

# Michael A Reynolds

## List of Publications by Year in descending order

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Version: 2024-02-01

22  
papers

567  
citations

759233

12  
h-index

752698

20  
g-index

22  
all docs

22  
docs citations

22  
times ranked

601  
citing authors

#	ARTICLE	IF	CITATIONS
1	A Simple and Rapid Method of Forming Double-Sided TiO <sub>2</sub> Nanotube Arrays. <i>ChemElectroChem</i> , 2022, 9, .	3.4	3
2	Quest CCS facility: Halite damage and injectivity remediation in CO <sub>2</sub> injection wells. <i>International Journal of Greenhouse Gas Control</i> , 2022, 119, 103718.	4.6	11
3	Lignin-Derived Non-Heme Iron and Manganese Complexes: Catalysts for the On-Demand Production of Chlorine Dioxide in Water under Mild Conditions. <i>Inorganic Chemistry</i> , 2021, 60, 2905-2913.	4.0	8
4	Suppressing Barium Sulfate Crystallization with Hydroxycitrate: A Dual Nucleation and Growth Inhibitor. <i>Chemistry of Materials</i> , 2021, 33, 6997-7007.	6.7	7
5	Alginate as a green inhibitor of barite nucleation and crystal growth. <i>Molecular Systems Design and Engineering</i> , 2021, 6, 508-519.	3.4	9
6	A Technical Playbook for Chemicals and Additives Used in the Hydraulic Fracturing of Shales. <i>Energy &amp; Fuels</i> , 2020, 34, 15106-15125.	5.1	32
7	Acidic Polysaccharides as Green Alternatives for Barite Scale Dissolution. <i>ACS Applied Materials &amp; Interfaces</i> , 2020, 12, 55434-55443.	8.0	11
8	Special Issue of <i>Energy &amp; Fuels</i> Honoring Michael T. Klein. <i>Energy &amp; Fuels</i> , 2020, 34, 15079-15081.	5.1	2
9	Room-Temperature Catalytic Treatment of High-Salinity Produced Water at Neutral pH. <i>Industrial &amp; Engineering Chemistry Research</i> , 2020, 59, 10356-10363.	3.7	2
10	Fit-for-purpose treatment goals for produced waters in shale oil and gas fields. <i>Water Research</i> , 2020, 173, 115467.	11.3	71
11	A microfluidic approach for probing hydrodynamic effects in barite scale formation. <i>Lab on A Chip</i> , 2019, 19, 1534-1544.	6.0	15
12	Highly-active nickel phosphide hydrotreating catalysts prepared in situ using nickel hypophosphite precursors. <i>Journal of Catalysis</i> , 2016, 335, 204-214.	6.2	56
13	Carbazole hydrodenitrogenation over nickel phosphide and Ni-rich bimetallic phosphide catalysts. <i>Applied Catalysis A: General</i> , 2014, 482, 221-230.	4.3	40
14	Ruthenium Derivatives of NiS <sub>2</sub> N <sub>2</sub> Complexes as Analogues of Bioorganometallic Reaction Centers. <i>Organometallics</i> , 2003, 22, 1619-1625.	2.3	48
15	Hydrogenation and Carbon-Sulfur Bond Hydrogenolysis of Benzothiophene Promoted by Re <sub>2</sub> (CO) <sub>10</sub> and H <sub>4</sub> Re <sub>4</sub> (CO) <sub>12</sub> . <i>Inorganic Chemistry</i> , 2003, 42, 2191-2193.	4.0	19
16	Re <sub>2</sub> (CO) <sub>10</sub> -Promoted S-Binding, C-S Bond Cleavage, and Hydrogenation of Benzothiophenes: Organometallic Models for the Hydrodesulfurization of Thiophenes. <i>Journal of the American Chemical Society</i> , 2002, 124, 1689-1697.	13.7	32
17	The Influence of Cyanide on the Carbonylation of Iron(II): Synthesis of Fe-SR-CN-CO Centers Related to the Hydrogenase Active Sites. <i>Journal of the American Chemical Society</i> , 2001, 123, 6933-6934.	13.7	83
18	Re <sub>2</sub> (CO) <sub>10</sub> -Mediated Carbon-Hydrogen and Carbon-Sulfur Bond Cleavage of Dibenzothiophene and 2,5-Dimethylthiophene. <i>Organometallics</i> , 2001, 20, 1071-1078.	2.3	55

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19	Organometallic models of catalytic hydrodesulfurization: Re <sub>2</sub> (CO) <sub>10</sub> -promoted cleavage of C–S bonds in benzothiophene. <i>Chemical Communications</i> , 2000, , 513-514.	4.1	20
20	Transition Metal Complexes of Chromium, Molybdenum, Tungsten, and Manganese Containing $\eta^1$ (S)-2,5-Dimethylthiophene, Benzothiophene, and Dibenzothiophene Ligands. <i>Organometallics</i> , 1999, 18, 4075-4081.	2.3	42
21	Minireview on the Evolution of Tetrathiometalate Salts as Protean Building Blocks of Catalysts and Materials for the Energy Transition: Recent Advances and Future Perspectives. <i>Energy &amp; Fuels</i> , 0, , .	5.1	0
22	Suppressing barite crystallization with organophosphorus compounds. <i>CrystEngComm</i> , 0, , .	2.6	1