## Patricia M Dove

## List of Publications by Year in descending order

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49 papers

7,892 citations

36 h-index 223531 46 g-index

50 all docs

50 docs citations

50 times ranked

7523 citing authors

#	Article	IF	CITATIONS
1	Crystallization by particle attachment in synthetic, biogenic, and geologic environments. Science, 2015, 349, aaa6760.	6.0	1,467
2	The Role of Mg2+ as an Impurity in Calcite Growth. Science, 2000, 290, 1134-1137.	6.0	638
3	Thermodynamics of Calcite Growth: Baseline for Understanding Biomineral Formation. , 1998, 282, 724-727.		448
4	Kinetics of quartz dissolution in electrolyte solutions using a hydrothermal mixed flow reactor. Geochimica Et Cosmochimica Acta, 1990, 54, 955-969.	1.6	422
5	Kinetics of calcite growth: surface processes and relationships to macroscopic rate laws. Geochimica Et Cosmochimica Acta, 2000, 64, 2255-2266.	1.6	388
6	Calcite precipitation mechanisms and inhibition by orthophosphate: In situ observations by Scanning Force Microscopy. Geochimica Et Cosmochimica Acta, 1993, 57, 705-714.	1.6	309
7	The dissolution kinetics of amorphous silica into sodium chloride solutions: effects of temperature and ionic strength. Geochimica Et Cosmochimica Acta, 2000, 64, 4193-4203.	1.6	265
8	Mechanisms of classical crystal growth theory explain quartz and silicate dissolution behavior. Proceedings of the National Academy of Sciences of the United States of America, 2005, 102, 15357-15362.	3.3	254
9	The dissolution kinetics of quartz in aqueous mixed cation solutions. Geochimica Et Cosmochimica Acta, 1999, 63, 3715-3727.	1.6	221
10	Surface charge density on silica in alkali and alkaline earth chloride electrolyte solutions. Geochimica Et Cosmochimica Acta, 2005, 69, 4963-4970.	1.6	202
11	Dissolution rate of quartz in lead and sodium electrolyte solutions between 25 and 300°C: Effect of the nature of surface complexes and reaction affinity. Geochimica Et Cosmochimica Acta, 1994, 58, 541-551.	1.6	197
12	The influence of the alkaline earth cations, magnesium, calcium, and barium on the dissolution kinetics of quartz. Geochimica Et Cosmochimica Acta, 1997, 61, 3329-3340.	1.6	192
13	Kinetics of amorphous silica dissolution and the paradox of the silica polymorphs. Proceedings of the National Academy of Sciences of the United States of America, 2008, 105, 9903-9908.	3.3	189
14	Dissolution kinetics of quartz in sodium chloride solutions: Analysis of existing data and a rate model for 25°C. Geochimica Et Cosmochimica Acta, 1992, 56, 4147-4156.	1.6	181
15	MATERIALS SCIENCE: Shaping Crystals with Biomolecules. Science, 2004, 306, 1301-1302.	6.0	174
16	Polysaccharide chemistry regulates kinetics of calcite nucleation through competition of interfacial energies. Proceedings of the National Academy of Sciences of the United States of America, 2013, 110, 9261-9266.	3.3	173
17	Carboxylated molecules regulate magnesium content of amorphous calcium carbonates during calcification. Proceedings of the National Academy of Sciences of the United States of America, 2009, 106, 21511-21516.	3.3	163
18	Morphological consequences of differential Mg <sup>2+</sup> incorporation at structurally distinct steps on calcite. American Mineralogist, 2004, 89, 714-720.	0.9	145

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19	Mineral/solution reaction rates in a mixed flow reactor: Wollastonite hydrolysis. Geochimica Et Cosmochimica Acta, 1986, 50, 2509-2516.	1.6	141
20	Kinetics of Silica Nucleation on Carboxyl- and Amine-Terminated Surfaces: Insights for Biomineralization. Journal of the American Chemical Society, 2009, 131, 5244-5250.	6.6	128
21	Reconciling disparate views of template-directed nucleation through measurement of calcite nucleation kinetics and binding energies. Proceedings of the National Academy of Sciences of the United States of America, 2014, 111, 1304-1309.	3.3	122
22	Crystal chemical controls on the dissolution kinetics of the isostructural sulfates: Celestite, anglesite, and barite. Geochimica Et Cosmochimica Acta, 1995, 59, 1907-1915.	1.6	121
23	Nanoscale effects of strontium on calcite growth: An in situ AFM study in the absence of vital effects. Geochimica Et Cosmochimica Acta, 2005, 69, 3017-3027.	1.6	120
24	Reversed calcite morphologies induced by microscopic growth kinetics: insight into biomineralization. Geochimica Et Cosmochimica Acta, 1999, 63, 2507-2512.	1.6	113
25	Compatible real-time rates of mineral dissolution by Atomic Force Microscopy (AFM). Chemical Geology, 1996, 127, 331-338.	1.4	112
26	Surface site-specific interactions of aspartate with calcite during dissolution; implications for biomineralization. American Mineralogist, 1997, 82, 878-887.	0.9	109
27	Microbially catalyzed dissolution of iron and aluminum oxyhydroxide mineral surface coatings. Geochimica Et Cosmochimica Acta, 1997, 61, 4467-4477.	1.6	96
28	Effects of temperature and transport conditions on calcite growth in the presence of Mg2+: Implications for paleothermometry. Geochimica Et Cosmochimica Acta, 2005, 69, 4227-4236.	1.6	89
29	Structural Development of Mercaptophenol Self-Assembled Monolayers and the Overlying Mineral Phase during Templated CaCO <sub>3</sub> Crystallization from a Transient Amorphous Film. Journal of the American Chemical Society, 2007, 129, 10370-10381.	6.6	89
30	New insights into mechanisms of biomolecular control on growth of inorganic crystals. CrystEngComm, 2007, 9, 1144.	1.3	77
31	Geochemical controls on the kinetics of quartz fracture at subcritical tensile stresses. Journal of Geophysical Research, 1995, 100, 22349-22359.	3.3	75
32	Influence of Ion-Associated Water on the Hydrolysis of Siâ^O Bonded Interactions. Journal of Physical Chemistry A, 2010, 114, 2534-2542.	1.1	65
33	Title is missing!. Aquatic Geochemistry, 2001, 7, 13-32.	1.5	64
34	Raman spectroscopic characterization of the magnesium content in amorphous calcium carbonates. Journal of Raman Spectroscopy, 2012, 43, 543-548.	1.2	57
35	Investigation of bacterial-mineral interactions using Fluid Tapping Modeâ,, Atomic Force Microscopy. Geochimica Et Cosmochimica Acta, 1996, 60, 2473-2480.	1.6	55
36	Isotopic tracer evidence for the amorphous calcium carbonate to calcite transformation by dissolution–reprecipitation. Geochimica Et Cosmochimica Acta, 2015, 165, 407-417.	1.6	51

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37	Molecular Dynamics of Ion Hydration in the Presence of Small Carboxylated Molecules and Implications for Calcification. Journal of Physical Chemistry B, 2010, 114, 10488-10495.	1.2	48
38	Nucleation on surfaces and in confinement. MRS Bulletin, 2016, 41, 388-392.	1.7	32
39	Metastable solubility and local structure of amorphous calcium carbonate (ACC). Geochimica Et Cosmochimica Acta, 2020, 289, 196-206.	1.6	27
40	A new method for <i>in situ</i> structural investigations of nano-sized amorphous and crystalline materials using mixed-flow reactors. Acta Crystallographica Section A: Foundations and Advances, 2019, 75, 758-765.	0.0	21
41	Nucleation Pathways in Electrolyte Solutions. , 2017, , 1-24.		14
42	Systematic dependence of kinetic and thermodynamic barriers to homogeneous silica nucleation on NaCl and amino acids. Journal of Materials Research, 2019, 34, 442-455.	1.2	13
43	A Mixed Flow Reactor Method to Synthesize Amorphous Calcium Carbonate Under Controlled Chemical Conditions. Methods in Enzymology, 2013, 532, 557-568.	0.4	10
44	Resolving the Control of Magnesium on Calcite Growth: Thermodynamic and Kinetic Consequences of Impurity Incorporation for Biomineral Formation. Materials Research Society Symposia Proceedings, 2000, 620, 1.	0.1	5
45	Biologically Inspired Silicification Process for Improving Mechanical Properties of Sand. Journal of Geotechnical and Geoenvironmental Engineering - ASCE, 2011, 137, 949-957.	1.5	4
46	Reply to Comment on "Kinetics of quartz dissolution in electrolyte solutions using a hydrothermal mixed flow reactor― Geochimica Et Cosmochimica Acta, 1992, 56, 4093.	1.6	1
47	Molecular Dynamics of Cation Hydration in the Presence of Carboxylated Molecules: Implications for Calcification. Materials Research Society Symposia Proceedings, 2011, 1301, 51.	0.1	1
48	Experimental creep behaviour and modelling of silicified sand. Proceedings of the Institution of Civil Engineers: Ground Improvement, 2013, 166, 115-124.	0.7	1
49	The Kinetics of Calcite Growth: Interpreting Chemical Affinity-Based Rate Laws Through the Lens of Direct Observation. Materials Research Society Symposia Proceedings, 2000, 620, 1.	0.1	0