

Divina A Navarro

List of Publications by Year in descending order

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Version: 2024-02-01

33
papers

1,715
citations

304743

22
h-index

414414

32
g-index

33
all docs

33
docs citations

33
times ranked

2596
citing authors

#	ARTICLE	IF	CITATIONS
1	Humic Acid-Induced Silver Nanoparticle Formation Under Environmentally Relevant Conditions. <i>Environmental Science & Technology</i> , 2011, 45, 3895-3901.	10.0	265
2	Assessing antibiotic sorption in soil: a literature review and new case studies on sulfonamides and macrolides. <i>Chemistry Central Journal</i> , 2014, 8, 5.	2.6	174
3	Influences of Chemical Properties, Soil Properties, and Solution pH on Soil-Water Partitioning Coefficients of Per- and Polyfluoroalkyl Substances (PFASs). <i>Environmental Science & Technology</i> , 2020, 54, 15883-15892.	10.0	171
4	Investigating uptake of water-dispersible CdSe/ZnS quantum dot nanoparticles by <i>Arabidopsis thaliana</i> plants. <i>Journal of Hazardous Materials</i> , 2012, 211-212, 427-435.	12.4	134
5	Ecological Risk Assessment of Nano-enabled Pesticides: A Perspective on Problem Formulation. <i>Journal of Agricultural and Food Chemistry</i> , 2018, 66, 6480-6486.	5.2	106
6	Bioavailability of silver and silver sulfide nanoparticles to lettuce (<i>Lactuca sativa</i>): Effect of agricultural amendments on plant uptake. <i>Journal of Hazardous Materials</i> , 2015, 300, 788-795.	12.4	98
7	Sorption of PFOA onto different laboratory materials: Filter membranes and centrifuge tubes. <i>Chemosphere</i> , 2019, 222, 671-678.	8.2	91
8	Natural Organic Matter-Mediated Phase Transfer of Quantum Dots in the Aquatic Environment. <i>Environmental Science & Technology</i> , 2009, 43, 677-682.	10.0	62
9	Increasing ionic strength and valency of cations enhance sorption through hydrophobic interactions of PFAS with soil surfaces. <i>Science of the Total Environment</i> , 2022, 817, 152975.	8.0	60
10	Cd Tolerance and Accumulation in the Aquatic Macrophyte, <i>Chara australis</i> : Potential Use for Charophytes in Phytoremediation. <i>Environmental Science & Technology</i> , 2011, 45, 5332-5338.	10.0	52
11	Sorptive remediation of perfluorooctanoic acid (PFOA) using mixed mineral and graphene/carbon-based materials. <i>Environmental Chemistry</i> , 2018, 15, 472.	1.5	44
12	Quantifying the Sensitivity of Soil Microbial Communities to Silver Sulfide Nanoparticles Using Metagenome Sequencing. <i>PLoS ONE</i> , 2016, 11, e0161979.	2.5	41
13	Impact of (nano)formulations on the distribution and wash-off of copper pesticides and fertilisers applied on citrus leaves. <i>Environmental Chemistry</i> , 2019, 16, 401.	1.5	37
14	Remobilisation of silver and silver sulphide nanoparticles in soils. <i>Environmental Pollution</i> , 2014, 193, 102-110.	7.5	36
15	Characterization and ecological risk assessment of nanoparticulate CeO ₂ as a diesel fuel catalyst. <i>Environmental Toxicology and Chemistry</i> , 2013, 32, 1896-1905.	4.3	35
16	Predicting partitioning of radiolabelled ¹⁴ C-PFOA in a range of soils using diffuse reflectance infrared spectroscopy. <i>Science of the Total Environment</i> , 2019, 686, 505-513.	8.0	35
17	Combined effects of cadmium and zinc on growth, tolerance, and metal accumulation in <i>Chara australis</i> and enhanced phytoextraction using EDTA. <i>Ecotoxicology and Environmental Safety</i> , 2013, 98, 236-243.	6.0	33
18	Differences in Soil Mobility and Degradability between Water-Dispersible CdSe and CdSe/ZnS Quantum Dots. <i>Environmental Science & Technology</i> , 2011, 45, 6343-6349.	10.0	31

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19	Sorption behaviour of per- and polyfluoroalkyl substances (PFASs) as affected by the properties of coastal estuarine sediments. <i>Science of the Total Environment</i> , 2020, 720, 137263.	8.0	28
20	Partitioning of hydrophobic CdSe quantum dots into aqueous dispersions of humic substances: Influence of capping-group functionality on the phase-transfer mechanism. <i>Journal of Colloid and Interface Science</i> , 2010, 348, 119-128.	9.4	27
21	Behaviour of fullerenes (C60) in the terrestrial environment: Potential release from biosolids-amended soils. <i>Journal of Hazardous Materials</i> , 2013, 262, 496-503.	12.4	27
22	Comparing the Leaching Behavior of Per- and Polyfluoroalkyl Substances from Contaminated Soils Using Static and Column Leaching Tests. <i>Environmental Science & Technology</i> , 2022, 56, 368-378.	10.0	24
23	An investigation into the long-term binding and uptake of PFOS, PFOA and PFHxS in soil " plant systems. <i>Journal of Hazardous Materials</i> , 2021, 404, 124065.	12.4	22
24	Fullerol as a Potential Pathway for Mineralization of Fullerene Nanoparticles in Biosolid-Amended Soils. <i>Environmental Science and Technology Letters</i> , 2016, 3, 7-12.	8.7	19
25	Mineralisation and release of 14C-graphene oxide (GO) in soils. <i>Chemosphere</i> , 2020, 238, 124558.	8.2	15
26	Assessment of Mobilization Potential of Per- and Polyfluoroalkyl Substances for Soil Remediation. <i>Environmental Science & Technology</i> , 2022, 56, 10030-10041.	10.0	12
27	Partitioning behavior and stabilization of hydrophobically coated HfO ₂ , ZrO ₂ and Hf _x Zr _{1-x} O ₂ nanoparticles with natural organic matter reveal differences dependent on crystal structure. <i>Journal of Hazardous Materials</i> , 2011, 196, 302-310.	12.4	9
28	Fate of radiolabeled C60 fullerenes in aged soils. <i>Environmental Pollution</i> , 2017, 221, 293-300.	7.5	9
29	Organic carbon and salinity affect desorption of PFAS from estuarine sediments. <i>Journal of Soils and Sediments</i> , 2022, 22, 1302-1314.	3.0	5
30	Method for extraction and analysis of per- and poly-fluoroalkyl substances in contaminated asphalt. <i>Analytical Methods</i> , 2022, 14, 1678-1689.	2.7	5
31	Mixed-Mode Remediation of Cadmium and Arsenate Ions Using Graphene-Based Materials. <i>Clean - Soil, Air, Water</i> , 2018, 46, 1800073.	1.1	3
32	Potential Application of Laser-Induced Breakdown Spectroscopy (LIBS) Data for the Determination of Cation Exchange Capacity (CEC) of Agricultural Soils. <i>ChemistrySelect</i> , 2020, 5, 3798-3804.	1.5	3
33	Fate of copper in soil: effect of agrochemical (nano)formulations and soil properties. <i>Environmental Science: Nano</i> , 0, , .	4.3	2