

# Ziyun Wang

## List of Publications by Year in descending order

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Version: 2024-02-01

60  
papers

8,607  
citations

87888

38  
h-index

138484

58  
g-index

64  
all docs

64  
docs citations

64  
times ranked

6735  
citing authors

#	ARTICLE	IF	CITATIONS
1	Seeing is believing: In-situ visualising dynamic evolution in CO <sub>2</sub> electrolysis. <i>Current Opinion in Electrochemistry</i> , 2022, 31, 100846.	4.8	5
2	Efficient electrosynthesis of n-propanol from carbon monoxide using a Ag@Ru-Cu catalyst. <i>Nature Energy</i> , 2022, 7, 170-176.	39.5	96
3	High carbon utilization in CO <sub>2</sub> reduction to multi-carbon products in acidic media. <i>Nature Catalysis</i> , 2022, 5, 564-570.	34.4	197
4	Accurate and Affordable Explicit Solvent Quantum Mechanics for Electrocatalysis Investigations. <i>Matter</i> , 2021, 4, 12-14.	10.0	4
5	Rational catalyst design for CO oxidation: a gradient-based optimization strategy. <i>Catalysis Science and Technology</i> , 2021, 11, 2604-2615.	4.1	5
6	Stabilizing Highly Active Ru Sites by Suppressing Lattice Oxygen Participation in Acidic Water Oxidation. <i>Journal of the American Chemical Society</i> , 2021, 143, 6482-6490.	13.7	204
7	Silica-copper catalyst interfaces enable carbon-carbon coupling towards ethylene electrosynthesis. <i>Nature Communications</i> , 2021, 12, 2808.	12.8	91
8	Gold-in-copper at low *CO coverage enables efficient electromethanation of CO <sub>2</sub> . <i>Nature Communications</i> , 2021, 12, 3387.	12.8	70
9	Coordination Number-Dependent Complete Oxidation of Methane on NiO Catalysts. <i>ACS Catalysis</i> , 2021, 11, 9837-9849.	11.2	9
10	Ternary Alloys Enable Efficient Production of Methoxylated Chemicals via Selective Electrocatalytic Hydrogenation of Lignin Monomers. <i>Journal of the American Chemical Society</i> , 2021, 143, 17226-17235.	13.7	43
11	Boride-derived oxygen-evolution catalysts. <i>Nature Communications</i> , 2021, 12, 6089.	12.8	51
12	Catalyst synthesis under CO <sub>2</sub> electroreduction favours faceting and promotes renewable fuels electrosynthesis. <i>Nature Catalysis</i> , 2020, 3, 98-106.	34.4	325
13	Tuning OH binding energy enables selective electrochemical oxidation of ethylene to ethylene glycol. <i>Nature Catalysis</i> , 2020, 3, 14-22.	34.4	120
14	High-valence metals improve oxygen evolution reaction performance by modulating 3d metal oxidation cycle energetics. <i>Nature Catalysis</i> , 2020, 3, 985-992.	34.4	390
15	Promoting CO <sub>2</sub> methanation via ligand-stabilized metal oxide clusters as hydrogen-donating motifs. <i>Nature Communications</i> , 2020, 11, 6190.	12.8	93
16	Enhanced multi-carbon alcohol electroproduction from CO via modulated hydrogen adsorption. <i>Nature Communications</i> , 2020, 11, 3685.	12.8	72
17	High-Rate and Efficient Ethylene Electrosynthesis Using a Catalyst/Promoter/Transport Layer. <i>ACS Energy Letters</i> , 2020, 5, 2811-2818.	17.4	106
18	Efficient electrically powered CO <sub>2</sub> -to-ethanol via suppression of deoxygenation. <i>Nature Energy</i> , 2020, 5, 478-486.	39.5	363

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19	Accelerated discovery of CO <sub>2</sub> electrocatalysts using active machine learning. <i>Nature</i> , 2020, 581, 178-183.	27.8	807
20	Enhanced Nitrate-to-Ammonia Activity on Copper–Nickel Alloys via Tuning of Intermediate Adsorption. <i>Journal of the American Chemical Society</i> , 2020, 142, 5702-5708.	13.7	638
21	Achieving rational design of alloy catalysts using a descriptor based on a quantitative structure–energy equation. <i>Chemical Communications</i> , 2020, 56, 3214-3217.	4.1	13
22	Molecular tuning of CO <sub>2</sub> -to-ethylene conversion. <i>Nature</i> , 2020, 577, 509-513.	27.8	682
23	Hydrogenation of benzoic acid to benzyl alcohol over Pt/SnO <sub>2</sub> . <i>Applied Catalysis A: General</i> , 2020, 593, 117420.	4.3	15
24	Efficient Methane Electrosynthesis Enabled by Tuning Local CO <sub>2</sub> Availability. <i>Journal of the American Chemical Society</i> , 2020, 142, 3525-3531.	13.7	154
25	Cooperative CO <sub>2</sub> -to-ethanol conversion via enriched intermediates at molecule–metal catalyst interfaces. <i>Nature Catalysis</i> , 2020, 3, 75-82.	34.4	390
26	Gas Surface Interaction and Surface Reactions. <i>Springer Handbooks</i> , 2020, , 905-928.	0.6	0
27	Dopant-tuned stabilization of intermediates promotes electrosynthesis of valuable C <sub>3</sub> products. <i>Nature Communications</i> , 2019, 10, 4807.	12.8	26
28	CO <sub>2</sub> Electroreduction from Carbonate Electrolyte. <i>ACS Energy Letters</i> , 2019, 4, 1427-1431.	17.4	141
29	Binding Site Diversity Promotes CO <sub>2</sub> Electroreduction to Ethanol. <i>Journal of the American Chemical Society</i> , 2019, 141, 8584-8591.	13.7	338
30	Identifying the general trend of activity of non-stoichiometric metal oxide phases for CO oxidation on Pd(111). <i>Science China Chemistry</i> , 2019, 62, 784-789.	8.2	13
31	Synergy of Single-Atom Ni <sub>1</sub> and Ru <sub>1</sub> Sites on CeO <sub>2</sub> for Dry Reforming of CH <sub>4</sub> . <i>Journal of the American Chemical Society</i> , 2019, 141, 7283-7293.	13.7	272
32	Evidence of the O–Pd–O and Pd–O <sub>4</sub> structure units as oxide seeds and their origin on Pd(211): revealing the mechanism of surface oxide formation. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 6499-6505.	2.8	7
33	Efficient electrocatalytic conversion of carbon monoxide to propanol using fragmented copper. <i>Nature Catalysis</i> , 2019, 2, 251-258.	34.4	188
34	Hydroxide promotes carbon dioxide electroreduction to ethanol on copper via tuning of adsorbed hydrogen. <i>Nature Communications</i> , 2019, 10, 5814.	12.8	201
35	Efficient upgrading of CO to C <sub>3</sub> fuel using asymmetric C-C coupling active sites. <i>Nature Communications</i> , 2019, 10, 5186.	12.8	127
36	Constraining CO coverage on copper promotes high-efficiency ethylene electroproduction. <i>Nature Catalysis</i> , 2019, 2, 1124-1131.	34.4	214

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37	Copper adparticle enabled selective electrosynthesis of n-propanol. Nature Communications, 2018, 9, 4614.	12.8	153
38	A robust fuel cell operated on nearly dry methane at 500 Å°C enabled by synergistic thermal catalysis and electrocatalysis. Nature Energy, 2018, 3, 1042-1050.	39.5	230
39	Copper nanocavities confine intermediates for efficient electrosynthesis of C3 alcohol fuels from carbon monoxide. Nature Catalysis, 2018, 1, 946-951.	34.4	354
40	Ionic-Liquid-Controlled Two-Dimensional Monolayer Bi <sub>2</sub> MoO <sub>6</sub> and Its Adsorption of Azo Molecules. ACS Applied Nano Materials, 2018, 1, 5083-5091.	5.0	19
41	First-Principles Determination of CO Adsorption and Desorption on Pt(111) in the Free Energy Landscape. Journal of Physical Chemistry C, 2018, 122, 21478-21483.	3.1	29
42	Temperature-Controlled Selectivity of Hydrogenation and Hydrodeoxygenation in the Conversion of Biomass Molecule by the Ru <sub>1</sub> /mpg-C <sub>3</sub> N <sub>4</sub> Catalyst. Journal of the American Chemical Society, 2018, 140, 11161-11164.	13.7	199
43	Ordered Porous Nitrogen-Doped Carbon Matrix with Atomically Dispersed Cobalt Sites as an Efficient Catalyst for Dehydrogenation and Transfer Hydrogenation of N-Heterocycles. Angewandte Chemie, 2018, 130, 11432-11436.	2.0	24
44	Ordered Porous Nitrogen-Doped Carbon Matrix with Atomically Dispersed Cobalt Sites as an Efficient Catalyst for Dehydrogenation and Transfer Hydrogenation of N-Heterocycles. Angewandte Chemie - International Edition, 2018, 57, 11262-11266.	13.8	165
45	Insights into the mechanism of electrochemical ozone production via water splitting on the Ni and Sb doped SnO <sub>2</sub> catalyst. Physical Chemistry Chemical Physics, 2017, 19, 3800-3806.	2.8	18
46	Formulating the bonding contribution equation in heterogeneous catalysis: a quantitative description between the surface structure and adsorption energy. Physical Chemistry Chemical Physics, 2017, 19, 5063-5069.	2.8	15
47	Non-Thermal Plasma Activation of Gold-Based Catalysts for Low-Temperature Water-Gas Shift Catalysis. Angewandte Chemie, 2017, 129, 5671-5675.	2.0	11
48	Non-Thermal Plasma Activation of Gold-Based Catalysts for Low-Temperature Water-Gas Shift Catalysis. Angewandte Chemie - International Edition, 2017, 56, 5579-5583.	13.8	77
49	A rational catalyst design of CO oxidation using the bonding contribution equation. Chemical Communications, 2017, 53, 8106-8109.	4.1	11
50	Understanding Catalytic Reactions over Zeolites: A Density Functional Theory Study of Selective Catalytic Reduction of NO <sub>x</sub> by NH <sub>3</sub> over Cu-SAPO-34. ACS Catalysis, 2016, 6, 7882-7891.	11.2	99
51	Towards rational catalyst design: a general optimization framework. Philosophical Transactions Series A, Mathematical, Physical, and Engineering Sciences, 2016, 374, 20150078.	3.4	22
52	Elucidating the mechanism and active site of the cyclohexanol dehydrogenation on copper-based catalysts: A density functional theory study. Surface Science, 2015, 640, 181-189.	1.9	38
53	Understanding complete oxidation of methane on spinel oxides at a molecular level. Nature Communications, 2015, 6, 7798.	12.8	237
54	Some Attempts in the Rational Design of Heterogeneous Catalysts Using Density Functional Theory Calculations. Topics in Catalysis, 2015, 58, 633-643.	2.8	15

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55	Possibility of designing catalysts beyond the traditional volcano curve: a theoretical framework for multi-phase surfaces. <i>Chemical Science</i> , 2015, 6, 5703-5711.	7.4	40
56	Extraordinary mechanical properties of monatomic C <sub>3</sub> N <sub>2</sub> chain. <i>Molecular Simulation</i> , 2015, 41, 256-261.	2.0	0
57	Activity and coke formation of nickel and nickel carbide in dry reforming: A deactivation scheme from density functional theory. <i>Journal of Catalysis</i> , 2014, 311, 469-480.	6.2	231
58	Identifying the trend of reactivity for sp <sup>2</sup> materials: an electron delocalization model from first principles calculations. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 9498.	2.8	30
59	Deep Separation of Benzene from Cyclohexane by Liquid Extraction Using Ionic Liquids as the Solvent. <i>Industrial &amp; Engineering Chemistry Research</i> , 2012, 51, 5559-5564.	3.7	66
60	Evaluation of the ionic liquids 1-alkyl-3-methylimidazolium hexafluorophosphate as a solvent for the extraction of benzene from cyclohexane: (Liquid+liquid) equilibria. <i>Journal of Chemical Thermodynamics</i> , 2012, 48, 145-149.	2.0	49