

Zachary A Piazza

List of Publications by Year in descending order

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13
papers

2,838
citations

687363

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1058476

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docs citations

15
times ranked

1683
citing authors

#	ARTICLE	IF	CITATIONS
1	Observation of an all-boron fullerene. <i>Nature Chemistry</i> , 2014, 6, 727-731.	13.6	724
2	Planar hexagonal B ₃₆ as a potential basis for extended single-atom layer boron sheets. <i>Nature Communications</i> , 2014, 5, 3113.	12.8	645
3	Understanding Boron through Size-Selected Clusters: Structure, Chemical Bonding, and Fluxionality. <i>Accounts of Chemical Research</i> , 2014, 47, 1349-1358.	15.6	474
4	B ₂₂ ⁺ and B ₂₃ ⁺ : All-Boron Analogues of Anthracene and Phenanthrene. <i>Journal of the American Chemical Society</i> , 2012, 134, 18065-18073.	13.7	198
5	Transition-Metal-Centered Nine-Membered Boron Rings: M ⁺ B ₉ and M ⁺ B ₉ ⁺ (M = Rh, Ir). <i>Journal of the American Chemical Society</i> , 2012, 134, 165-168.	13.7	157
6	Complexes between Planar Boron Clusters and Transition Metals: A Photoelectron Spectroscopy and Ab Initio Study of CoB ₁₂ ⁺ and RhB ₁₂ ⁺ . <i>Journal of Physical Chemistry A</i> , 2014, 118, 8098-8105.	2.5	143
7	A combined photoelectron spectroscopy and ab initio study of the quasi-planar B ₂₄ ⁺ cluster. <i>Journal of Chemical Physics</i> , 2013, 139, 144307.	3.0	128
8	A photoelectron spectroscopy and <i>ab initio</i> study of B ₂₁ ⁺ : Negatively charged boron clusters continue to be planar at 21. <i>Journal of Chemical Physics</i> , 2012, 136, 104310.	3.0	127
9	A photoelectron spectroscopy and ab initio study of the structures and chemical bonding of the B ₂₅ ⁺ cluster. <i>Journal of Chemical Physics</i> , 2014, 141, 034303.	3.0	65
10	B ₂₇ ⁺ : Appearance of the smallest planar boron cluster containing a hexagonal vacancy. <i>Journal of Chemical Physics</i> , 2015, 142, 204305.	3.0	60
11	Geometrical requirements for transition-metal-centered aromatic boron wheels: the case of VB ₁₀ ⁺ . <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 13663.	2.8	50
12	The electronic structure and chemical bonding in gold dihydride: AuH ₂ ⁺ and AuH ₂ . <i>Chemical Science</i> , 2012, 3, 3286.	7.4	49
13	Bond-bending isomerism of Au ₂ I ₃ ⁺ : competition between covalent bonding and aurophilicity. <i>Chemical Science</i> , 2016, 7, 475-481.	7.4	16