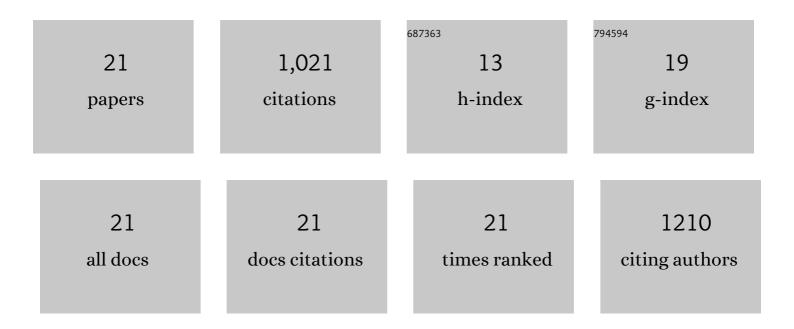
Bilal Acemioglu

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Removal of a reactive dye using NaOH-activated biochar prepared from peanut shell by pyrolysis process. International Journal of Coal Preparation and Utilization, 2022, 42, 671-693.	2.1	33
2	Use of peanut shell-based polyurethane type rigid foam in removing remazol orange RGB dye from aqueous solution. International Journal of Chemistry and Technology, 2020, 4, 79-89.	0.6	0
3	Adsorption of basic red 2 by activated biomass charcoal in batch and column systems. International Journal of Chemistry and Technology, 2019, 3, 136-145.	0.6	1
4	Adsorption of safranin-O dye by peanut shell-based polyurethane type foam. International Journal of Chemistry and Technology, 2018, 2, 95-104.	0.6	11
5	Kinetics, thermodynamics and isotherm studies of malachite green adsorption by modified orange peel. International Journal of Chemistry and Technology, 2017, 1, 58-66.	0.6	7
6	Competitive Removal of Malachite Green and Rhodamine B Using Clinoptilolite in a Two-dye System. Clays and Clay Minerals, 2016, 64, 299-313.	1.3	8
7	Adsorption of Basic green 4 from aqueous solution by olive pomace and commercial activated carbon: process design, isotherm, kinetic and thermodynamic studies. Desalination and Water Treatment, 2016, 57, 16653-16669.	1.0	24
8	Investigation of the adsorption kinetics of methylene blue onto cotton wastes. Turkish Journal of Chemistry, 2014, 38, 454-469.	1.2	6
9	Removal of Remazol Brilliant Blue R From Aqueous Solution by Pirina Pretreated with Nitric Acid and Commercial Activated Carbon. Water, Air, and Soil Pollution, 2014, 225, 1.	2.4	27
10	Sorption of remazol brilliant blue R onto polyurethaneâ€ŧype foam prepared from peanut shell. Journal of Applied Polymer Science, 2013, 127, 4340-4351.	2.6	26
11	Removal of methylene blue from aqueous solution using cotton stalk, cotton waste and cotton dust. Journal of Hazardous Materials, 2010, 183, 421-427.	12.4	110
12	Removal of Methylene Blue from Aqueous Solution Using Cotton Stalk: As a Bioadsorbent. , 2010, , 899-906.		1
13	Adsorption of Cr(III), Ni(II), Zn(II), Co(II) ions onto phenolated wood resin. Journal of Applied Polymer Science, 2006, 101, 2838-2846.	2.6	28
14	Adsorption of Congo red from aqueous solution onto calcium-rich fly ash. Journal of Colloid and Interface Science, 2004, 274, 371-379.	9.4	288
15	Removal of Fe(II) ions from aqueous solution by Calabrian pine bark wastes. Bioresource Technology, 2004, 93, 99-102.	9.6	47
16	Copper (II) adsorption from aqueous solution by herbaceous peat. Journal of Colloid and Interface Science, 2004, 269, 303-309.	9.4	202
17	A KINETIC STUDY OF SULFURIC ACID-CATALYZED LIQUEFACTION OF WOOD INTO PHENOL. Chemical Engineering Communications, 2004, 191, 968-980.	2.6	20
18	Kinetics of wood phenolysis in the presence of HCL as catalyst. Journal of Applied Polymer Science, 2002, 85, 1098-1103.	2.6	12

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#	Article	IF	CITATIONS
19	Solvent effect on nonradiative process of pyronin B in protic and aprotic solvent systems. Journal of Luminescence, 2002, 97, 153-160.	3.1	23
20	Equilibrium Studies on Adsorption of Cu(II) from Aqueous Solution onto Cellulose. Journal of Colloid and Interface Science, 2001, 243, 81-84.	9.4	79
21	Solvent effect on the ground and excited state dipole moments of fluorescein. Computational and Theoretical Chemistry, 2001, 548, 165-171.	1.5	68