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List of Publications by Year
in descending order

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58 papers	810 citations	471509 17 h-index	610901 24 g-index
59 all docs	59 docs citations	59 times ranked	861 citing authors

#	Article	IF	Citations
1	Glycation of Î±-synuclein hampers its binding to synaptic-like vesicles and its driving effect on their fusion. Cellular and Molecular Life Sciences, 2022, 79, .	5.4	9
2	Cu2+, Ca2+, and methionine oxidation expose the hydrophobic Î±-synuclein NAC domain. International Journal of Biological Macromolecules, 2021, 169, 251-263.	7.5	8
3	Frataxins Emerge as New Players of the Intracellular Antioxidant Machinery. Antioxidants, 2021, 10, 315.	5.1	5
4	Understanding metal binding in neuromedin C. Inorganica Chimica Acta, 2020, 499, 119197.	2.4	2
5	Unraveling the NaCl Concentration Effect on the First Stages of Î±-Synuclein Aggregation. Biomacromolecules, 2020, 21, 5200-5212.	5.4	8
6	Unravelling the effect ofN(Îµ)-(carboxyethyl)lysine on the conformation, dynamics and aggregation propensity of Î±-synuclein. Chemical Science, 2020, 11, 3332-3344.	7.4	13
7	How Does Pyridoxamine Inhibit the Formation of Advanced Glycation End Products? The Role of Its Primary Antioxidant Activity. Antioxidants, 2019, 8, 344.	5.1	25
8	Nitration and Glycation Diminish the Î±-Synuclein Role in the Formation and Scavenging of Cu²⁺-Catalyzed Reactive Oxygen Species. ACS Chemical Neuroscience, 2019, 10, 2919-2930.	3.5	15
9	A Coarse-Grained Molecular Dynamics Approach to the Study of the Intrinsically Disordered Protein Î±-Synuclein. Journal of Chemical Information and Modeling, 2019, 59, 1458-1471.	5.4	44
10	Does glycation really distort the peptide Î±-helicity?. International Journal of Biological Macromolecules, 2019, 129, 254-266.	7.5	5
11	A density functional theory study of the freeâ€radical scavenging activity of aminoguanidine. Comparison with its reactive carbonyl compound and metal scavenging activities. International Journal of Quantum Chemistry, 2019, 119, e25911.	2.0	5
12	A Systematic DFT Study of Some Plausible Zn(II) and Al(III) Interaction Sites in N-Terminally Acetylated Î±-Synuclein. Journal of Physical Chemistry A, 2018, 122, 690-699.	2.5	9
13	Formation mechanism of glyoxal-DNA adduct, a DNA cross-link precursor. International Journal of Biological Macromolecules, 2017, 98, 664-675.	7.5	8
14	Glycation of Lysozyme by Glycolaldehyde Provides New Mechanistic Insights in Diabetes-Related Protein Aggregation. ACS Chemical Biology, 2017, 12, 1152-1162.	3.4	16
15	New insights into human farnesyl pyrophosphate synthase inhibition by second-generation bisphosphonate drugs. Journal of Computer-Aided Molecular Design, 2017, 31, 675-688.	2.9	3
16	Copper(II) Binding Sites in N-Terminally Acetylated Î±-Synuclein: A Theoretical Rationalization. Journal of Physical Chemistry A, 2017, 121, 5711-5719.	2.5	14
17	Ortho-methylated 3-hydroxypyridines hinder hen egg-white lysozyme fibrillogenesis. Scientific Reports, 2015, 5, 12052.	3.3	18
18	Conformational ensembles of neuromedin C reveal a progressive coil-helix transition within a binding-induced folding mechanism. RSC Advances, 2015, 5, 83074-83088.	3.6	6

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19	Trapping a salt-dependent unfolding intermediate of the marginally stable protein Yfh1. Frontiers in Molecular Biosciences, 2014, 1, 13.	3.5	9
20	Mechanistic Insights in Glycation-Induced Protein Aggregation. Biomacromolecules, 2014, 15, 3449-3462.	5.4	51
21	The hydrophobic substituent in aminophospholipids affects the formation kinetics of their Schiff bases. Bioorganic and Medicinal Chemistry Letters, 2013, 23, 2202-2206.	2.2	0
22	Formation of Schiff Bases of O-Phosphorylethanolamine and O-Phospho-<scp>d</scp>,<scp>l</scp>-serine with Pyridoxal 5â€²-Phosphate. Experimental and Theoretical Studies. Journal of Physical Chemistry A, 2012, 116, 1897-1905.	2.5	19
23	Towards a detailed description of pyridoxamine tautomeric species. New Journal of Chemistry, 2012, 36, 1751.	2.8	5
24	Phenol Group in Pyridoxamine Acts as a Stabilizing Element for Its Carbinolamines and Schiff Bases. Chemistry and Biodiversity, 2011, 8, 1318-1332.	2.1	12
25	Understanding non-enzymatic aminophospholipid glycation and its inhibition. Polar head features affect the kinetics of Schiff base formation. Bioorganic and Medicinal Chemistry, 2011, 19, 4536-4543.	3.0	21
26	Theoretical and experimental study of the vertical excitation energies in the ionic and tautomeric forms of 4-aminomethylpyridine. Journal of Photochemistry and Photobiology A: Chemistry, 2010, 209, 19-26.	3.9	4
27	Impact of the ionic forms on the UVâ€“Vis spectra 2â€hydroxybenzylamine. A TDâ€DFT study. International Journal of Quantum Chemistry, 2010, 110, 2179-2191.	2.0	2
28	A comparative study of the chemical reactivity of pyridoxamine, Ac-Phe-Lys and Ac-Cys with various glycating carbonyl compounds. Amino Acids, 2009, 36, 437-448.	2.7	21
29	Unexpected isomeric equilibrium in pyridoxamine Schiff bases. Bioorganic Chemistry, 2009, 37, 26-32.	4.1	23
30	The pyridoxamine action on Amadori compounds: A reexamination of its scavenging capacity and chelating effect. Bioorganic and Medicinal Chemistry, 2008, 16, 5557-5569.	3.0	46
31	Kinetic Study of the Reaction of Glycolaldehyde with Two Glycation Target Models. Annals of the New York Academy of Sciences, 2008, 1126, 235-240.	3.8	13
32	Pyridoxamine, a scavenger agent of carbohydrates. International Journal of Chemical Kinetics, 2007, 39, 154-167.	1.6	25
33	Inhibition of Glycosylation Processes: the Reaction between Pyridoxamine and Glucose. Chemistry and Biodiversity, 2005, 2, 964-975.	2.1	26
34	Theoretical study of thiolysis in penicillins and cephalosporines. International Journal of Chemical Kinetics, 2005, 37, 434-443.	1.6	6
35	Thiol-catalysed hydrolysis of cephalosporins and possible rate-limiting amine anion expulsion. Journal of Physical Organic Chemistry, 2004, 17, 521-528.	1.9	17
36	Photo-Induced Processes in Vitamin B6 Compounds. Chemistry and Biodiversity, 2004, 1, 1073-1090.	2.1	18

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37	FT-IR study of pyridoxamine 5â€² phosphate. Biochimica Et Biophysica Acta - Proteins and Proteomics, 2003, 1647, 83-87.	2.3	7
38	Molecular modelling studies on Henryâ€“Michaelis complexes of a class-C Î²-lactamase and Î²-lactam compounds. Computational and Theoretical Chemistry, 2002, 578, 19-28.	1.5	9
39	The role of a Î²-proton transfer donor in the degradation of benzylpenicillin. Journal of Molecular Catalysis A, 2001, 175, 3-16.	4.8	8
40	Kinetic and Molecular-Modelling Study of the Interaction betweenStaphylococcus aureus PC1 Enzyme and Imipenem. Helvetica Chimica Acta, 2001, 84, 3366-3379.	1.6	3
41	Electrostatic and structural similarity of classical and non-classical lactam compounds. Journal of Computer-Aided Molecular Design, 2001, 15, 819-833.	2.9	4
42	Ab initio study of the alkaline hydrolysis of a thio-Î²-lactam structure. Chemical Physics Letters, 2000, 326, 304-310.	2.6	10
43	Thiol-catalysed hydrolysis of benzylpenicillin. Perkin Transactions II RSC, 2000, , 1521-1525.	1.1	17
44	Theoretical Study of the Alkaline Hydrolysis of a Bicyclic Aza-Î²-lactam. Journal of Physical Chemistry B, 2000, 104, 11389-11394.	2.6	21
45	Kinetic and Molecular-Modelling Studies of Reactions of a Class-AÎ²-Lactamase with Compounds Bearing a Methoxy Group on theÎ²-Lactam Ring. Helvetica Chimica Acta, 1999, 82, 1274-1288.	1.6	10
46	Theoretical Study of the Alkaline Hydrolysis of an Oxo-Î²-Lactam Structure. Journal of Physical Chemistry A, 1999, 103, 8879-8884.	2.5	20
47	Alkaline hydrolysis of N-methylazetidin-2-one. Hydration effects. Computational and Theoretical Chemistry, 1998, 426, 313-321.	1.5	12
48	Chemical Reactivity of Penicillins and Cephalosporins. Intramolecular Involvement of the Acyl-Amido Side Chain. Journal of Organic Chemistry, 1998, 63, 9052-9060.	3.2	21
49	Î²-Lactamase-catalysed hydrolysis of cephalexin: evolution of the cephalosporoate intermediate. Journal of the Chemical Society Perkin Transactions II, 1997, , 2439-2444.	0.9	9
50	Study of the Alkaline Hydrolysis of the Azetidin-2-one Ring byab initioMethods: Influence of the solvent. Helvetica Chimica Acta, 1997, 80, 739-747.	1.6	27
51	The Degradation Mechanism of an Oral Cephalosporin: Cefaclor. Helvetica Chimica Acta, 1996, 79, 1793-1802.	1.6	9
52	pH Dependence of and kinetic solvent isotope effects on the methanolysis and hydrolysis of .beta.-lactams catalyzed by class C .beta.-lactamase. Journal of the American Chemical Society, 1995, 117, 12092-12095.	13.7	34
53	Penicillin 3-aldehyde is a good substrate and not an inhibitor of Î²-lactamases A and C. Journal of the Chemical Society Perkin Transactions II, 1995, , 869-870.	0.9	1
54	Alkaline Hydrolysis of Cefotaxime. A HPLC and 1HnmR Study. Journal of Pharmaceutical Sciences, 1994, 83, 322-327.	3.3	13

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55	Degradation of Cephaloridine on Alkaline Hydrolysis. Helvetica Chimica Acta, 1993, 76, 1619-1625.	1.6	5
56	HPLC and1H-NMR Studies of Alkaline Hydrolysis of Some 7-(Oxyiminoacyl)cephalosporins. Helvetica Chimica Acta, 1993, 76, 2789-2802.	1.6	11
57	Alkaline hydrolysis of cephaloridine: An1Hï£¿NMR study. Temperature dependence of the rate constants. International Journal of Chemical Kinetics, 1993, 25, 865-874.	1.6	3
58	Theoretical calculations of ?-lactam antibiotics. Theoretica Chimica Acta, 1993, 86, 229-239.	0.8	25