

# Susan A Carroll

## List of Publications by Year in descending order

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80  
papers

3,958  
citations

101543

36  
h-index

118850

62  
g-index

81  
all docs

81  
docs citations

81  
times ranked

3455  
citing authors

#	ARTICLE	IF	CITATIONS
1	Direct effects of CO <sub>2</sub> and temperature on silicate weathering: Possible implications for climate control. <i>Geochimica Et Cosmochimica Acta</i> , 1994, 58, 1853-1856.	3.9	212
2	X-ray absorption spectroscopic study of Fe reference compounds for the analysis of natural sediments. <i>American Mineralogist</i> , 2004, 89, 572-585.	1.9	210
3	CO <sub>2</sub> -induced dissolution of low permeability carbonates. Part I: Characterization and experiments. <i>Advances in Water Resources</i> , 2013, 62, 370-387.	3.8	148
4	X-Ray Absorption Spectroscopy of Strontium(II) Coordination. <i>Journal of Colloid and Interface Science</i> , 2000, 222, 198-212.	9.4	141
5	Review: Role of chemistry, mechanics, and transport on well integrity in CO <sub>2</sub> storage environments. <i>International Journal of Greenhouse Gas Control</i> , 2016, 49, 149-160.	4.6	141
6	Dependence of labradorite dissolution kinetics on CO <sub>2</sub> (aq), Al(aq), and temperature. <i>Chemical Geology</i> , 2005, 217, 213-225.	3.3	138
7	Cesium migration in Hanford sediment: a multisite cation exchange model based on laboratory transport experiments. <i>Journal of Contaminant Hydrology</i> , 2003, 67, 219-246.	3.3	136
8	Rock-Water Interactions Controlling Zinc, Cadmium, and Lead Concentrations in Surface Waters and Sediments, U.S. Tri-State Mining District. 1. Molecular Identification Using X-ray Absorption Spectroscopy. <i>Environmental Science &amp; Technology</i> , 1998, 32, 943-955.	10.0	124
9	Transformation of meta-stable calcium silicate hydrates to tobermorite: reaction kinetics and molecular structure from XRD and NMR spectroscopy. <i>Geochemical Transactions</i> , 2009, 10, 1.	0.7	120
10	CO <sub>2</sub> -induced dissolution of low permeability carbonates. Part II: Numerical modeling of experiments. <i>Advances in Water Resources</i> , 2013, 62, 388-408.	3.8	111
11	Amorphous silica precipitation (60 to 120°C): comparison of laboratory and field rates. <i>Geochimica Et Cosmochimica Acta</i> , 1998, 62, 1379-1396.	3.9	110
12	Key factors for determining groundwater impacts due to leakage from geologic carbon sequestration reservoirs. <i>International Journal of Greenhouse Gas Control</i> , 2014, 29, 153-168.	4.6	107
13	Reactivity of Mount Simon Sandstone and the Eau Claire Shale Under CO <sub>2</sub> Storage Conditions. <i>Environmental Science &amp; Technology</i> , 2013, 47, 252-261.	10.0	102
14	Rock-Water Interactions Controlling Zinc, Cadmium, and Lead Concentrations in Surface Waters and Sediments, U.S. Tri-State Mining District. 2. Geochemical Interpretation. <i>Environmental Science &amp; Technology</i> , 1998, 32, 956-965.	10.0	93
15	Evaporite Caprock Integrity: An Experimental Study of Reactive Mineralogy and Pore-Scale Heterogeneity during Brine-CO <sub>2</sub> Exposure. <i>Environmental Science &amp; Technology</i> , 2013, 47, 262-268.	10.0	91
16	Rates of mineral dissolution under CO <sub>2</sub> storage conditions. <i>Chemical Geology</i> , 2015, 399, 134-144.	3.3	91
17	Chemical and Mechanical Properties of Wellbore Cement Altered by CO <sub>2</sub> -Rich Brine Using a Multianalytical Approach. <i>Environmental Science &amp; Technology</i> , 2013, 47, 1745-1752.	10.0	87
18	X-Ray Absorption Spectroscopy of Strontium(II) Coordination. <i>Journal of Colloid and Interface Science</i> , 2000, 222, 184-197.	9.4	84

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19	Metal Speciation and Bioavailability in Contaminated Estuary Sediments, Alameda Naval Air Station, California. <i>Environmental Science &amp; Technology</i> , 2000, 34, 3665-3673.	10.0	82
20	Geochemical detection of carbon dioxide in dilute aquifers. <i>Geochemical Transactions</i> , 2009, 10, 4.	0.7	77
21	Permeability of Wellbore-Cement Fractures Following Degradation by Carbonated Brine. <i>Rock Mechanics and Rock Engineering</i> , 2013, 46, 455-464.	5.4	73
22	Evaluation of silica-water surface chemistry using NMR spectroscopy. <i>Geochimica Et Cosmochimica Acta</i> , 2002, 66, 913-926.	3.9	70
23	Experimental Study of Cement - Sandstone/Shale - Brine - CO <sub>2</sub> Interactions. <i>Geochemical Transactions</i> , 2011, 12, 9.	0.7	65
24	Water Challenges for Geologic Carbon Capture and Sequestration. <i>Environmental Management</i> , 2010, 45, 651-661.	2.7	64
25	Development of scaling parameters to describe CO <sub>2</sub> -rock interactions within Weyburn-Midale carbonate flow units. <i>International Journal of Greenhouse Gas Control</i> , 2013, 16, S185-S193.	4.6	63
26	Experimental calibration of a numerical model describing the alteration of cement/caprock interfaces by carbonated brine. <i>International Journal of Greenhouse Gas Control</i> , 2014, 22, 176-188.	4.6	62
27	Direct electrolytic dissolution of silicate minerals for air CO <sub>2</sub> mitigation and carbon-negative H <sub>2</sub> production. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2013, 110, 10095-10100.	7.1	61
28	Mechanical and hydraulic coupling in cement-caprock interfaces exposed to carbonated brine. <i>International Journal of Greenhouse Gas Control</i> , 2014, 25, 109-120.	4.6	59
29	Kinetics of chlorite dissolution at elevated temperatures and CO <sub>2</sub> conditions. <i>Chemical Geology</i> , 2013, 347, 1-8.	3.3	56
30	Interactions of U(VI), Nd, and Th(IV) at the Calcite-Solution Interface. <i>Radiochimica Acta</i> , 1992, 58-59, 245-252.	1.2	50
31	Managing geologic CO <sub>2</sub> storage with pre-injection brine production: a strategy evaluated with a model of CO <sub>2</sub> injection at Sn�hvit. <i>Energy and Environmental Science</i> , 2016, 9, 1504-1512.	30.8	50
32	Speciation and fate of trace metals in estuarine sediments under reduced and oxidized conditions, Sealane Lagoon, Alameda Naval Air Station (USA). <i>Geochemical Transactions</i> , 2002, 3, 1.	0.7	47
33	Trace Metal Source Terms in Carbon Sequestration Environments. <i>Environmental Science &amp; Technology</i> , 2013, 47, 322-329.	10.0	46
34	Surface complexation model for strontium sorption to amorphous silica and goethite. <i>Geochemical Transactions</i> , 2008, 9, 2.	0.7	45
35	Influence of Chemical, Mechanical, and Transport Processes on Wellbore Leakage from Geologic CO <sub>2</sub> Storage Reservoirs. <i>Accounts of Chemical Research</i> , 2017, 50, 1829-1837.	15.6	39
36	Effect of thermal stress on wellbore integrity during CO <sub>2</sub> injection. <i>International Journal of Greenhouse Gas Control</i> , 2018, 77, 14-26.	4.6	38

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37	Pre-injection brine production in CO <sub>2</sub> storage reservoirs: An approach to augment the development, operation, and performance of CCS while generating water. <i>International Journal of Greenhouse Gas Control</i> , 2016, 54, 499-512.	4.6	35
38	Development and calibration of a reactive transport model for carbonate reservoir porosity and permeability changes based on CO <sub>2</sub> core-flood experiments. <i>International Journal of Greenhouse Gas Control</i> , 2017, 57, 73-88.	4.6	35
39	Effect of solution saturation state and temperature on diopside dissolution. <i>Geochemical Transactions</i> , 2007, 8, 3.	0.7	34
40	Mineral-Solution Interactions in the U(VI)-CO <sub>2</sub> -H <sub>2</sub> O System. <i>Radiochimica Acta</i> , 1991, 52-53, 187-194.	1.2	33
41	Applicability of aquifer impact models to support decisions at CO <sub>2</sub> sequestration sites. <i>International Journal of Greenhouse Gas Control</i> , 2016, 52, 319-330.	4.6	33
42	Incorporating reaction-rate dependence in reaction-front models of wellbore-cement/carbonated-brine systems. <i>International Journal of Greenhouse Gas Control</i> , 2017, 59, 160-171.	4.6	30
43	Quantification of Key Long-term Risks at CO <sub>2</sub> Sequestration Sites: Latest Results from US DOE's National Risk Assessment Partnership (NRAP) Project. <i>Energy Procedia</i> , 2014, 63, 4816-4823.	1.8	29
44	Muscovite dissolution kinetics as a function of pH at elevated temperature. <i>Chemical Geology</i> , 2017, 466, 149-158.	3.3	27
45	Assessment of geophysical monitoring methods for detection of brine and CO <sub>2</sub> leakage in drinking water aquifers. <i>International Journal of Greenhouse Gas Control</i> , 2019, 90, 102803.	4.6	26
46	Transport and detection of carbon dioxide in dilute aquifers. <i>Energy Procedia</i> , 2009, 1, 2111-2118.	1.8	24
47	Value of information methodology for assessing the ability of electrical resistivity to detect CO <sub>2</sub> /brine leakage into a shallow aquifer. <i>International Journal of Greenhouse Gas Control</i> , 2013, 18, 101-113.	4.6	24
48	Illite dissolution kinetics from 100 to 280 °C and pH 3 to 9. <i>Geochimica Et Cosmochimica Acta</i> , 2017, 209, 9-23.	3.9	22
49	A risk-based approach to evaluating the Area of Review and leakage risks at CO <sub>2</sub> storage sites. <i>International Journal of Greenhouse Gas Control</i> , 2020, 93, 102884.	4.6	22
50	Assessment of two-phase flow on the chemical alteration and sealing of leakage pathways in cemented wellbores. <i>International Journal of Greenhouse Gas Control</i> , 2018, 69, 72-80.	4.6	21
51	Real Time 3D Observations of Portland Cement Carbonation at CO <sub>2</sub> Storage Conditions. <i>Environmental Science &amp; Technology</i> , 2020, 54, 8323-8332.	10.0	21
52	Multiscale modeling of CO <sub>2</sub> -induced carbonate dissolution: From core to meter scale. <i>International Journal of Greenhouse Gas Control</i> , 2019, 88, 272-289.	4.6	19
53	Precipitation of Nd-Ca carbonate solid solution at 25 °C. <i>Geochimica Et Cosmochimica Acta</i> , 1993, 57, 3383-3393.	3.9	17
54	Determination of Diffusion Profiles in Altered Wellbore Cement Using X-ray Computed Tomography Methods. <i>Environmental Science &amp; Technology</i> , 2014, 48, 7094-7100.	10.0	17

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55	Chlorite dissolution kinetics at pH 3–10 and temperature to 275 °C. <i>Chemical Geology</i> , 2016, 421, 55-64.	3.3	16
56	Downhole pressure and chemical monitoring for CO <sub>2</sub> and brine leak detection in aquifers above a CO <sub>2</sub> storage reservoir. <i>International Journal of Greenhouse Gas Control</i> , 2019, 91, 102812.	4.6	15
57	Merits of pressure and geochemical data as indicators of CO <sub>2</sub> /brine leakage into a heterogeneous, sedimentary aquifer. <i>International Journal of Greenhouse Gas Control</i> , 2016, 52, 237-249.	4.6	14
58	Wellbore integrity in carbon sequestration environments: 1. Experimental study of Cement–Sandstone/Shale–Brine–CO <sub>2</sub> . <i>Energy Procedia</i> , 2011, 4, 5186-5194.	1.8	13
59	Wellbore integrity at the Krechba Carbon Storage Site, In Salah, Algeria: 2. Reactive transport modeling of geochemical interactions near the Cement–Formation interface. <i>Energy Procedia</i> , 2011, 4, 5195-5202.	1.8	13
60	Calibration of NMR porosity to estimate permeability in carbonate reservoirs. <i>International Journal of Greenhouse Gas Control</i> , 2019, 87, 19-26.	4.6	13
61	Risk-based Monitoring Network Design for Geologic Carbon Storage Sites. <i>Energy Procedia</i> , 2017, 114, 4345-4356.	1.8	12
62	A Review of Well Integrity Based on Field Experience at Carbon Utilization and Storage Sites. <i>International Journal of Greenhouse Gas Control</i> , 2022, 113, 103533.	4.6	12
63	The formation of metastable aluminosilicates in the Al–Si–H <sub>2</sub> O system: Results from solution chemistry and solid-state NMR spectroscopy. <i>Geochimica Et Cosmochimica Acta</i> , 2011, 75, 6080-6093.	3.9	10
64	Probing the Surface Structure of Divalent Transition Metals Using Surface Specific Solid-State NMR Spectroscopy. <i>Environmental Science &amp; Technology</i> , 2012, 46, 2806-2812.	10.0	10
65	Experiments and modeling of variably permeable carbonate reservoir samples in contact with CO <sub>2</sub> -acidified brines. <i>Energy Procedia</i> , 2014, 63, 3126-3137.	1.8	10
66	Fracture-scale model of immiscible fluid flow. <i>Physical Review E</i> , 2013, 87, 013012.	2.1	9
67	Managing Geologic CO <sub>2</sub> Storage with Pre-injection Brine Production in Tandem Reservoirs. <i>Energy Procedia</i> , 2017, 114, 4757-4764.	1.8	9
68	Reduced Order Models for Prediction of Groundwater Quality Impacts from CO <sub>2</sub> and Brine Leakage. <i>Energy Procedia</i> , 2014, 63, 4875-4883.	1.8	8
69	Impact of Chemical and Mechanical Processes on Leakage from Damaged Wells in CO <sub>2</sub> Storage Sites. <i>Environmental Science &amp; Technology</i> , 2020, 54, 1196-1203.	10.0	6
70	The Role of Wellbore Remediation on the Evolution of Groundwater Quality from CO <sub>2</sub> and Brine Leakage. <i>Energy Procedia</i> , 2014, 63, 4799-4806.	1.8	4
71	Framing Monitoring Needs to Detect Leakage from Wells to the Overburden. <i>Energy Procedia</i> , 2017, 114, 3628-3635.	1.8	4
72	Experimental investigation of cement, Topopah Spring tuff, and water interactions at 200 °C. <i>Applied Geochemistry</i> , 1998, 13, 571-579.	3.0	3

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73	Calibration of NMR well logs from carbonate reservoirs with laboratory NMR measurements and $^{13}\text{C}$ NMR. Energy Procedia, 2014, 63, 3089-3096.	1.8	3
74	Assessment of Thermal Stress on Well Integrity as a Function of Size and Material Properties. Energy Procedia, 2017, 114, 5241-5248.	1.8	3
75	Boiling Temperature and Reversed Deliquescence Relative Humidity Measurements for Mineral Assemblages in the NaCl + NaNO <sub>3</sub> + KNO <sub>3</sub> + Ca(NO <sub>3</sub> ) <sub>2</sub> + H <sub>2</sub> O System. Journal of Solution Chemistry, 2006, 35, 1187-1215.	1.2	2
76	Fabrication and Transport of Double Emulsion Microcapsules for Applications in Unconventional Resources. , 2015, , .		2
77	Non-invasive measurement of proppant pack deformation. International Journal of Rock Mechanics and Minings Sciences, 2016, 87, 39-47.	5.8	2
78	Validation of a reactive transport model for predicting changes in porosity and permeability in carbonate core samples. International Journal of Greenhouse Gas Control, 2019, 90, 102797.	4.6	2
79	Application of the Aquifer Impact Model to support decisions at a CO <sub>2</sub> sequestration site. , 2017, 7, 1020-1034.		1
80	Evaporative Evolution of Carbonate-Rich Brines from Synthetic Topopah Spring Tuff Pore Water, YuccaMountain, NV. Materials Research Society Symposia Proceedings, 2004, 824, 481.	0.1	0