

Vitaly Solovyov

List of Publications by Year in descending order

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68
papers

1,851
citations

257450

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265206

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71
all docs

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docs citations

71
times ranked

1443
citing authors

#	ARTICLE	IF	CITATIONS
1	ISIDA - Platform for Virtual Screening Based on Fragment and Pharmacophoric Descriptors. <i>Current Computer-Aided Drug Design</i> , 2008, 4, 191-198.	1.2	173
2	Substructural fragments: an universal language to encode reactions, molecular and supramolecular structures. <i>Journal of Computer-Aided Molecular Design</i> , 2005, 19, 693-703.	2.9	166
3	Exhaustive QSPR Studies of a Large Diverse Set of Ionic Liquids: How Accurately Can We Predict Melting Points?. <i>Journal of Chemical Information and Modeling</i> , 2007, 47, 1111-1122.	5.4	129
4	Correlation of blood-brain penetration using structural descriptors. <i>Bioorganic and Medicinal Chemistry</i> , 2006, 14, 4888-4917.	3.0	80
5	Modeling of Ion Complexation and Extraction Using Substructural Molecular Fragments. <i>Journal of Chemical Information and Computer Sciences</i> , 2000, 40, 847-858.	2.8	76
6	Crown Ether-Ammonium Complexes: Binding Mechanisms and Solvent Effects. <i>European Journal of Organic Chemistry</i> , 1999, 1999, 1847-1856.	2.4	72
7	Solvent Effects on Crown Ether Complexations. <i>Journal of Organic Chemistry</i> , 1996, 61, 5221-5226.	3.2	69
8	Benchmarking of Linear and Nonlinear Approaches for Quantitative Structure-Property Relationship Studies of Metal Complexation with Ionophores. <i>Journal of Chemical Information and Modeling</i> , 2006, 46, 808-819.	5.4	68
9	Quantitative Structure-Property Relationship Modeling of β -Cyclodextrin Complexation Free Energies. <i>Journal of Chemical Information and Computer Sciences</i> , 2004, 44, 529-541.	2.8	66
10	Determination of Successive Complexation Constants in an Ionic Liquid: Complexation of UO_2^{2+} with NO_3^- in $\text{C}_{4\text{-mim}}\text{Tf}_2\text{N}$ Studied by UV-Vis Spectroscopy. <i>Journal of Physical Chemistry B</i> , 2010, 114, 4276-4282.	2.6	60
11	Skin Permeation Rate as a Function of Chemical Structure. <i>Journal of Medicinal Chemistry</i> , 2006, 49, 3305-3314.	6.4	49
12	In Silico-Design of New Uranyl Extractants Based on Phosphoryl-Containing Podands: QSPR Studies, Generation and Screening of Virtual Combinatorial Library, and Experimental Tests. <i>Journal of Chemical Information and Computer Sciences</i> , 2004, 44, 1365-1382.	2.8	44
13	Assessment of the Macrocyclic Effect for the Complexation of Crown-Ethers with Alkali Cations Using the Substructural Molecular Fragments Method. <i>Journal of Chemical Information and Computer Sciences</i> , 2002, 42, 812-829.	2.8	41
14	QSAR modeling of blood:air and tissue:air partition coefficients using theoretical descriptors. <i>Bioorganic and Medicinal Chemistry</i> , 2005, 13, 6450-6463.	3.0	40
15	Anti-HIV Activity of HEPT, TIBO, and Cyclic Urea Derivatives: Structure-Property Studies, Focused Combinatorial Library Generation, and Hits Selection Using Substructural Molecular Fragments Method. <i>Journal of Chemical Information and Computer Sciences</i> , 2003, 43, 1703-1719.	2.8	39
16	The impact of alicyclic substituents on the extraction ability of new family of 1,10-phenanthroline-2,9-diamides. <i>RSC Advances</i> , 2020, 10, 26022-26033.	3.6	34
17	Complexation of phosphoryl-containing mono-, bi- and tri-podands with alkali cations in acetonitrile. Structure of the complexes and binding selectivity. <i>Journal of the Chemical Society Perkin Transactions II</i> , 1998, , 1489-1498.	0.9	33
18	Stochastic versus Stepwise Strategies for Quantitative Structure-Activity Relationship Generation How Much Effort May the Mining for Successful QSAR Models Take?. <i>Journal of Chemical Information and Modeling</i> , 2007, 47, 927-939.	5.4	32

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19	Interpretability of SAR/QSAR Models of any Complexity by Atomic Contributions. <i>Molecular Informatics</i> , 2012, 31, 639-642.	2.5	32
20	Conformation of 18-Crown-5 and Its Influence on Complexation with Alkali and Ammonium Cations: Why 18-Crown-5 Binds More Than 1000 Times Weaker Than 18C6. <i>Journal of Organic Chemistry</i> , 1996, 61, 8113-8116.	3.2	29
21	Steric and Stereoelectronic Effects in Aza Crown Ether Complexes[1]. <i>European Journal of Organic Chemistry</i> , 1998, 1998, 1379-1389.	2.4	28
22	Structure-property modeling of metal binders using molecular fragments. <i>Russian Chemical Bulletin</i> , 2004, 53, 1434-1445.	1.5	27
23	Quantitative Structure-Property Relationship (QSPR) Modeling of Normal Boiling Point Temperature and Composition of Binary Azeotropes. <i>Industrial & Engineering Chemistry Research</i> , 2011, 50, 14162-14167.	3.7	25
24	Successful <i>In Silico</i> -Design of New Efficient Uranyl Binders. <i>Solvent Extraction and Ion Exchange</i> , 2007, 25, 433-462.	2.0	24
25	Mining Chemical Reactions Using Neighborhood Behavior and Condensed Graphs of Reactions Approaches. <i>Journal of Chemical Information and Modeling</i> , 2012, 52, 2325-2338.	5.4	24
26	Supramolecular complexes: Determination of stability constants on the basis of various experimental methods. <i>Protection of Metals and Physical Chemistry of Surfaces</i> , 2015, 51, 1-35.	1.1	23
27	TOWARDS AN INFORMATION SYSTEM ON SOLVENT EXTRACTION. <i>Solvent Extraction and Ion Exchange</i> , 2001, 19, 791-837.	2.0	22
28	QSPR Modeling of the AmIII/EuIII Separation Factor: How Far Can we Predict ?. <i>Solvent Extraction and Ion Exchange</i> , 2007, 25, 1-26.	2.0	21
29	QSPR ensemble modelling of alkaline-earth metal complexation. <i>Journal of Inclusion Phenomena and Macrocyclic Chemistry</i> , 2013, 76, 159-171.	1.6	21
30	New Approach for Accurate QSPR Modeling of Metal Complexation: Application to Stability Constants of Complexes of Lanthanide Ions Ln ³⁺ , Ag ⁺ , Zn ²⁺ , Cd ²⁺ and Hg ²⁺ with Organic Ligands in Water. <i>Macrocyclic Chemistry</i> , 2012, 5, 404-410.	0.5	21
31	Stability constants of complexes of Zn ²⁺ , Cd ²⁺ , and Hg ²⁺ with organic ligands: QSPR consensus modeling and design of new metal binders. <i>Journal of Inclusion Phenomena and Macrocyclic Chemistry</i> , 2012, 72, 309-321.	1.6	19
32	QSPR ensemble modelling of the 1:1 and 1:2 complexation of Co ²⁺ , Ni ²⁺ , and Cu ²⁺ with organic ligands: relationships between stability constants. <i>Journal of Computer-Aided Molecular Design</i> , 2014, 28, 549-564.	2.9	19
33	Individual Hydrogen Bond Strength QSPR Modelling with ISIDA Local Descriptors: a Step Towards Polyfunctional Molecules. <i>Molecular Informatics</i> , 2014, 33, 477-487.	2.5	19
34	<i>In Silico</i> ; Design of Potential Anti-HIV Actives Using Fragment Descriptors. <i>Combinatorial Chemistry and High Throughput Screening</i> , 2005, 8, 403-416.	1.1	18
35	The complexation of metal ions with various organic ligands in water: prediction of stability constants by QSPR ensemble modelling. <i>Journal of Inclusion Phenomena and Macrocyclic Chemistry</i> , 2015, 83, 89-101.	1.6	17
36	Structure-property modelling of complex formation of strontium with organic ligands in water. <i>Journal of Structural Chemistry</i> , 2006, 47, 298-311.	1.0	16

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37	Complexation of Mn ²⁺ , Fe ²⁺ , Y ³⁺ , La ³⁺ , Pb ²⁺ , and UO ₂ ²⁺ with Organic Ligands: QSPR Ensemble Modeling of Stability Constants. <i>Industrial & Engineering Chemistry Research</i> , 2012, 51, 13482-13489.	3.7	16
38	Coordination Numbers of Central Atoms in Coordination Compounds. <i>Russian Journal of Coordination Chemistry/Koordinatsionnaya Khimiya</i> , 2003, 29, 660-668.	1.0	13
39	Computer-aided design of new metal binders. <i>Radiochimica Acta</i> , 2008, 96, 505-511.	1.2	13
40	Predictive Models for Halogenâ€bond Basicity of Binding Sites of Polyfunctional Molecules. <i>Molecular Informatics</i> , 2016, 35, 70-80.	2.5	12
41	QSPR modeling of potentiometric sensitivity towards heavy metal ions for polymeric membrane sensors. <i>Sensors and Actuators B: Chemical</i> , 2019, 301, 126941.	7.8	11
42	Complexation of the new tetrakis[methyl(diphenylphosphorylated)] cyclen derivative with transition metals: First examples of octacoordinate zinc(II) and cobalt(II) complexes with cyclen molecules. <i>Inorganica Chimica Acta</i> , 2018, 478, 250-259.	2.4	10
43	2,4,6-Tris[2-(diphenylphosphoryl)-4-ethylphenoxy]-1,3,5-triazine: A new ligand for lithium binding. <i>Inorganica Chimica Acta</i> , 2019, 497, 119095.	2.4	10
44	Predictive Models for the Free Energy of Hydrogen Bonded Complexes with Single and Cooperative Hydrogen Bonds. <i>Molecular Informatics</i> , 2016, 35, 629-638.	2.5	9
45	QSPR Modeling of Potentiometric Mg ²⁺ /Ca ²⁺ Selectivity for PVCâ€plasticized Sensor Membranes. <i>Electroanalysis</i> , 2020, 32, 792-798.	2.9	9
46	Complexation of lithium and sodium cations with \hat{I}^2 -phosphorylate ethers, modelling terminal groups of organophosphorus podands. An experimental and theoretical study. <i>Journal of Molecular Structure</i> , 1992, 271, 311-325.	3.6	8
47	Classification of Metal Binders by Na ⁻ Bayes Classifier on the Base of Molecular Fragment Descriptors and Ensemble Modeling. <i>Molecular Informatics</i> , 2019, 38, e1900002.	2.5	8
48	Design of phosphoryl containing podands with Li ⁺ /Na ⁺ selectivity using machine learning. <i>SAR and QSAR in Environmental Research</i> , 2021, 32, 521-539.	2.2	7
49	Predictive cartography of metal binders using generative topographic mapping. <i>Journal of Computer-Aided Molecular Design</i> , 2017, 31, 701-714.	2.9	6
50	Predictive Models for HOMO and LUMO Energies of Nâ€Donor Heterocycles as Ligands for Lanthanides Separation. <i>Molecular Informatics</i> , 2018, 37, e1800025.	2.5	6
51	Machine learning analysis of microwave dielectric properties for seven structure types: The role of the processing and composition. <i>Journal of Physics and Chemistry of Solids</i> , 2021, 156, 110178.	4.0	6
52	Thermodynamics and selectivity of complexation of lithium and sodium thiocyanates with phosphorus-containing podands and compounds modeling the terminal groups of these podands. <i>Russian Chemical Bulletin</i> , 1994, 43, 1493-1499.	1.5	5
53	Complexation of gallium(III) nitrate with 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetrakis(methylenephosphonic acid). <i>Russian Chemical Bulletin</i> , 2018, 67, 2184-2187.	1.5	5
54	3D molecular fragment descriptors for structureâ€property modeling: predicting the free energies for the complexation between antipodal guests and \hat{I}^2 -cyclodextrins. <i>Journal of Inclusion Phenomena and Macrocyclic Chemistry</i> , 2017, 89, 167-175.	1.6	4

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55	Thermodynamic radii of lanthanide ions derived from metal–ligand complexes stability constants. <i>Journal of Inclusion Phenomena and Macrocyclic Chemistry</i> , 2020, 98, 69-78.	1.6	4
56	Determination of the purity, melting point and heat of melting of cyclic polyethers. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1986, 35, 632-633.	0.0	2
57	Phosphorus-containing podands. 6. Calorimetric study of complexation of 1,17-bis (diphenylphosphinyl)-3,6,9,12,15-pentaoxaheptadecane with alkali and alkaline-earth metal salts in acetonitrile. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1991, 40, 497-502.	0.0	2
58	Synthesis, Complexation Properties, and Evaluation of New Aminodiphosphonic Acids as Vector Molecules for ⁶⁸ Ga Radiopharmaceuticals. <i>Molecules</i> , 2021, 26, 2357.	3.8	2
59	Prediction of Stability Constants of metal-ligand Complexes Using Thermodynamic Radii of Metal Ions. <i>Comments on Inorganic Chemistry</i> , 2023, 43, 16-33.	5.2	2
60	Evaluation of the electron-donor and electron-acceptor functions of ionized atoms and groups in biologically active substances on the basis of thermodynamic data. <i>Pharmaceutical Chemistry Journal</i> , 1984, 18, 327-331.	0.8	1
61	Phosphorus-containing podands. 4. Complexation of calcium chloride in alcohol at 298 K by solution calorimetry. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1989, 38, 731-734.	0.0	1
62	Determination of standard enthalpies of complexation of calcium salts with 18-crown-6 by means of scanning calorimetry. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1986, 35, 1591-1593.	0.0	0
63	Calorimetric study of complexation of calcium chloride and nitrate with 18-crown-6 in ethanol at 298.15 K. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1987, 36, 545-547.	0.0	0
64	Calorimetric study of the interaction of calcium salts with some benzo-crown ethers. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1988, 37, 660-662.	0.0	0
65	Equilibrium constants and complexation enthalpies and entropies of lithium, sodium, potassium, ammonium, and calcium thiocyanates with benzo-12-crown-4 in acetonitrile. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1988, 37, 2163-2165.	0.0	0
66	Ternary mixed-ligand complexes of copper(II) with α -aminoalkylphosphonic acids and amino acids. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1991, 40, 492-496.	0.0	0
67	Thermodynamic study of hydrogen-bond formation by 2-chloro-2-fluoro-2-nitroarylethanol with hexa-methylphosphoric triamide and acetone. <i>Bulletin of the Academy of Sciences of the USSR Division of Chemical Science</i> , 1991, 40, 57-61.	0.0	0
68	Quantitative Structure–Property Relationship Modeling of β -Cyclodextrin Complexation Free Energies. <i>ChemInform</i> , 2004, 35, no.	0.0	0