

# Songhu Yuan

## List of Publications by Year in descending order

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76  
papers

3,727  
citations

126907

33  
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133252

59  
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77  
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77  
docs citations

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times ranked

2912  
citing authors

#	ARTICLE	IF	CITATIONS
1	Production of Abundant Hydroxyl Radicals from Oxygenation of Subsurface Sediments. <i>Environmental Science &amp; Technology</i> , 2016, 50, 214-221.	10.0	286
2	Electrolytic Manipulation of Persulfate Reactivity by Iron Electrodes for Trichloroethylene Degradation in Groundwater. <i>Environmental Science &amp; Technology</i> , 2014, 48, 656-663.	10.0	224
3	Mechanisms of hydroxyl radical production from abiotic oxidation of pyrite under acidic conditions. <i>Geochimica Et Cosmochimica Acta</i> , 2016, 172, 444-457.	3.9	198
4	Pd-Catalytic In Situ Generation of $H_2O_2$ from $H_2$ and $O_2$ Produced by Water Electrolysis for the Efficient Electro-Fenton Degradation of Rhodamine B. <i>Environmental Science &amp; Technology</i> , 2011, 45, 8514-8520.	10.0	192
5	Oxidizing Impact Induced by Mackinawite (FeS) Nanoparticles at Oxidic Conditions due to Production of Hydroxyl Radicals. <i>Environmental Science &amp; Technology</i> , 2016, 50, 11646-11653.	10.0	168
6	An integrated catalyst of Pd supported on magnetic Fe <sub>3</sub> O <sub>4</sub> nanoparticles: Simultaneous production of H <sub>2</sub> O <sub>2</sub> and Fe <sup>2+</sup> for efficient electro-Fenton degradation of organic contaminants. <i>Water Research</i> , 2014, 48, 190-199.	11.3	129
7	Mechanisms of electron transfer from structural Fe(II) in reduced nontronite to oxygen for production of hydroxyl radicals. <i>Geochimica Et Cosmochimica Acta</i> , 2018, 223, 422-436.	3.9	118
8	Electrocatalytic activity of Pd-loaded Ti/TiO <sub>2</sub> nanotubes cathode for TCE reduction in groundwater. <i>Water Research</i> , 2013, 47, 3573-3582.	11.3	113
9	Formation, Aggregation, and Deposition Dynamics of NOM-Iron Colloids at Anoxic-Oxic Interfaces. <i>Environmental Science &amp; Technology</i> , 2017, 51, 12235-12245.	10.0	105
10	Contaminant Degradation by $^{\bullet}OH$ during Sediment Oxygenation: Dependence on Fe(II) Species. <i>Environmental Science &amp; Technology</i> , 2020, 54, 2975-2984.	10.0	101
11	Production of hydroxyl radicals from abiotic oxidation of pyrite by oxygen under circumneutral conditions in the presence of low-molecular-weight organic acids. <i>Geochimica Et Cosmochimica Acta</i> , 2017, 218, 153-166.	3.9	100
12	Efficient Degradation of TCE in Groundwater Using Pd and Electro-generated $H_2$ and $O_2$ : A Shift in Pathway from Hydrodechlorination to Oxidation in the Presence of Ferrous Ions. <i>Environmental Science &amp; Technology</i> , 2012, 46, 3398-3405.	10.0	99
13	Efficient degradation of contaminants of emerging concerns by a new electro-Fenton process with Ti/MMO cathode. <i>Chemosphere</i> , 2013, 93, 2796-2804.	8.2	89
14	Oxidation of trichloroethylene by the hydroxyl radicals produced from oxygenation of reduced nontronite. <i>Water Research</i> , 2017, 113, 72-79.	11.3	85
15	Microbial reduction and precipitation of vanadium (V) in groundwater by immobilized mixed anaerobic culture. <i>Bioresource Technology</i> , 2015, 192, 410-417.	9.6	79
16	Oxidative Degradation of Organic Contaminants by FeS in the Presence of $O_2$ . <i>Environmental Science &amp; Technology</i> , 2020, 54, 4091-4101.	10.0	76
17	Efficient reduction of Cr(VI) in groundwater by a hybrid electro-Pd process. <i>Water Research</i> , 2014, 48, 326-334.	11.3	73
18	Formation and Transport of Cr(III)-NOM-Fe Colloids upon Reaction of Cr(VI) with NOM-Fe(II) Colloids at Anoxic-Oxic Interfaces. <i>Environmental Science &amp; Technology</i> , 2020, 54, 4256-4266.	10.0	73

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19	Mechanisms of hydroxyl radicals production from pyrite oxidation by hydrogen peroxide: Surface versus aqueous reactions. <i>Geochimica Et Cosmochimica Acta</i> , 2018, 238, 394-410.	3.9	66
20	Effect of reduced humic acid on the transport of ferrihydrite nanoparticles under anoxic conditions. <i>Water Research</i> , 2017, 109, 347-357.	11.3	61
21	Geochemical Stability of Dissolved Mn(III) in the Presence of Pyrophosphate as a Model Ligand: Complexation and Disproportionation. <i>Environmental Science &amp; Technology</i> , 2019, 53, 5768-5777.	10.0	57
22	Electrochemically Induced Oxidative Precipitation of Fe(II) for As(III) Oxidation and Removal in Synthetic Groundwater. <i>Environmental Science &amp; Technology</i> , 2014, 48, 5145-5153.	10.0	55
23	Oxidizing Capacity of Iron Electrocoagulation Systems for Refractory Organic Contaminant Transformation. <i>Environmental Science &amp; Technology</i> , 2019, 53, 12629-12638.	10.0	55
24	A three-electrode column for Pd-catalytic oxidation of TCE in groundwater with automatic pH-regulation and resistance to reduced sulfur compound foiling. <i>Water Research</i> , 2013, 47, 269-278.	11.3	51
25	Effect of Coexisting Fe(III) (oxyhydr)oxides on Cr(VI) Reduction by Fe(II)-Bearing Clay Minerals. <i>Environmental Science &amp; Technology</i> , 2019, 53, 13767-13775.	10.0	49
26	Electrochemically Induced Dual Reactive Barriers for Transformation of TCE and Mixture of Contaminants in Groundwater. <i>Environmental Science &amp; Technology</i> , 2012, 46, 12003-12011.	10.0	42
27	Production of Hydroxyl radicals from oxygenation of simulated AMD due to CaCO <sub>3</sub> -induced pH increase. <i>Water Research</i> , 2017, 111, 118-126.	11.3	40
28	Water Table Fluctuations Regulate Hydrogen Peroxide Production and Distribution in Unconfined Aquifers. <i>Environmental Science &amp; Technology</i> , 2020, 54, 4942-4951.	10.0	40
29	Integration of water collection and purification on cactus- and beetle-inspired eco-friendly superwetttable materials. <i>Water Research</i> , 2021, 206, 117759.	11.3	40
30	Abiotic degradation of methyl parathion by manganese dioxide: Kinetics and transformation pathway. <i>Chemosphere</i> , 2016, 150, 90-96.	8.2	37
31	Regulation of Electrochemically Generated Ferrous Ions from an Iron Cathode for Pd-Catalytic Transformation of MTBE in Groundwater. <i>Environmental Science &amp; Technology</i> , 2013, 47, 7918-7926.	10.0	36
32	A New Mechanism in Electrochemical Process for Arsenic Oxidation: Production of H <sub>2</sub> O <sub>2</sub> from Anodic O <sub>2</sub> Reduction on the Cathode under Automatically Developed Alkaline Conditions. <i>Environmental Science &amp; Technology</i> , 2015, 49, 5689-5696.	10.0	36
33	Impact of Fe(II) oxidation in the presence of iron-reducing bacteria on subsequent Fe(III) bio-reduction. <i>Science of the Total Environment</i> , 2018, 639, 1007-1014.	8.0	34
34	Iron-Anode Enhanced Sand Filter for Arsenic Removal from Tube Well Water. <i>Environmental Science &amp; Technology</i> , 2017, 51, 889-896.	10.0	33
35	Interplay between iron species transformation and hydroxyl radicals production in soils and sediments during anoxic-oxic cycles. <i>Geoderma</i> , 2020, 370, 114351.	5.1	32
36	Ligand-Enhanced Electron Utilization for Trichloroethylene Degradation by <sup>•</sup> OH during Sediment Oxygenation. <i>Environmental Science &amp; Technology</i> , 2021, 55, 7044-7051.	10.0	32

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37	Impact of Redox Reactions on Colloid Transport in Saturated Porous Media: An Example of Ferrihydrite Colloids Transport in the Presence of Sulfide. <i>Environmental Science &amp; Technology</i> , 2016, 50, 10968-10977.	10.0	31
38	Citrate-enhanced release of arsenic during pyrite oxidation at circumneutral conditions. <i>Water Research</i> , 2017, 109, 245-252.	11.3	31
39	Electrogeneration of H <sub>2</sub> for Pd-catalytic hydrodechlorination of 2,4-dichlorophenol in groundwater. <i>Chemosphere</i> , 2012, 87, 1097-1104.	8.2	28
40	Cu-catalytic generation of reactive oxidizing species from H <sub>2</sub> and O <sub>2</sub> produced by water electrolysis for electro-fenton degradation of organic contaminants. <i>Chemical Engineering Journal</i> , 2013, 233, 117-123.	12.7	28
41	Transformation and removal of arsenic in groundwater by sequential anodic oxidation and electrocoagulation. <i>Journal of Contaminant Hydrology</i> , 2014, 164, 299-307.	3.3	28
42	Responses of the Microbial Community Structure in Fe(II)-Bearing Sediments to Oxygenation: The Role of Reactive Oxygen Species. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 738-747.	2.7	27
43	Real-time evaluation of natural organic matter deposition processes onto model environmental surfaces. <i>Water Research</i> , 2018, 129, 231-239.	11.3	26
44	Electrochemical transformation of trichloroethylene in aqueous solution by electrode polarity reversal. <i>Water Research</i> , 2014, 67, 267-275.	11.3	25
45	Production of hydroxyl radicals from Fe(II) oxygenation induced by groundwater table fluctuations in a sand column. <i>Science of the Total Environment</i> , 2017, 584-585, 41-47.	8.0	25
46	Destabilization of emulsions by natural minerals. <i>Journal of Hazardous Materials</i> , 2011, 192, 1882-1885.	12.4	23
47	Anoxic storage regenerates reactive Fe(II) in reduced nontronite with short-term oxidation. <i>Geochimica Et Cosmochimica Acta</i> , 2019, 257, 96-109.	3.9	23
48	Effect of in situ generated iron oxyhydroxide coatings on FeS oxygenation and resultant hydroxyl radical production for contaminant degradation. <i>Chemical Engineering Journal</i> , 2020, 394, 124961.	12.7	22
49	Electrochemically induced oxidative removal of As(III) from groundwater in a dual-anode sand column. <i>Journal of Hazardous Materials</i> , 2016, 305, 41-50.	12.4	20
50	Oxygenation of acid sulfate soils stimulates CO <sub>2</sub> emission: Roles of acidic dissolution and hydroxyl radical oxidation. <i>Chemical Geology</i> , 2020, 533, 119437.	3.3	20
51	Benzene promotes microbial Fe(III) reduction and flavins secretion. <i>Geochimica Et Cosmochimica Acta</i> , 2019, 264, 92-104.	3.9	19
52	Cr(VI) Formation from Cr <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (OH) <sub>3</sub> Induced by Mn(II) Oxidation on the Surface of Cr <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (OH) <sub>3</sub> . <i>ACS Earth and Space Chemistry</i> , 2020, 4, 1558-1564.	2.7	19
53	Sulfide drives hydroxyl radicals production in oxic ferric oxyhydroxides environments. <i>Chemosphere</i> , 2019, 234, 450-460.	8.2	15
54	Reduced nontronite-activated H <sub>2</sub> O <sub>2</sub> for contaminants degradation: The beneficial role of clayed fractions in ISCO treatments. <i>Journal of Hazardous Materials</i> , 2020, 386, 121945.	12.4	15

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55	Electrolytic groundwater circulation well for trichloroethylene degradation in a simulated aquifer. <i>Science China Technological Sciences</i> , 2021, 64, 251-260.	4.0	15
56	Arsenic oxidation and immobilization in acid mine drainage in karst areas. <i>Science of the Total Environment</i> , 2020, 727, 138629.	8.0	14
57	Glucose oxidase modified Fenton reactions for in-situ ROS generation and potential application in groundwater remediation. <i>Chemosphere</i> , 2020, 253, 126648.	8.2	14
58	Mechanistic Insight into Humic Acid-Enhanced Hydroxyl Radical Production from Fe(II)-Bearing Clay Mineral Oxygenation. <i>Environmental Science &amp; Technology</i> , 2021, 55, 13366-13375.	10.0	14
59	Hydrodechlorination of TCE in a circulated electrolytic column at high flow rate. <i>Chemosphere</i> , 2016, 144, 59-64.	8.2	13
60	Attenuation of Fe(III)-reducing bacteria during table fluctuation of groundwater containing Fe <sup>2+</sup> . <i>Science of the Total Environment</i> , 2019, 694, 133660.	8.0	13
61	Oxidation of Fe(II) by Flavins under Anoxic Conditions. <i>Environmental Science &amp; Technology</i> , 2020, 54, 11622-11630.	10.0	13
62	Effect of dam on iron species distribution and transformation in riparian zones. <i>Journal of Hydrology</i> , 2022, 610, 127869.	5.4	12
63	Field tests of in-well electrolysis removal of arsenic from high phosphate and iron groundwater. <i>Science of the Total Environment</i> , 2018, 644, 1630-1640.	8.0	11
64	Model-based analysis of dissolved oxygen supply to aquifers within riparian zones during river level fluctuations: Dynamics and influencing factors. <i>Journal of Hydrology</i> , 2021, 598, 126460.	5.4	10
65	Pd-catalytic hydrodechlorination of chlorinated hydrocarbons in groundwater using H <sub>2</sub> produced by a dual-anode system. <i>Water Research</i> , 2015, 86, 74-81.	11.3	9
66	Optimization Strategies for in Situ Groundwater Remediation by a Vertical Circulation Well Based on Particle Tracking and Node-Dependent Finite Difference Methods. <i>Water Resources Research</i> , 2020, 56, e2020WR027396.	4.2	8
67	Effects of Reduced Sulfur Compounds on Pd-Catalytic Hydrodechlorination of Trichloroethylene in Groundwater by Cathodic H <sub>2</sub> under Electrochemically Induced Oxidizing Conditions. <i>Environmental Science &amp; Technology</i> , 2013, 47, 130904143021003.	10.0	7
68	Response to Comment on "Electrolytic Manipulation of Persulfate Reactivity by Iron Electrodes for TCE Degradation in Groundwater". <i>Environmental Science &amp; Technology</i> , 2014, 48, 4632-4633.	10.0	7
69	Distribution of Arsenite-Oxidizing Bacteria and its Correlation with Temperature in Hot Springs of the Tibetan-Yunnan Geothermal Zone in Western China. <i>Geomicrobiology Journal</i> , 2015, 32, 482-493.	2.0	7
70	Fe(II) oxygenation inhibits bacterial Mn(II) oxidation by <i>P. putida</i> MnB1 in groundwater under O <sub>2</sub> -perturbed conditions. <i>Journal of Hazardous Materials</i> , 2022, 435, 128972.	12.4	6
71	Response to Comment on "Production of Abundant Hydroxyl Radicals from Oxygenation of Subsurface Sediments". <i>Environmental Science &amp; Technology</i> , 2016, 50, 4890-4891.	10.0	5
72	Groundwater Circulation Enhanced Electrobioremediation of 1,2-Dichloroethane in a Simulated Heterogeneous Aquifer. <i>Environmental Engineering Science</i> , 2022, 39, 606-615.	1.6	5

#	ARTICLE	IF	CITATIONS
73	Adsorption Mechanism of Humic Acid on Cu/Fe Bimetallic Particles and Its Influence on the Reduction of Nitrobenzene in Groundwater. <i>Water, Air, and Soil Pollution</i> , 2014, 225, 1.	2.4	4
74	Redox transformation of structural iron in nontronite induced by quinones under anoxic conditions. <i>Science of the Total Environment</i> , 2021, 801, 149637.	8.0	3
75	Effects of riboflavin and desferrioxamine B on Fe(II) oxidation by O <sub>2</sub> . <i>Fundamental Research</i> , 2022, 2, 208-217.	3.3	3
76	Hydroxylamine promoted Fe(III) reduction in H <sub>2</sub> O <sub>2</sub> /soil systems for phenol degradation. <i>Environmental Science and Pollution Research</i> , 2022, 29, 30285-30296.	5.3	3