

Thom H Dunning

List of Publications by Year in descending order

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63
papers

57,394
citations

186265

28
h-index

118850

62
g-index

64
all docs

64
docs citations

64
times ranked

20798
citing authors

#	ARTICLE	IF	CITATIONS
1	Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen. <i>Journal of Chemical Physics</i> , 1989, 90, 1007-1023.	3.0	27,560
2	Electron affinities of the first-row atoms revisited. Systematic basis sets and wave functions. <i>Journal of Chemical Physics</i> , 1992, 96, 6796-6806.	3.0	13,437
3	Gaussian basis sets for use in correlated molecular calculations. III. The atoms aluminum through argon. <i>Journal of Chemical Physics</i> , 1993, 98, 1358-1371.	3.0	8,623
4	Benchmark calculations with correlated molecular wave functions. IV. The classical barrier height of the $\text{H}+\text{H}_2 \rightarrow \text{H}_2+\text{H}$ reaction. <i>Journal of Chemical Physics</i> , 1994, 100, 7410-7415.	3.0	1,636
5	Gaussian basis sets for use in correlated molecular calculations. X. The atoms aluminum through argon revisited. <i>Journal of Chemical Physics</i> , 2001, 114, 9244-9253.	3.0	1,463
6	Gaussian basis sets for use in correlated molecular calculations. IX. The atoms gallium through krypton. <i>Journal of Chemical Physics</i> , 1999, 110, 7667-7676.	3.0	1,309
7	Gaussian basis sets for use in correlated molecular calculations. VII. Valence, core-valence, and scalar relativistic basis sets for Li, Be, Na, and Mg. <i>Theoretical Chemistry Accounts</i> , 2011, 128, 69-82.	1.4	536
8	Generalized valence bond description of bonding in low-lying states of molecules. <i>Accounts of Chemical Research</i> , 1973, 6, 368-376.	15.6	467
9	NWChem: Past, present, and future. <i>Journal of Chemical Physics</i> , 2020, 152, 184102.	3.0	425
10	Variational transition state theory and tunneling for a heavy-“light”-heavy reaction using an ab initio potential energy surface. $37\text{Cl}+\text{H}(\text{D}) \rightarrow 35\text{Cl}+\text{H}(\text{D})$. <i>Journal of Chemical Physics</i> , 1983, 78, 4400-4413.	3.0	242
11	A theoretical study of the potential energy surface for $\text{OH}+\text{H}_2$. <i>Journal of Chemical Physics</i> , 1980, 72, 1303-1311.	3.0	206
12	Benchmark calculations with correlated molecular wave functions. VIII. Bond energies and equilibrium geometries of the CH_n and C_2H_n ($n=1-4$) series. <i>Journal of Chemical Physics</i> , 1997, 106, 4119-4140.	3.0	116
13	Valence correlation in the s^2d^n , sd^{n+1} , and $dn+2$ states of the first-row transition metal atoms. <i>Journal of Chemical Physics</i> , 1981, 75, 3466-3476.	3.0	113
14	Theoretical studies of the reactions of HCN with atomic hydrogen. <i>Journal of Chemical Physics</i> , 1985, 82, 2280-2294.	3.0	102
15	SO_2 revisited: Impact of tight d augmented correlation consistent basis sets on structure and energetics. <i>Journal of Chemical Physics</i> , 2003, 119, 11712-11714.	3.0	83
16	Theory of Hypervalency: Recoupled Pair Bonding in SF_6 ($n=6$). <i>Journal of Physical Chemistry A</i> , 2009, 113, 7915-7926.	2.5	81
17	Insights into the Perplexing Nature of the Bonding in C_2 from Generalized Valence Bond Calculations. <i>Journal of Chemical Theory and Computation</i> , 2014, 10, 195-201.	5.3	66
18	Theoretical characterization of the potential energy surface of the ground state of the HCO system. <i>Journal of Chemical Physics</i> , 1980, 73, 2304-2309.	3.0	65

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19	Theoretical characterization of negative ions. Calculation of the electron affinities of carbon, oxygen, and fluorine. <i>Journal of Chemical Physics</i> , 1982, 76, 6046-6056.	3.0	64
20	Reaction dynamics for O(3P)+H ₂ and D ₂ . IV. Reduced dimensionality quantum and quasiclassical rate constants with an adiabatic incorporation of the bending motion. <i>Journal of Chemical Physics</i> , 1984, 81, 1739-1752.	3.0	49
21	The First Row Anomaly and Recoupled Pair Bonding in the Halides of the Late p-Block Elements. <i>Accounts of Chemical Research</i> , 2013, 46, 359-368.	15.6	47
22	Theoretical studies of the O+H ₂ reaction. <i>Journal of Chemical Physics</i> , 1980, 72, 2894-2896.	3.0	42
23	Bonding in ClF _n (<i>n</i> = 1-7) Molecules: Further Insight into the Electronic Structure of Hypervalent Molecules and Recoupled Pair Bonds. <i>Journal of Physical Chemistry A</i> , 2009, 113, 12645-12654.	2.5	42
24	A comparison between polar covalent bonding and hypervalent recoupled pair bonding in diatomic chalcogen halide species {O,S,Se} Å— {F,Cl,Br}. <i>Molecular Physics</i> , 2009, 107, 991-998.	1.7	42
25	Insights into the Electronic Structure of Molecules from Generalized Valence Bond Theory. <i>Journal of Physical Chemistry A</i> , 2016, 120, 1763-1778.	2.5	42
26	Recoupled Pair Bonding in PF _n (<i>n</i> = 1-5). <i>Journal of Physical Chemistry A</i> , 2010, 114, 8845-8851.	2.5	41
27	From NWChem to NWChemEx: Evolving with the Computational Chemistry Landscape. <i>Chemical Reviews</i> , 2021, 121, 4962-4998.	47.7	39
28	Insights into the Electronic Structure of Ozone and Sulfur Dioxide from Generalized Valence Bond Theory: Bonding in O ₃ and SO ₂ . <i>Journal of Physical Chemistry A</i> , 2015, 119, 7683-7694.	2.5	37
29	Bonding in SCl _n (<i>n</i> = 1-6): A Quantum Chemical Study. <i>Journal of Physical Chemistry A</i> , 2011, 115, 4757-4764.	2.5	26
30	Spin-Coupled Generalized Valence Bond Theory: New Perspectives on the Electronic Structure of Molecules and Chemical Bonds. <i>Journal of Physical Chemistry A</i> , 2021, 125, 2021-2050.	2.5	26
31	Bonding and Isomerism in SF _n Cl (<i>n</i> = 1-6): A Quantum Chemical Study. <i>Journal of Physical Chemistry A</i> , 2011, 115, 329-341.	2.5	25
32	Bonding in Sulfur "Oxygen Compounds" HSO/SOH and SOO/OSO: An Example of Recoupled Pair Bonding. <i>Journal of Chemical Theory and Computation</i> , 2013, 9, 4444-4452.	5.3	24
33	Bonding in FSSF ₃ : Breakdown in Bond Length-Strength Correlations and Implications for SF ₂ Dimerization. <i>Journal of Physical Chemistry Letters</i> , 2013, 4, 3139-3143.	4.6	24
34	Hypervalency and recoupled pair bonding in the p-block elements. <i>Computational and Theoretical Chemistry</i> , 2011, 963, 7-12.	2.5	23
35	Fundamental aspects of recoupled pair bonds. I. Recoupled pair bonds in carbon and sulfur monofluoride. <i>Journal of Chemical Physics</i> , 2015, 142, 034113.	3.0	19
36	Orbital Hybridization in Modern Valence Bond Wave Functions: Methane, Ethylene, and Acetylene. <i>Journal of Physical Chemistry A</i> , 2020, 124, 204-214.	2.5	18

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37	Generalized Valence Bond Description of the Ground States ($X^{1\Sigma^+}$) of Homonuclear Pnictogen Diatomic Molecules: N_2 , P_2 , and As_2 . <i>Journal of Chemical Theory and Computation</i> , 2015, 11, 2496-2507.	5.3	17
38	The nature of the SO bond of chlorinated sulfur-oxygen compounds. <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	15
39	Fundamental aspects of recoupled pair bonds. II. Recoupled pair bond dyads in carbon and sulfur difluoride. <i>Journal of Chemical Physics</i> , 2015, 142, 034114.	3.0	15
40	Generalized Valence Bond Description of Chalcogen-Nitrogen Compounds. II. NO, F(NO), and H(NO). <i>Journal of Physical Chemistry A</i> , 2015, 119, 1456-1463.	2.5	13
41	Generalized Valence Bond Description of Chalcogen-Nitrogen Compounds. I. NS, F(NS), and H(NS). <i>Journal of Physical Chemistry A</i> , 2015, 119, 1446-1455.	2.5	13
42	Variations in the Nature of Triple Bonds: The N_2 , HCN, and HC_2H Series. <i>Journal of Physical Chemistry A</i> , 2016, 120, 4526-4533.	2.5	13
43	SiH_2 , a critical study. <i>Molecular Physics</i> , 2004, 102, 2597-2606.	1.7	12
44	Why edge inversion? Theoretical characterization of the bonding in the transition states for inversion in $F_nNH(3\hat{n})$ and $F_nPH(3\hat{n})$ ($n=3$). <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	12
45	Valence Bond and Molecular Orbital: Two Powerful Theories that Nicely Complement One Another. <i>Journal of Chemical Education</i> , 2021, 98, 3617-3620.	2.3	12
46	Spin-Coupled Generalized Valence Bond Description of Group 14 Species: The Carbon, Silicon and Germanium Hydrides, XH_n ($n=4$). <i>Journal of Physical Chemistry A</i> , 2019, 123, 2401-2419.	2.5	11
47	High Level ab Initio Calculations for ClF_n ($n=6$) Ions: Refining the Recoupled Pair Bonding Model. <i>Journal of Physical Chemistry A</i> , 2013, 117, 4251-4266.	2.5	10
48	Insights into the Electronic Structure of Disulfur Tetrafluoride Isomers from Generalized Valence Bond Theory. <i>Journal of Physical Chemistry A</i> , 2014, 118, 10117-10126.	2.5	10
49	The electronic structure of the two lowest states of CuC. <i>Journal of Chemical Physics</i> , 2008, 129, 174306.	3.0	9
50	Effects of Ligand Electronegativity on Recoupled Pair Bonds with Application to Sulfurane Precursors. <i>Journal of Physical Chemistry A</i> , 2014, 118, 5709-5719.	2.5	9
51	Insights into the Electronic Structure of Ozone and Sulfur Dioxide from Generalized Valence Bond Theory: Addition of Hydrogen Atoms. <i>Journal of Physical Chemistry A</i> , 2016, 120, 2720-2726.	2.5	9
52	Bonding in PF_2Cl , PF_3Cl , and PF_4Cl : insight into isomerism and apicophilicity from ab initio calculations and the recoupled pair bonding model. <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	8
53	Generalized Valence Bond Description of Chalcogen-Nitrogen Compounds. III. Why the NO-OH and NS-OH Bonds Are So Different. <i>Journal of Physical Chemistry A</i> , 2016, 120, 6846-6850.	2.5	7
54	Resolving a puzzling anomaly in the spin-coupled generalized valence bond description of benzene. <i>Journal of Computational Chemistry</i> , 2020, 41, 1421-1426.	3.3	6

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55	Fundamental Aspects of Recoupled Pair Bonds. III. The Frustrated Recoupled Pair Bond in Oxygen Monofluoride. <i>Journal of Physical Chemistry A</i> , 2016, 120, 9607-9611.	2.5	5
56	The nature of the chemical bond and the role of non-dynamical and dynamical correlation in Be ₂ . <i>Journal of Chemical Physics</i> , 2020, 152, 214111.	3.0	5
57	High level ab initio calculations on ClF _n ⁺ (n = 1-6): Recoupled pair bonding involving a closed-shell central ion. <i>Computational and Theoretical Chemistry</i> , 2017, 1116, 73-85.	2.5	4
58	A cautionary tale: Problems in the valence-CASSCF description of the ground state (X ¹ Σ ⁺) of BF. <i>Journal of Chemical Physics</i> , 2020, 153, 114113.	3.0	4
59	Quantum chemical calculations using the floating point systems, Inc. Model 164 attached processor. <i>International Journal of Quantum Chemistry</i> , 1983, 24, 613-622.	2.0	2
60	Reply to "Comment on "Insights into the Electronic Structure of Ozone and Sulfur Dioxide from Generalized Valence Bond Theory: Bonding in O ₃ and SO ₂ ". <i>Journal of Physical Chemistry A</i> , 2016, 120, 171-172.	2.5	2
61	New Insights into the Remarkable Difference between CH ₅ ⁺ and SiH ₅ ⁺ . <i>Journal of Physical Chemistry A</i> , 2021, 125, 7414-7424.	2.5	2
62	Dynamical electron correlation and the chemical bond. I. Covalent bonds in AH and AF (A = B-F). <i>Journal of Chemical Physics</i> , 0, , .	3.0	2
63	Nature of the Bonding in the Bifluoride Anion, FHF ⁻ . <i>Journal of Physical Chemistry Letters</i> , 2021, 12, 7293-7298.	4.6	1