

# Yasuo Matubara

## List of Publications by Year in descending order

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Version: 2024-02-01

20  
papers

752  
citations

758635

12  
h-index

794141

19  
g-index

21  
all docs

21  
docs citations

21  
times ranked

1179  
citing authors

#	ARTICLE	IF	CITATIONS
1	A Small yet Complete Framework for a Potentiostat, Galvanostat, and Electrochemical Impedance Spectrometer. <i>Journal of Chemical Education</i> , 2021, 98, 3362-3370.	1.1	11
2	A Small All-in-One Photon-Counting Device for Measuring Luminescence Decays to Determine the Lifetimes of Photoexcited Materials. <i>Journal of Chemical Education</i> , 2020, 97, 300-304.	1.1	3
3	A Bi-functional Second Coordination Sphere for Electrocatalytic CO <sub>2</sub> Reduction: The Concerted Improvement by a Local Proton Source and Local Coulombic Interactions. <i>Chemistry Letters</i> , 2020, 49, 315-317.	0.7	11
4	Standard Electrode Potentials for Electrochemical Hydrogen Production, Carbon Dioxide Reduction, and Oxygen Reduction Reactions in <i>N,N</i> -Dimethylacetamide. <i>Chemistry Letters</i> , 2020, 49, 915-917.	0.7	5
5	Unified Benchmarking of Electrocatalysts in Noninnocent Second Coordination Spheres for CO <sub>2</sub> Reduction. <i>ACS Energy Letters</i> , 2019, 4, 1999-2004.	8.8	29
6	Thermodynamic Cycles Relevant to Hydrogenation of CO <sub>2</sub> to Formic Acid in Water and Acetonitrile. <i>Chemistry Letters</i> , 2019, 48, 627-629.	0.7	9
7	Boundary Temperatures at Which Ionic Liquid Solutions Dissolving an Electroactive Ion Start to Exhibit a Colligative Behavior. <i>Chemistry Letters</i> , 2019, 48, 925-927.	0.7	0
8	Standard Electrode Potentials for the Reduction of CO <sub>2</sub> to CO in Acetonitrile/Water Mixtures Determined Using a Generalized Method for Proton-Coupled Electron-Transfer Reactions. <i>ACS Energy Letters</i> , 2017, 2, 1886-1891.	8.8	53
9	Experimental Insight into the Thermodynamics of the Dissolution of Electrolytes in Room-Temperature Ionic Liquids: From the Mass Action Law to the Absolute Standard Chemical Potential of a Proton. <i>ACS Omega</i> , 2016, 1, 1393-1411.	1.6	16
10	Striking Differences in Properties of Geometric Isomers of [Ir(tpy)(ppy)H] <sup>+</sup> : Experimental and Computational Studies of their Hydricities, Interaction with CO <sub>2</sub> , and Photochemistry. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 14128-14132.	7.2	51
11	Thermodynamic Aspects of Electrocatalytic CO <sub>2</sub> Reduction in Acetonitrile and with an Ionic Liquid as Solvent or Electrolyte. <i>ACS Catalysis</i> , 2015, 5, 6440-6452.	5.5	162
12	Hydride Reduction of NAD(P) <sup>+</sup> Model Compounds with a Ru(II) Hydrido Complex. <i>Organometallics</i> , 2015, 34, 5530-5539.	1.1	13
13	Reactivity of a fac-ReCl(±-diimine)(CO) <sub>3</sub> complex with an NAD <sup>+</sup> model ligand toward CO <sub>2</sub> reduction. <i>Chemical Communications</i> , 2014, 50, 728-730.	2.2	22
14	Electrocatalytic CO <sub>2</sub> Reduction with a Homogeneous Catalyst in Ionic Liquid: High Catalytic Activity at Low Overpotential. <i>Journal of Physical Chemistry Letters</i> , 2014, 5, 2033-2038.	2.1	108
15	Formation of $\pi$ -Coordinated Dihydropyridine Ruthenium(II) Complexes by Hydride Transfer from Ruthenium(II) to Pyridinium Cations. <i>Organometallics</i> , 2013, 32, 6162-6165.	1.1	11
16	Photochemistry of fac-[Re(bpy)(CO) <sub>3</sub> Cl]. <i>Chemistry - A European Journal</i> , 2012, 18, 15722-15734.	1.7	74
17	Thermodynamic and Kinetic Hydricity of Ruthenium(II) Hydride Complexes. <i>Journal of the American Chemical Society</i> , 2012, 134, 15743-15757.	6.6	117
18	Development of an Efficient and Durable Photocatalytic System for Hydride Reduction of an NAD(P) <sup>+</sup> Model Compound Using a Ruthenium(II) Complex Based on Mechanistic Studies. <i>Journal of the American Chemical Society</i> , 2010, 132, 10547-10552.	6.6	35

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19	Quantitative Photochemical Formation of [Ru(tpy)(bpy)H] <sup>+</sup> . Inorganic Chemistry, 2009, 48, 10138-10145.	1.9	12
20	Colloidal platinum nanoparticles dispersed by polyvinylpyrrolidone and poly(diallyldimethylammonium chloride) with high catalytic activity for hydrogen production based on formate decomposition. Sustainable Energy and Fuels, 0, , .	2.5	0