

# Linghao Kong

## List of Publications by Year in descending order

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Version: 2024-02-01

33  
papers

965  
citations

430874

18  
h-index

454955

30  
g-index

33  
all docs

33  
docs citations

33  
times ranked

755  
citing authors

#	ARTICLE	IF	CITATIONS
1	Mechanisms of Sb(III) Oxidation by Pyrite-Induced Hydroxyl Radicals and Hydrogen Peroxide. <i>Environmental Science &amp; Technology</i> , 2015, 49, 3499-3505.	10.0	144
2	Calculation and application of Sb toxicity coefficient for potential ecological risk assessment. <i>Science of the Total Environment</i> , 2018, 610-611, 167-174.	8.0	112
3	Mechanisms of Sb(III) Photooxidation by the Excitation of Organic Fe(III) Complexes. <i>Environmental Science &amp; Technology</i> , 2016, 50, 6974-6982.	10.0	63
4	Kinetics and Mechanism of Photopromoted Oxidative Dissolution of Antimony Trioxide. <i>Environmental Science &amp; Technology</i> , 2014, 48, 14266-14272.	10.0	53
5	Removal of Arsenic from Strongly Acidic Wastewater Using Phosphorus Pentasulfide As Precipitant: UV-Light Promoted Sulfuration Reaction and Particle Aggregation. <i>Environmental Science &amp; Technology</i> , 2018, 52, 4794-4801.	10.0	46
6	Photopromoted oxidative dissolution of stibnite. <i>Applied Geochemistry</i> , 2015, 61, 53-61.	3.0	44
7	Mechanisms of UV-Light Promoted Removal of As(V) by Sulfide from Strongly Acidic Wastewater. <i>Environmental Science &amp; Technology</i> , 2017, 51, 12583-12591.	10.0	42
8	Dynamic flow and pollution of antimony from polyethylene terephthalate (PET) fibers in China. <i>Science of the Total Environment</i> , 2021, 771, 144643.	8.0	39
9	Removal of Chloride Ions from Strongly Acidic Wastewater Using Cu(O)/Cu(II): Efficiency Enhancement by UV Irradiation and the Mechanism for Chloride Ions Removal. <i>Environmental Science &amp; Technology</i> , 2019, 53, 383-389.	10.0	36
10	Specific H <sub>2</sub> S Release from Thiosulfate Promoted by UV Irradiation for Removal of Arsenic and Heavy Metals from Strongly Acidic Wastewater. <i>Environmental Science &amp; Technology</i> , 2020, 54, 14076-14084.	10.0	33
11	Adsorption of antimony on kaolinite as a function of time, pH, HA and competitive anions. <i>Environmental Earth Sciences</i> , 2016, 75, 1.	2.7	30
12	UV-Improved Removal of Chloride Ions from Strongly Acidic Wastewater Using Bi <sub>2</sub> O <sub>3</sub> : Efficiency Enhancement and Mechanisms. <i>Environmental Science &amp; Technology</i> , 2019, 53, 10371-10378.	10.0	30
13	UV-Light-Induced Aggregation of Arsenic and Metal Sulfide Particles in Acidic Wastewater: The Role of Free Radicals. <i>Environmental Science &amp; Technology</i> , 2018, 52, 10719-10727.	10.0	28
14	Sulfate radical-based removal of chloride ion from strongly acidic wastewater: Kinetics and mechanism. <i>Journal of Hazardous Materials</i> , 2021, 410, 124540.	12.4	27
15	Photo-induced dissolution of Bi <sub>2</sub> O <sub>3</sub> during photocatalysis reactions: Mechanisms and inhibition method. <i>Journal of Hazardous Materials</i> , 2021, 412, 125267.	12.4	23
16	A review on the removal of Cl(-) with high concentration from industrial wastewater: Approaches and mechanisms. <i>Science of the Total Environment</i> , 2022, 824, 153909.	8.0	23
17	UV light irradiation improves the aggregation and settling performance of metal sulfide particles in strongly acidic wastewater. <i>Water Research</i> , 2019, 163, 114860.	11.3	22
18	A novel precipitant for the selective removal of fluoride ion from strongly acidic wastewater: Synthesis, efficiency, and mechanism. <i>Journal of Hazardous Materials</i> , 2021, 403, 124039.	12.4	20

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19	Removal of Cl(-I) from strongly acidic wastewater containing Cu(II) by complexation-precipitation using thiourea: Efficiency enhancement by ascorbic acid. <i>Journal of Hazardous Materials</i> , 2021, 402, 123836.	12.4	18
20	Influences of Particles and Aquatic Colloids on the Oxidation of Sb(III) in Natural Water. <i>ACS Earth and Space Chemistry</i> , 2020, 4, 661-671.	2.7	17
21	Removal of Ni(II) from strongly acidic wastewater by chelating precipitation and recovery of NiO from the precipitates. <i>Journal of Environmental Sciences</i> , 2021, 104, 365-375.	6.1	16
22	Calcium sulfide-organosilicon complex for sustained release of H <sub>2</sub> S in strongly acidic wastewater: Synthesis, mechanism and efficiency. <i>Journal of Hazardous Materials</i> , 2022, 421, 126745.	12.4	14
23	Mechanism for Photopromoted Release of Vanadium from Vanadium Titano-Magnetite. <i>Environmental Science &amp; Technology</i> , 2018, 52, 1954-1962.	10.0	12
24	Recovery of Re(VII) from strongly acidic wastewater using sulphide: Acceleration by UV irradiation and the underlying mechanism. <i>Journal of Hazardous Materials</i> , 2021, 416, 126233.	12.4	12
25	Reductive Removal and Recovery of As(V) and As(III) from Strongly Acidic Wastewater by a UV/Formic Acid Process. <i>Environmental Science &amp; Technology</i> , 2022, 56, 9732-9743.	10.0	12
26	Removal of Cl(̂I) from strongly acidic wastewater using NaBiO <sub>3</sub> : A process of simultaneous oxidation and precipitation. <i>Desalination</i> , 2020, 491, 114566.	8.2	10
27	Hydrophilicity/hydrophobicity of metal sulfide particles as a determinator of aggregation performance in wastewater. <i>Journal of Water Process Engineering</i> , 2021, 40, 101900.	5.6	10
28	H <sub>2</sub> S release rate strongly affects particle size and settling performance of metal sulfides in acidic wastewater: The role of homogeneous and heterogeneous nucleation. <i>Journal of Hazardous Materials</i> , 2022, 438, 129484.	12.4	9
29	The mechanism for promoted oxygenation of V(IV) by goethite: Positive effect of surface hydroxyl groups. <i>Journal of Hazardous Materials</i> , 2019, 369, 254-260.	12.4	7
30	Clean and effective removal of Cl(-I) from strongly acidic wastewater by PbO <sub>2</sub> . <i>Journal of Environmental Sciences</i> , 2022, 120, 1-8.	6.1	5
31	Chemical solidification/stabilization of arsenic sulfide and oxide mixed wastes using elemental sulfur: Efficiencies, mechanisms and long-term stabilization enhancement by dicyclopentadiene. <i>Journal of Hazardous Materials</i> , 2021, 419, 126390.	12.4	4
32	The Recycling of Acid Wastewater with High Concentrations of Organic Matter: Recovery of H <sub>2</sub> SO <sub>4</sub> and Preparation of Activated Carbon. <i>Water (Switzerland)</i> , 2022, 14, 183.	2.7	3
33	Improving removal rate and efficiency of As(V) by sulfide from strongly acidic wastewater in a modified photochemical reactor. <i>Environmental Technology (United Kingdom)</i> , 2022, 43, 2329-2341.	2.2	1