

# Wilhelm Hans Eugen Schwarz

## List of Publications by Year in descending order

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Version: 2024-02-01

40  
papers

1,168  
citations

361413

20  
h-index

395702

33  
g-index

44  
all docs

44  
docs citations

44  
times ranked

1079  
citing authors

#	ARTICLE	IF	CITATIONS
1	The Lewis electron-pair bonding model: modern energy decomposition analysis. <i>Nature Reviews Chemistry</i> , 2019, 3, 48-63.	30.2	197
2	Measuring Orbitals: Provocation or Reality?. <i>Angewandte Chemie - International Edition</i> , 2006, 45, 1508-1517.	13.8	79
3	Uranyl-Glycine-Water Complexes in Solution: Comprehensive Computational Modeling of Coordination Geometries, Stabilization Energies, and Luminescence Properties. <i>Inorganic Chemistry</i> , 2011, 50, 2082-2093.	4.0	68
4	A diuranium carbide cluster stabilized inside a C80 fullerene cage. <i>Nature Communications</i> , 2018, 9, 2753.	12.8	63
5	Das Kombinierte Ni <sub>1/2</sub> herungsverfahren. <i>Theoretica Chimica Acta</i> , 1968, 11, 307-324.	0.8	59
6	The Lewis electron-pair bonding model: the physical background, one century later. <i>Nature Reviews Chemistry</i> , 2019, 3, 35-47.	30.2	52
7	On the maximum bond multiplicity of carbon: unusual C <sub>4</sub> quadruple bonding in molecular CUO. <i>Chemical Science</i> , 2012, 3, 2786.	7.4	49
8	On the Upper Limits of Oxidation States in Chemistry. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 3242-3245.	13.8	46
9	Nonspherical atomic ground-state densities and chemical deformation densities from x-ray scattering. <i>Journal of Chemical Physics</i> , 1990, 92, 4956-4969.	3.0	44
10	An 18-Electron System Containing a Superheavy Element: Theoretical Studies of Sg@Au <sub>12</sub> . <i>Inorganic Chemistry</i> , 2015, 54, 3695-3701.	4.0	42
11	Theoretical Investigations of Geometry, Electronic Structure and Stability of UO <sub>6</sub> : Octahedral Uranium Hexoxide and Its Isomers <sup>â€</sup> . <i>Journal of Physical Chemistry A</i> , 2010, 114, 8837-8844.	2.5	39
12	Geometries, electronic structures, and excited states of UN <sub>2</sub> , NUO <sup>+</sup> , and UO <sub>2</sub> <sup>2+</sup> : a combined CCSD(T), RAS/CASPT2 and TDDFT study. <i>Theoretical Chemistry Accounts</i> , 2011, 129, 467-481.	1.4	39
13	Bonding trends across the series of tricarbonato-actinyl anions [(AnO <sub>2</sub> ) <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> ] <sup>4-</sup> (An = Uâ€“Cm): the plutonium turn. <i>Dalton Transactions</i> , 2017, 46, 2542-2550.	3.3	34
14	How Much Can Density Functional Approximations (DFA) Fail? The Extreme Case of the FeO <sub>4</sub> Species. <i>Journal of Chemical Theory and Computation</i> , 2016, 12, 1525-1533.	5.3	33
15	Understanding the Uniqueness of 2p Elements in Periodic Tables. <i>Chemistry - A European Journal</i> , 2020, 26, 15558-15564.	3.3	31
16	Probing the Electronic Structure and Chemical Bonding of Mono-Uranium Oxides with Different Oxidation States: UO <sub>x</sub> ( <i>x</i> = 1, 2, 3, 4) and UO <sub>x</sub> ( <i>x</i> = 1, 2, 3). <i>Journal of Physical Chemistry A</i> , 2016, 120, 1084-1096.	2.5	29
17	Antibond Breakingâ€”The Formation and Decomposition of He@Adamantane: Descriptions, Explanations, and Meaning of Concepts. <i>Chemistry - A European Journal</i> , 2010, 16, 9107-9116.	3.3	27
18	Observations and Descriptions versus Explanationsâ€”An Example: Does Nature, Does Theory Know About Steric Hindrance?. <i>Chemistry - A European Journal</i> , 2012, 18, 4470-4479.	3.3	25

#	ARTICLE	IF	CITATIONS
19	On structure and bonding of lanthanoid trifluorides LnF <sub>3</sub> (Ln = La to Lu). Physical Chemistry Chemical Physics, 2013, 15, 7839.	2.8	25
20	Theoretical Studies on Hexanuclear Oxometalates [M <sub>6</sub> L <sub>19</sub> ] <sup>iq</sup> (M = Cr, Mo, W, Sg, Nd, U). Electronic Structures, Oxidation States, Aromaticity, and Stability. Inorganic Chemistry, 2015, 54, 7171-7180.	4.0	24
21	Understanding Periodic and Non-periodic Chemistry in Periodic Tables. Frontiers in Chemistry, 2020, 8, 813.	3.6	22
22	Excited States and Absorption Spectra of UF <sub>6</sub> : A RASPT2 Theoretical Study with Spin-Orbit Coupling. Journal of Chemical Theory and Computation, 2011, 7, 3223-3231.	5.3	19
23	Physical origin of chemical periodicities in the system of elements. Pure and Applied Chemistry, 2019, 91, 1969-1999.	1.9	17
24	Äoer Oxidationszahl-Übergrenzen in der Chemie. Angewandte Chemie, 2018, 130, 3297-3300.	2.0	15
25	Structures and stability of AnO <sub>4</sub> isomers, An = Pu, Am, and Cm: a relativistic density functional study. Physical Chemistry Chemical Physics, 2014, 16, 8997-9001.	2.8	13
26	100th Anniversary of Bohr's Model of the Atom. Angewandte Chemie - International Edition, 2013, 52, 12228-12238.	13.8	11
27	On Two Different Objectives of the Concepts of Ionic Radii. Chemistry - A European Journal, 2013, 19, 14758-14767.	3.3	11
28	Excited-State Chemistry: Photocatalytic Methanol Oxidation by Uranyl@Zeolite through Oxygen-Centered Radicals. Inorganic Chemistry, 2020, 59, 6287-6300.	4.0	11
29	Diversity of Chemical Bonding and Oxidation States in MS <sub>4</sub> Molecules of Group-8 Elements. Chemistry - A European Journal, 2017, 23, 10580-10589.	3.3	7
30	The Good Reasons for a Standard Periodic Table of the Chemical Elements. Zeitschrift Fur Anorganische Und Allgemeine Chemie, 2022, 648, .	1.2	4
31	< i>Cis</i>- and < i>trans</i>-binding influences in [NUO <sub>4</sub> ·(N <sub>2</sub> ) <sub>n</sub> ] <sup>+</sup> . Journal of Chemical Physics, 2022, 157, .	3.0	4
32	Some solved problems of the periodic system of chemical elements. International Journal of Quantum Chemistry, 2009, 110, NA-NA.	2.0	3
33	Correspondence on "Core Electron Topologies in Chemical Compounds: Case Study of Carbon versus Silicon". Angewandte Chemie, 2019, 131, 10512-10515.	2.0	3
34	Metal Oxo-Fluoride Molecules OnMF <sub>2</sub> (M = Mn and Fe; n = 1-4) and O <sub>2</sub> MnF: Matrix Infrared Spectra and Quantum Chemistry. Inorganic Chemistry, 2021, 60, 7687-7696.	4.0	3
35	Electronic Structure and Spectroscopic Properties of Group-7 Tri-Oxo-Halides MO <sub>3</sub> X (M = Tj ETQq1 1.0784314 <sub>3</sub> rgBT / Over).	4.0	3
36	Correspondence on "Core Electron Topologies in Chemical Compounds: Case Study of Carbon versus Silicon". Angewandte Chemie - International Edition, 2019, 58, 10404-10407.	13.8	2

#	ARTICLE	IF	CITATIONS
37	In memoriam Hermann Hartmann, founder of TCA, on the occasion of his 100th birthday. <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	1
38	Editorial: Celebrating the International Year of the Periodic Table: Beyond Mendeleev 150. <i>Frontiers in Chemistry</i> , 2020, 8, 610869.	3.6	0
39	<i>d</i> - and <i>s</i> -orbital populations in the <i>d</i> block: unbound atoms in physical vacuum versus chemical elements in condensed matter. A Dronskowski-population analysis. <i>Zeitschrift Fur Naturforschung - Section B Journal of Chemical Sciences</i> , 2021, 76, 547-557.	0.7	0
40	On the highest oxidation states of the actinoids in $\text{AnO}_{4}$ molecules ( $\text{An}=\text{Ac}$ )	3.3	0