

Artur Terzyk

List of Publications by Year in descending order

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207
papers

5,719
citations

94433

37
h-index

110387

64
g-index

209
all docs

209
docs citations

209
times ranked

5536
citing authors

#	ARTICLE	IF	CITATIONS
1	Study of the selection mechanism of heavy metal (Pb ²⁺ , Cu ²⁺ , Ni ²⁺ , and Cd ²⁺) adsorption on clinoptilolite. <i>Journal of Colloid and Interface Science</i> , 2006, 304, 21-28.	9.4	510
2	The influence of activated carbon surface chemical composition on the adsorption of acetaminophen (paracetamol) in vitro. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 177, 23-45.	4.7	370
3	Further insights into the role of carbon surface functionalities in the mechanism of phenol adsorption. <i>Journal of Colloid and Interface Science</i> , 2003, 268, 301-329.	9.4	247
4	Ammonium sorption from aqueous solutions by the natural zeolite Transcarpathian clinoptilolite studied under dynamic conditions. <i>Journal of Colloid and Interface Science</i> , 2005, 284, 408-415.	9.4	121
5	MOF materials as therapeutic agents, drug carriers, imaging agents and biosensors in cancer biomedicine: Recent advances and perspectives. <i>Progress in Materials Science</i> , 2021, 117, 100743.	32.8	120
6	Water adsorption on carbons – Critical review of the most popular analytical approaches. <i>Advances in Colloid and Interface Science</i> , 2008, 137, 82-143.	14.7	109
7	Molecular properties and intermolecular forces – factors balancing the effect of carbon surface chemistry in adsorption of organics from dilute aqueous solutions. <i>Journal of Colloid and Interface Science</i> , 2004, 275, 9-29.	9.4	106
8	Porous structure of natural and modified clinoptilolites. <i>Journal of Colloid and Interface Science</i> , 2006, 297, 77-85.	9.4	85
9	Grand Canonical Monte Carlo Simulation Study of Methane Adsorption at an Open Graphite Surface and in Slitlike Carbon Pores at 273 K. <i>Langmuir</i> , 2005, 21, 5639-5646.	3.5	83
10	Estimation of the pore-size distribution function from the nitrogen adsorption isotherm. Comparison of density functional theory and the method of Do and co-workers. <i>Carbon</i> , 2003, 41, 1113-1125.	10.3	78
11	The influence of activated carbon surface chemical composition on the adsorption of acetaminophen (paracetamol) in vitro. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2000, 163, 135-150.	4.7	77
12	Thermally modified active carbon as a support for catalysts for NH ₃ synthesis. <i>Carbon</i> , 1996, 34, 403-409.	10.3	76
13	Developing the solution analogue of the Toth adsorption isotherm equation. <i>Journal of Colloid and Interface Science</i> , 2003, 266, 473-476.	9.4	75
14	The Chemistry of Bioconjugation in Nanoparticles-Based Drug Delivery System. <i>Advances in Condensed Matter Physics</i> , 2015, 2015, 1-27.	1.1	75
15	What kind of pore size distribution is assumed in the Dubinin – Astakhov adsorption isotherm equation?. <i>Carbon</i> , 2002, 40, 2879-2886.	10.3	73
16	Searching the most optimal model of water sorption on foodstuffs in the whole range of relative humidity. <i>Food Research International</i> , 2009, 42, 1203-1214.	6.2	72
17	How realistic is the pore size distribution calculated from adsorption isotherms if activated carbon is composed of fullerene-like fragments?. <i>Physical Chemistry Chemical Physics</i> , 2007, 9, 5919.	2.8	70
18	Estimating the pore size distribution of activated carbons from adsorption data of different adsorbates by various methods. <i>Journal of Colloid and Interface Science</i> , 2004, 273, 39-63.	9.4	66

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19	Pillared graphene as a gas separation membrane. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 17027.	2.8	65
20	Carbon Dioxide Adsorption-Induced Deformation of Microporous Carbons. <i>Journal of Physical Chemistry C</i> , 2010, 114, 5126-5133.	3.1	61
21	The general mechanism of water sorption on foodstuffs – Importance of the multitemperature fitting of data and the hierarchy of models. <i>Journal of Food Engineering</i> , 2007, 82, 528-535.	5.2	60
22	Thermodynamic properties of benzene adsorbed in activated carbons and multi-walled carbon nanotubes. <i>Chemical Physics Letters</i> , 2006, 421, 409-414.	2.6	59
23	New correlations between the composition of the surface layer of carbon and its physicochemical properties exposed while paracetamol is adsorbed at different temperatures and pH. <i>Journal of Colloid and Interface Science</i> , 2003, 257, 13-30.	9.4	58
24	Thermodynamics of Hydrogen Adsorption in Slit-like Carbon Nanopores at 77 K. Classical versus Path-Integral Monte Carlo Simulations. <i>Langmuir</i> , 2007, 23, 3666-3672.	3.5	56
25	Ultra-long carbon nanotube-paraffin composites of record thermal conductivity and high phase change enthalpy among paraffin-based heat storage materials. <i>Journal of Energy Storage</i> , 2021, 36, 102396.	8.1	52
26	Applicability of the generalised D TM Arcy and Watt model to description of water sorption on pineapple and other foodstuffs. <i>Journal of Food Engineering</i> , 2007, 79, 718-723.	5.2	50
27	Catalytic conversion of ethanol on carbon catalysts. <i>Carbon</i> , 1994, 32, 265-271.	10.3	49
28	Displacement of Methane by Coadsorbed Carbon Dioxide Is Facilitated In Narrow Carbon Nanopores. <i>Journal of Physical Chemistry C</i> , 2012, 116, 13640-13649.	3.1	48
29	The comparative characterization of structural heterogeneity of mesoporous activated carbon fibers (ACFs). <i>Applied Surface Science</i> , 2003, 206, 67-77.	6.1	44
30	Hyper-parallel tempering Monte Carlo simulations of Ar adsorption in new models of microporous non-graphitizing activated carbon: effect of microporosity. <i>Journal of Physics Condensed Matter</i> , 2007, 19, 406208.	1.8	43
31	Heterogeneous Do ^{Do} model of water adsorption on carbons. <i>Journal of Colloid and Interface Science</i> , 2005, 290, 1-13.	9.4	42
32	State of Hydrogen in Idealized Carbon Slitlike Nanopores at 77 K. <i>Langmuir</i> , 2006, 22, 1970-1972.	3.5	42
33	Cryogenic Separation of Hydrogen Isotopes in Single-Walled Carbon and Boron-Nitride Nanotubes: Insight into the Mechanism of Equilibrium Quantum Sieving in Quasi-One-Dimensional Pores. <i>Journal of Physical Chemistry B</i> , 2008, 112, 8275-8284.	2.6	42
34	Synergetic effect of carbon nanopore size and surface oxidation on CO ₂ capture from CO ₂ /CH ₄ mixtures. <i>Journal of Colloid and Interface Science</i> , 2013, 397, 144-153.	9.4	42
35	The new correlation between microporosity of strictly microporous activated carbons and fractal dimension on the basis of the Polanyi ^{Do} Dubin theory of adsorption. <i>Carbon</i> , 2001, 39, 267-278.	10.3	41
36	Improvement of the Derjaguin ^{Broekhoff} de Boer Theory for Capillary Condensation/Evaporation of Nitrogen in Mesoporous Systems and Its Implications for Pore Size Analysis of MCM-41 Silicas and Related Materials. <i>Langmuir</i> , 2005, 21, 1827-1833.	3.5	40

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37	Fractal dimension of microporous carbon on the basis of Polanyi-Dubinin theory of adsorption. Part IV. The comparative analysis of two alternative solutions of the overall adsorption isotherm equation for microporous fractal solids. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1999, 152, 293-313.	4.7	39
38	Toward Solving the Unstable Linear Fredholm Equation of the First Kind: A New Procedure Called the Adsorption Stochastic Algorithm (ASA) and Its Properties. <i>Langmuir</i> , 2003, 19, 4253-4268.	3.5	38
39	Controlling enzymatic activity by immobilization on graphene oxide. <i>Die Naturwissenschaften</i> , 2017, 104, 36.	1.6	37
40	Nanoscale Water Contact Angle on Polytetrafluoroethylene Surfaces Characterized by Molecular Dynamics-Atomic Force Microscopy Imaging. <i>Langmuir</i> , 2018, 34, 4526-4534.	3.5	37
41	Using in-situ adsorption dilatometry for assessment of micropore size distribution in monolithic carbons. <i>Carbon</i> , 2016, 103, 263-272.	10.3	36
42	Can carbon surface oxidation shift the pore size distribution curve calculated from Ar, N ₂ and CO ₂ adsorption isotherms? Simulation results for a realistic carbon model. <i>Journal of Physics Condensed Matter</i> , 2009, 21, 315005.	1.8	35
43	Fullerene-Intercalated Graphene Nano-Containers – Mechanism of Argon Adsorption and High-Pressure CH ₄ and CO ₂ Storage Capacities. <i>Adsorption Science and Technology</i> , 2009, 27, 281-296.	3.2	35
44	Molecular dynamics simulation insight into the mechanism of phenol adsorption at low coverages from aqueous solutions on microporous carbons. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 812-817.	2.8	35
45	Super-sieving effect in phenol adsorption from aqueous solutions on nanoporous carbon beads. <i>Carbon</i> , 2018, 135, 12-20.	10.3	34
46	Influence of activated carbon surface oxygen functionalities on SO ₂ physisorption – Simulation and experiment. <i>Chemical Physics Letters</i> , 2013, 578, 85-91.	2.6	32
47	Fractal dimension of microporous carbon on the basis of Polanyi-Dubinin theory of adsorption. Dubinin-Radushkevich adsorption isotherm equation. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1996, 119, 175-181.	4.7	30
48	New phosphorus-containing spherical carbon adsorbents as promising materials in drug adsorption and release. <i>Journal of Colloid and Interface Science</i> , 2011, 354, 891-894.	9.4	30
49	Removal of internal caps during hydrothermal treatment of bamboo-like carbon nanotubes and application of tubes in phenol adsorption. <i>Journal of Colloid and Interface Science</i> , 2012, 381, 36-42.	9.4	30
50	Molecular simulation aided nanoporous carbon design for highly efficient low-concentrated formaldehyde capture. <i>Carbon</i> , 2017, 124, 152-160.	10.3	30
51	Water Adsorption Property of Hierarchically Nanoporous Detonation Nanodiamonds. <i>Langmuir</i> , 2017, 33, 11180-11188.	3.5	28
52	The Impact of Carbon Surface Composition on the Diffusion and Adsorption of Paracetamol at Different Temperatures and at Neutral pH. <i>Journal of Colloid and Interface Science</i> , 2000, 230, 219-222.	9.4	27
53	What Is the Value of Water Contact Angle on Silicon?. <i>Materials</i> , 2020, 13, 1554.	2.9	27
54	Fractal dimension of microporous carbon on the basis of the Polanyi-Dubinin theory of adsorption. Part 2: Dubinin-Astakhov adsorption isotherm equation. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1997, 126, 67-73.	4.7	26

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55	Adsorption Studies of Cu ¹¹ from Aqueous/Acidic Solutions on to Bentonite. Adsorption Science and Technology, 1999, 17, 441-458.	3.2	26
56	Optimal Single-Walled Carbon Nanotube Vessels for Short-Term Reversible Storage of Carbon Dioxide at Ambient Temperatures. Journal of Physical Chemistry C, 2010, 114, 21465-21473.	3.1	26
57	Screening of carbonaceous nanoporous materials for capture of nerve agents. Physical Chemistry Chemical Physics, 2013, 15, 291-298.	2.8	25
58	New insights into the ideal adsorbed solution theory. Physical Chemistry Chemical Physics, 2015, 17, 7232-7247.	2.8	25
59	Morphologically disordered pore model for characterization of micro-mesoporous carbons. Carbon, 2017, 111, 358-370.	10.3	25
60	Fractal dimension of microporous carbon on the basis of first solution of a laplace transform using an incomplete gamma function. Computers & Chemistry, 1996, 20, 427-430.	1.2	24
61	Bimodal pore size distributions for carbons: Experimental results and computational studies. Journal of Colloid and Interface Science, 2007, 310, 205-216.	9.4	24
62	The influence of carbon surface oxygen groups on Dubinin's Astakhov equation parameters calculated from CO ₂ adsorption isotherm. Journal of Physics Condensed Matter, 2010, 22, 085003.	1.8	24
63	Methane-Induced Deformation of Porous Carbons: From Normal to High-Pressure Operating Conditions. Journal of Physical Chemistry C, 2012, 116, 1740-1747.	3.1	24
64	Fractal dimension of microporous carbon on the basis of the Polanyi-Dubinin theory of adsorption. Part 3: Adsorption and adsorption thermodynamics in the micropores of fractal carbons. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1998, 136, 245-261.	4.7	23
65	The Application of a CONTIN Package for the Evaluation of Micropore Size Distribution Functions. Langmuir, 2002, 18, 5406-5413.	3.5	23
66	Effect of the Carbon Surface Layer Chemistry on Benzene Adsorption from the Vapor Phase and from Dilute Aqueous Solutions. Langmuir, 2005, 21, 12257-12267.	3.5	23
67	The impact of carbon surface chemical composition on the adsorption of phenol determined at the real oxic and anoxic conditions. Applied Surface Science, 2007, 253, 5752-5755.	6.1	23
68	Simple model of adsorption on external surface of carbon nanotubes—a new analytical approach basing on molecular simulation data. Adsorption, 2010, 16, 197-213.	3.0	23
69	BET surface area of carbonaceous adsorbents—Verification using geometric considerations and GCMC simulations on virtual porous carbon models. Applied Surface Science, 2010, 256, 5204-5209.	6.1	23
70	Phenol adsorption on closed carbon nanotubes. Journal of Colloid and Interface Science, 2011, 361, 288-292.	9.4	23
71	First Molecular Dynamics simulation insight into the mechanism of organics adsorption from aqueous solutions on microporous carbons. Chemical Physics Letters, 2011, 515, 102-108.	2.6	22
72	Simulating the effect of carbon nanotube curvature on adsorption of polycyclic aromatic hydrocarbons. Adsorption, 2011, 17, 1-4.	3.0	22

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73	Nanotube-mediated efficiency of cisplatin anticancer therapy. <i>Carbon</i> , 2014, 70, 46-58.	10.3	22
74	THE SIMPLE PROCEDURE OF THE CALCULATION OF DIFFUSION COEFFICIENT FOR ADSORPTION ON SPHERICAL AND CYLINDRICAL ADSORBENT PARTICLES. <i>Separation Science and Technology</i> , 2001, 36, 513-525.	2.5	21
75	Impact of the carbon pore size and topology on the equilibrium quantum sieving of hydrogen isotopes at zero coverage and finite pressures. <i>Journal of Physics Condensed Matter</i> , 2009, 21, 144210.	1.8	21
76	Carbon Molecular Sieves: Reconstruction of Atomistic Structural Models with Experimental Constraints. <i>Journal of Physical Chemistry C</i> , 2014, 118, 12996-13007.	3.1	21
77	New forcefield for water nanodroplet on a graphene surface. <i>Chemical Physics Letters</i> , 2017, 674, 98-102.	2.6	21
78	The structural parameters of microporous solid, including fractal dimension, on the basis of the potential theory of adsorption – the general solution. <i>Computers & Chemistry</i> , 1997, 21, 83-87.	1.2	20
79	Some remarks on the calculation of the pore size distribution function of activated carbons. <i>Journal of Colloid and Interface Science</i> , 2006, 300, 453-474.	9.4	20
80	Simple models of adsorption in nanotubes. <i>Journal of Colloid and Interface Science</i> , 2006, 295, 310-317.	9.4	20
81	Adsorption from aqueous solutions on opened carbon nanotubes – organic compounds speed up delivery of water from inside. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 9341.	2.8	20
82	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. <i>Adsorption Science and Technology</i> , 2000, 18, 477-508.	3.2	19
83	New relationships between the characteristic energy of adsorption and the average effective diameter of carbon slit-like micropores – the dependence on the type of an adsorbate. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 177, 57-68.	4.7	19
84	Ar, CCl ₄ and C ₆ H ₆ adsorption outside and inside of the bundles of multi-walled carbon nanotubes – simulation study. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 4982.	2.8	19
85	Enhanced adsorption of paracetamol on closed carbon nanotubes by formation of nanoaggregates: Carbon nanotubes as potential materials in hot-melt drug deposition-experiment and simulation. <i>Journal of Colloid and Interface Science</i> , 2012, 376, 209-216.	9.4	19
86	Atomic-scale molecular models of oxidized activated carbon fibre nanoregions: Examining the effects of oxygen functionalities on wet formaldehyde adsorption. <i>Carbon</i> , 2020, 165, 67-81.	10.3	19
87	Testing isotherm models and recovering empirical relationships for adsorption in microporous carbons using virtual carbon models and grand canonical Monte Carlo simulations. <i>Journal of Physics Condensed Matter</i> , 2008, 20, 385212.	1.8	18
88	Some Remarks on the Classification of Water Vapor Sorption Isotherms and Blahovec and Yanniotis Isotherm Equation. <i>Drying Technology</i> , 2011, 29, 984-991.	3.1	18
89	Structural properties of amorphous diamond-like carbon: percolation, cluster, and pair correlation analysis. <i>RSC Advances</i> , 2012, 2, 4292.	3.6	18
90	Cytotoxic or Not? Disclosing the Toxic Nature of Carbonaceous Nanomaterials through Nano – Bio Interactions. <i>Materials</i> , 2020, 13, 2060.	2.9	18

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91	Nanovehicles as a novel target strategy for hyperthermic intraperitoneal chemotherapy: a multidisciplinary study of peritoneal carcinomatosis. <i>Oncotarget</i> , 2015, 6, 22776-22798.	1.8	18
92	The Normalization of the Micropore-Size Distribution Function in the Polanyi–Dubinin Type of Adsorption Isotherm Equations. <i>Journal of Colloid and Interface Science</i> , 2000, 227, 482-494.	9.4	17
93	Describing Adsorption of Paracetamol from Aqueous Solution on Carbons While Utilizing the Most Widespread Isotherm Models—The Impact of Surface Carbonyl and Basic Groups. <i>Journal of Colloid and Interface Science</i> , 2002, 247, 507-510.	9.4	17
94	The Simple Procedure of the Calculation of Diffusion Coefficient for Adsorption on Spherical and Cylindrical Adsorbent Particles—Experimental Verification. <i>Journal of Colloid and Interface Science</i> , 2002, 249, 256-261.	9.4	17
95	The effect of carbon surface chemical composition on the adsorption of acetanilide. <i>Journal of Colloid and Interface Science</i> , 2004, 272, 59-75.	9.4	17
96	Microscopic model of carbonaceous nanoporous molecular sieves—anomalous transport in molecularly confined spaces. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 11351.	2.8	17
97	Simulating the changes in carbon structure during the burn-off process. <i>Journal of Colloid and Interface Science</i> , 2011, 360, 211-219.	9.4	17
98	Water Nanodroplet on a Hydrocarbon “Carpet”—The Mechanism of Water Contact Angle Stabilization by Airborne Contaminations on Graphene, Au, and PTFE Surfaces. <i>Langmuir</i> , 2019, 35, 420-427.	3.5	17
99	Revisiting Wetting, Freezing, and Evaporation Mechanisms of Water on Copper. <i>ACS Applied Materials & Interfaces</i> , 2021, 13, 37893-37903.	8.0	17
100	Energetics of water adsorption and immersion on carbons. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 1999, 148, 271-281.	4.7	16
101	Comments on “An Isotherm Equation for Adsorption on Fractal Surfaces of Heterogeneous Porous Materials”. <i>Langmuir</i> , 1999, 15, 285-288.	3.5	16
102	Two-dimensional gas and vacancy solution approaches in the thermodynamic description of adsorption equilibrium. <i>Journal of Colloid and Interface Science</i> , 2005, 282, 335-339.	9.4	16
103	New approach to determination of surface heterogeneity of adsorbents and catalysts from the temperature programmed desorption (TPD) technique: One step beyond the condensation approximation (CA) method. <i>Journal of Colloid and Interface Science</i> , 2005, 291, 334-344.	9.4	16
104	Improvement of the Derjaguin-Broekhoff-de Boer Theory for the Capillary Condensation/Evaporation of Nitrogen in Spherical Cavities and Its Application for the Pore Size Analysis of Silicas with Ordered Cage-like Mesopores. <i>Langmuir</i> , 2005, 21, 10530-10536.	3.5	16
105	Hydrothermal opening of multiwall carbon nanotube with H ₂ O ₂ solution. <i>Chemical Physics Letters</i> , 2009, 482, 316-319.	2.6	16
106	Intrinsic D ₂ /H ₂ Selectivity of NaX Zeolite: Interplay between Adsorption and Kinetic Factors. <i>Journal of Physical Chemistry C</i> , 2015, 119, 15373-15380.	3.1	16
107	New relationships between the characteristic energy of adsorption and the average effective diameter of carbon slit-like micropores. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2002, 201, 17-30.	4.7	15
108	The evaluation of the surface heterogeneity of carbon blacks from the lattice density functional theory. <i>Carbon</i> , 2004, 42, 1813-1823.	10.3	15

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109	Simulation of SF ₆ adsorption on the bundles of single walled carbon nanotubes. <i>Microporous and Mesoporous Materials</i> , 2012, 154, 51-55.	4.4	15
110	Separation of CO ₂ –CH ₄ mixtures on defective single walled carbon nanohorns – tip does matter. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 16468.	2.8	15
111	Nuclear Quantum Effects in the Layering and Diffusion of Hydrogen Isotopes in Carbon Nanotubes. <i>Journal of Physical Chemistry Letters</i> , 2015, 6, 3367-3372.	4.6	15
112	Impact of an adsorbed phase nonideality in the calculation of the filling pressure of carbon slit-like micropores. <i>Carbon</i> , 2004, 42, 573-583.	10.3	14
113	Properties of Phenol Confined in Realistic Carbon Micropore Model: Experiment and Simulation. <i>Journal of Physical Chemistry C</i> , 2015, 119, 19987-19995.	3.1	14
114	Dynamics of effusive and diffusive gas separation on pillared graphene. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 17018-17023.	2.8	14
115	Selective carboxylation versus layer-by-layer unsheathing of multi-walled carbon nanotubes: new insights from the reaction with boiling nitrating mixture. <i>RSC Advances</i> , 2019, 9, 37608-37613.	3.6	14
116	Hedgehog-like structure, PVDF- carbon nanohorn hybrid membranes for improved removal of VOCs from water. <i>Chemical Engineering Journal</i> , 2022, 438, 135574.	12.7	14
117	Energetics of water adsorption and immersion on carbons. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2001, 179, 39-55.	4.7	13
118	Adsorption of Biologically Active Compounds from Aqueous Solutions on to Commercial Unmodified Activated Carbons. Part V. The Mechanism of the Physical and Chemical Adsorption of Phenol. <i>Adsorption Science and Technology</i> , 2003, 21, 539-585.	3.2	13
119	Description of benzene adsorption in slit-like pores. Theoretical foundations of the improved Horvath–Kawazoe method. <i>Carbon</i> , 2004, 42, 851-864.	10.3	13
120	Thermodynamics of the CMMS Approach and Carbon Surface Chemistry in SO ₂ Adsorption. <i>Langmuir</i> , 2006, 22, 6887-6892.	3.5	13
121	Static and thermodynamic properties of low-density supercritical 4He – breakdown of the Feynman–Hibbs approximation. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 9182.	2.8	13
122	Detecting adsorption space in carbon nanotubes by benzene uptake. <i>Journal of Colloid and Interface Science</i> , 2013, 391, 74-85.	9.4	13
123	Water nanodroplet on a graphene surface – a new old system. <i>Journal of Physics Condensed Matter</i> , 2016, 28, 495002.	1.8	13
124	Numerical Analysis of the Horvath–Kawazoe Equation – The Adsorption of Nitrogen, Argon, Benzene, Carbon Tetrachloride and Sulphur Hexafluoride. <i>Adsorption Science and Technology</i> , 2002, 20, 295-305.	3.2	12
125	Parameterization of the corrected Dubinin–Serpinsky adsorption isotherm equation. <i>Journal of Colloid and Interface Science</i> , 2005, 291, 600-605.	9.4	12
126	Carbon surface chemical composition in para-nitrophenol adsorption determined under real oxic and anoxic conditions. <i>Journal of Colloid and Interface Science</i> , 2008, 320, 40-51.	9.4	12

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127	Conscious Changes of Carbon Nanotubes Cytotoxicity by Manipulation with Selected Nanofactors. <i>Applied Biochemistry and Biotechnology</i> , 2015, 176, 730-741.	2.9	12
128	Nano-Structured Carbon Matrixes Obtained from Chitin and Chitosan by a Novel Method. <i>Journal of Nanoscience and Nanotechnology</i> , 2016, 16, 2623-2631.	0.9	12
129	Modeling of the Hysteresis Phenomena in Finite-Sized Slitlike Nanopores. Revision of the Recent Results by Rigorous Numerical Analysis. <i>Langmuir</i> , 2005, 21, 6613-6627.	3.5	11
130	Benzene adsorption on carbonaceous materials: The influence of pore structure on the state of the adsorbate. <i>Applied Surface Science</i> , 2006, 253, 2525-2539.	6.1	11
131	Nanoporous Quantum Filters: Inside Vapor-Liquid Transitions of Quantum Fluids in Nanopores. <i>Journal of Physical Chemistry B</i> , 2010, 114, 5047-5052.	2.6	11
132	Constant Pressure Path Integral Gibbs Ensemble Monte Carlo Method. <i>Journal of Chemical Theory and Computation</i> , 2013, 9, 2922-2929.	5.3	11
133	Synthesis of carbon nanotubes and nanotube forests on copper catalyst. <i>Materials Research Express</i> , 2014, 1, 035040.	1.6	11
134	Correlation between the catalytic and electrocatalytic properties of nitrogen-doped carbon nanooxions and the polarity of the carbon surface: Experimental and theoretical investigations. <i>Carbon</i> , 2019, 151, 120-129.	10.3	11
135	Carbon nanohorn improved durable PVDF membranes - The future of membrane distillation and desalination. <i>Desalination</i> , 2021, 511, 115117.	8.2	11
136	Are nanohedgehogs thirsty? Toward new superhydrophobic and anti-icing carbon nanohorn-polymer hybrid surfaces. <i>Chemical Engineering Journal</i> , 2022, 446, 137126.	12.7	11
137	Molecular dynamics of zigzag single walled carbon nanotube immersion in water. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 5621.	2.8	10
138	Virtual Porous Carbons. , 2012, , 61-104.		10
139	The first atomistic modelling-aided reproduction of morphologically defective single walled carbon nanohorns. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 1232-1240.	2.8	10
140	Applicability of molecular simulations for modelling the adsorption of the greenhouse gas CF ₄ on carbons. <i>Journal of Physics Condensed Matter</i> , 2013, 25, 015004.	1.8	10
141	MD simulation of organics adsorption from aqueous solution in carbon slit-like pores. Foundations of the pore blocking effect. <i>Journal of Physics Condensed Matter</i> , 2014, 26, 055008.	1.8	10
142	New application of carbon nanotubes in haemostatic dressing filled with anticancer substance. <i>Biomedicine and Pharmacotherapy</i> , 2015, 69, 349-354.	5.6	10
143	Water at Curved Carbon Surface: Mechanisms of Adsorption Revealed by First Calorimetric Study. <i>Journal of Physical Chemistry C</i> , 2015, 119, 2703-2715.	3.1	10
144	Switchable hydrophobicity/hydrophilicity of a HOPG surface - Comment on the paper by Y. Wei and C.Q. Jia, <i>Carbon</i> , 87 (2015) 10-17. <i>Carbon</i> , 2017, 115, 571-573.	10.3	10

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145	Stability of coordination polymers in water: state of the art and towards a methodology for nonporous materials. <i>Adsorption</i> , 2019, 25, 1-11.	3.0	10
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