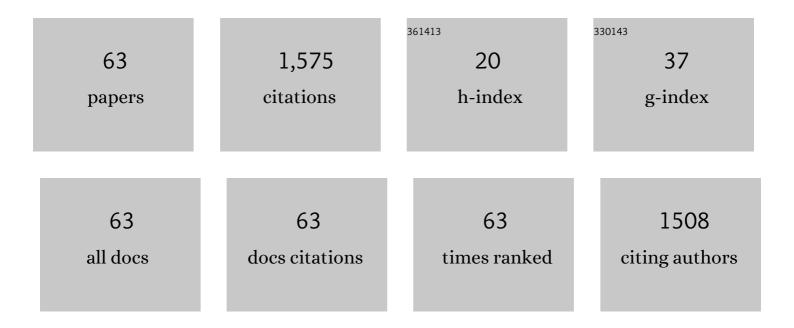
L M Da Silva

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Electrochemical Behavior of Symmetric Electrical Double-Layer Capacitors and Pseudocapacitors and Identification of Transport Anomalies in the Interconnected Ionic and Electronic Phases Using the Impedance Technique. Nanomaterials, 2022, 12, 676.	4.1	7
2	Direct observation of the CO2 formation and C–H consumption of carbon electrode in an aqueous neutral electrolyte supercapacitor by in-situ FTIR and Raman. Journal of Energy Chemistry, 2022, 71, 488-496.	12.9	10
3	In-situ electrochemical and operando Raman techniques to investigate the effect of porosity in different carbon electrodes in organic electrolyte supercapacitors. Journal of Energy Storage, 2022, 50, 104219.	8.1	10
4	Characterization of porous cobalt hexacyanoferrate and activated carbon electrodes under dynamic polarization conditions in a sodium-ion pseudocapacitor. Journal of Energy Chemistry, 2021, 54, 53-62.	12.9	20
5	Multi-walled carbon nanotubes and activated carbon composite material as electrodes for electrochemical capacitors. Journal of Energy Storage, 2021, 33, 100738.	8.1	20
6	Robust, freestanding, and bendable multi-walled carbon nanotube buckypapers as electrode materials for quasi-solid-state potassium-ion supercapacitors. Diamond and Related Materials, 2021, 115, 108354.	3.9	8
7	Ragone Plots for Electrochemical Double‣ayer Capacitors. Batteries and Supercaps, 2021, 4, 1291-1303.	4.7	36
8	Pseudocapacitive behaviour of iron oxides supported on carbon nanofibers as a composite electrode material for aqueous-based supercapacitors. Journal of Energy Storage, 2021, 42, 103052.	8.1	17
9	Charge-storage mechanism of highly defective NiO nanostructures on carbon nanofibers in electrochemical supercapacitors. Nanoscale, 2021, 13, 9590-9605.	5.6	14
10	Niobium pentoxide nanoparticles decorated graphene as electrode material in aqueous-based supercapacitors: Accurate determination of the working voltage window and the analysis of the distributed capacitance in the time domain. Journal of Energy Storage, 2021, 44, 103371.	8.1	16
11	New Insights on the Sodium Water-in-Salt Electrolyte and Carbon Electrode Interface from Electrochemistry and Operando Raman Studies. ACS Applied Materials & Interfaces, 2021, 13, 61139-61153.	8.0	10
12	Reviewing the fundamentals of supercapacitors and the difficulties involving the analysis of the electrochemical findings obtained for porousÂelectrode materials. Energy Storage Materials, 2020, 27, 555-590.	18.0	179
13	Pseudo-capacitive behavior of multi-walled carbon nanotubes decorated with nickel and manganese (hydr)oxides nanoparticles. Journal of Energy Storage, 2020, 31, 101583.	8.1	13
14	A rational experimental approach to identify correctly the working voltage window of aqueous-based supercapacitors. Scientific Reports, 2020, 10, 19195.	3.3	35
15	Study of the aging process of nanostructured porous carbon-based electrodes in electrochemical capacitors filled with aqueous or organic electrolytes. Journal of Energy Storage, 2020, 28, 101249.	8.1	15
16	Niobium pentoxide nanoparticles @ multi-walled carbon nanotubes and activated carbon composite material as electrodes for electrochemical capacitors. Energy Storage Materials, 2019, 22, 311-322.	18.0	34
17	Double-pulse chronoamperometry using short times for the kinetic study of simple quasi-reversible electrochemical reactions at low overpotentials. Journal of Electroanalytical Chemistry, 2019, 848, 113291.	3.8	2
18	Tungsten oxide and carbide composite synthesized by hot filament chemical deposition as electrodes in aqueous-based electrochemical capacitors. Journal of Energy Storage, 2019, 26, 100905.	8.1	9

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19	Antimicrobial alumina nanobiostructures of disulfide- and triazole-linked peptides: Synthesis, characterization, membrane interactions and biological activity. Colloids and Surfaces B: Biointerfaces, 2019, 177, 94-104.	5.0	17
20	Core-niobium pentoxide carbon-shell nanoparticles decorating multiwalled carbon nanotubes as electrode for electrochemical capacitors. Journal of Power Sources, 2019, 434, 226737.	7.8	23
21	How to Measure and Calculate Equivalent Series Resistance of Electric Double-Layer Capacitors. Molecules, 2019, 24, 1452.	3.8	68
22	Highly stable nickel-aluminum alloy current collectors and highly defective multi-walled carbon nanotubes active material for neutral aqueous-based electrochemical capacitors. Journal of Energy Storage, 2019, 23, 116-127.	8.1	18
23	Nickel oxide nanoparticles supported onto oriented multi-walled carbon nanotube as electrodes for electrochemical capacitors. Electrochimica Acta, 2019, 298, 468-483.	5.2	50
24	Environmentally Friendly Functionalization of Porous Carbon Electrodes for Aqueous-Based Electrochemical Capacitors. IEEE Nanotechnology Magazine, 2019, 18, 73-82.	2.0	10
25	Supercapacitive properties, anomalous diffusion, and porous behavior of nanostructured mixed metal oxides containing Sn, Ru, and Ir. Electrochimica Acta, 2019, 295, 302-315.	5.2	10
26	Enhanced ferroelectricity and conductance in iron-doped polystyrene sulfonate. Journal of Non-Crystalline Solids, 2019, 503-504, 103-109.	3.1	1
27	Nanobiostructure of fibrous-like alumina functionalized with an analog of the BP100 peptide: Synthesis, characterization and biological applications. Colloids and Surfaces B: Biointerfaces, 2018, 163, 275-283.	5.0	14
28	Fabrication and characterisation of a mixed oxide-covered mesh electrode composed of NiCo2O4 and its capability of generating hydroxyl radicals during the oxygen evolution reaction in electrolyte-free water. Journal of Solid State Electrochemistry, 2018, 22, 1289-1302.	2.5	7
29	Degradation of paracetamol in a bubble column reactor with ozone generated in electrolyte-free water using a solid polymer electrolyte filter-press electrochemical reactor. Journal of Solid State Electrochemistry, 2018, 22, 1349-1363.	2.5	3
30	Revisiting ideal gases and proposal of a simple experiment for determining atmospheric pressure in the laboratory. Quimica Nova, 2018, , .	0.3	1
31	Surface and Electrochemical Properties of Radially Oriented Multiwalled Carbon Nanotubes Grown on Stainless Steel Mesh. Journal of the Electrochemical Society, 2018, 165, A3684-A3696.	2.9	18
32	Synthesis of chitosan-stabilised bimetallic nanoparticles containing Fe and Ni and the reductive degradation of nimesulide. Ecletica Quimica, 2018, 43, 10.	0.5	1
33	Alternative method to obtain the Tafel plot for simple electrode reactions using batch injection analysis coupled with multiple-pulse amperometric detection. Electrochimica Acta, 2017, 242, 180-186.	5.2	13
34	An environmentally friendly electrochemical reactor for the degradation of organic pollutants in the total absence of a liquid electrolyte: A case study using diclofenac as a model pollutant. Journal of Environmental Chemical Engineering, 2017, 5, 3873-3881.	6.7	6
35	Synthesis, Characterization, and Application of Fe–Ni Bimetallic Nanoparticles for the Reductive Degradation of Nimesulide. Clean - Soil, Air, Water, 2017, 45, .	1.1	2
36	Fabrication and characterisation of mixed oxide-covered mesh electrodes of nominal composition Ni(x)Co(1 â^'x)O supported on stainless-steel prepared by thermal decomposition using the slow cooling rate method. Electrochimica Acta, 2016, 194, 127-135.	5.2	7

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37	Characterisation of silica-supported Fe–Ni bimetallic nanoparticles and kinetic study of reductive degradation of the drug nimesulide. Journal of Environmental Chemical Engineering, 2016, 4, 4354-4365.	6.7	14
38	Electrochemical ozone production using electrolyte-free water for environmental applications. Journal of Environmental Chemical Engineering, 2016, 4, 418-427.	6.7	21
39	Preparation, characterization, and application in biosensors of functionalized platforms with poly(4-aminobenzoic acid). Journal of Materials Science, 2015, 50, 1103-1116.	3.7	12
40	Simultaneous Determination of Paracetamol and Ibuprofen in Pharmaceutical Samples by Differential Pulse Voltammetry Using a Boron-Doped Diamond Electrode. Journal of the Brazilian Chemical Society, 2014, , .	0.6	14
41	Application of oxide fine-mesh electrodes composed of Sb-SnO 2 for the electrochemical oxidation of Cibacron Marine FG using an SPE filter-press reactor. Electrochimica Acta, 2014, 146, 714-732.	5.2	20
42	Fabrication and characterization of oxide fine-mesh electrodes composed of Sb-SnO2 and study of oxygen evolution from the electrolysis of electrolyte-free water in a solid polymer electrolyte filter-press cell: Possibilities for the combustion of organic pollutants. Electrochimica Acta, 2014, 121, 1-14.	5.2	13
43	Decolorization, Degradation and Toxicity of Dye Solutions Containing Orange Cassafix® CA-2R after UV/H2O2 Oxidation under Laminar Flow Conditions. Journal of Advanced Oxidation Technologies, 2014, 17, .	0.5	1
44	Electrochemical impedance spectroscopy study of the oxygen evolution reaction on a gas-evolving anode composed of lead dioxide microfibers. Electrochimica Acta, 2013, 90, 332-343.	5.2	40
45	Determinação de nimesulida por análise por injeção em fluxo com detecção amperométrica de múltiplos pulsos. Quimica Nova, 2013, 36, 1296-1302.	0.3	19
46	Fabrication and characterization of a porous gas-evolving anode constituted of lead dioxide microfibers electroformed on a carbon cloth substrate. Electrochimica Acta, 2012, 70, 365-374.	5.2	19
47	Influência das condições de resfriamento sobre as propriedades superficiais e eletroquÃmicas de anodos dimensionalmente estáveis. Quimica Nova, 2011, 34, 200-205.	0.3	9
48	Synthesis, characterization and electrochemical behavior of the vanadium pentoxide/cetyl pyridinium chloride hybrid material. Journal of Solid State Electrochemistry, 2010, 14, 305-312.	2.5	5
49	Characterization of an electrochemical reactor for the ozone production in electrolyte-free water. Journal of Applied Electrochemistry, 2010, 40, 855-864.	2.9	35
50	Chemical Reduction of Hexavalent Chromium Present in Contaminated Soil using a Packedâ€bed Column Reactor. Clean - Soil, Air, Water, 2009, 37, 858-865.	1.1	7
51	Reduction of Hexavalent Chromium in Soil and Ground Water Using Zero-Valent Iron Under Batch and Semi-Batch Conditions. Water, Air, and Soil Pollution, 2009, 197, 49-60.	2.4	63
52	Chemical Reduction of Hexavalent Chromium and Its Immobilisation Under Batch Conditions Using a Slurry Reactor. Water, Air, and Soil Pollution, 2009, 203, 305-315.	2.4	11
53	Application of electrochemically generated ozone to the discoloration and degradation of solutions containing the dye Reactive Orange 122. Journal of Hazardous Materials, 2009, 164, 10-17.	12.4	47
54	Electrochemical Ozone Production as an Environmentally Friendly Technology for Water Treatment. Clean - Soil, Air, Water, 2008, 36, 34-44.	1.1	23

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55	Electrochemical investigation of the passive behaviour of biomaterials based on Ag–Sn and Cu–Zn–Al in carbonate buffer in the absence and presence of chloride. Journal of Applied Electrochemistry, 2007, 37, 961-969.	2.9	6
56	Improvement of the electrochemical properties of "as-grown―boron-doped polycrystalline diamond electrodes deposited on tungsten wires using ethanol. Journal of Solid State Electrochemistry, 2007, 11, 1449-1457.	2.5	27
57	Surface, kinetics and electrocatalytic properties of the Ti/(Ti+Ru+Ce)O2-system for the oxygen evolution reaction in alkaline medium. Electrochimica Acta, 2006, 51, 2809-2818.	5.2	27
58	Characterisation of a laboratory electrochemical ozonation system and its application in advanced oxidation processes. Journal of Applied Electrochemistry, 2006, 36, 523-530.	2.9	33
59	Surface, kinetics and electrocatalytic properties of Ti/(IrO2 + Ta2O5) electrodes, prepared using controlled cooling rate, for ozone production. Electrochimica Acta, 2004, 49, 3977-3988.	5.2	94
60	Electrochemical impedance spectroscopy study during accelerated life test of conductive oxides: Ti/(Ru + Ti + Ce)O2-system. Electrochimica Acta, 2004, 49, 4893-4906.	5.2	45
61	Electrochemical ozone production: influence of the supporting electrolyte on kinetics and current efficiency. Electrochimica Acta, 2003, 48, 699-709.	5.2	146
62	Investigation of surface properties of Ru-based oxide electrodes containing Ti, Ce and Nb. Electrochimica Acta, 2003, 48, 1885-1891.	5.2	27
63	Green processes for environmental application. Electrochemical ozone production. Pure and Applied Chemistry, 2001, 73, 1871-1884.	1.9	73