screening of carbonaceous nanoporous materials for capture of nerve agents. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 291-298.	2.8	25
79	Constant Pressure Path Integral Gibbs Ensemble Monte Carlo Method. <i>Journal of Chemical Theory and Computation</i> , 2013, 9, 2922-2929.	5.3	11
80	Detecting adsorption space in carbon nanotubes by benzene uptake. <i>Journal of Colloid and Interface Science</i> , 2013, 391, 74-85.	9.4	13
81	Applicability of molecular simulations for modelling the adsorption of the greenhouse gas CF <sub>4</sub> on carbons. <i>Journal of Physics Condensed Matter</i> , 2013, 25, 015004.	1.8	10
82	Separation of CO <sub>2</sub> –CH <sub>4</sub> mixtures on defective single walled carbon nanohorns – tip does matter. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 16468.	2.8	15
83	Simulation of SF <sub>6</sub> adsorption on the bundles of single walled carbon nanotubes. <i>Microporous and Mesoporous Materials</i> , 2012, 154, 51-55.	4.4	15
84	Cryogenic Noble Gas Separation without Distillation: The Effect of Carbon Surface Curvature on Adsorptive Separation. <i>Journal of Physical Chemistry C</i> , 2012, 116, 19363-19371.	3.1	6
85	Methane-Induced Deformation of Porous Carbons: From Normal to High-Pressure Operating Conditions. <i>Journal of Physical Chemistry C</i> , 2012, 116, 1740-1747.	3.1	24
86	Virtual Porous Carbons. , 2012, , 61-104.		10
87	Displacement of Methane by Coadsorbed Carbon Dioxide Is Facilitated In Narrow Carbon Nanopores. <i>Journal of Physical Chemistry C</i> , 2012, 116, 13640-13649.	3.1	48
88	Structural properties of amorphous diamond-like carbon: percolation, cluster, and pair correlation analysis. <i>RSC Advances</i> , 2012, 2, 4292.	3.6	18
89	Enhanced adsorption of paracetamol on closed carbon nanotubes by formation of nanoaggregates: Carbon nanotubes as potential materials in hot-melt drug deposition-experiment and simulation. <i>Journal of Colloid and Interface Science</i> , 2012, 376, 209-216.	9.4	19
90	Removal of internal caps during hydrothermal treatment of bamboo-like carbon nanotubes and application of tubes in phenol adsorption. <i>Journal of Colloid and Interface Science</i> , 2012, 381, 36-42.	9.4	30

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91	Quantum fluctuations increase the self-diffusive motion of para-hydrogen in narrow carbon nanotubes. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 9824.	2.8	4
92	Pillared graphene as a gas separation membrane. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 17027.	2.8	65
93	First Molecular Dynamics simulation insight into the mechanism of organics adsorption from aqueous solutions on microporous carbons. <i>Chemical Physics Letters</i> , 2011, 515, 102-108.	2.6	22
94	Cryogenic Helium Adsorbed in Zeolite Rho: Inside Localization Controlled Self-Diffusion of Confined Quantum Particles. <i>Journal of Physical Chemistry C</i> , 2011, 115, 18105-18110.	3.1	3
95	Molecular dynamics of zigzag single walled carbon nanotube immersion in water. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 5621.	2.8	10
96	Simulating the effect of carbon nanotube curvature on adsorption of polycyclic aromatic hydrocarbons. <i>Adsorption</i> , 2011, 17, 1-4.	3.0	22
97	New phosphorus-containing spherical carbon adsorbents as promising materials in drug adsorption and release. <i>Journal of Colloid and Interface Science</i> , 2011, 354, 891-894.	9.4	30
98	Simulating the changes in carbon structure during the burn-off process. <i>Journal of Colloid and Interface Science</i> , 2011, 360, 211-219.	9.4	17
99	New model describing adsorption from liquid binary mixtures of nonelectrolytes with limited and unlimited miscibility of components. <i>Journal of Colloid and Interface Science</i> , 2011, 359, 512-519.	9.4	3
100	Phenol adsorption on closed carbon nanotubes. <i>Journal of Colloid and Interface Science</i> , 2011, 361, 288-292.	9.4	23
101	The influence of the carbon surface chemical composition on Dubininâ€™Astakhov equation parameters calculated from SF <sub>6</sub> adsorption dataâ€™ grand canonical Monte Carlo simulation. <i>Journal of Physics Condensed Matter</i> , 2011, 23, 395005.	1.8	5
102	Some Remarks on the Classification of Water Vapor Sorption Isotherms and Blahovec and Yanniotis Isotherm Equation. <i>Drying Technology</i> , 2011, 29, 984-991.	3.1	18
103	Simple model of adsorption on external surface of carbon nanotubesâ€™ a new analytical approach basing on molecular simulation data. <i>Adsorption</i> , 2010, 16, 197-213.	3.0	23
104	The system of carbon tetrachloride and closed carbon nanotubes analyzed by a combination of molecular simulations, analytical modeling, and adsorption calorimetry. <i>Journal of Colloid and Interface Science</i> , 2010, 349, 321-330.	9.4	6
105	BET surface area of carbonaceous adsorbentsâ€™ Verification using geometric considerations and GCMC simulations on virtual porous carbon models. <i>Applied Surface Science</i> , 2010, 256, 5204-5209.	6.1	23
106	Surface area of closed carbon nanotubes determined from room temperature measurements of alcohols adsorption. <i>Chemical Physics Letters</i> , 2010, 499, 141-145.	2.6	5
107	The influence of carbon surface oxygen groups on Dubininâ€™Astakhov equation parameters calculated from CO <sub>2</sub> adsorption isotherm. <i>Journal of Physics Condensed Matter</i> , 2010, 22, 085003.	1.8	24
108	Microscopic model of carbonaceous nanoporous molecular sievesâ€™ anomalous transport in molecularly confined spaces. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 11351.	2.8	17

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109	Carbon Dioxide Adsorption-Induced Deformation of Microporous Carbons. <i>Journal of Physical Chemistry C</i> , 2010, 114, 5126-5133.	3.1	61
110	Nanoporous Quantum Filters: Inside Vapor-Liquid Transitions of Quantum Fluids in Nanopores. <i>Journal of Physical Chemistry B</i> , 2010, 114, 5047-5052.	2.6	11
111	Optimal Single-Walled Carbon Nanotube Vessels for Short-Term Reversible Storage of Carbon Dioxide at Ambient Temperatures. <i>Journal of Physical Chemistry C</i> , 2010, 114, 21465-21473.	3.1	26
112	Activated carbon immersed in water—the origin of linear correlation between enthalpy of immersion and oxygen content studied by molecular dynamics simulation. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 10701.	2.8	7
113	Molecular dynamics simulation insight into the mechanism of phenol adsorption at low coverages from aqueous solutions on microporous carbons. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 812-817.	2.8	35
114	Adsorption potential distributions for carbons having defined pore structure—GCMC simulations of the effect of heterogeneity. <i>Adsorption</i> , 2009, 15, 99-113.	3.0	6
115	Hydrothermal opening of multiwall carbon nanotube with H <sub>2</sub> O <sub>2</sub> solution. <i>Chemical Physics Letters</i> , 2009, 482, 316-319.	2.6	16
116	Frequency-Dependent Diffusion Constant of Quantum Fluids from Path Integral Monte Carlo and Tikhonov's Regularizing Functional. <i>Journal of Chemical Theory and Computation</i> , 2009, 5, 1990-1996.	5.3	5
117	Can carbon surface oxidation shift the pore size distribution curve calculated from Ar, N <sub>2</sub> and CO <sub>2</sub> adsorption isotherms? Simulation results for a realistic carbon model. <i>Journal of Physics Condensed Matter</i> , 2009, 21, 315005.	1.8	35
118	Ar, CCl <sub>4</sub> and C <sub>6</sub> H <sub>6</sub> adsorption outside and inside of the bundles of multi-walled carbon nanotubes—simulation study. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 4982.	2.8	19
119	Searching the most optimal model of water sorption on foodstuffs in the whole range of relative humidity. <i>Food Research International</i> , 2009, 42, 1203-1214.	6.2	72
120	Impact of the carbon pore size and topology on the equilibrium quantum sieving of hydrogen isotopes at zero coverage and finite pressures. <i>Journal of Physics Condensed Matter</i> , 2009, 21, 144210.	1.8	21
121	Fullerene-Intercalated Graphene Nano-Containers—Mechanism of Argon Adsorption and High-Pressure CH <sub>4</sub> and CO <sub>2</sub> Storage Capacities. <i>Adsorption Science and Technology</i> , 2009, 27, 281-296.	3.2	35
122	Adsorption from aqueous solutions on opened carbon nanotubes—organic compounds speed up delivery of water from inside. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 9341.	2.8	20
123	Static and thermodynamic properties of low-density supercritical 4He—breakdown of the Feynman-Hibbs approximation. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 9182.	2.8	13
124	Carbon surface chemical composition in para-nitrophenol adsorption determined under real oxic and anoxic conditions. <i>Journal of Colloid and Interface Science</i> , 2008, 320, 40-51.	9.4	12
125	Water adsorption on carbons—Critical review of the most popular analytical approaches. <i>Advances in Colloid and Interface Science</i> , 2008, 137, 82-143.	14.7	109
126	Argon adsorption in channel-like mesoporous carbons at 77K: Grand Canonical Monte Carlo simulations and pore size analysis. <i>Microporous and Mesoporous Materials</i> , 2008, 116, 665-669.	4.4	6



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127	Cryogenic Separation of Hydrogen Isotopes in Single-Walled Carbon and Boron-Nitride Nanotubes: Insight into the Mechanism of Equilibrium Quantum Sieving in Quasi-One-Dimensional Pores. <i>Journal of Physical Chemistry B</i> , 2008, 112, 8275-8284.	2.6	42
128	Heterogeneity on high-resolution $\hat{\pm}$ plots for carbon nanotubesâ€”GCMC study. <i>Physical Chemistry Chemical Physics</i> , 2008, 10, 4551.	2.8	5
129	Testing isotherm models and recovering empirical relationships for adsorption in microporous carbons using virtual carbon models and grand canonical Monte Carlo simulations. <i>Journal of Physics Condensed Matter</i> , 2008, 20, 385212.	1.8	18
130	One-Step Steam Pyrolysis Preparation and Characterization of Spherical Carbon Adsorbents Obtained from Ion-Exchange Resins. <i>Adsorption Science and Technology</i> , 2008, 26, 407-413.	3.2	1
131	How realistic is the pore size distribution calculated from adsorption isotherms if activated carbon is composed of fullerene-like fragments?. <i>Physical Chemistry Chemical Physics</i> , 2007, 9, 5919.	2.8	70
132	Hyper-parallel tempering Monte Carlo simulations of Ar adsorption in new models of microporous non-graphitizing activated carbon: effect of microporosity. <i>Journal of Physics Condensed Matter</i> , 2007, 19, 406208.	1.8	43
133	Thermodynamics of Hydrogen Adsorption in Slit-like Carbon Nanopores at 77 K. Classical versus Path-Integral Monte Carlo Simulations. <i>Langmuir</i> , 2007, 23, 3666-3672.	3.5	56
134	Impact of the interaction with the positive charge in adsorption of benzene and other organic compounds from aqueous solutions on carbons. <i>Applied Surface Science</i> , 2007, 253, 4006-4009.	6.1	6
135	The impact of carbon surface chemical composition on the adsorption of phenol determined at the real oxic and anoxic conditions. <i>Applied Surface Science</i> , 2007, 253, 5752-5755.	6.1	23
136	Applicability of the generalised Dâ€™Arcy and Watt model to description of water sorption on pineapple and other foodstuffs. <i>Journal of Food Engineering</i> , 2007, 79, 718-723.	5.2	50
137	Bimodal pore size distributions for carbons: Experimental results and computational studies. <i>Journal of Colloid and Interface Science</i> , 2007, 310, 205-216.	9.4	24
138	Effective diffusion coefficient determination within cylindrical granules of adsorbents using a direct simulation method. <i>Journal of Colloid and Interface Science</i> , 2007, 313, 449-453.	9.4	6
139	The general mechanism of water sorption on foodstuffs â€“ Importance of the multitemperature fitting of data and the hierarchy of models. <i>Journal of Food Engineering</i> , 2007, 82, 528-535.	5.2	60
140	Thermodynamics of the CMMS Approach and Carbon Surface Chemistry in SO <sub>2</sub> Adsorption. <i>Langmuir</i> , 2006, 22, 6887-6892.	3.5	13
141	State of Hydrogen in Idealized Carbon Slitlike Nanopores at 77 K. <i>Langmuir</i> , 2006, 22, 1970-1972.	3.5	42
142	Grand Canonical Monte Carlo Simulation Study of Hydrogen Storage in Ordered Mesoporous Carbons at 303 K. <i>Adsorption Science and Technology</i> , 2006, 24, 411-426.	3.2	4
143	Thermodynamic properties of benzene adsorbed in activated carbons and multi-walled carbon nanotubes. <i>Chemical Physics Letters</i> , 2006, 421, 409-414.	2.6	59
144	Benzene adsorption on carbonaceous materials: The influence of pore structure on the state of the adsorbate. <i>Applied Surface Science</i> , 2006, 253, 2525-2539.	6.1	11

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145	Corrected thermodynamic description of adsorption via formalism of the theory of volume filling of micropores. <i>Journal of Colloid and Interface Science</i> , 2006, 298, 66-73.	9.4	7
146	Some remarks on the calculation of the pore size distribution function of activated carbons. <i>Journal of Colloid and Interface Science</i> , 2006, 300, 453-474.	9.4	20
147	Study of the selection mechanism of heavy metal (Pb <sup>2+</sup> , Cu <sup>2+</sup> , Ni <sup>2+</sup> , and Cd <sup>2+</sup> ) adsorption on clinoptilolite. <i>Journal of Colloid and Interface Science</i> , 2006, 304, 21-28.	9.4	510
148	Porous structure of natural and modified clinoptilolites. <i>Journal of Colloid and Interface Science</i> , 2006, 297, 77-85.	9.4	85
149	Simple models of adsorption in nanotubes. <i>Journal of Colloid and Interface Science</i> , 2006, 295, 310-317.	9.4	20
150	Two-dimensional gas and vacancy solution approaches in the thermodynamic description of adsorption equilibrium. <i>Journal of Colloid and Interface Science</i> , 2005, 282, 335-339.	9.4	16
151	New approach to determination of surface heterogeneity of adsorbents and catalysts from the temperature programmed desorption (TPD) technique: One step beyond the condensation approximation (CA) method. <i>Journal of Colloid and Interface Science</i> , 2005, 291, 334-344.	9.4	16
152	Heterogeneous Do-Do model of water adsorption on carbons. <i>Journal of Colloid and Interface Science</i> , 2005, 290, 1-13.	9.4	42
153	Parameterization of the corrected Dubinin-Serpinsky adsorption isotherm equation. <i>Journal of Colloid and Interface Science</i> , 2005, 291, 600-605.	9.4	12
154	Ammonium sorption from aqueous solutions by the natural zeolite Transcarpathian clinoptilolite studied under dynamic conditions. <i>Journal of Colloid and Interface Science</i> , 2005, 284, 408-415.	9.4	121
155	Modeling of the Hysteresis Phenomena in Finite-Sized Slitlike Nanopores. Revision of the Recent Results by Rigorous Numerical Analysis. <i>Langmuir</i> , 2005, 21, 6613-6627.	3.5	11
156	Improvement of the Derjaguin-Broekhoff-de Boer Theory for the Capillary Condensation/Evaporation of Nitrogen in Spherical Cavities and Its Application for the Pore Size Analysis of Silicas with Ordered Cage-like Mesopores. <i>Langmuir</i> , 2005, 21, 10530-10536.	3.5	16
157	Grand Canonical Monte Carlo Simulation Study of Methane Adsorption at an Open Graphite Surface and in Slitlike Carbon Pores at 273 K. <i>Langmuir</i> , 2005, 21, 5639-5646.	3.5	83
158	Improvement of the Derjaguin-Broekhoff-de Boer Theory for Capillary Condensation/Evaporation of Nitrogen in Mesoporous Systems and Its Implications for Pore Size Analysis of MCM-41 Silicas and Related Materials. <i>Langmuir</i> , 2005, 21, 1827-1833.	3.5	40
159	Effect of the Carbon Surface Layer Chemistry on Benzene Adsorption from the Vapor Phase and from Dilute Aqueous Solutions. <i>Langmuir</i> , 2005, 21, 12257-12267.	3.5	23
160	Description of benzene adsorption in slit-like pores. Theoretical foundations of the improved Horvath-Kawazoe method. <i>Carbon</i> , 2004, 42, 851-864.	10.3	13
161	Estimating the pore size distribution of activated carbons from adsorption data of different adsorbates by various methods. <i>Journal of Colloid and Interface Science</i> , 2004, 273, 39-63.	9.4	66
162	The effect of carbon surface chemical composition on the adsorption of acetanilide. <i>Journal of Colloid and Interface Science</i> , 2004, 272, 59-75.	9.4	17

#	ARTICLE	IF	CITATIONS
163	Molecular properties and intermolecular forces factors balancing the effect of carbon surface chemistry in adsorption of organics from dilute aqueous solutions. <i>Journal of Colloid and Interface Science</i> , 2004, 275, 9-29.	9.4	106
164	The applicability of the numerical algorithm for the evaluation of isosteric heat of adsorption. <i>Carbon</i> , 2004, 42, 53-58.	10.3	8
165	Impact of an adsorbed phase nonideality in the calculation of the filling pressure of carbon slit-like micropores. <i>Carbon</i> , 2004, 42, 573-583.	10.3	14
166	The evaluation of the surface heterogeneity of carbon blacks from the lattice density functional theory. <i>Carbon</i> , 2004, 42, 1813-1823.	10.3	15
167	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. Part VI. The Mechanism of the Physical and Chemical Adsorption of Acetanilide. <i>Adsorption Science and Technology</i> , 2004, 22, 353-376.	3.2	7
168	Estimation of the pore-size distribution function from the nitrogen adsorption isotherm. Comparison of density functional theory and the method of Do and co-workers. <i>Carbon</i> , 2003, 41, 1113-1125.	10.3	78
169	New correlations between the composition of the surface layer of carbon and its physicochemical properties exposed while paracetamol is adsorbed at different temperatures and pH. <i>Journal of Colloid and Interface Science</i> , 2003, 257, 13-30.	9.4	58
170	Developing the solution analogue of the Toth adsorption isotherm equation. <i>Journal of Colloid and Interface Science</i> , 2003, 266, 473-476.	9.4	75
171	Further insights into the role of carbon surface functionalities in the mechanism of phenol adsorption. <i>Journal of Colloid and Interface Science</i> , 2003, 268, 301-329.	9.4	247
172	The comparative characterization of structural heterogeneity of mesoporous activated carbon fibers (ACFs). <i>Applied Surface Science</i> , 2003, 206, 67-77.	6.1	44
173	Toward Solving the Unstable Linear Fredholm Equation of the First Kind: A New Procedure Called the Adsorption Stochastic Algorithm (ASA) and Its Properties. <i>Langmuir</i> , 2003, 19, 4253-4268.	3.5	38
174	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. Part IV. Do the Properties of Amphoteric Carbon Surface Layers Influence the Adsorption of Paracetamol at Acidic pH Levels?. <i>Adsorption Science and Technology</i> , 2003, 21, 93-123.	3.2	9
175	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. Part V. The Mechanism of the Physical and Chemical Adsorption of Phenol. <i>Adsorption Science and Technology</i> , 2003, 21, 539-585.	3.2	13
176	Numerical Analysis of the Horvath-Kawazoe Equation The Adsorption of Nitrogen, Argon, Benzene, Carbon Tetrachloride and Sulphur Hexafluoride. <i>Adsorption Science and Technology</i> , 2002, 20, 295-305.	3.2	12
177	The Application of a CONTIN Package for the Evaluation of Micropore Size Distribution Functions. <i>Langmuir</i> , 2002, 18, 5406-5413.	3.5	23
178	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. Part III. Theoretical Description of Paracetamol Adsorption Data at Neutral pH. <i>Adsorption Science and Technology</i> , 2002, 20, 63-81.	3.2	6
179	Describing Adsorption of Paracetamol from Aqueous Solution on Carbons While Utilizing the Most Widespread Isotherm Models The Impact of Surface Carbonyl and Basic Groups. <i>Journal of Colloid and Interface Science</i> , 2002, 247, 507-510.	9.4	17
180	The Simple Procedure of the Calculation of Diffusion Coefficient for Adsorption on Spherical and Cylindrical Adsorbent Particles Experimental Verification. <i>Journal of Colloid and Interface Science</i> , 2002, 249, 256-261.	9.4	17

#	ARTICLE	IF	CITATIONS
181	Homogeneous and Heterogeneous Micropore Structures in Carbonaceous Adsorbentsâ€”Twenty Years Later. <i>Journal of Colloid and Interface Science</i> , 2002, 254, 242-249.	9.4	4
182	New relationships between the characteristic energy of adsorption and the average effective diameter of carbon slit-like micropores. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2002, 201, 17-30.	4.7	15
183	What kind of pore size distribution is assumed in the Dubininâ€™Astakhov adsorption isotherm equation?. <i>Carbon</i> , 2002, 40, 2879-2886.	10.3	73
184	THE SIMPLE PROCEDURE OF THE CALCULATION OF DIFFUSION COEFFICIENT FOR ADSORPTION ON SPHERICAL AND CYLINDRICAL ADSORBENT PARTICLES. <i>Separation Science and Technology</i> , 2001, 36, 513-525.	2.5	21
185	The influence of activated carbon surface chemical composition on the adsorption of acetaminophen (paracetamol) in vitro. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 177, 23-45.	4.7	370
186	New relationships between the characteristic energy of adsorption and the average effective diameter of carbon slit-like micropores â€™ the dependence on the type of an adsorbate. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 177, 57-68.	4.7	19
187	Energetics of water adsorption and immersion on carbons. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 179, 39-55.	4.7	13
188	A Simple Method of the Determination of the Structural Heterogeneity of Microporous Solids. <i>Journal of Colloid and Interface Science</i> , 2001, 236, 387-390.	9.4	0
189	Characterization of Microporous Carbon Materials by Means of a New Gamma-Type Adsorption Isotherm Equation. <i>Journal of Colloid and Interface Science</i> , 2001, 243, 300-305.	9.4	5
190	The Comparative Analysis of the Properties of Two Micropore-Size Distribution Functions: The Pfeiferâ€™Avnir Function and the Gamma-Type One. <i>Journal of Colloid and Interface Science</i> , 2001, 244, 439-443.	9.4	4
191	The new correlation between microporosity of strictly microporous activated carbons and fractal dimension on the basis of the Polanyiâ€™Dubinin theory of adsorption. <i>Carbon</i> , 2001, 39, 267-278.	10.3	41
192	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. <i>Adsorption Science and Technology</i> , 2000, 18, 477-508.	3.2	19
193	The Normalization of the Micropore-Size Distribution Function in the Polanyiâ€™Dubinin Type of Adsorption Isotherm Equations. <i>Journal of Colloid and Interface Science</i> , 2000, 227, 482-494.	9.4	17
194	The Impact of Carbon Surface Composition on the Diffusion and Adsorption of Paracetamol at Different Temperatures and at Neutral pH. <i>Journal of Colloid and Interface Science</i> , 2000, 230, 219-222.	9.4	27
195	The influence of activated carbon surface chemical composition on the adsorption of acetaminophen (paracetamol) in vitro. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2000, 163, 135-150.	4.7	77
196	Energetics of water adsorption and immersion on carbons. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1999, 148, 271-281.	4.7	16
197	Fractal dimension of microporous carbon on the basis of Polanyiâ€™Dubinin theory of adsorption. Part IV. The comparative analysis of two alternative solutions of the overall adsorption isotherm equation for microporous fractal solids. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1999, 152, 293-313.	4.7	39
198	Comments on â€™An Isotherm Equation for Adsorption on Fractal Surfaces of Heterogeneous Porous Materialsâ€™. <i>Langmuir</i> , 1999, 15, 285-288.	3.5	16

#	ARTICLE	IF	CITATIONS
199	Adsorption Studies of Cu <sup>11</sup> from Aqueous/Acidic Solutions on to Bentonite. Adsorption Science and Technology, 1999, 17, 441-458.	3.2	26
200	Porosity of carbon films applied to chemical sensor construction. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1998, 132, 127-135.	4.7	3
201	Fractal dimension of microporous carbon on the basis of the Polanyi-Dubinin theory of adsorption. Part 3: Adsorption and adsorption thermodynamics in the micropores of fractal carbons. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1998, 136, 245-261.	4.7	23
202	Fractal dimension of microporous carbon on the basis of the Polanyi-Dubinin theory of adsorption. Part 2: Dubinin-Astakhov adsorption isotherm equation. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1997, 126, 67-73.	4.7	26
203	The structural parameters of microporous solid, including fractal dimension, on the basis of the potential theory of adsorption – the general solution. Computers & Chemistry, 1997, 21, 83-87.	1.2	20
204	Thermally modified active carbon as a support for catalysts for NH <sub>3</sub> synthesis. Carbon, 1996, 34, 403-409.	10.3	76
205	Fractal dimension of microporous carbon on the basis of first solution of a laplace transform using an incomplete gamma function. Computers & Chemistry, 1996, 20, 427-430.	1.2	24
206	Fractal dimension of microporous carbon on the basis of Polanyi-Dubinin theory of adsorption. Dubinin-Radushkevich adsorption isotherm equation. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1996, 119, 175-181.	4.7	30
207	Catalytic conversion of ethanol on carbon catalysts. Carbon, 1994, 32, 265-271.	10.3	49