Catherine A Mccammon

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/325772/publications.pdf

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258 papers

11,758 citations

²⁶⁶³⁰ 56 h-index

95 g-index

267 all docs

267 docs citations

times ranked

267

6596 citing authors

#	Article	IF	CITATIONS
1	The Redox State of Earth's Mantle. Annual Review of Earth and Planetary Sciences, 2008, 36, 389-420.	11.0	856
2	Experimental evidence for the existence of iron-rich metal in the Earth's lower mantle. Nature, 2004, 428, 409-412.	27.8	500
3	The oxidation state of the mantle and the extraction of carbon from Earth's interior. Nature, 2013, 493, 84-88.	27.8	371
4	Perovskite as a possible sink for ferric iron in the lower mantle. Nature, 1997, 387, 694-696.	27.8	300
5	<i>MossA</i> : a program for analyzing energy-domain Mössbauer spectra from conventional and synchrotron sources. Journal of Applied Crystallography, 2012, 45, 329-331.	4.5	219
6	Systematic iron isotope variations in mantle rocks and minerals: The effects of partial melting and oxygen fugacity. Earth and Planetary Science Letters, 2005, 235, 435-452.	4.4	206
7	Iron Partitioning and Density Changes of Pyrolite in Earth's Lower Mantle. Science, 2010, 327, 193-195.	12.6	197
8	Iron Isotope Fractionation and the Oxygen Fugacity of the Mantle. Science, 2004, 304, 1656-1659.	12.6	173
9	The < sup > 57 < /sup > Fe Synchrotron Mössbauer Source at the ESRF. Journal of Synchrotron Radiation, 2012, 19, 559-569.	2.4	171
10	Structural complexity of simple Fe2O3 at high pressures and temperatures. Nature Communications, 2016, 7, 10661.	12.8	161
11	GEOCHEMISTRY: The Paradox of Mantle Redox. Science, 2005, 308, 807-808.	12.6	156
12	Fe-N system at high pressure reveals a compound featuring polymeric nitrogen chains. Nature Communications, 2018, 9, 2756.	12.8	153
13	The effect of water activity on the oxidation and structural state of Fe in a ferro-basaltic melt. Geochimica Et Cosmochimica Acta, 2005, 69, 5071-5085.	3.9	151
14	Stable intermediate-spin ferrous iron in lower-mantle perovskite. Nature Geoscience, 2008, 1, 684-687.	12.9	150
15	Structure and elasticity of single-crystal (Mg,Fe)O and a new method of generating shear waves for gigahertz ultrasonic interferometry. Journal of Geophysical Research, 2002, 107, ECV 4-1.	3.3	149
16	Oxygen fugacity, temperature reproducibility, and H2O contents of nominally anhydrous piston-cylinder experiments using graphite capsules. American Mineralogist, 2008, 93, 1838-1844.	1.9	148
17	A redox profile of the Slave mantle and oxygen fugacity control in the cratonic mantle. Contributions To Mineralogy and Petrology, 2004, 148, 55-68.	3.1	146
18	Mössbauer Spectroscopy of Minerals. AGU Reference Shelf, 0, , 332-347.	0.6	142

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19	Intermediate-spin ferrous iron in lowermost mantle post-perovskite and perovskite. Nature Geoscience, 2008, 1, 688-691.	12.9	131
20	The Effect of Alumina on the Electrical Conductivity of Silicate Perovskite. , 1998, 282, 922-924.		126
21	Oxidation of the Kaapvaal lithospheric mantle driven by metasomatism. Contributions To Mineralogy and Petrology, 2009, 157, 491-504.	3.1	122
22	Ferric Iron Content of Mineral Inclusions in Diamonds from Sã Luiz: A View into the Lower Mantle. Science, 1997, 278, 434-436.	12.6	113
23	High Poisson's ratio of Earth's inner core explained by carbon alloying. Nature Geoscience, 2015, 8, 220-223.	12.9	113
24	Structural systematics of hydrous ringwoodite and water in Earth's interior. American Mineralogist, 2003, 88, 1402-1407.	1.9	110
25	Fractionation of oxygen and iron isotopes by partial melting processes: Implications for the interpretation of stable isotope signatures in mafic rocks. Earth and Planetary Science Letters, 2009, 283, 156-166.	4.4	110
26	Electrical conductivity of orthopyroxene and plagioclase in the lower crust. Contributions To Mineralogy and Petrology, 2012, 163, 33-48.	3.1	106
27	9. Insights into Phase Transformations from Mössbauer Spectroscopy. , 2000, , 241-264.		105
28	Lattice thermal conductivity of lower mantle minerals and heat flux from Earth's core. Proceedings of the National Academy of Sciences of the United States of America, 2011, 108, 17901-17904.	7.1	103
29	Hydrogen solubility and speciation in natural, gem-quality chromian diopside. American Mineralogist, 2004, 89, 941-949.	1.9	101
30	Deep magma ocean formation set the oxidation state of Earth's mantle. Science, 2019, 365, 903-906.	12.6	99
31	Sulfur solubility in reduced mafic silicate melts: Implications for the speciation and distribution of sulfur on Mercury. Earth and Planetary Science Letters, 2016, 448, 102-114.	4.4	98
32	The oxygen fugacity at which graphite or diamond forms from carbonate-bearing melts in eclogitic rocks. Contributions To Mineralogy and Petrology, 2015, 169, 1.	3.1	96
33	Tibetan chromitites: Excavating the slab graveyard. Geology, 2015, 43, 179-182.	4.4	94
34	Mantle plumes are oxidised. Earth and Planetary Science Letters, 2019, 527, 115798.	4.4	85
35	Stability of iron-bearing carbonates in the deep Earth's interior. Nature Communications, 2017, 8, 15960.	12.8	84
36	Effect of water on the electrical conductivity of lower crustal clinopyroxene. Journal of Geophysical Research, 2011, 116, .	3.3	82

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37	Precise Moessbauer milliprobe determination of ferric iron in rock-forming minerals and limitations of electron microprobe analysis. American Mineralogist, 1999, 84, 78-85.	1.9	81
38	Pressure dependence of H solubility in magnesiow $\tilde{A}\frac{1}{4}$ stite up to 25 GPa: Implications for the storage of water in the Earth's lower mantle. Geophysical Research Letters, 2002, 29, 89-1-89-4.	4.0	79
39	X-ray diffraction and $ exttt{M} ilde{A}\P$ ssbauer spectroscopy study of fcc iron hydride FeH at high pressures and implications for the composition of the Earth's core. Earth and Planetary Science Letters, 2011, 307, 409-414.	4.4	78
40	Pitfalls in geothermobarometry of eclogites: Fe 3+ and changes in the mineral chemistry of omphacite at ultrahigh pressures. Contributions To Mineralogy and Petrology, 2004, 147, 305-318.	3.1	77
41	Melting processes and mantle sources of lavas on Mercury. Earth and Planetary Science Letters, 2016, 439, 117-128.	4.4	77
42	GEOPHYSICS: Deep Diamond Mysteries. Science, 2001, 293, 813-814.	12.6	75
43	High-pressure behavior of iron carbide (Fe ₇ C ₃) at inner core conditions. Journal of Geophysical Research, 2011, 116, .	3.3	75
44	Magnetic exchange bias of more than 1 Tesla in a natural mineral intergrowth. Nature Nanotechnology, 2007, 2, 631-634.	31.5	74
45	Rate of hydrogen–iron redox exchange in silicate melts and glasses. Geochimica Et Cosmochimica Acta, 2003, 67, 2427-2441.	3.9	73
46	Electrical conductivities of pyrope-almandine garnets up to 19 GPa and 1700 ÂC. American Mineralogist, 2006, 91, 1371-1377.	1.9	73
47	Oxygen vacancy ordering in CaTiO ₃ –CaFeO _{2.5} perovskites: From isolated defects to infinite sheets. Phase Transitions, 1999, 69, 133-146.	1.3	71
48	The compositional variability of eudialyte-group minerals. Mineralogical Magazine, 2011, 75, 87-115.	1.4	69
49	A Mössbauer milliprobe: Practical considerations. Hyperfine Interactions, 1994, 92, 1235-1239.	0.5	68
50	The stability of magnesite in the transition zone and the lower mantle as function of oxygen fugacity. Geophysical Research Letters, 2011, 38, n/a-n/a.	4.0	67
51	A biogeochemical–hydrological framework for the role of redox-active compounds in aquatic systems. Nature Geoscience, 2021, 14, 264-272.	12.9	67
52	The structure of the metallic high-pressure Fe3O4polymorph: experimental and theoretical study. Journal of Physics Condensed Matter, 2003, 15, 7697-7706.	1.8	65
53	Arrested kinetic Li isotope fractionation at the margin of the IlÃmaussaq complex, South Greenland: Evidence for open-system processes during final cooling of peralkaline igneous rocks. Chemical Geology, 2007, 246, 207-230.	3.3	62
54	Portable laser-heating system for diamond anvil cells. Journal of Synchrotron Radiation, 2009, 16, 737-741.	2.4	61

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55	Changes in tourmaline composition during magmatic and hydrothermal processes leading to tin-ore deposition: The Cornubian Batholith, SW England. Ore Geology Reviews, 2017, 83, 215-234.	2.7	61
56	Effect of iron oxidation state on the electrical conductivity of the Earth's lower mantle. Nature Communications, 2013, 4, 1427.	12.8	60
57	Electronic spin states of ferric and ferrous iron in the lower-mantle silicate perovskite. American Mineralogist, 2012, 97, 592-597.	1.9	58
58	Importance of Correlation Effects in hcp Iron Revealed by a Pressure-Induced Electronic Topological Transition. Physical Review Letters, 2013, 110, 117206.	7.8	58
59	High-pressure spectroscopic study of siderite (FeCO ₃) with a focus on spin crossover. American Mineralogist, 2015, 100, 2670-2681.	1.9	57
60	Evidence for H2O-bearing fluids in the lower mantle from diamond inclusion. Lithos, 2016, 265, 237-243.	1.4	57
61	Pressure-Induced Magnetization in FeO: Evidence from Elasticity and Mössbauer Spectroscopy. Physical Review Letters, 2004, 93, 215502.	7.8	55
62	Charge-ordering transition in iron oxide Fe4O5 involving competing dimer and trimer formation. Nature Chemistry, 2016, 8, 501-508.	13.6	54
63	Shock-induced metallic iron nanoparticles in olivine-rich Martian meteorites. Earth and Planetary Science Letters, 2007, 262, 37-49.	4.4	53
64	Effect of Pressure on the Composition of the Lower Mantle End Member FexO. Science, 1993, 259, 66-68.	12.6	51
65	Low ferric iron content of (Mg,Fe)O at high pressures and temperatures. Geophysical Research Letters, 1998, 25, 1589-1592.	4.0	51
66	Effect of hydration on the single-crystal elasticity of Fe-bearing wadsleyite to 12 GPa. American Mineralogist, 2011, 96, 1606-1612.	1.9	51
67	Structure and density of molten fayalite at high pressure. Geochimica Et Cosmochimica Acta, 2013, 118, 118-128.	3.9	51
68	Portable double-sided laser-heating system for Mössbauer spectroscopy and X-ray diffraction experiments at synchrotron facilities with diamond anvil cells. Review of Scientific Instruments, 2012, 83, 124501.	1.3	50
69	Discovery of Fe7O9: a new iron oxide with a complex monoclinic structure. Scientific Reports, 2016, 6, 32852.	3.3	50
70	Fe3+-rich augite and high electrical conductivity in the deep lithosphere. Geology, 2012, 40, 131-134.	4.4	49
71	Multidisciplinary Constraints on the Abundance of Diamond and Eclogite in the Cratonic Lithosphere. Geochemistry, Geophysics, Geosystems, 2018, 19, 2062-2086.	2.5	49
72	Oxidized iron in garnets from the mantle transition zone. Nature Geoscience, 2018, 11, 144-147.	12.9	48

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73	Effects of the Iron Content and Redox State on the Structure of Sodium Borosilicate Glasses: A <scp>R</scp> aman, <scp>M</scp> össbauer and Boron Kâ€Edge <scp>XANES</scp> Spectroscopy Study. Journal of the American Ceramic Society, 2012, 95, 962-971.	3.8	47
74	Solubility of hydrogen and ferric iron in rutile and TiO2(II): Implications for phase assemblages during ultrahigh-pressure metamorphism and for the stability of silica polymorphs in the lower mantle. Geophysical Research Letters, 2004, 31, .	4.0	45
75	Cooling history of lunar Mg-suite gabbronorite 76255, troctolite 76535 and Stillwater pyroxenite SC-936: The record in exsolution and ordering in pyroxenes. Geochimica Et Cosmochimica Acta, 2006, 70, 6068-6078.	3.9	45
76	Acoustic velocities of pure and ironâ€bearing magnesium silicate perovskite measured to 25 GPa and 1200 K. Geophysical Research Letters, 2012, 39, .	4.0	45
77	Coupled Interactions between Volatile Activity and Fe Oxidation State during Arc Crustal Processes. Journal of Petrology, 2015, 56, 795-814.	2.8	45
78	Evidence for ionic conductivity in lower mantle (Mg,Fe)(Si,Al)O3perovskite. Journal of Geophysical Research, 2002, 107, ECV 11-1-ECV 11-7.	3.3	44
79	Effect of non-hydrostatic conditions on the elastic behaviour of magnetite: an in situ single-crystal X-ray diffraction study. Physics and Chemistry of Minerals, 2007, 34, 627-635.	0.8	44
80	Short-range order and Fe clustering in <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:mrow><mml:msub><mml:mrow><mml:mrow><mml:mtext>Mg</mml:mtext></mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mrow><mml:mro< td=""><td>ow>3.2nml:r</td><td>mn∦4</td></mml:mro<></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:mrow></mml:msub></mml:mrow></mml:math>	ow> 3.2 nml:r	mn∦4
81	Low-spin Fe2+ in silicate perovskite and a possible layer at the base of the lower mantle. Physics of the Earth and Planetary Interiors, 2010, 180, 215-221.	1.9	44
82	Crystal chemistry of the tsumcorite-group minerals. New data on ferrilotharmeyerite, tsumcorite, thometzekite, mounanaite, helmutwinklerite, and a redefinition of gartrellite. European Journal of Mineralogy, 1998, 10, 179-206.	1.3	44
83	Highâ€Pressure NiAsâ€Type Modification of FeN. Angewandte Chemie - International Edition, 2017, 56, 7302-7306.	13.8	43
84	Effect of redox on Fe–Mg–Mn exchange between olivine and melt and an oxybarometer for basalts. Contributions To Mineralogy and Petrology, 2020, 175, 1.	3.1	42
85	Iron-rich perovskite in the Earth's lower mantle. Earth and Planetary Science Letters, 2011, 309, 179-184.	4.4	41
86	Structurally hidden magnetic transitions in Fe <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:msub><mml:mrow></mml:mrow><mml:mn>3</mml:mn></mml:msub></mml:math> C at high pressures. Physical Review B, 2012, 85, .	3.2	41
87	Lower mantle electrical conductivity based on measurements of Al, $\hat{a}\in\%$ Fe-bearing perovskite under lower mantle conditions. Earth and Planetary Science Letters, 2014, 393, 165-172.	4.4	41
88	Oxidation potential in the Earth's lower mantle as recorded by ferropericlase inclusions in diamond. Earth and Planetary Science Letters, 2015, 417, 49-56.	4.4	40
89	Seismically invisible water in Earth's transition zone?. Earth and Planetary Science Letters, 2018, 498, 9-16.	4.4	40
90	Magnesium silicate perovskite and effect of iron oxidation state on its bulk sound velocity at the conditions of the lower mantle. Earth and Planetary Science Letters, 2014, 393, 182-186.	4.4	39

#	Article	IF	CITATIONS
91	Structural properties of ferromagnesian cordierites. American Mineralogist, 2001, 86, 66-79.	1.9	38
92	Comparison of two electron probe microanalysis techniques to determine ferric iron in synthetic w $\tilde{A}^{1}/4$ stite samples. European Journal of Mineralogy, 2000, 12, 63-71.	1.3	37
93	Mössbauer spectroscopic determination of Fe3+/Fe2+ in synthetic basaltic glass: a test of empirical fO2 equations under superliquidus and subliquidus conditions. Contributions To Mineralogy and Petrology, 2004, 147, 565-580.	3.1	37
94	A Mössbauer study of oxygen vacancy and cation distribution in 6H-BaTi1â^'xFexO3â^'x/2. Journal of Solid State Chemistry, 2004, 177, 262-267.	2.9	37
95	The magnesiowi; ½ stite: iron equilibrium and its implications for the activity-composition relations of (Mg,Fe) 2SiO4 olivine solid solutions. Contributions To Mineralogy and Petrology, 2003, 146, 308-325.	3.1	36
96	Metamorphic Na- and OH-rich disordered dravite with tetrahedral boron, associated with omphacite, from Syros, Greece: chemistry and structure. European Journal of Mineralogy, 2004, 16, 817-823.	1.3	35
97	Local Oxygen-Vacancy Ordering and Twinned Octahedral Tilting Pattern in the Bi _{0.81} Pb _{0.19} FeO _{2.905} Cubic Perovskite. Chemistry of Materials, 2012, 24, 1378-1385.	6.7	35
98	Magnetism in cold subducting slabs at mantle transition zone depths. Nature, 2019, 570, 102-106.	27.8	33
99	Ferric/ferrous iron ratios in sodic amphiboles: M�ssbauer analysis, stoichiometry-based model calculations and the high-resolution microanalytical flank method. Contributions To Mineralogy and Petrology, 2000, 140, 135-147.	3.1	32
100	The effect of silica on ferric/ferrous ratio in silicate melts: An experimental study using Mossbauer spectroscopy. American Mineralogist, 2010, 95, 545-555.	1.9	31
101	Effect of high pressure on the crystal structure and electronic properties of magnetite below 25 GPa. American Mineralogist, 2012, 97, 128-133.	1.9	31
102	Stability of Fe,Al-bearing bridgmanite in the lower mantle and synthesis of pure Fe-bridgmanite. Science Advances, 2016, 2, e1600427.	10.3	31
103	The transition from short-range to long-range ordering of oxygen vacancies in CaFexTi1-xO3-x/2 perovskites. Physical Chemistry Chemical Physics, 2000, 2, 3933-3941.	2.8	30
104	Oxidation state of the lower mantle: In situ observations of the iron electronic configuration in bridgmanite at extreme conditions. Earth and Planetary Science Letters, 2015, 423, 78-86.	4.4	30
105	Petrogenesis of the Rifted Southern Victoria Land Lithospheric Mantle, Antarctica, Inferred from Petrography, Geochemistry, Thermobarometry and Oxybarometry of Peridotite and Pyroxenite Xenoliths from the Mount Morning Eruptive Centre. Journal of Petrology, 2015, 56, 193-226.	2.8	30
106	Spin transition of ferric iron in the NAL phase: Implications for the seismic heterogeneities of subducted slabs in the lower mantle. Earth and Planetary Science Letters, 2016, 434, 91-100.	4.4	30
107	Hydrogenation of C60 at 2GPa pressure and high temperature. Chemical Physics, 2006, 325, 445-451.	1.9	29

Iron oxidation state of<mml:math xmlns:mml="http://www.w3.org/1998/Math/MathML"

display="inline"><mml:mrow><mml:msub><mml:mrow><mml:mtext>FeTiO</mml:mtext></mml:mrow><mml:mrow><mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp</mml:mpsp

#	Article	IF	Citations
109	The behaviour of ferric iron during partial melting of peridotite. Geochimica Et Cosmochimica Acta, 2018, 239, 235-254.	3.9	29
110	Microscopic properties to macroscopic behaviour: The influence of iron electronic state. Journal of Mineralogical and Petrological Sciences, 2006, 101, 130-144.	0.9	28
111	XANES study of the oxidation state of Cr in lower mantle phases: Periclase and magnesium silicate perovskite. American Mineralogist, 2007, 92, 966-972.	1.9	28
112	Moldavites from the Cheb Basin, Czech Republic. Geochimica Et Cosmochimica Acta, 2009, 73, 1145-1179.	3.9	27
113	High-pressure experimental and computational XANES studies of(Mg,Fe)(Si,Al)O3perovskite and (Mg,Fe)O ferropericlase as in the Earth's lower mantle. Physical Review B, 2009, 79, .	3.2	27
114	Tetrahedral occupancy of ferric iron in (Mg,Fe)O: Implications for point defects in the Earth's lower mantle. Physics of the Earth and Planetary Interiors, 2010, 180, 179-188.	1.9	27
115	Iron spin state in silicate perovskite at conditions of the Earth's deep interior. High Pressure Research, 2013, 33, 663-672.	1.2	27
116	Abnormal Elasticity of Feâ€Bearing Bridgmanite in the Earth's Lower Mantle. Geophysical Research Letters, 2018, 45, 4725-4732.	4.0	27
117	Structure, ordering and cation interactions in Ca-free P2 $1\/$ c clinopyroxenes. Physics and Chemistry of Minerals, 1998, 25, 249-258.	0.8	26
118	Ferric iron content of ferropericlase as a function of composition, oxygen fugacity, temperature and pressure: Implications for redox conditions during diamond formation in the lower mantle. Earth and Planetary Science Letters, 2013, 365, 7-16.	4.4	26
119	Tidal charts of the Indian Ocean north of 15°S. Journal of Geophysical Research, 1977, 82, 5993-5998.	3.3	25
120	Comparative compressibility of hydrous wadsleyite and ringwoodite: Effect of H ₂ O and implications for detecting water in the transition zone. Journal of Geophysical Research: Solid Earth, 2015, 120, 8259-8280.	3.4	25
121	Redox-induced lower mantle density contrast and effect on mantle structure and primitive oxygen. Nature Geoscience, 2016, 9, 723-727.	12.9	25
122	Diamond destruction and growth during mantle metasomatism: An experimental study of diamond resorption features. Earth and Planetary Science Letters, 2019, 506, 493-506.	4.4	25
123	Pressure-induced phase transition in Mg0.8Fe0.2O ferropericlase. Physics and Chemistry of Minerals, 2006, 33, 35-44.	0.8	24
124	Microstructural investigations on strongly stained olivines of the chassignite NWA 2737 and implications for its shock history. Earth and Planetary Science Letters, 2010, 300, 255-263.	4.4	24
125	Ferric iron and water incorporation in wadsleyite under hydrous and oxidizing conditions: A XANES, Mossbauer, and SIMS study. American Mineralogist, 2012, 97, 1483-1493.	1.9	24
126	Portable double-sided pulsed laser heating system for time-resolved geoscience and materials science applications. Review of Scientific Instruments, 2017, 88, 084501.	1.3	24

#	Article	lF	CITATIONS
127	Crystal chemistry of iron-containing perovskites. Phase Transitions, 1996, 58, 1-26.	1.3	23
128	High-Pressure Behavior of Perovskite: <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:msub>FeTiO<mml:mn></mml:mn></mml:msub></mml:math> Disso into <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:mo< td=""><td>ciation</td><td></td></mml:mo<></mml:math>	ciation	

#	Article	IF	Citations
145	Order–disorder–reorder process in thermally treated dolomite samples: a combined powder and single-crystal X-ray diffraction study. Physics and Chemistry of Minerals, 2012, 39, 319-328.	0.8	19
146	Electronic spin state of Fe,Al-containing MgSiO3 perovskite at lower mantle conditions. Lithos, 2014, 189, 167-172.	1.4	19
147	Water, iron, redox environment: effects on the wadsleyite–ringwoodite phase transition. Contributions To Mineralogy and Petrology, 2015, 170, 1.	3.1	19
148	Intersite partitioning of Mg and Fe in Ca-free high-pressure C2/c clinopyroxene. American Mineralogist, 1997, 82, 923-930.	1.9	18
149	Displacive phase transitions and spontaneous strains in oxygen deficient CaFexTi1-xO3-x/2perovskites (Olexle0.40). Journal of Physics Condensed Matter, 2000, 12, 3661-3670.	1.8	18
150	POLYAKOVITE-(Ce), (REE,Ca)4 (Mg,Fe2+) (Cr3+,Fe3+)2 (Ti,Nb)2 Si4 O22, A NEW METAMICT MINERAL SPECIES FROM THE ILMEN MOUNTAINS, SOUTHERN URALS, RUSSIA: MINERAL DESCRIPTION AND CRYSTAL CHEMISTRY. Canadian Mineralogist, 2001, 39, 1095-1104.	1.0	18
151	Incorporation of Fe and Al in MgSiO3 perovskite: An investigation by 27Al and 29Si NMR spectroscopy. American Mineralogist, 2012, 97, 1955-1964.	1.9	18
152	Single crystal synthesis of δ-(Al,Fe)OOH. American Mineralogist, 2017, 102, 1953-1956.	1.9	18
153	Saline aqueous fluid circulation in mantle wedge inferred from olivine wetting properties. Nature Communications, 2019, 10, 5557.	12.8	18
154	Effects of Temperature and Pressure on the Mössbauer Spectra of Models for the [4Fe-4S]2+ Clusters of Ironâ^Sulfur Proteins and the Structure of [PPh4]2[Fe4S4(SCH2CO2C2H5)4]. Inorganic Chemistry, 1999, 38, 4256-4261.	4.0	17
155	FeO and MnO high-pressure phase diagrams: relations between structural and magnetic properties. Phase Transitions, 2007, 80, 1151-1163.	1.3	17
156	Synchrotron Mössbauer Source technique for in situ measurement of iron-bearing inclusions in natural diamonds. Lithos, 2016, 265, 328-333.	1.4	17
157	Comparative study of the influence of pulsed and continuous wave laser heating on the mobilization of carbon and its chemical reaction with iron in a diamond anvil cell. Journal of Applied Physics, 2019, 125, .	2.5	17
158	FERRIAN WINCHITE FROM THE ILMEN MOUNTAINS, SOUTHERN URALS, RUSSIA, AND SOME PROBLEMS WITH THE CURRENT SCHEME FOR AMPHIBOLE NOMENCLATURE. Canadian Mineralogist, 2001, 39, 171-177.	1.0	16
159	A simultaneous deformation and diffusion experiment: Quantifying the role of deformation in enhancing metamorphic reactions. Earth and Planetary Science Letters, 2009, 278, 386-394.	4.4	16
160	Mott transition in <mml:math xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:msub><mml:mi mathvariant="normal">CaFe</mml:mi><mml:mn>2</mml:mn></mml:msub><mml:msub><mml:mi mathvariant="normal">O<mml:mi><mml:mn>4</mml:mn></mml:mi></mml:mi></mml:msub></mml:math> at around 50 GPa.	3.2	16
161	Physical Review B, 2013, 88, . Crystal chemistry of Fe3+-bearing (Mg, Fe)SiO3 perovskite: a single-crystal X-ray diffraction study. Physics and Chemistry of Minerals, 2014, 41, 409-417.	0.8	16
162	lron spin state in silicate glass at high pressure: Implications for melts in the Earth's lower mantle. Earth and Planetary Science Letters, 2014, 385, 130-136.	4.4	16

#	Article	IF	CITATIONS
163	Fe3+ partitioning systematics between orthopyroxene and garnet in mantle peridotite xenoliths and implications for thermobarometry of oxidized and reduced mantle rocks. Contributions To Mineralogy and Petrology, 2015, 169, 1.	3.1	16
164	The composition and redox state of bridgmanite in the lower mantle as a function of oxygen fugacity. Geochimica Et Cosmochimica Acta, 2021, 303, 110-136.	3.9	16
165	Chemical Stability of FeOOH at High Pressure and Temperature, and Oxygen Recycling in Early Earth History**. European Journal of Inorganic Chemistry, 2021, 2021, 3048-3053.	2.0	16
166	The magnetic structure of bernalite, Fe(OH)3. Journal of Magnetism and Magnetic Materials, 1996, 152, 33-39.	2.3	15
167	Iron–magnesium alloying at high pressures and temperatures. Journal of Physics Condensed Matter, 2004, 16, S1143-S1150.	1.8	15
168	Effect of Lone-Electron-Pair Cations on the Orientation of Crystallographic Shear Planes in Anion-Deficient Perovskites. Inorganic Chemistry, 2013, 52, 10009-10020.	4.0	15
169	Stability and Solubility of the FeAlO ₃ Component in Bridgmanite at Uppermost Lower Mantle Conditions. Journal of Geophysical Research: Solid Earth, 2020, 125, e2019JB018447.	3.4	15
170	Serpentinization-Driven H2 Production From Continental Break-Up to Mid-Ocean Ridge Spreading: Unexpected High Rates at the West Iberia Margin. Frontiers in Earth Science, 2021, 9, .	1.8	15
171	A MÃ \P ssbauer effect investigation of the magnetic behaviour of (Fe, Co)S1 + x solid solutions. Journal of Physics and Chemistry of Solids, 1982, 43, 431-437.	4.0	14
172	Description and crystal structure of a new mineral – plimerite, ZnFe3+4(PO4)3(OH)5 – the Zn-analogue of rockbridgeite and frondelite, from Broken Hill, New South Wales, Australia. Mineralogical Magazine, 2009, 73, 131-148.	1.4	14
173	Angular, spectral, and temporal properties of nuclear radiation from a <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:msup><mml:mrow></mml:mrow><mml:mn>57</mml:mn></mml:msup></mml:math> Fe synchrotron Mössbauer source. Physical Review A. 2012. 86	2.5	14
174	Fe-rich and As-bearing vesuvianite and wiluite from Kozlov, Czech Republic. American Mineralogist, 2013, 98, 1330-1337.	1.9	14
175	Quantification of water in majoritic garnet. American Mineralogist, 2015, 100, 1084-1092.	1.9	14
176	Highâ€Pressure Phase Transition of Iron: A Combined Magnetic Remanence and Mössbauer Study. Geochemistry, Geophysics, Geosystems, 2017, 18, 4646-4654.	2.5	14
177	Bridgmanite is nearly dry at the top of the lower mantle. Earth and Planetary Science Letters, 2021, 570, 117088.	4.4	14
178	Effect of MgO, Y2O3, and Fe2O3 on silicothermal synthesis and sintering of X-sialon. An XRD, multinuclear MAS NMR and 57Fe Mössbauer study. Journal of the European Ceramic Society, 2000, 20, 1975-1985.	5.7	13
179	Cation partitioning in an unusual strontian potassicrichterite from Siberia: Rietveld structure refinement and MA¶ssbauer spectroscopy. Mineralogical Magazine, 2000, 64, 19-23.	1.4	13
180	Magnetic properties of synthetic <i>P</i> 2 ₁ / <i>c</i> (Mg-Fe)SiO ₃ clinopyroxenes as observed from their low-temperature Mössbauer spectra and from SQUID magnetization measurements. American Mineralogist, 2001, 86, 957-964.	1.9	13

#	Article	IF	CITATIONS
181	High-pressure behavior of FeOCl. Physical Review B, 2013, 88, .	3.2	13
182	Probing nonequivalent sites in iron phosphide Fe2P and its mechanism of phase transition. European Physical Journal B, 2013, 86, 1.	1.5	13
183	Water in Transition Zone and Lower Mantle Minerals. Geophysical Monograph Series, 0, , 57-68.	0.1	13
184	Time differentiated nuclear resonance spectroscopy coupled with pulsed laser heating in diamond anvil cells. Review of Scientific Instruments, 2015, 86, 114501.	1.3	13
185	Experimental constraint on grain-scale fluid connectivity in subduction zones. Earth and Planetary Science Letters, 2020, 552, 116610.	4.4	13
186	Equation of state, bonding character, and phase transition of cubanite, CuFe ₂ S ₃ , studied from 0 to 5 GPa. American Mineralogist, 1995, 80, 1-8.	1.9	13
187	Pressure-induced structural phase transition of the iron end-member of ringwoodite (Â-Fe2SiO4) investigated by X-ray diffraction and Mossbauer spectroscopy. American Mineralogist, 2011, 96, 833-840.	1.9	12
188	Twoâ€stage spin transition of iron in FeAlâ€bearing phase D at lower mantle. Journal of Geophysical Research: Solid Earth, 2016, 121, 6411-6420.	3.4	12
189	Decomposition of ferropericlase (Mg0.80Fe0.20)O at high pressures and temperatures. Journal of Alloys and Compounds, 2005, 390, 41-45.	5.5	11
190	The equation of state of wadsleyite solid solutions: Constraining the effects of anisotropy and crystal chemistry. American Mineralogist, 2017, 102, 2494-2504.	1.9	11
191	Experimental investigation of FeCO3 (siderite) stability in Earth's lower mantle using XANES spectroscopy. American Mineralogist, 2019, 104, 1083-1091.	1.9	11
192	$ ext{M} ilde{A} extsf{q}$ ssbauer spectroscopy of quenched high-pressure phases: Investigating the Earth's interior. Hyperfine Interactions, 1994, 90, 89-105.	0.5	10
193	Monoclinic FeO at high pressures. Zeitschrift Fur Kristallographie - Crystalline Materials, 2008, 223, 461-464.	0.8	10
194	Identification of Mackinawite and Constraints on Its Electronic Configuration Using Mössbauer Spectroscopy. Minerals (Basel, Switzerland), 2020, 10, 1090.	2.0	10
195	Electronic transitions of iron in almandine-composition glass to 91 GPa. American Mineralogist, 2016, 101, 1659-1667.	1.9	9
196	The high-pressure behavior of spherocobaltite (CoCO3): a single crystal Raman spectroscopy and XRD study. Physics and Chemistry of Minerals, 2018, 45, 59-68.	0.8	9
197	Transport of melt and volatiles in magmas inferred from kinetic experiments on the partial melting of granitic rocks. Lithos, 2018, 318-319, 434-447.	1.4	9
198	Seismic detectability of carbonates in the deep Earth: A nuclear inelastic scattering study. American Mineralogist, 2020, 105, 325-332.	1.9	9

#	Article	IF	CITATIONS
199	THE CRYSTAL CHEMISTRY OF POTASSIC-FERRISADANAGAITE. Canadian Mineralogist, 2000, 38, 669-674.	1.0	9
200	Brendelite, (Bi,Pb)2Fe3+,2+O2(OH)(PO4), a new mineral from Schneeberg, Germany: Description and crystal structure. Mineralogy and Petrology, 1998, 63, 263-277.	1.1	8
201	xmins:mmi="nttp://www.w3.org/1998/Nath/Nath/Niat	nn al:2 ni> </td <td>mានl:mrow> <</td>	mានl:mrow> <
202	Ferrorhodonite, CaMn3Fe[Si5O15], a new mineral species from Broken Hill, New South Wales, Australia. Physics and Chemistry of Minerals, 2017, 44, 323-334.	0.8	8
203	Sound velocities of skiagite–iron–majorite solid solution to 56ÂGPa probed by nuclear inelastic scattering. Physics and Chemistry of Minerals, 2018, 45, 397-404.	0.8	8
204	Effect of Fe 3+ on Phase Relations in the Lower Mantle: Implications for Redox Melting in Stagnant Slabs. Journal of Geophysical Research: Solid Earth, 2019, 124, 12484-12497.	3.4	8
205	Oxygen Vacancy Substitution Linked to Ferric Iron in Bridgmanite at 27ÂGPa. Geophysical Research Letters, 2020, 47, e2019GL086296.	4.0	8
206	The CaGeO3–Ca3Fe2Ge3O12 garnet join: an experimental study. Physics and Chemistry of Minerals, 2005, 32, 197-207.	0.8	7
207	Mechanochemical activation of mixtures of wolframite (FeWO4) with carbon, studied by 57Fe M¶ssbauer spectroscopy. Journal of the European Ceramic Society, 2006, 26, 2581-2585.	5.7	7
208	The influence of solid solution on elastic wave velocity determination in (Mg,Fe)O using nuclear inelastic scattering. Physics of the Earth and Planetary Interiors, 2014, 229, 16-23.	1.9	7
209	Spin transition of ferric iron in the calciumâ€ferrite type aluminous phase. Journal of Geophysical Research: Solid Earth, 2017, 122, 5935-5944.	3.4	7
210	Single-crystal elasticity of iron-bearing phase E and seismic detection of water in Earth's upper mantle. American Mineralogist, 2019, 104, 1526-1529.	1.9	7
211	A new (Mg0.5Fe0.53+)(Si0.5Al0.53+)O3 LiNbO3-type phase synthesized at lower mantle conditions. American Mineralogist, 2019, 104, 1213-1216.	1.9	7
212	Carbon-Bearing Phases throughout Earth's Interior. , 2019, , 66-88.		7
213	Effects of composition and pressure on electronic states of iron in bridgmanite. American Mineralogist, 2020, 105, 1030-1039.	1.9	7
214	Spin state and electronic environment of iron in basaltic glass in the lower mantle. American Mineralogist, 2017, 102, 2106-2112.	1.9	7
215	The crystal structures of Fe-bearing MgCO ₃ <i>sp</i> <and<i>sp³-carbonates at 98â€GPa from single-crystal X-ray diffraction using synchrotron radiation. Acta Crystallographica Section E: Crystallographic Communications, 2020, 76, 715-719.</and<i>	0.5	7
216	Crystal field and charge transfer spectrum of (Mg, Fe)SiO3 majorite. Physics and Chemistry of Minerals, 1996, 23, 94.	0.8	6

#	Article	IF	CITATIONS
217	Title is missing!. , 2000, 126, 241-245.		6
218	Crystal-structure refinement of Na-bearing clinopyroxenes from mantle-derived eclogite xenoliths. American Mineralogist, 2007, 92, 1242-1245.	1.9	6
219	High-temperature Mossbauer spectroscopy: A probe for the relaxation time of Fe species in silicate melts and glasses. American Mineralogist, 2010, 95, 1701-1707.	1.9	6
220	Local environment and valence state of iron in microinclusions in fibrous diamonds: X-ray absorption and Mössbauer data. Russian Geology and Geophysics, 2010, 51, 1262-1266.	0.7	6
221	Nuclear forward scattering by the 68.7 keV state of ⁷³ Ge in CaGeO ₃ and GeO ₂ . Europhysics Letters, 2013, 104, 17006.	2.0	6
222	High-pressure synthesis of skiagite-majorite garnet and investigation of its crystal structure. American Mineralogist, 2015, 100, 2650-2654.	1.9	6
223	Coupled substitution of Fe3+and H+for Si in wadsleyite: A study by polarized infrared and Mössbauer spectroscopies and single-crystal X-ray diffraction. American Mineralogist, 2016, 101, 1236-1239.	1.9	6
224	Sound velocities of bridgmanite from density of states determined by nuclear inelastic scattering and first-principles calculations. Progress in Earth and Planetary Science, 2016, 3, .	3.0	6
225	Evidence for a pressure-induced spin transition in olivine-type LiFePO4 triphylite. Physical Review B, 2018, 97, .	3.2	6
226	Fate of Hydrocarbons in Iron-Bearing Mineral Environments during Subduction. Minerals (Basel,) Tj ETQq0 0 0 rg	BT /Overlo 2.0	ck 10 Tf 50 3
227	Discovery of Elgoresyite, (Mg,Fe)5Si2O9: Implications for Novel Iron-Magnesium Silicates in Rocky Planetary Interiors. ACS Earth and Space Chemistry, 2021, 5, 2124-2130.	2.7	6
228	The Effect of Feâ€Al Substitution on the Crystal Structure of MgSiO ₃ Bridgmanite. Journal of Geophysical Research: Solid Earth, 2021, 126, e2021JB021936.	3.4	6
229	How to stop a molecular rotator. $M\tilde{A}\P$ ssbauer spectroscopic studies on (îbenzene)(îcyclopentadienyl)iron(II) hexafluorophosphate in the presence and absence of high pressure. Chemical Communications, 1996, , 11-12.	4.1	5
230	A Mössbauer study of the color of Roman pottery from the Leptiminus archaeological site, Tunisia. Geoarchaeology - an International Journal, 2002, 17, 863-874.	1.5	5
231	Deep Earth carbon reactions through time and space. American Mineralogist, 2020, 105, 22-27.	1.9	5
232	Pressure Destabilizes Oxygen Vacancies in Bridgmanite. Journal of Geophysical Research: Solid Earth, 2021, 126, .	3.4	5
233	Anisotropic mean-squared-displacement tensor in cubic almandine garnet: a single crystal 57Fe Mössbauer study. Physics and Chemistry of Minerals, 2012, 39, 561-575.	0.8	4
234	Effect of composition on compressibility of skiagite-Fe-majorite garnet. American Mineralogist, 2017, 102, 184-191.	1.9	4

#	Article	IF	CITATIONS
235	Local Structure of Ferroic Iron Formates at Low Temperature and High Pressure Studied by Mössbauer Spectroscopy. Journal of Physical Chemistry C, 2019, 123, 21676-21684.	3.1	4
236	THE CRYSTAL STRUCTURE OF GLADIUSITE, (Fe2+,Mg)4Fe3+2 (PO4)(OH)11(H2O). Canadian Mineralogist, 2001, 39, 1121-1130.	1.0	4
237	Structural evolution in a pyrolitic magma ocean under mantle conditions. Earth and Planetary Science Letters, 2022, 584, 117473.	4.4	4
238	Response. Science, 1993, 261, 924-925.	12.6	3
239	Magnetic defect structure of iron-rich Fe x O. Hyperfine Interactions, 1994, 94, 1989-1993.	0.5	3
240	High-pressure in situ investigation of cubanite (CuFe2S3): Electronic structure. Hyperfine Interactions, 1994, 93, 1511-1514.	0.5	3
241	Trigonal distortion of ferropericlase (Mg0.8Fe0.2)O at high pressures. Doklady Physics, 2005, 50, 343-345.	0.7	3
242	Structural studies of New Zealand pounamu using Mössbauer spectroscopy and electron paramagnetic resonance. Journal of the Royal Society of New Zealand, 2005, 35, 385-398.	1.9	3
243	Thermal equation of state of synthetic orthoferrosilite at lunar pressures and temperatures. Physics and Chemistry of Minerals, 2013, 40, 691-703.	0.8	3
244	Annealing of metamict gadolinite-(Y): X-ray diffraction, Raman, IR, and Mössbauer spectroscopy. Zeitschrift Fur Kristallographie - Crystalline Materials, 2019, 234, 587-593.	0.8	3
245	Cramping a Molecular Rollerball. Investigation of the Effect of Pressure on the Mössbauer Spectra of Three Cyclopentadienyl(arene)iron(II) Salts. Inorganic Chemistry, 1997, 36, 4017-4023.	4.0	2
246	$M\tilde{A}\P$ ssbauer Spectroscopy in the Geosciences: Highlights and Perspectives. Hyperfine Interactions, 2002, 144/145, 289-296.	0.5	2
247	Reply to "Comments on â€~Spin crossover in (Mg,Fe)O: A Mössbauer effect study with an alternative interpretation of x-ray emission spectroscopy data' ― Physical Review B, 2007, 75, .	3.2	2
248	Titanium-Rich Magnesio-Hastingsite Macrocrysts In A Camptonite Dike, Lafarge Quarry, Montreal Island, Québec: Early Crystallization In A Pseudo-Unary System. Canadian Mineralogist, 2016, 54, 65-78.	1.0	2
249	Ferri-kaersutite, NaCa ₂ (Mg ₃ TiFe ³⁺)(Si ₆ Al ₂)O ₂₂ O< a new oxo-amphibole from Harrow Peaks, Northern Victoria Land, Antarctica. American Mineralogist, 2016. 101. 461-468.	_{2<td>sub>,</td>}	sub>,
250	The Effect of Pulsed Laser Heating on the Stability of Ferropericlase at High Pressures. Minerals (Basel, Switzerland), 2020, 10, 542.	2.0	2
251	Mössbauer Spectroscopy with High Spatial Resolution: Spotlight on Geoscience. Topics in Applied Physics, 2021, , 221-266.	0.8	2
252	Eine NiAsâ€artige Hochdruckmodifikation von FeN. Angewandte Chemie, 2017, 129, 7408-7412.	2.0	2

#	Article	IF	CITATIONS
253	A reversed redox gradient in Earth's mantle transition zone. Earth and Planetary Science Letters, 2021, 575, 117181.	4.4	1
254	Oxygen Vacancies in Perovskite and Related Structures: Implications for the Lower Mantle. Materials Research Society Symposia Proceedings, 2002, 718, 1.	0.1	1
255	XANES study of spin crossover in Fe-bearing silicate perovskite. Phase Transitions, 2009, 82, 336-343.	1.3	O
256	Vancomycin Use in Patients Discharged from the Emergency Department. Annals of Emergency Medicine, 2013, 62, S63-S64.	0.6	0
257	Sustainable oxygen evolution catalysis – electrochemical generation of mössbauerite <i>via</i> corrosion engineering of steel. Materials Advances, 2021, 2, 5650-5656.	5.4	O
258	Mössbauer Spectroscopy in the Geosciences: Highlights and Perspectives., 2003,, 289-296.		0