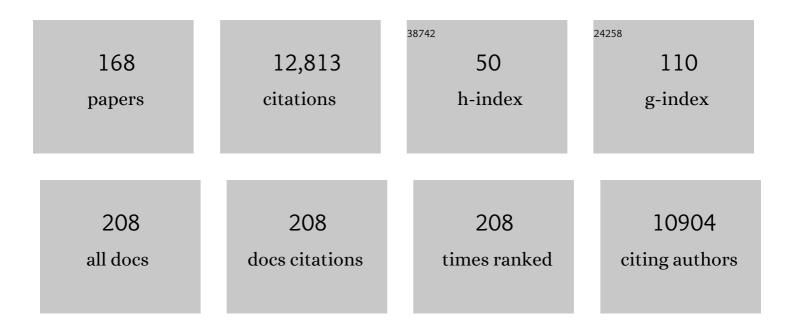
## Matthias Arenz

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	The gas diffusion electrode setup as a testing platform for evaluating fuel cell catalysts: A comparative RDEâ€GDE study. Electrochemical Science Advances, 2023, 3, .	2.8	6
2	Surfactant-free Ir nanoparticles synthesized in ethanol: Catalysts for the oxygen evolution reaction. Materials Letters, 2022, 308, 131209.	2.6	7
3	Benchmarking Fuel Cell Electrocatalysts Using Gas Diffusion Electrodes: Inter-lab Comparison and Best Practices. ACS Energy Letters, 2022, 7, 816-826.	17.4	58
4	Tracking the Catalyst Layer Depth-Dependent Electrochemical Degradation of a Bimodal Pt/C Fuel Cell Catalyst: A Combined <i>Operando</i> Small- and Wide-Angle X-ray Scattering Study. ACS Catalysis, 2022, 12, 2077-2085.	11.2	15
5	On the electro-oxidation of small organic molecules: Towards a fuel cell catalyst testing platform based on gas diffusion electrode setups. Journal of Power Sources, 2022, 522, 230979.	7.8	5
6	On the electrooxidation of glucose on gold: Towards an electrochemical glucaric acid production as value-added chemical. Electrochimica Acta, 2022, 410, 140023.	5.2	16
7	Electrochemical Reduction of CO <sub>2</sub> on Au Electrocatalysts in a Zeroâ€Gap, Halfâ€Cell Gas Diffusion Electrode Setup: a Systematic Performance Evaluation and Comparison to an Hâ€cell Setup**. ChemElectroChem, 2022, 9, .	3.4	17
8	(Digital Presentation) Small Angle X-Ray Scattering Studies Investigating the Degradation of Electrocatalysts. ECS Meeting Abstracts, 2022, MA2022-01, 1169-1169.	0.0	0
9	(Digital Presentation) Improved Nanocatalysts, Supported IrO <sub>x</sub> and IrRuO <sub>x</sub> , for Enhanced Oxygen Evolution Reaction. ECS Meeting Abstracts, 2022, MA2022-01, 2327-2327.	0.0	0
10	L-cysteine oxidation on Pt and Au rotating disk electrodes: Insights on mixed controlled kinetics. Journal of Electroanalytical Chemistry, 2021, 880, 114920.	3.8	3
11	Self-supported Pt–CoO networks combining high specific activity with high surface area for oxygen reduction. Nature Materials, 2021, 20, 208-213.	27.5	139
12	Commercial Spirits for Surfactant-Free Syntheses of Electro-Active Platinum Nanoparticles. Sustainable Chemistry, 2021, 2, 1-7.	4.7	8
13	The Gas Diffusion Electrode Setup as Straightforward Testing Device for Proton Exchange Membrane Water Electrolyzer Catalysts. Jacs Au, 2021, 1, 247-251.	7.9	50
14	Beyond Active Site Design: A Surfactantâ€Free Toolbox Approach for Optimized Supported Nanoparticle Catalysts. ChemCatChem, 2021, 13, 1692-1705.	3.7	23
15	Insights from <i>In Situ</i> Studies on the Early Stages of Platinum Nanoparticle Formation. Journal of Physical Chemistry Letters, 2021, 12, 3224-3231.	4.6	11
16	Operando SAXS study of a Pt/C fuel cell catalyst with an X-ray laboratory source. Journal Physics D: Applied Physics, 2021, 54, 294004.	2.8	6
17	Surfactant-free synthesis of size controlled platinum nanoparticles: Insights from in situ studies. Applied Surface Science, 2021, 549, 149263.	6.1	18
18	Elucidating Pt-Based Nanocomposite Catalysts for the Oxygen Reduction Reaction in Rotating Disk Electrode and Gas Diffusion Electrode Measurements. ACS Catalysis, 2021, 11, 7584-7594.	11.2	11

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19	The Oxygen Reduction Reaction on Pt: Why Particle Size and Interparticle Distance Matter. ACS Catalysis, 2021, 11, 7144-7153.	11.2	49
20	Anion Dependent Particle Size Control of Platinum Nanoparticles Synthesized in Ethylene Glycol. Nanomaterials, 2021, 11, 2092.	4.1	6
21	Bayesian Optimization of Highâ€Entropy Alloy Compositions for Electrocatalytic Oxygen Reduction**. Angewandte Chemie, 2021, 133, 24346-24354.	2.0	22
22	Surfactant-free colloidal strategies for highly dispersed and active supported IrO2 catalysts: Synthesis and performance evaluation for the oxygen evolution reaction. Journal of Catalysis, 2021, 401, 54-62.	6.2	14
23	Bayesian Optimization of Highâ€Entropy Alloy Compositions for Electrocatalytic Oxygen Reduction**. Angewandte Chemie - International Edition, 2021, 60, 24144-24152.	13.8	61
24	Bifunctional Pt-IrO <sub>2</sub> Catalysts for the Oxygen Evolution and Oxygen Reduction Reactions: Alloy Nanoparticles versus Nanocomposite Catalysts. ACS Catalysis, 2021, 11, 820-828.	11.2	50
25	Electrochemical Reduction of CO2 on Au Electrocatalysts Under Commercially Relevant Condition. ECS Meeting Abstracts, 2021, MA2021-02, 1727-1727.	0.0	Ο
26	Testing fuel cell catalysts under more realistic reaction conditions: accelerated stress tests in a gas diffusion electrode setup. JPhys Energy, 2020, 2, 024003.	5.3	29
27	Self-supported nanostructured iridium-based networks as highly active electrocatalysts for oxygen evolution in acidic media. Journal of Materials Chemistry A, 2020, 8, 1066-1071.	10.3	43
28	Structural Changes of Au(111) Singleâ€Crystal Electrode Surface in Ionic Liquids. ChemElectroChem, 2020, 7, 501-508.	3.4	8
29	On the facile and accurate determination of the Pt content in standard carbon supported Pt fuel cell catalysts. Analytica Chimica Acta, 2020, 1101, 41-49.	5.4	3
30	Synthesis of Iridium Nanocatalysts for Water Oxidation in Acid: Effect of the Surfactant. ChemCatChem, 2020, 12, 1282-1287.	3.7	31
31	Carbon-Supported Platinum Electrocatalysts Probed in a Gas Diffusion Setup with Alkaline Environment: How Particle Size and Mesoscopic Environment Influence the Degradation Mechanism. ACS Catalysis, 2020, 10, 13040-13049.	11.2	18
32	Environment Matters: CO <sub>2</sub> RR Electrocatalyst Performance Testing in a Gas-Fed Zero-Gap Electrolyzer. ACS Catalysis, 2020, 10, 13096-13108.	11.2	55
33	Teaching old precursors new tricks: Fast room temperature synthesis of surfactant-free colloidal platinum nanoparticles. Journal of Colloid and Interface Science, 2020, 577, 319-328.	9.4	20
34	Particle Size Effect on Platinum Dissolution: Practical Considerations for Fuel Cells. ACS Applied Materials & Interfaces, 2020, 12, 25718-25727.	8.0	48
35	Solventâ€Dependent Growth and Stabilization Mechanisms of Surfactantâ€Free Colloidal Pt Nanoparticles. Chemistry - A European Journal, 2020, 26, 9012-9023.	3.3	26
36	UV-induced syntheses of surfactant-free precious metal nanoparticles in alkaline methanol and ethanol. Nanoscale Advances, 2020, 2, 2288-2292.	4.6	15

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37	Influence of Anion Chaotropicity on the SO <sub>2</sub> Oxidation Reaction: When Spectator Species Determine the Reaction Pathway. ChemElectroChem, 2020, 7, 1843-1850.	3.4	8
38	Comparative study of cytotoxicity by platinum nanoparticles and ions in vitro systems based on fish cell lines. Toxicology in Vitro, 2020, 66, 104859.	2.4	9
39	(Invited) The Toolbox Concept for the Synthesis of Surfactant-Free Colloidal Nanoparticles as Electrocatalysts. ECS Transactions, 2020, 97, 443-455.	0.5	4
40	A New Approach to Probe the Degradation of Fuel Cell Catalysts under Realistic Conditions: Combining Tests in a Gas Diffusion Electrode Setup with Small Angle X-ray Scattering. Journal of the Electrochemical Society, 2020, 167, 134515.	2.9	29
41	The Dissolution Dilemma for Low Pt Loading Polymer Electrolyte Membrane Fuel Cell Catalysts. Journal of the Electrochemical Society, 2020, 167, 164501.	2.9	32
42	(Invited) The Tool-Box Concept for the Synthesis of Surfactant-Free Colloidal Nanoparticles As Electrocatalysts. ECS Meeting Abstracts, 2020, MA2020-01, 1140-1140.	0.0	0
43	Opportunities and Knowledge Gaps of SO <sub>2</sub> Electrocatalytic Oxidation for H <sub>2</sub> Electrochemical Generation. ACS Catalysis, 2019, 9, 8136-8143.	11.2	22
44	Monovalent Alkali Cations: Simple and Eco-Friendly Stabilizers for Surfactant-Free Precious Metal Nanoparticle Colloids. ACS Sustainable Chemistry and Engineering, 2019, 7, 13680-13686.	6.7	29
45	Oxygen Reduction Reaction on Polycrystalline Platinum: On the Activity Enhancing Effect of Polyvinylidene Difluoride. Surfaces, 2019, 2, 69-77.	2.3	3
46	Controlled Synthesis of Surfactantâ€Free Waterâ€Dispersible Colloidal Platinum Nanoparticles by the Co4Cat Process. ChemSusChem, 2019, 12, 1229-1239.	6.8	27
47	Accelerated Durability Test for Highâ€Surfaceâ€Area Oxyhydroxide Nickel Supported on Raney Nickel as Catalyst for the Alkaline Oxygen Evolution Reaction. ChemPhysChem, 2019, 20, 3147-3153.	2.1	11
48	Electrochemical Synthesis of High-Value Chemicals: Detection of Key Reaction Intermediates and Products Combining Gas Chromatography–Mass Spectrometry and <i>in Situ</i> Infrared Spectroscopy. Journal of Physical Chemistry C, 2019, 123, 12762-12772.	3.1	3
49	Sputtered Platinum Thin-films for Oxygen Reduction in Gas Diffusion Electrodes: A Model System for Studies under Realistic Reaction Conditions. Surfaces, 2019, 2, 336-348.	2.3	27
50	Examining the Structure Sensitivity of the Oxygen Evolution Reaction on Pt Singleâ€Crystal Electrodes: A Combined Experimental and Theoretical Study. ChemPhysChem, 2019, 20, 3154-3162.	2.1	11
51	Enhanced Oxygen Reduction Reaction on Fe/N/C Catalyst in Acetate Buffer Electrolyte. ACS Catalysis, 2019, 9, 3082-3089.	11.2	32
52	Testing a Silver Nanowire Catalyst for the Selective COâ,, Reduction in a Gas Diffusion Electrode Half-cell Setup Enabling High Mass Transport Conditions. Chimia, 2019, 73, 922.	0.6	18
53	Catalyst design criteria and fundamental limitations in the electrochemical synthesis of dimethyl carbonate. Green Chemistry, 2019, 21, 6200-6209.	9.0	6
54	Catalyst Development for Water/CO2 Co-electrolysis. Chimia, 2019, 73, 707.	0.6	5

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55	Ir nanoparticles with ultrahigh dispersion as oxygen evolution reaction (OER) catalysts: synthesis and activity benchmarking. Catalysis Science and Technology, 2019, 9, 6345-6356.	4.1	61
56	Mechanism of Electrochemical L ysteine Oxidation on Pt. ChemElectroChem, 2019, 6, 1009-1013.	3.4	7
57	Support-free nanostructured Pt Cu electrocatalyst for the oxygen reduction reaction prepared by alternating magnetron sputtering. Journal of Power Sources, 2019, 413, 432-440.	7.8	12
58	Electrochemically Generated Copper Carbonyl for Selective Dimethyl Carbonate Synthesis. ACS Catalysis, 2019, 9, 859-866.	11.2	15
59	Halideâ€Induced Leaching of Pt Nanoparticles – Manipulation of Particle Size by Controlled Ostwald Ripening. ChemNanoMat, 2019, 5, 462-471.	2.8	17
60	PEMFC Catalyst Testing: From RDE to GDE Setup. ECS Meeting Abstracts, 2019, , .	0.0	0
61	The Interparticle Distance Effect on Transient Platinum Dissolution: Degradation at High and Low Loadings. ECS Meeting Abstracts, 2019, , .	0.0	0
62	Benchmarking high surface area electrocatalysts in a gas diffusion electrode: measurement of oxygen reduction activities under realistic conditions. Energy and Environmental Science, 2018, 11, 988-994.	30.8	147
63	Influence of the Electrode and Chaotropicity of the Electrolyte on the Oscillatory Behavior of the Electrocatalytic Oxidation of SO <sub>2</sub> . Journal of Physical Chemistry C, 2018, 122, 1243-1247.	3.1	11
64	Electrochemical stability of subnanometer Pt clusters. Electrochimica Acta, 2018, 277, 211-217.	5.2	18
65	On the Preparation and Testing of Fuel Cell Catalysts Using the Thin Film Rotating Disk Electrode Method. Journal of Visualized Experiments, 2018, , .	0.3	20
66	Size effect studies in catalysis: a simple surfactant-free synthesis of sub 3Ânm Pd nanocatalysts supported on carbon. RSC Advances, 2018, 8, 33794-33797.	3.6	7
67	Spatially Localized Synthesis and Structural Characterization of Platinum Nanocrystals Obtained Using UV Light. ACS Omega, 2018, 3, 10351-10356.	3.5	13
68	[Mo 3 S 13 ] 2â^' Cluster Decorated Sulfurâ€doped Reduced Graphene Oxide as Noble Metalâ€Free Catalyst for Hydrogen Evolution Reaction in Polymer Electrolyte Membrane Electrolyzers. ChemElectroChem, 2018, 5, 2672-2680.	3.4	15
69	Colloids for Catalysts: A Concept for the Preparation of Superior Catalysts of Industrial Relevance. Angewandte Chemie, 2018, 130, 12518-12521.	2.0	12
70	Colloids for Catalysts: A Concept for the Preparation of Superior Catalysts of Industrial Relevance. Angewandte Chemie - International Edition, 2018, 57, 12338-12341.	13.8	53
71	Kinetics, Assembling, and Conformation Control of L ysteine Adsorption on Pt Investigated by in situ FTIR Spectroscopy and QCMâ€Đ. ChemPhysChem, 2018, 19, 2340-2348.	2.1	13
72	Investigating Particle Size Effects in Catalysis by Applying a Size-Controlled and Surfactant-Free Synthesis of Colloidal Nanoparticles in Alkaline Ethylene Glycol: Case Study of the Oxygen Reduction Reaction on Pt. ACS Catalysis, 2018, 8, 6627-6635.	11.2	119

<ul> <li>NanoElectrocatalysis: From Basic Research to Applications in Energy Conversion. Chimia, 2018, 72 276-285.</li> <li>Enhanced Oxygen Reduction Activity by Selective Anion Adsorption on Non-Precious-Metal Catalys ACS Catalysis, 2018, 8, 7104-7112.</li> <li>Particle Size Effect Vs. Particle Proximity Effect: Systematic Study on ORR Activity of High Surface Area Pt/C Catalysts for Polymer Electrolyte Membrane Fuel Cells. ECS Meeting Abstracts, 2018,</li> <li>Nanoparticles in a box: a concept to isolate, store and re-use colloidal surfactant-free precious met nanoparticles. Journal of Materials Chemistry A, 2017, 5, 6140-6145.</li> <li>UV&amp;&amp; Huduced Synthesis and Stabilization of Surfactant&amp; Free Colloidal Pt Nanoparticles with Cont Particle Size in Ethylene Clycol. ChemNanoMat, 2017, 3, 89-93.</li> <li>pH matters: The influence of the catalyst ink on the oxygen reduction activity determined in thin firotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li>Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; Amp; Interfaces, 2017, 9, 38176-38180.</li> <li>Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	0.6	
<ul> <li>ACS Catalysis, 2018, 8, 7104-7112.</li> <li>Particle Size Effect Vs. Particle Proximity Effect: Systematic Study on ORR Activity of High Surface Area Pt/C Catalysts for Polymer Electrolyte Membrane Fuel Cells. ECS Meeting Abstracts, 2018, , .</li> <li>Nanoparticles in a box: a concept to isolate, store and re-use colloidal surfactant-free precious met nanoparticles. Journal of Materials Chemistry A, 2017, 5, 6140-6145.</li> <li>UVâ€hduced Synthesis and Stabilization of Surfactantâ€Free Colloidal Pt Nanoparticles with Cont Particle Size in Ethylene Clycol. ChemNanoMat, 2017, 3, 89-93.</li> <li>pH matters: The influence of the catalyst ink on the oxygen reduction activity determined in thin fi rotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li>Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li>Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>		2
<ul> <li>Area Pt/C Catalysts for Polymer Electrolyte Membrane Fuel Cells. ECS Meeting Abstracts, 2018, , .</li> <li>Nanoparticles in a box: a concept to isolate, store and re-use colloidal surfactant-free precious met nanoparticles. Journal of Materials Chemistry A, 2017, 5, 6140-6145.</li> <li>UVâ€hduced Synthesis and Stabilization of Surfactantâ€Free Colloidal Pt Nanoparticles with Cont Particle Size in Ethylene Glycol. ChemNanoMat, 2017, 3, 89-93.</li> <li>pH matters: The influence of the catalyst ink on the oxygen reduction activity determined in thin firotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li>Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li>Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	ts. 11.2	53
<ul> <li><sup>76</sup> nanoparticles. Journal of Materials Chemistry A, 2017, 5, 6140-6145.</li> <li><sup>77</sup> UVâ€Induced Synthesis and Stabilization of Surfactantâ€Free Colloidal Pt Nanoparticles with Cont Particle Size in Ethylene Glycol. ChemNanoMat, 2017, 3, 89-93.</li> <li><sup>78</sup> pH matters: The influence of the catalyst ink on the oxygen reduction activity determined in thin fi rotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li><sup>79</sup> Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li><sup>80</sup> Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	0.0	0
<ul> <li>Particle Size in Éthylene Glycol. ChemNanoMat, 2017, 3, 89-93.</li> <li>pH matters: The influence of the catalyst ink on the oxygen reduction activity determined in thin firotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li>Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li>Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	al 10.3	37
<ul> <li><sup>78</sup> rotating disk electrode measurements. Journal of Power Sources, 2017, 353, 19-27.</li> <li><sup>79</sup> Accessing the Inaccessible: Analyzing the Oxygen Reduction Reaction in the Diffusion Limit. ACS Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li><sup>80</sup> Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	rolled 2.8	30
<ul> <li>Applied Materials &amp; amp; Interfaces, 2017, 9, 38176-38180.</li> <li>Design and Application of External Reference Electrode for Kinetic Studies at Elevated Temperature and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.</li> </ul>	m 7.8	51
and Pressures. Journal of the Electrochemical Society, 2017, 164, F821-F824.	8.0	21
	S 2.9	1
<ul> <li>Single Graphene Layer on Pt(111) Creates Confined Electrochemical Environment via Selective Ion</li> <li>Transport. Angewandte Chemie, 2017, 129, 13063-13067.</li> </ul>	2.0	1
<ul> <li>Single Graphene Layer on Pt(111) Creates Confined Electrochemical Environment via Selective Ion</li> <li>Transport. Angewandte Chemie - International Edition, 2017, 56, 12883-12887.</li> </ul>	13.8	32
<ul> <li>Tetrahexahedral Pt Nanoparticles: Comparing the Oxygen Reduction Reaction under Transient vs</li> <li>Steady-State Conditions. ACS Catalysis, 2017, 7, 1-6.</li> </ul>	11.2	25
84 Synthesis Mechanism and Influence of Light on Unprotected Platinum Nanoparticles Synthesis at Room Temperature. ChemNanoMat, 2016, 2, 104-107.	2.8	15
<sup>85</sup> Inâ€Situ FTIR Spectroscopy: Probing the Electrochemical Interface during the Oxygen Reduction Reaction on a Commercial Platinum Highâ€Surfaceâ€Area Catalyst. ChemCatChem, 2016, 8, 1125	-1131. <sup>3.7</sup>	19
86 Evaluation of temperature and electrolyte concentration dependent Oxygen solubility and diffusivi in phosphoric acid. Electrochimica Acta, 2016, 209, 399-406.	ty 5.2	18
87 Fuel cell catalyst degradation: Identical location electron microscopy and related methods. Nano Energy, 2016, 29, 299-313.	16.0	62
88 On the oxygen reduction reaction in phosphoric acid electrolyte: Evidence of significantly increased inhibition at steady state conditions. Electrochimica Acta, 2016, 204, 78-83.	5.2	21
89 The colloidal tool-box approach for fuel cell catalysts: utilizing graphitized carbon supports. Electrochimica Acta, 2016, 197, 221-227.	5.2	3
The colloidal tool-box approach for fuel cell catalysts: Systematic study of perfluorosulfonate-ionomer impregnation and Pt loading. Catalysis Today, 2016, 262, 82-89.	4.4	23

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91	Oxygen Reduction Reaction on Pt Overlayers Deposited onto a Gold Film: Ligand, Strain, and Ensemble Effect. ACS Catalysis, 2016, 6, 671-676.	11.2	79
92	On the structural composition and stability of Fe–N–C catalysts prepared by an intermediate acid leaching. Journal of Solid State Electrochemistry, 2016, 20, 969-981.	2.5	39
93	On the Need of Improved Accelerated Degradation Protocols (ADPs): Examination of Platinum Dissolution and Carbon Corrosion in Half-Cell Tests. Journal of the Electrochemical Society, 2016, 163, F1510-F1514.	2.9	112
94	Cold nanoparticles assembled with dithiocarbamate-anchored molecular wires. Scientific Reports, 2015, 5, 15273.	3.3	11
95	Equilibrium coverage of OH ad in correlation with platinum catalyzed fuel cell reactions in HClO 4. Electrochemistry Communications, 2015, 53, 41-44.	4.7	21
96	On the influence of hydronium and hydroxide ion diffusion on the hydrogen and oxygen evolution reactions in aqueous media. Electrochimica Acta, 2015, 158, 13-17.	5.2	16
97	1-Naphthylamine functionalized Pt nanoparticles: electrochemical activity and redox chemistry occurring on one surface. New Journal of Chemistry, 2015, 39, 2557-2564.	2.8	12
98	On the influence of hydronium and hydroxide ion diffusion on the hydrogen and oxygen evolution reactions in aqueous media. Electrochimica Acta, 2015, 159, 66-70.	5.2	12
99	Electrochemical Stability and Postmortem Studies of Pt/SiC Catalysts for Polymer Electrolyte Membrane Fuel Cells. ACS Applied Materials & Interfaces, 2015, 7, 6153-6161.	8.0	25
100	Surface Chemistry of "Unprotected―Nanoparticles: A Spectroscopic Investigation on Colloidal Particles. Journal of Physical Chemistry C, 2015, 119, 17655-17661.	3.1	64
101	Gas diffusion electrode setup for catalyst testing in concentrated phosphoric acid at elevated temperatures. Review of Scientific Instruments, 2015, 86, 024102.	1.3	33
102	Structural disordering of de-alloyed Pt bimetallic nanocatalysts: the effect on oxygen reduction reaction activity and stability. Physical Chemistry Chemical Physics, 2015, 17, 28044-28053.	2.8	14
103	From single crystal model catalysts to systematic studies of supported nanoparticles. Surface Science, 2015, 631, 278-284.	1.9	23
104	Design and test of a flexible electrochemical setup for measurements in aqueous electrolyte solutions at elevated temperature and pressure. Review of Scientific Instruments, 2014, 85, 085105.	1.3	19
105	Investigating the activity enhancement on PtxCo1â^'x alloys induced by a combined strain and ligand effect. Journal of Power Sources, 2014, 245, 908-914.	7.8	27
106	Stabilizing Catalytically Active Nanoparticles by Ligand Linking: Toward Three-Dimensional Networks with High Catalytic Surface Area. Langmuir, 2014, 30, 5564-5573.	3.5	25
107	PtxCo1â^'x alloy NPs prepared by colloidal tool-box synthesis: The effect of de-alloying on the oxygen reduction reaction activity. International Journal of Hydrogen Energy, 2014, 39, 9143-9148.	7.1	7
108	Core-shell TiO2@C: towards alternative supports as replacement for high surface area carbon for PEMFC catalysts. Electrochimica Acta, 2014, 139, 21-28.	5.2	39

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109	Comparative degradation study of carbon supported proton exchange membrane fuel cell electrocatalysts – The influence of the platinum to carbon ratio on the degradation rate. Journal of Power Sources, 2014, 261, 14-22.	7.8	163
110	The particle proximity effect: from model to high surface area fuel cell catalysts. RSC Advances, 2014, 4, 14971.	3.6	70
111	Activity inhibition and its mitigation in high temperature proton exchange membrane fuel cells: The role of phosphoric acid, ammonium trifluoromethanesulfonate, and polyvinylidene difluoride. Journal of Power Sources, 2014, 272, 1072-1077.	7.8	15
112	The effect of particle proximity on the oxygen reduction rate of size-selected platinum clusters. Nature Materials, 2013, 12, 919-924.	27.5	327
113	Investigating the corrosion of high surface area carbons during start/stop fuel cell conditions: A Raman study. Electrochimica Acta, 2013, 114, 455-461.	5.2	65
114	On the influence of the Pt to carbon ratio on the degradation of high surface area carbon supported PEM fuel cell electrocatalysts. Electrochemistry Communications, 2013, 34, 153-156.	4.7	57
115	Pt based PEMFC catalysts prepared from colloidal particle suspensions – a toolbox for model studies. Physical Chemistry Chemical Physics, 2013, 15, 3602.	2.8	64
116	Probing Degradation by IL-TEM: The Influence of Stress Test Conditions on the Degradation Mechanism. Journal of the Electrochemical Society, 2013, 160, F608-F615.	2.9	96
117	Design, development, and demonstration of a fully LabVIEW controlled <i>in situ</i> electrochemical Fourier transform infrared setup combined with a wall-jet electrode to investigate the electrochemical interface of nanoparticulate electrocatalysts under reaction conditions. Review of Scientific Instruments. 2013. 84. 074103.	1.3	10
118	Comparative IL-TEM Study Concerning the Degradation of Carbon Supported Pt-Based Electrocatalysts. Journal of the Electrochemical Society, 2012, 159, B677-B682.	2.9	46
119	Comparative DEMS study on the electrochemical oxidation of carbon blacks. Journal of Power Sources, 2012, 217, 392-399.	7.8	50
120	Establishing the potential dependent equilibrium oxide coverage on platinum in alkaline solution and its influence on the oxygen reduction. Journal of Power Sources, 2012, 217, 262-267.	7.8	20
121	Identical-location TEM investigations of Pt/C electrocatalyst degradation at elevated temperatures. Journal of Electroanalytical Chemistry, 2011, 662, 355-360.	3.8	98
122	The Particle Size Effect on the Oxygen Reduction Reaction Activity of Pt Catalysts: Influence of Electrolyte and Relation to Single Crystal Models. Journal of the American Chemical Society, 2011, 133, 17428-17433.	13.7	461
123	IL-TEM investigations on the degradation mechanism of Pt/C electrocatalysts with different carbon supports. Energy and Environmental Science, 2011, 4, 234-238.	30.8	124
124	Comparative Study of the Degradation of Different Carbon Supported Pt Fuel Cell Catalysts. ECS Transactions, 2011, 41, 811-816.	0.5	2
125	A DEMS study on the electrochemical oxidation of a high surface area carbon black. Electrochemistry Communications, 2011, 13, 1473-1473.	4.7	48
126	Investigation of the Oxygen Reduction Activity on Silver – A Rotating Disc Electrode Study. Fuel Cells, 2010. 10. 575-581.	2.4	99

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127	Electrochemically induced nanocluster migration. Electrochimica Acta, 2010, 56, 810-816.	5.2	59
128	AuPt core–shell nanocatalysts with bulk Pt activity. Electrochemistry Communications, 2010, 12, 1487-1489.	4.7	50
129	Temperature Dependent CO Oxidation Mechanisms on Size-Selected Clusters. Journal of Physical Chemistry C, 2010, 114, 1651-1654.	3.1	76
130	Size-selected clusters as heterogeneous model catalysts under applied reaction conditions. Physical Chemistry Chemical Physics, 2010, 12, 10288.	2.8	81
131	Influence of the Electrolyte on the Particle Size Effect of the Oxygen Reduction Reaction on Pt Nanoparticles. ECS Transactions, 2009, 25, 455-462.	0.5	4
132	Stability of Pt Alloy High Surface Area Catalysts. ECS Transactions, 2009, 25, 555-563.	0.5	1
133	Adsorbateâ€Induced Surface Segregation for Core–Shell Nanocatalysts. Angewandte Chemie - International Edition, 2009, 48, 3529-3531.	13.8	295
134	Log on for new catalysts. Nature Chemistry, 2009, 1, 518-519.	13.6	62
135	The influence of electrochemical annealing in CO saturated solution on the catalytic activity of Pt nanoparticles. Electrochimica Acta, 2009, 54, 5018-5022.	5.2	27
136	Degradation of Carbon-Supported Pt Bimetallic Nanoparticles by Surface Segregation. Journal of the American Chemical Society, 2009, 131, 16348-16349.	13.7	182
137	Non-destructive transmission electron microscopy study of catalyst degradation under electrochemical treatment. Journal of Power Sources, 2008, 185, 734-739.	7.8	150
138	Fuel cell catalyst degradation on the nanoscale. Electrochemistry Communications, 2008, 10, 1144-1147.	4.7	309
139	Measurement of oxygen reduction activities via the rotating disc electrode method: From Pt model surfaces to carbon-supported high surface area catalysts. Electrochimica Acta, 2008, 53, 3181-3188.	5.2	888
140	Dual pulsed-beam controlled mole fraction studies of the catalytic oxidation of CO on supported Pd nanocatalysts. Journal of Catalysis, 2008, 255, 234-240.	6.2	27
141	Analysis of the Impact of Individual Glass Constituents on Electrocatalysis on Pt Electrodes in Alkaline Solution. Journal of the Electrochemical Society, 2008, 155, P78.	2.9	63
142	Micromechanical sensor for studying heats of surface reactions, adsorption, and cluster deposition processes. Review of Scientific Instruments, 2007, 78, 054101.	1.3	12
143	Trends in electrocatalysis on extended and nanoscale Pt-bimetallic alloy surfaces. Nature Materials, 2007, 6, 241-247.	27.5	2,902
144	Cluster Chemistry:  Size-Dependent Reactivity Induced by Reverse Spill-Over. Journal of the American Chemical Society, 2007, 129, 9635-9639.	13.7	52

#	Article	IF	CITATIONS
145	Chapter 1 Size effects in the chemistry of small clusters. Chemical Physics of Solid Surfaces, 2007, , 1-51.	0.3	15
146	Factors in gold nanocatalysis: oxidation of CO in the non-scalable size regime. Topics in Catalysis, 2007, 44, 145-158.	2.8	190
147	The polymerization of acetylene on supported metal clusters. Low Temperature Physics, 2006, 32, 1097-1103.	0.6	10
148	Monodispersed cluster-assembled materials. Materials Today, 2006, 9, 48-49.	14.2	9
149	CO Combustion on Supported Gold Clusters. ChemPhysChem, 2006, 7, 1871-1879.	2.1	121
150	CO surface electrochemistry on Pt-nanoparticles: A selective review. Electrochimica Acta, 2005, 50, 5144-5154.	5.2	154
151	The Effect of the Particle Size on the Kinetics of CO Electrooxidation on High Surface Area Pt Catalysts. Journal of the American Chemical Society, 2005, 127, 6819-6829.	13.7	514
152	Carbon-supported Pt–Sn electrocatalysts for the anodic oxidation of H2, CO, and H2/CO mixtures.Part II: The structure–activity relationship. Journal of Catalysis, 2005, 232, 402-410.	6.2	156
153	In situ CO oxidation on well characterized Pt3Sn(hkl) surfaces: A selective review. Surface Science, 2005, 576, 145-157.	1.9	71
154	The Impact of Geometric and Surface Electronic Properties of Pt-Catalysts on the Particle Size Effect in Electrocatalysis. Journal of Physical Chemistry B, 2005, 109, 14433-14440.	2.6	613
155	Surface (electro-)chemistry on Pt(111) modified by a Pseudomorphic Pd monolayer. Surface Science, 2004, 573, 57-66.	1.9	56
156	Temperature-Induced Deposition Method for Anchoring Metallic Nanoparticles onto Reflective Substrates for in Situ Electrochemical Infrared Spectroscopy. Journal of Physical Chemistry B, 2004, 108, 17915-17920.	2.6	16
157	Surface Electrochemistry of CO on Reconstructed Gold Single Crystal Surfaces Studied by Infrared Reflection Absorption Spectroscopy and Rotating Disk Electrode. Journal of the American Chemical Society, 2004, 126, 10130-10141.	13.7	93
158	Anion Adsorption, CO Oxidation, and Oxygen Reduction Reaction on a Au(100) Surface:Â The pH Effect. Journal of Physical Chemistry B, 2004, 108, 625-634.	2.6	143
159	Preferential oxidation of carbon monoxide adsorbed on Pd submonolayer films deposited on Pt(100). Electrochemistry Communications, 2003, 5, 809-813.	4.7	24
160	The effect of specific chloride adsorption on the electrochemical behavior of ultrathin Pd films deposited on Pt() in acid solution. Surface Science, 2003, 523, 199-209.	1.9	70
161	The Oxygen Reduction Reaction on Thin Palladium Films Supported on a Pt(111) Electrode. Journal of Physical Chemistry B, 2003, 107, 9813-9819.	2.6	70
162	The electro-oxidation of formic acid on Pt–Pd single crystal bimetallic surfaces. Physical Chemistry Chemical Physics, 2003, 5, 4242-4251.	2.8	203

#	Article	IF	CITATIONS
163	Surface Chemistry on Bimetallic Alloy Surfaces:Â Adsorption of Anions and Oxidation of CO on Pt3Sn(111). Journal of the American Chemical Society, 2003, 125, 2736-2745.	13.7	127
164	CO adsorption and kinetics on well-characterized Pd films on Pt() in alkaline solutions. Surface Science, 2002, 506, 287-296.	1.9	71
165	Oxygen electrocatalysis in alkaline electrolyte: Pt(hkl), Au(hkl) and the effect of Pd-modification. Electrochimica Acta, 2002, 47, 3765-3776.	5.2	209
166	In-situ Characterization of Metal/Electrolyte Interfaces: Sulfate Adsorption on Cu(111). Physica Status Solidi A, 2001, 187, 63-74.	1.7	24
167	Cyanide adlayers on Pt(111) in chloride containing electrolytes studied by in situ, ex situ IRAS and LEED. Surface Science, 2000, 461, 98-106.	1.9	19
168	Sulfate adsorption on Cu(111) studied by in-situ IRRAS and STM: revealing the adsorption site and desorption behavior. Surface Science, 1999, 442, 215-222.	1.9	49