## Debanjan Dhar

List of Publications by Year in descending order

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933447 1281871 1,069 11 10 11 citations h-index g-index papers 11 11 11 1329 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Copper–Oxygen Complexes Revisited: Structures, Spectroscopy, and Reactivity. Chemical Reviews, 2017, 117, 2059-2107.	47.7	505
2	Hydrogen Atom Abstraction from Hydrocarbons by a Copper(III)-Hydroxide Complex. Journal of the American Chemical Society, 2015, 137, 1322-1329.	13.7	149
3	Perturbing the Copper(III)–Hydroxide Unit through Ligand Structural Variation. Journal of the American Chemical Society, 2016, 138, 356-368.	13.7	100
4	Reactivity of the copper( <scp>iii</scp> )-hydroxide unit with phenols. Chemical Science, 2017, 8, 1075-1085.	7.4	60
5	Mechanistic Dichotomy in Proton-Coupled Electron-Transfer Reactions of Phenols with a Copper Superoxide Complex. Journal of the American Chemical Society, 2019, 141, 5470-5480.	13.7	55
6	Formally Copper(III)–Alkylperoxo Complexes as Models of Possible Intermediates in Monooxygenase Enzymes. Journal of the American Chemical Society, 2017, 139, 10220-10223.	13.7	52
7	Determination of the Cu(III)–OH Bond Distance by Resonance Raman Spectroscopy Using a Normalized Version of Badger's Rule. Journal of the American Chemical Society, 2017, 139, 4477-4485.	13.7	50
8	Nickel(II) Complex of a Hexadentate Ligand with Two <i>&gt;o</i> -lminosemiquinonato(1â^') Ï€-Radical Units and Its Monocation and Dication. Inorganic Chemistry, 2016, 55, 5759-5771.	4.0	36
9	Effects of Charged Ligand Substituents on the Properties of the Formally Copper(III)-Hydroxide ([CuOH] <sup>2+</sup> ) Unit. Inorganic Chemistry, 2018, 57, 9794-9806.	4.0	30
10	Redox-Induced Structural Reorganization Dictates Kinetics of Cobalt(III) Hydride Formation via Proton-Coupled Electron Transfer. Journal of the American Chemical Society, 2021, 143, 3393-3406.	13.7	24
11	Quantitative Effects of Disorder on Chemically Modified Amorphous Carbon Electrodes. ACS Applied Energy Materials, 2020, 3, 8038-8047.	5.1	8