Frank Weinhold

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/6979280/publications.pdf

Version: 2024-02-01

229 papers

57,241 citations

69 h-index 227 g-index

277 all docs

277 docs citations

times ranked

277

23843 citing authors

#	Article	IF	CITATIONS
1	Intermolecular interactions from a natural bond orbital, donor-acceptor viewpoint. Chemical Reviews, 1988, 88, 899-926.	23.0	15,946
2	Natural population analysis. Journal of Chemical Physics, 1985, 83, 735-746.	1.2	8,757
3	Natural hybrid orbitals. Journal of the American Chemical Society, 1980, 102, 7211-7218.	6.6	4,505
4	Natural bond orbital analysis of nearâ€Hartree–Fock water dimer. Journal of Chemical Physics, 1983, 78, 4066-4073.	1,2	2,775
5	Natural localized molecular orbitals. Journal of Chemical Physics, 1985, 83, 1736-1740.	1.2	2,019
6	Analysis of the geometry of the hydroxymethyl radical by the "different hybrids for different spins― natural bond orbital procedure. Computational and Theoretical Chemistry, 1988, 169, 41-62.	1.5	1,905
7	<i>NBO 6.0</i> : Natural bond orbital analysis program. Journal of Computational Chemistry, 2013, 34, 1429-1437.	1.5	1,269
8	Natural bond orbital methods. Wiley Interdisciplinary Reviews: Computational Molecular Science, 2012, 2, 1-42.	6.2	1,153
9	NATURAL BOND ORBITALS AND EXTENSIONS OF LOCALIZED BONDING CONCEPTS. Chemistry Education Research and Practice, 2001, 2, 91-104.	1.4	732
10	Electronic Basis of Improper Hydrogen Bonding:Â A Subtle Balance of Hyperconjugation and Rehybridization. Journal of the American Chemical Society, 2003, 125, 5973-5987.	6.6	700
11	Natural resonance theory: I. General formalism. Journal of Computational Chemistry, 1998, 19, 593-609.	1.5	621
12	What is NBO analysis and how is it useful? International Reviews in Physical Chemistry, 2016, 35, 399-440.	0.9	585
13	Metric geometry of equilibrium thermodynamics. Journal of Chemical Physics, 1975, 63, 2479-2483.	1.2	560
14	Natural bond orbital analysis: A critical overview of relationships to alternative bonding perspectives. Journal of Computational Chemistry, 2012, 33, 2363-2379.	1.5	515
15	Natural resonance theory: II. Natural bond order and valency. Journal of Computational Chemistry, 1998, 19, 610-627.	1.5	480
16	Natural bond orbital analysis of molecular interactions: Theoretical studies of binary complexes of HF, H2O, NH3, N2, O2, F2, CO, and CO2 with HF, H2O, and NH3. Journal of Chemical Physics, 1986, 84, 5687-5705.	1.2	474
17	Natural resonance theory: III. Chemical applications. Journal of Computational Chemistry, 1998, 19, 628-646.	1.5	460
18	The Natural Bond Orbital Lewis Structure Concept for Molecules, Radicals, and Radical Ions., 1988,, 227-236.		392

#	Article	IF	CITATIONS
19	Resonance properties of complex-rotated hamiltonians. Molecular Physics, 1978, 36, 1613-1630.	0.8	369
20	Collagen Stability:Â Insights from NMR Spectroscopic and Hybrid Density Functional Computational Investigations of the Effect of Electronegative Substituents on Prolyl Ring Conformations. Journal of the American Chemical Society, 2002, 124, 2497-2505.	6.6	318
21	Quantum-mechanical studies on the origin of barriers to internal rotation about single bonds. Journal of the American Chemical Society, 1979, 101, 1700-1709.	6.6	312
22	Natural chemical shielding analysis of nuclear magnetic resonance shielding tensors from gauge-including atomic orbital calculations. Journal of Chemical Physics, 1997, 107, 1173-1184.	1.2	258
23	Natural bond orbital analysis of steric interactions. Journal of Chemical Physics, 1997, 107, 5406-5421.	1.2	257
24	<i>NBO 7.0</i> : New vistas in localized and delocalized chemical bonding theory. Journal of Computational Chemistry, 2019, 40, 2234-2241.	1.5	249
25	A new twist on molecular shape. Nature, 2001, 411, 539-541.	13.7	234
26	Structure and spectroscopy of (HCN)n clusters: Cooperative and electronic delocalization effects in C–Hâ‹â‹N hydrogen bonding. Journal of Chemical Physics, 1995, 103, 333-347.	1.2	220
27	Nature of H-bonding in clusters, liquids, and enzymes: an ab initio, natural bond orbital perspective. Computational and Theoretical Chemistry, 1997, 398-399, 181-197.	1.5	219
28	What is a hydrogen bond? Mutually consistent theoretical and experimental criteria for characterizing H-bonding interactions. Molecular Physics, 2012, 110, 565-579.	0.8	195
29	Metric geometry of equilibrium thermodynamics. II. Scaling, homogeneity, and generalized Gibbs–Duhem relations. Journal of Chemical Physics, 1975, 63, 2484-2487.	1.2	179
30	Natural Bond Orbital Analysis of Internal Rotation Barriers and Related Phenomena. Israel Journal of Chemistry, 1991, 31, 277-285.	1.0	179
31	Natural steric analysis of internal rotation barriers. International Journal of Quantum Chemistry, 1999, 72, 269-280.	1.0	179
32	Natural steric analysis: Ab initiovan der Waals radii of atoms and ions. Journal of Chemical Physics, 1997, 107, 5422-5432.	1.2	173
33	Antiâ€Electrostatic Hydrogen Bonds. Angewandte Chemie - International Edition, 2014, 53, 11214-11217.	7.2	171
34	On the role of d orbitals in sulfur hexafluoride. Journal of the American Chemical Society, 1986, 108, 3586-3593.	6.6	168
35	Rebuttal to the Bickelhaupt–Baerends Case for Steric Repulsion Causing the Staggered Conformation of Ethane. Angewandte Chemie - International Edition, 2003, 42, 4188-4194.	7.2	160
36	Quantum cluster equilibrium theory of liquids: General theory and computer implementation. Journal of Chemical Physics, 1998, 109, 367-372.	1.2	153

3

#	Article	IF	Citations
37	Dynamic polarizabilities of twoâ€electron atoms, with rigorous upper and lower bounds. Journal of Chemical Physics, 1976, 65, 4913-4926.	1.2	136
38	The Nature of the Silicon–Oxygen Bond. Organometallics, 2011, 30, 5815-5824.	1.1	135
39	Moment-theory investigations of photoabsorption and dispersion profiles in atoms and ions. Physical Review A, 1976, 14, 1042-1056.	1.0	133
40	Experimental and theoretical determination of the temperature dependence of deuteron and oxygen quadrupole coupling constants of liquid water. Journal of Chemical Physics, 1995, 103, 6941-6950.	1.2	132
41	Spectroscopic Evidence for Clusters of Likeâ€Charged Ions in Ionic Liquids Stabilized by Cooperative Hydrogen Bonding. ChemPhysChem, 2016, 17, 458-462.	1.0	115
42	Origin of Methyl Internal Rotation Barriers. Accounts of Chemical Research, 1999, 32, 983-993.	7.6	113
43	Nine questions on energy decomposition analysis. Journal of Computational Chemistry, 2019, 40, 2248-2283.	1.5	113
44	Understanding barriers to internal rotation in substituted toluenes and their cations. Journal of Chemical Physics, 1995, 102, 6787-6805.	1.2	110
45	Quantum Cluster Equilibrium Theory of Liquids:Â Temperature Dependence of Hydrogen Bonding in LiquidN-Methylacetamide Studied by IR Spectra. Journal of Physical Chemistry B, 1998, 102, 9312-9318.	1.2	110
46	Quantum cluster equilibrium theory of liquids: Illustrative application to water. Journal of Chemical Physics, 1998, 109, 373-384.	1.2	107
47	Metric geometry of equilibrium thermodynamics. V. Aspects of heterogeneous equilibrium. Journal of Chemical Physics, 1976, 65, 559-564.	1.2	105
48	NaturalJ-Coupling Analysis:Â Interpretation of ScalarJ-Couplings in Terms of Natural Bond Orbitals. Journal of the American Chemical Society, 2001, 123, 12026-12036.	6.6	102
49	Investigation of the differences in stability of the OCâ‹â‹â‹HF and COâ‹â‹â‹HF complexes. Journal of Ch Physics, 1985, 82, 2679-2687.	emical 1.2	101
50	The role of delocalization in benzene. Journal of the American Chemical Society, 1993, 115, 10952-10957.	6.6	100
51	Theoretical study of hydrogen bonding in liquid and gaseous N-methylformamide. Journal of Chemical Physics, 1997, 107, 499-507.	1.2	99
52	Origin of Trans-Bent Geometries in Maximally Bonded Transition Metal and Main Group Molecules. Journal of the American Chemical Society, 2006, 128, 7335-7345.	6.6	99
53	Reduced Density Matrices of Atoms and Molecules. II. On the Nâ€Representability Problem. Journal of Chemical Physics, 1967, 47, 2298-2311.	1.2	97
54	What is a hydrogen bond? Resonance covalency in the supramolecular domain. Chemistry Education Research and Practice, 2014, 15, 276-285.	1.4	97

#	Article	IF	Citations
55	Hyperconjugative Interactions in Permethylated Siloxanes and Ethers: The Nature of the SiO Bond. Journal of the American Chemical Society, 2013, 135, 5762-5767.	6.6	94
56	Reduced Density Matrices of Atoms and Molecules. I. The 2 Matrix of Doubleâ€Occupancy, Configurationâ€Interaction Wavefunctions for Singlet States. Journal of Chemical Physics, 1967, 46, 2752-2758.	1.2	93
57	NMR Investigations ofClostridium pasteurianumRubredoxin. Origin of Hyperfine1H,2H,13C, and15N NMR Chemical Shifts in Ironâ^'Sulfur Proteins As Determined by Comparison of Experimental Data with Hybrid Density Functional Calculationsâ€. Journal of the American Chemical Society, 1998, 120, 4806-4814.	6.6	93
58	Efficient optimization of natural resonance theory weightings and bond orders by gramâ€based convex programming. Journal of Computational Chemistry, 2019, 40, 2028-2035.	1.5	91
59	Natural bond critical point analysis: Quantitative relationships between natural bond orbitalâ€based and QTAIMâ€based topological descriptors of chemical bonding. Journal of Computational Chemistry, 2012, 33, 2440-2449.	1.5	89
60	Metric geometry of equilibrium thermodynamics. III. Elementary formal structure of a vectorâ€algebraic representation of equilibrium thermodynamics. Journal of Chemical Physics, 1975, 63, 2488-2495.	1.2	87
61	Observation of an Eclipsed Csp3-CH3 Bond in a Tricyclic Orthoamide; Experimental and Theoretical Evidence for CHâƒ>O Hydrogen Bonds. Angewandte Chemie International Edition in English, 1987, 26, 1175-1177.	4.4	87
62	Metric geometry of equilibrium thermodynamics. IV. Vectorâ€algebraic evaluation of thermodynamic derivatives. Journal of Chemical Physics, 1975, 63, 2496-2501.	1.2	86
63	Experimental and theoretical studies of hydrogen bonding in neat, liquid formamide. Journal of Chemical Physics, 1995, 102, 5118-5125.	1.2	85
64	CHEMISTRY: High Bond Orders in Metal-Metal Bonding. Science, 2007, 316, 61-63.	6.0	81
65	Upper and Lower Bounds to Quantum-Mechanical Properties. Advances in Quantum Chemistry, 1972, , 299-331.	0.4	80
66	Quantum cluster equilibrium theory of liquids: Freezing of QCE/3-21G water to tetrakaidecahedral "Bucky-ice― Journal of Chemical Physics, 1999, 110, 508-515.	1.2	75
67	Quantum cluster equilibrium theory of liquids: molecular clusters and thermodynamics of liquid ethanol. Molecular Physics, 1999, 97, 465-477.	0.8	75
68	Some remarks on nonorthogonal orbitals in quantum chemistry. Computational and Theoretical Chemistry, 1988, 165, 189-202.	1.5	70
69	Blue-Shifted and Red-Shifted Hydrogen Bonds in Hypervalent Rare-Gas FRgâ^'H···Y Sandwiches. Journal of Physical Chemistry A, 2004, 108, 4720-4730.	1.1	70
70	Criteria of Accuracy of Approximate Wavefunctions. Journal of Mathematical Physics, 1970, 11, 2127-2138.	0.5	69
71	Imaginaryâ€frequency polarizability and van der Waals force constants of twoâ€electron atoms, with rigorous bounds. Journal of Chemical Physics, 1977, 66, 191-198.	1.2	67
72	Dipole oscillator strengths, with rigorous limits of error, for He andLi+. Physical Review A, 1974, 9, 118-128.	1.0	66

#	Article	IF	CITATIONS
73	Transferability of natural bond orbitals. Journal of the American Chemical Society, 1988, 110, 368-372.	6.6	65
74	Protonation of Rhenium Alkyne Complexes Produces î-3-Allyl Rhenium Complexes via Observable 1-Metallacyclopropene Intermediates. Journal of the American Chemical Society, 1998, 120, 12500-12511.	6.6	65
7 5	Structure of Liquid N-Methylacetamide:  Temperature Dependence of NMR Chemical Shifts and Quadrupole Coupling Constants. Journal of Physical Chemistry A, 1997, 101, 8861-8870.	1.1	64
76	Calculation of Upper and Lower Bounds to Oscillator Strengths. Journal of Chemical Physics, 1971, 54, 1874-1881.	1.2	61
77	Complex-coordinate studies of helium autoionizing resonances. International Journal of Quantum Chemistry, 1978, 14, 727-736.	1.0	61
78	Bayâ€type H···H "bonding―in cisâ€2â€butene and related species: QTAIM versus NBO description. Journa Computational Chemistry, 2014, 35, 1499-1508.	al of 1.5	59
79	A theoretical model of bonding in hyperlithiated carbon compounds. Journal of the American Chemical Society, 1985, 107, 1919-1921.	6.6	58
80	Common Textbook and Teaching Misrepresentations of Lewis Structures. Journal of Chemical Education, 1995, 72, 583.	1,1	57
81	Conceptual model of "through-bonds" interactions. Journal of the American Chemical Society, 1976, 98, 4392-4393.	6.6	56
82	Photoelectron spectroscopy of free radicals with cmâ^1 resolution: The benzyl cation. Journal of Chemical Physics, 1991, 95, 8665-8668.	1.2	56
83	Structures of the aluminum oxides studied by ab initio methods with natural bond orbital analysis. Journal of Chemical Physics, 1992, 97, 3420-3430.	1.2	55
84	High Harmonic Generation Spectra of Neutral Helium by the Complex-Scaled (t,t′) Method: Role of Dynamical Electron Correlation. Physical Review Letters, 1997, 78, 2100-2103.	2.9	54
85	Solvent and concentration dependence of the hydroxyl chemical shift of methanol. Molecular Physics, 1998, 93, 145-151.	0.8	54
86	Resonance Character of Hydrogenâ€bonding Interactions in Water and Other Hâ€bonded Species. Advances in Protein Chemistry, 2005, 72, 121-155.	4.4	53
87	Joint treatment of ab initio and experimental data in molecular force field calculations with Tikhonov's method of regularization. Journal of Chemical Physics, 1994, 100, 1414-1424.	1.2	52
88	Rabbit-ears hybrids, VSEPR sterics, and other orbital anachronisms. Chemistry Education Research and Practice, 2014, 15, 417-434.	1.4	52
89	Temperature dependence of hydrogen bonding in neat, liquid formamide. Journal of Chemical Physics, 1995, 103, 3636-3642.	1.2	49
90	Radical hydrogen bonding: Origin of stability of radical-molecule complexes. Journal of Chemical Physics, 2007, 127, 164102.	1.2	49

#	Article	IF	CITATIONS
91	$3c/4e\ \ddot{ }f\ \dot{ }$,-Type Long-Bonding: A Novel Transitional Motif toward the Metallic Delocalization Limit. Inorganic Chemistry, 2013, 52, 5154-5166.	1.9	48
92	Valence and extra-valence orbitals in main group and transition metal bonding. Journal of Computational Chemistry, 2007, 28, 198-203.	1.5	46
93	Bonding Analysis of TM(cAAC) ₂ (TM = Cu, Ag, and Au) and the Importance of Reference State. Organometallics, 2015, 34, 3442-3449.	1.1	46
94	Resonance Theory Reboot. Journal of the American Chemical Society, 2019, 141, 4156-4166.	6.6	45
95	Bond-antibond analysis of internal rotation barriers in glyoxal and related molecules: WhereINDO fails. International Journal of Quantum Chemistry, 1981, 19, 781-791.	1.0	43
96	Calculations of one-, two- and three-bond nuclear spin-spin couplings in a model peptide and correlations with experimental data. Journal of Biomolecular NMR, 1994, 4, 519-542.	1.6	43
97	Structure of Magnesium Cluster Grignard Reagents. Inorganic Chemistry, 1995, 34, 2980-2983.	1.9	43
98	Quantum cluster equilibrium theory of liquids: temperature dependent chemical shifts, quadrupole coupling constants and vibrational frequencies in liquid ethanol. Molecular Physics, 1999, 97, 479-486.	0.8	43
99	Trans-Hydrogen-Bondh2JNNandh1JNHCouplings in the DNA Aâ^'T Base Pair:Â Natural Bond Orbital Analysis. Journal of the American Chemical Society, 2002, 124, 1190-1191.	6.6	43
100	Some remarks on the C–H bond dipole moment. Journal of Chemical Physics, 1986, 84, 2428-2430.	1.2	42
101	Chemical Bonding as a Superposition Phenomenon. Journal of Chemical Education, 1999, 76, 1141.	1.1	42
102	Improved General Understanding of the Hydrogenâ€Bonding Phenomena: A Reply. Angewandte Chemie - International Edition, 2015, 54, 2600-2602.	7.2	42
103	Theoretical Studies of Protium/Deuterium Fractionation Factors and Cooperative Hydrogen Bonding in Peptides. Journal of the American Chemical Society, 1995, 117, 9619-9624.	6.6	41
104	Quadrupole coupling constants in linear (HCN)nclusters: Theoretical and experimental evidence for cooperativity effects in C–Hâ‹â‹â‹N hydrogen bonding. Journal of Chemical Physics, 1995, 103, 348-352.	1.2	40
105	Predicting the Ionic Product of Water. Scientific Reports, 2017, 7, 10244.	1.6	40
106	Antisymmetrization effects in bondâ€orbital models of internal rotation barriers. Journal of Chemical Physics, 1980, 72, 2866-2868.	1.2	39
107	Comment on "Observation of alkaline earth complexes M(CO) ₈ (M = Ca, Sr, or Ba) that mimic transition metals― Science, 2019, 365, .	6.0	39
108	Theoretical Prediction of Robust Second-Row Oxyanion Clusters in the Metastable Domain of Antielectrostatic Hydrogen Bonding. Inorganic Chemistry, 2018, 57, 2035-2044.	1.9	38

#	Article	IF	CITATIONS
109	Comment on "Natural Bond Orbitals and the Nature of the Hydrogen Bond― Journal of Physical Chemistry A, 2018, 122, 724-732.	1.1	38
110	Boron oxides: Ab initiostudies with natural bond orbital analysis. Journal of Chemical Physics, 1993, 98, 1329-1335.	1.2	37
111	Electron-correlation effects in the positions and widths of two-electron autoionizing resonances. Physical Review A, 1979, 20, 27-31.	1.0	35
112	Syn and anti bent hydrazine radical cations. Effect of .sigma.,.pi. mixing on spectral properties. Journal of the American Chemical Society, 1985, 107, 143-149.	6.6	34
113	Photodissociation of (CO)+2: Theoretical studies of ground 2Bu and excited 2Bg potential energy surfaces. Journal of Chemical Physics, 1987, 87, 392-410.	1.2	34
114	Quantum cluster equilibrium theory of liquids part I: Molecular clusters and thermodynamics of liquid ammonia. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1998, 102, 197-204.	0.9	33
115	Torsion-vibration interactions in hydrogen peroxide. 1. Calculation of the trans barrier for hydroxyls overtone excitations up to $\nu=8$. The Journal of Physical Chemistry, 1988, 92, 4295-4306.	2.9	32
116	Effective Oâ€17 quadrupole moments for the calibrated computation of quadrupole coupling parameters at different levels of theory. Journal of Chemical Physics, 1996, 105, 8223-8230.	1.2	32
117	Theoretical studies of O2?:(H2O)n clusters. Journal of Computational Chemistry, 1986, 7, 294-305.	1.5	31
118	Threshold photoionization spectra of benzyl radical: Cation vibrational states and ab initio calculations. Journal of Chemical Physics, 1996, 104, 8886-8895.	1.2	31
119	Hyperfine-Shifted ¹³ C Resonance Assignments in an IronⴴSulfur Protein with Quantum Chemical Verification: Aliphatic CⴴH···S 3-Centerⴴ4-Electron Interactions. Journal of the American Chemical Society, 2011, 133, 1310-1316.	6.6	31
120	Inadequacies of the Point-Dipole Approximation for Describing Electronâ^'Nuclear Interactions in Paramagnetic Proteins:  Hybrid Density Functional Calculations and the Analysis of NMR Relaxation of High-Spin Iron(III) Rubredoxin. Journal of Physical Chemistry B, 1998, 102, 8300-8305.	1.2	30
121	Torsion-vibration interactions in overtone excited states of hydrogen peroxide. The Journal of Physical Chemistry, 1986, 90, 6405-6408.	2.9	29
122	Eine C _{sp} ³ â€gebundene Methylgruppe in ekliptischer Konformation; experimenteller und theoretischer Nachweis von CH … Oâ€Wasserstoffbrücken. Angewandte Chemie, 1987, 99, 1216-1218.	1.6	29
123	Estimates of ? and ? torsion angles in proteins from one-, two- and three-bond nuclear spin-spin couplings: Application to staphylococcal nuclease. Journal of Biomolecular NMR, 1994, 4, 543-551.	1.6	29
124	Quantum cluster equilibrium theory of liquids part II: Temperature dependent chemical shifts, quadrupole coupling constants and vibrational frequencies in liquid ammonia. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1998, 102, 205-212.	0.9	29
125	Critical test of quantum cluster equilibrium theory: Formic acid at B3LYP/6-31+G* hybrid density functional level. Journal of Chemical Physics, 1998, 109, 5945-5947.	1.2	29
126	Diamagnetism of helium. Journal of Chemical Physics, 2000, 113, 8667-8670.	1.2	28

#	Article	IF	CITATIONS
127	Improved Lower Bounds to the Overlap Integral of an Approximate Wavefunction with the True Wavefunction. Journal of Chemical Physics, 1967, 46, 2448-2449.	1.2	27
128	Variational Upper and Lower Bounds to Dipole Transition Moments. Physical Review Letters, 1970, 25, 907-909.	2.9	27
129	Dynamic polarizabilities of metastable 2 1,3S excited states of He and Li+, with rigorous upper and lower bounds. Journal of Chemical Physics, 1977, 66, 185-190.	1.2	27
130	Criteria of accuracy of resonance eigenvalues. International Journal of Quantum Chemistry, 1980, 17, 1201-1211.	1.0	26
131	Quantum cluster equilibrium theory of liquids: light and heavy QCE/3-21G model water. Physical Chemistry Chemical Physics, 2000, 2, 1613-1619.	1.3	26
132	Molecular Composition of Liquid Sulfur. Angewandte Chemie - International Edition, 2002, 41, 3199-3202.	7.2	26
133	Insight into the Mechanism of the Michael Reaction. ChemPhysChem, 2016, 17, 2022-2034.	1.0	26
134	Lower bounds to expectation values. Journal of Physics A, 1968, 1, 305-313.	1.6	25
135	Relative accuracy of length and velocity forms in oscillator-strength calculations. Physical Review A, 1974, 10, 1457-1463.	1.0	25
136	The principle of maximum overlap. Journal of the American Chemical Society, 1976, 98, 3745-3749.	6.6	25
137	Natural bond orbitals in multiconfigurational expansions: Local treatment of electron correlation in molecules. Journal of Chemical Physics, 1992, 97, 1095-1108.	1.2	25
138	Polyion Covalency: Exotic Species from the Unexplored World of Electrostatically Shielded Molecular Ion Chemistry. Angewandte Chemie - International Edition, 2017, 56, 14577-14581.	7.2	25
139	New Formulas for Lower Bounds to Expectation Values. Physical Review, 1969, 183, 142-147.	2.7	24
140	Geometric representation of equilibrium thermodynamics. Accounts of Chemical Research, 1976, 9, 236-240.	7.6	24
141	Torsion-vibration interactions in hydrogen peroxide. 2. Natural bond orbital analysis. The Journal of Physical Chemistry, 1988, 92, 4306-4313.	2.9	24
142	Calculation of nuclear spin-spin coupling constants with ab initio molecular orbital wave functions. The Journal of Physical Chemistry, 1993, 97, 11657-11665.	2.9	24
143	Quantum Chemical Calculations on Structural Models of the Catalytic Site of Chymotrypsin:Â Comparison of Calculated Results with Experimental Data from NMR Spectroscopy. Journal of the American Chemical Society, 2002, 124, 14373-14381.	6.6	24
144	News from the Periodic Table: An Introduction to "Periodicity Symbols, Tables, and Models for Higher-Order Valency and Donor–Acceptor Kinships". Journal of Chemical Education, 2007, 84, 1145.	1.1	24

#	Article	IF	CITATIONS
145	Mass polarization and Breit-Pauli corrections for the polarizability of helium-4. The Journal of Physical Chemistry, 1982, 86, 1111-1116.	2.9	23
146	Comments on "ls It Time To Retire the Hybrid Atomic Orbital?― Journal of Chemical Education, 2012, 89, 570-572.	1.1	23
147	Resonance Character of Copper/Silver/Gold Bonding in Small Moleculeâ‹â‹â‹MX (X=F, Cl, Br,) Tj ETQq1	1 0.784314 1.0	4 rgBT /Ove <mark>rl</mark> o
148	Quantum Cluster Equilibrium Theory of Liquids: Isotopically substituted QCE/3-21G Model Water. Zeitschrift Fur Physikalische Chemie, 2002, 216, .	1.4	21
149	The Path to Natural Bond Orbitals. Israel Journal of Chemistry, 2022, 62, .	1.0	21
150	On the dipole moment of three identical spherical atoms. Molecular Physics, 1978, 35, 1205-1210.	0.8	20
151	Resonance Bonding Patterns of Peroxide Chemistry: Cyclic Three-Center Hyperbonding in "Phosphadioxiraneâ€Intermediates. Journal of the American Chemical Society, 2006, 128, 11850-11859.	6.6	20
152	18â€electron rule and the 3c/4e hyperbonding saturation limit. Journal of Computational Chemistry, 2016, 37, 237-241.	1.5	20
153	Variational calculation of continuum corrections to overlap. Journal of Chemical Physics, 1973, 59, 355-362.	1.2	19
154	3c/4e î,,'-type long-bonding competes with ï‰-bonding in noble-gas hydrides HNgY (Ng = He, Ne, Ar, Kr, Xe, Rn;) Tj FTQq0	0 0 rgBT /Ove
155	Natural resonance theory: II. Natural bond order and valency. Journal of Computational Chemistry, 1998, 19, 610-627.	1.5	19
156	What Is the Nature of Supramolecular Bonding? Comprehensive NBO/NRT Picture of Halogen and Pnicogen Bonding in RPH2···IF/FI Complexes (R = CH3, OH, CF3, CN, NO2). Molecules, 2019, 24, 2090.	1.7	18
157	Variational Extensions of Lower Bounds to Expectation Values. Physical Review A, 1970, 1, 122-124.	1.0	17
158	Calculations of one-, two- and three-bond nuclear spin-spin couplings in a model peptide and correlations with experimental data. Journal of Biomolecular NMR, 1995, 5, 332.	1.6	17
159	Accurate Structure and Dynamics of the Metal-Site of Paramagnetic Metalloproteins from NMR Parameters Using Natural Bond Orbitals. Journal of the American Chemical Society, 2012, 134, 4670-4682.	6.6	17
160	Improved General Understanding of the Hydrogenâ€Bonding Phenomena: A Reply. Angewandte Chemie, 2015, 127, 2636-2638.	1.6	17
161	9-(9-Borabicyclo[3.3.1]nonyl)-9-azabicyclo[3.3.1]nonane radical cation: a failure of Bredt's rule kinetic stabilization. Journal of Organic Chemistry, 1980, 45, 2116-2119.	1.7	16
162	Coupling of internal rotations in propanelike molecules. International Journal of Quantum Chemistry, 1982, 21, 487-509.	1.0	16

#	Article	IF	CITATIONS
163	Addendum: Antiâ€Electrostatic Hydrogen Bonds. Angewandte Chemie - International Edition, 2014, 53, 12992-12992.	7.2	16
164	Ab Initio Calculations of Protium/Deuterium Fractionation Factors in O2H5+ Clusters. The Journal of Physical Chemistry, 1995, 99, 8013-8016.	2.9	15
165	Ab initioand regularized force fields of haloethanes: CH3CH2Cl, CH3CHCl2, CH3CF2Cl, and CH3CFCl2. Journal of Chemical Physics, 1998, 109, 7286-7299.	1.2	15
166	Nuclear motion and Breit–Pauli corrections to the diamagnetism of atomic helium. Journal of Chemical Physics, 2002, 117, 3243-3247.	1.2	15
167	Kinetics and Mechanism of Water Cluster Equilibria. Journal of Physical Chemistry B, 2014, 118, 7792-7798.	1.2	15
168	Upper and lower bounds to quantum-mechanical sum rules. Journal of Physics A, 1968, 1, 655-660.	1.6	14
169	Upper and lower bounds to the van der Waals interactions between atoms. Journal of Physics B: Atomic and Molecular Physics, 1969, 2, 517-520.	1.6	14
170	Lower Bounds to Expectation Values: Two-Electron Atoms. Canadian Journal of Chemistry, 1973, 51, 260-264.	0.6	14
171	Experimental and theoretical study of the relaxation of vibrationally excited HF by NO and CO. Journal of Chemical Physics, 1989, 91, 1688-1696.	1.2	14
172	Dickinson Energy of H2+. Journal of Chemical Physics, 1971, 54, 530-532.	1.2	13
173	Use of Hueckel Methodology with ab Initio Molecular Orbitals: Polarizabilities and Prediction of Organic Reactions. Journal of the American Chemical Society, 1994, 116, 1579-1580.	6.6	13
174	Resonance Natural Bond Orbitals: Efficient Semilocalized Orbitals for Computing and Visualizing Reactive Chemical Processes. Journal of Chemical Theory and Computation, 2019, 15, 916-921.	2.3	13
175	Natural resonance theory: I. General formalism. , 1998, 19, 593.		13
176	Inequalities for Multipole Dispersion Interactions. Journal of Chemical Physics, 1969, 50, 4136-4137.	1.2	12
177	Bounds to the lifetime of the Ar XVII2ÂS3state. Physical Review A, 1975, 11, 442-445.	1.0	12
178	G+K 1Σ+g doubleâ€minimum excited state of H2. Journal of Chemical Physics, 1977, 66, 303-305.	1.2	12
179	Complex-coordinate calculations with complex basis sets. Physical Review A, 1981, 24, 1254-1259.	1.0	12
180	Efficient evaluation of poly-electron populations in natural bond orbital analysis. Chemical Physics Letters, 2018, 711, 23-26.	1.2	12

#	Article	IF	Citations
181	Improved lower bounds to expectation values. Journal of Physics A, 1968, 1, 535-538.	1.6	11
182	Natural resonance theory. I. General formalism. Journal of Computational Chemistry, 2000, 21, 411-413.	1.5	11
183	Natural Bond Orbital Theory of Pseudo-Jahn–Teller Effects. Journal of Physical Chemistry A, 2018, 122, 4490-4498.	1.1	11
184	Pauling's Conceptions of Hybridization and Resonance in Modern Quantum Chemistry. Molecules, 2021, 26, 4110.	1.7	11
185	Characterization of the Methoxy Carbonyl Radical Formed via Photolysis of Methyl Chloroformate at 193.3 nm. Journal of Physical Chemistry A, 2007, 111, 1762-1770.	1.1	10
186	Why Do Cumulene Ketones Kink?. Journal of Organic Chemistry, 2017, 82, 12238-12245.	1.7	10
187	6 Natural bond orbital theory: Discovering chemistry with NBO7., 2021,, 129-156.		10
188	NBO/NRT Two-State Theory of Bond-Shift Spectral Excitation. Molecules, 2020, 25, 4052.	1.7	9
189	Variational Wavefunctions for H2+. Journal of Chemical Physics, 1972, 56, 3798-3801.	1.2	8
190	The 2Ag excited state of (CO)+2. Journal of Chemical Physics, 1988, 88, 1467-1468.	1.2	8
191	Tests of quantum cluster equilibrium (QCE)-based computational methods for describing formic acid clustering. Molecular Physics, 2003, 101, 1147-1153.	0.8	8
192	Determination of the Conformation of 2-Hydroxy- and 2-Aminobenzoic Acid Dimers Using 13C NMR and Density Functional Theory/Natural Bond Order Analysis:  The Central Importance of the Carboxylic Acid Carbon. Journal of Physical Chemistry A, 2006, 110, 8832-8839.	1.1	8
193	Remark on lower bounds to eigenvalues. International Journal of Quantum Chemistry, 1969, 3, 371-373.	1.0	7
194	On a formula of Braun and Rebane for variational bounds to overlap. International Journal of Quantum Chemistry, 1971, 5, 215-220.	1.0	7
195	Experimental and theoretical spin-spin coupling constants for [15N] formamide. Molecular Physics, 2002, 100, 2807-2814.	0.8	7
196	Isotopeâ€Sensitive Degenerate [1,3]â€Hydrogen Migration versus Competitive Enol–Keto Tautomerization. Chemistry - A European Journal, 2009, 15, 11815-11819.	1.7	7
197	Polyion Covalency: Exotic Species from the Unexplored World of Electrostatically Shielded Molecular Ion Chemistry. Angewandte Chemie, 2017, 129, 14769-14773.	1.6	7
198	The role of hyperconjugation in the unusual conformation of thymine: A natural bond orbital analysis. Computational and Theoretical Chemistry, 2018, 1130, 58-62.	1.1	7

#	Article	IF	Citations
199	Natural resonance theory of chemical reactivity, with illustrative application to intramolecular Claisen rearrangement. Tetrahedron, 2018, 74, 4799-4804.	1.0	7
200	To Be or Not to Be: Demystifying the 2ndâ€Quantized Picture of Complex Electronic Configuration Patterns in Chemistry with Natural Polyâ€Electron Population Analysis. Journal of Computational Chemistry, 2019, 40, 1509-1520.	1.5	7
201	Quantum Cluster Equilibrium. Letters in Mathematical Physics, 2014, , 77-96.	0.4	7
202	Comments on Purser's Article: "Lewis Structures are Models for Predicting Molecular Structure, Not Electronic Structure". Journal of Chemical Education, 2005, 82, 527.	1.1	6
203	Natural Bond–Bond Polarizability: A Hýckel-Like Electronic Delocalization Index. Journal of Organic Chemistry, 2013, 78, 1844-1850.	1.7	6
204	Rabbit ears concepts of water lone pairs: a reply to comments of Hiberty, Danovich, and Shaik. Chemistry Education Research and Practice, 2015, 16, 694-696.	1.4	6
205	Substituted Ortho-Benzynes: Properties of the Triple Bond. Journal of Organic Chemistry, 2020, 85, 9905-9914.	1.7	6
206	Anti-Electrostatic Pi-Hole Bonding: How Covalency Conquers Coulombics. Molecules, 2022, 27, 377.	1.7	6
207	Upper and Lower Bounds to Excited States of Twoâ€Electron Atoms. Journal of Chemical Physics, 1972, 57, 1738-1745.	1.2	5
208	Geometrical Aspects of Equilibrium Thermodynamics. Theoretical Chemistry, 1978, 3, 15-54.	0.2	5
209	Bivariational calculations of bounds on complex-frequency polarizabilities. Journal of Chemical Physics, 1978, 68, 2915.	1.2	5
210	Remarks on the calculation of upper and lower limits to quantum-mechanical properties. International Journal of Quantum Chemistry, 1971, 5, 721-728.	1.0	5
211	Quantum cluster equilibrium theory of liquids: temperature dependent chemical shifts, quadrupole coupling constants and vibrational frequencies in liquid ethanol. Molecular Physics, 1999, 97, 479-486.	0.8	5
212	Variation-perturbation approach to complex eigenvalues of resonance states. The Journal of Physical Chemistry, 1979, 83, 1517-1520.	2.9	4
213	Natural hybrid orbitals: Ab initio SCF and CI results for CO and NiCO. International Journal of Quantum Chemistry, 1980, 18, 201-209.	1.0	4
214	Chlorine dioxide: An exception that proves the rules of localized chemical bonding. Journal of Chemical Physics, 2022, 156, 124303.	1.2	4
215	The ABCs of Multiple Bonding. Science, 2007, 318, 746-746.	6.0	3
216	Quantitative Theoretical Predictions and Qualitative Bonding Analysis of the Divinylborinium System and Its Al, Ga, In, and Tl Congeners. Inorganic Chemistry, 2018, 57, 7851-7859.	1.9	3

#	Article	IF	Citations
217	Natural resonance theory: III. Chemical applications. , 1998, 19, 628.		3
218	Can a Wanzlick-like equilibrium exist between dicoordinate borylenes and diborenes?. Chemical Science, $0, , .$	3.7	3
219	Constraints on the values of force constants for molecular force field models based on ab initio calculations. Journal of Molecular Structure, 1997, 410-411, 457-461.	1.8	2
220	Die molekulare Zusammensetzung des fl�ssigen Schwefels. Angewandte Chemie, 2002, 114, 3331-3335.	1.6	2
221	Comment on "Superposition of Waves or Densities: Which Is the Nature of Chemical Resonance?―[<i>)J. Comput. Chem</i> . 2021, <i>42</i> , 412–417]. Journal of Computational Chemistry, 2021, 42, 1338-1340.	1.5	2
222	Time-Conjugation in a Unified Quantum Theory for Hermitian and Non-Hermitian Electronic Systems under Time-Reversal Symmetry. Symmetry, 2021, 13, 808.	1.1	2
223	Natural steric analysis of internal rotation barriers. , 1999, 72, 269.		2
224	Spectroscopic Evidence for Clusters of Likeâ€Charged Ions in Ionic Liquids Stabilized by Cooperative Hydrogen Bonding. ChemPhysChem, 2016, 17, 447-447.	1.0	1
225	Natural resonance theory: II. Natural bond order and valency. , 1998, 19, 610.		1
226	Partitioning Technique for Determinantal Equations. , 1976, , 307-314.		1
227	High-Density "Windowpane―Coordination Patterns of Water Clusters and Their NBO/NRT Characterization. Molecules, 2022, 27, 4218.	1.7	1
228	Supramolecular bonding. , 2003, , 579-709.		0
229	Sulfur Tetrahydride and Allied Superhydride Clusters: When Resonance Takes Precedence. Chemistry - A European Journal, 2021, 27, 6748-6759.	1.7	0