

Sang Soo Lee

List of Publications by Year in descending order

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49
papers

1,984
citations

218677

26
h-index

243625

44
g-index

50
all docs

50
docs citations

50
times ranked

2265
citing authors

#	ARTICLE	IF	CITATIONS
1	Trivalent ion overcharging on electrified graphene. <i>Journal of Physics Condensed Matter</i> , 2022, 34, 144001.	1.8	3
2	Impact of Ion-Ion Correlations on the Adsorption of M(III) (M = Am, Eu, Y) onto Muscovite (001) in the Presence of Sulfate. <i>Journal of Physical Chemistry C</i> , 2022, 126, 1400-1410.	3.1	3
3	Emergent Behavior at the Calcite-Water Interface during Reactive Transport in a Simple Microfluidic Channel. <i>ACS Earth and Space Chemistry</i> , 2022, 6, 861-870.	2.7	4
4	Ion correlations drive charge overscreening and heterogeneous nucleation at solid-aqueous electrolyte interfaces. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2021, 118, .	7.1	28
5	Replacement of Calcium Carbonate Polymorphs by Cerussite. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 2433-2441.	2.7	9
6	Pb Sorption at the Barite (001)-Water Interface. <i>Journal of Physical Chemistry C</i> , 2020, 124, 22035-22045.	3.1	9
7	Molecular-scale origins of wettability at petroleum-brine-carbonate interfaces. <i>Scientific Reports</i> , 2020, 10, 20507.	3.3	5
8	Nonclassical Behavior in Competitive Ion Adsorption at a Charged Solid-Water Interface. <i>Journal of Physical Chemistry Letters</i> , 2020, 11, 4029-4035.	4.6	10
9	Epitaxial Growth of Gibbsite Sheets on the Basal Surface of Muscovite Mica. <i>Journal of Physical Chemistry C</i> , 2019, 123, 27615-27627.	3.1	10
10	Effect of Anions on the Changes in the Structure and Adsorption Mechanism of Zirconium Species at the Muscovite (001)-Water Interface. <i>Journal of Physical Chemistry C</i> , 2019, 123, 16699-16710.	3.1	7
11	Mapping Three-dimensional Dissolution Rates of Calcite Microcrystals: Effects of Surface Curvature and Dissolved Metal Ions. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 833-843.	2.7	40
12	Effect of pH on the Formation of Gibbsite-Layer Films at the Muscovite (001)-Water Interface. <i>Journal of Physical Chemistry C</i> , 2019, 123, 6560-6571.	3.1	14
13	Oxidation induced strain and defects in magnetite crystals. <i>Nature Communications</i> , 2019, 10, 703.	12.8	40
14	Dissolution Kinetics of Epitaxial Cadmium Carbonate Overgrowths on Dolomite. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 212-220.	2.7	3
15	Simultaneous Adsorption and Incorporation of Sr ²⁺ at the Barite (001)-Water Interface. <i>Journal of Physical Chemistry C</i> , 2019, 123, 1194-1207.	3.1	21
16	Cathodic Corrosion at the Bismuth-Ionic Liquid Electrolyte Interface under Conditions for CO ₂ Reduction. <i>Chemistry of Materials</i> , 2018, 30, 2362-2373.	6.7	38
17	Evolution of Strain in Heteroepitaxial Cadmium Carbonate Overgrowths on Dolomite. <i>Crystal Growth and Design</i> , 2018, 18, 2871-2882.	3.0	6
18	Templating Growth of a Pseudomorphic Lepidocrocite Microshell at the Calcite-Water Interface. <i>Chemistry of Materials</i> , 2018, 30, 700-707.	6.7	4

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19	Pb ²⁺ Calcite Interactions under Far-from-Equilibrium Conditions: Formation of Micropylamids and Pseudomorphic Growth of Cerussite. <i>Journal of Physical Chemistry C</i> , 2018, 122, 2238-2247.	3.1	23
20	Arsenic uptake in bacterial calcite. <i>Geochimica Et Cosmochimica Acta</i> , 2018, 222, 642-654.	3.9	20
21	Effect of nitrogen passivation on interface composition and physical stress in SiO ₂ /SiC(4H) structures. <i>Applied Physics Letters</i> , 2018, 113, .	3.3	12
22	Heteroepitaxial growth of cadmium carbonate at dolomite and calcite surfaces: Mechanisms and rates. <i>Geochimica Et Cosmochimica Acta</i> , 2017, 205, 360-380.	3.9	28
23	Stern Layer Structure and Energetics at Mica-Water Interfaces. <i>Journal of Physical Chemistry C</i> , 2017, 121, 9402-9412.	3.1	119
24	Hydration Structure of the Barite (001) Water Interface: Comparison of X-ray Reflectivity with Molecular Dynamics Simulations. <i>Journal of Physical Chemistry C</i> , 2017, 121, 12236-12248.	3.1	38
25	Real-time observation of cation exchange kinetics and dynamics at the muscovite-water interface. <i>Nature Communications</i> , 2017, 8, 15826.	12.8	61
26	Heterogeneous Nucleation and Growth of Barium Sulfate at Organic Water Interfaces: Interplay between Surface Hydrophobicity and Ba ²⁺ Adsorption. <i>Langmuir</i> , 2016, 32, 5277-5284.	3.5	53
27	Surface Charge of the Calcite (104) Terrace Measured by Rb ⁺ Adsorption in Aqueous Solutions Using Resonant Anomalous X-ray Reflectivity. <i>Journal of Physical Chemistry C</i> , 2016, 120, 15216-15223.	3.1	24
28	X-ray Analyses of Lead Adsorption on the (001), (110), and (012) Hematite Surfaces. <i>Environmental Science & Technology</i> , 2016, 50, 12283-12291.	10.0	55
29	A Comparison of Adsorption, Reduction, and Polymerization of the Plutonyl(VI) and Uranyl(VI) Ions from Solution onto the Muscovite Basal Plane. <i>Langmuir</i> , 2016, 32, 10473-10482.	3.5	8
30	Structural Characterization of Aluminum (Oxy)hydroxide Films at the Muscovite (001) Water Interface. <i>Langmuir</i> , 2016, 32, 477-486.	3.5	14
31	Replacement of Calcite (CaCO ₃) by Cerussite (PbCO ₃). <i>Environmental Science & Technology</i> , 2016, 50, 12984-12991.	10.0	51
32	Rb ⁺ Adsorption at the Quartz(101) Aqueous Interface: Comparison of Resonant Anomalous X-ray Reflectivity with ab Initio Calculations. <i>Journal of Physical Chemistry C</i> , 2015, 119, 4778-4788.	3.1	34
33	Effects of the background electrolyte on Th(IV) sorption to muscovite mica. <i>Geochimica Et Cosmochimica Acta</i> , 2015, 165, 280-293.	3.9	11
34	X-ray-driven reaction front dynamics at calcite-water interfaces. <i>Science</i> , 2015, 349, 1330-1334.	12.6	69
35	Hydration layer structure at solid-water interfaces. <i>MRS Bulletin</i> , 2014, 39, 1056-1061.	3.5	65
36	Surface-Mediated Formation of Pu(IV) Nanoparticles at the Muscovite-Electrolyte Interface. <i>Environmental Science & Technology</i> , 2013, 47, 14178-14184.	10.0	27

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37	Changes in adsorption free energy and speciation during competitive adsorption between monovalent cations at the muscovite (001)-water interface. <i>Geochimica Et Cosmochimica Acta</i> , 2013, 123, 416-426.	3.9	57
38	Optimizing a flow-through X-ray transmission cell for studies of temporal and spatial variations of ion distributions at mineral-water interfaces. <i>Journal of Synchrotron Radiation</i> , 2013, 20, 125-136.	2.4	17
39	Investigation of Structure, Adsorption Free Energy, and Overcharging Behavior of Trivalent Yttrium Adsorbed at the Muscovite(001)-Water Interface. <i>Journal of Physical Chemistry C</i> , 2013, 117, 23738-23749.	3.1	36
40	Adsorption of Plutonium Oxide Nanoparticles. <i>Langmuir</i> , 2012, 28, 2620-2627.	3.5	27
41	Nanoscale Perturbations of Room Temperature Ionic Liquid Structure at Charged and Uncharged Interfaces. <i>ACS Nano</i> , 2012, 6, 9818-9827.	14.6	151
42	Monovalent Ion Adsorption at the Muscovite (001)-Solution Interface: Relationships among Ion Coverage and Speciation, Interfacial Water Structure, and Substrate Relaxation. <i>Langmuir</i> , 2012, 28, 8637-8650.	3.5	128
43	Heavy Metal Sorption at the Muscovite (001)-Fulvic Acid Interface. <i>Environmental Science & Technology</i> , 2011, 45, 9574-9581.	10.0	35
44	Application of eggshell waste for the immobilization of cadmium and lead in a contaminated soil. <i>Environmental Geochemistry and Health</i> , 2011, 33, 31-39.	3.4	119
45	Effects of natural and calcined oyster shells on Cd and Pb immobilization in contaminated soils. <i>Environmental Earth Sciences</i> , 2010, 61, 1301-1308.	2.7	178
46	Hydrated Cation Speciation at the Muscovite (001)-Water Interface. <i>Langmuir</i> , 2010, 26, 16647-16651.	3.5	126
47	Competitive adsorption of strontium and fulvic acid at the muscovite-solution interface observed with resonant anomalous X-ray reflectivity. <i>Geochimica Et Cosmochimica Acta</i> , 2010, 74, 1762-1776.	3.9	47
48	Enhanced Uptake and Modified Distribution of Mercury(II) by Fulvic Acid on the Muscovite (001) Surface. <i>Environmental Science & Technology</i> , 2009, 43, 5295-5300.	10.0	43
49	Distribution of barium and fulvic acid at the mica-solution interface using in-situ X-ray reflectivity. <i>Geochimica Et Cosmochimica Acta</i> , 2007, 71, 5763-5781.	3.9	53